

## APPENDIX A

### PERHITUNGAN NERACA MASSA

Tulang kering yang diolah adalah 13500 kg/hari

Komposisi bahan penyusun tulang adalah :

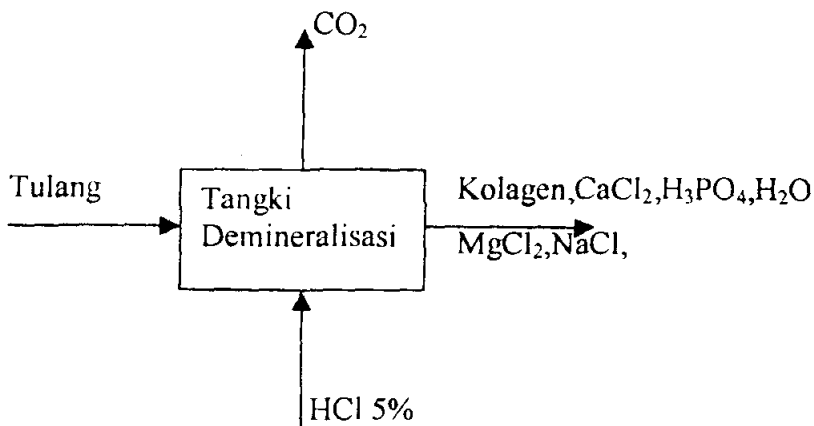
- $\text{Ca}_3(\text{PO}_4)_2$  = 57,35 %
- $\text{CaCO}_3$  = 3,85 %
- $\text{Mg}_3(\text{PO}_4)_2$  = 2,05 %
- $\text{Na}_2\text{CO}_3$  = 3,45 %
- Kolagen = 33,30 %

#### I. Unit Penyediaan Bahan Baku

##### Tangki Demineralisasi (F-140)

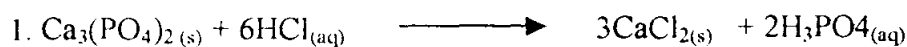
$$\begin{aligned}\text{Jumlah : } \text{Ca}_3(\text{PO}_4)_2 &= 57,35 \% \times 13500 = 7742,2500 \text{ kg} \\ \text{CaCO}_3 &= 3,85 \% \times 13500 = 519,7500 \text{ kg} \\ \text{Mg}_3(\text{PO}_4)_2 &= 2,05 \% \times 13500 = 276,7500 \text{ kg} \\ \text{Na}_2\text{CO}_3 &= 3,45 \% \times 13500 = 465,7500 \text{ kg} \\ \text{Kolagen} &= 33,30 \% \times 13500 = \underline{4495,5000 \text{ kg}} + \\ &= 13500,0000 \text{ kg}\end{aligned}$$

Terjadi proses penghilangan mineral dengan menambah HCl 5 % (othmer,1980)



Data : BM	$\text{Ca}_3(\text{PO}_4)_2$	= 310,19	$\text{CaCl}_2$	= 110,994
	$\text{CaCO}_3$	= 100,91	$\text{MgCl}_2$	= 95,234
	$\text{Mg}_3(\text{PO}_4)_2$	= 262,91	$\text{NaCl}$	= 58,448
	$\text{Na}_2\text{CO}_3$	= 105,993	$\text{H}_3\text{PO}_4$	= 97,999
	$\text{HCl}$	= 36,465	$\text{H}_2\text{O}$	= 18,016
	$\text{CO}_2$	= 44,011	Kolagen	= 2417,562
	Gelatin	= 2435,578		

Reaksi :



$$\text{Ca}_3(\text{PO}_4)_2 = \frac{7742,25}{310,19} = 24,9597 \text{ kmol}$$

$$\text{HCl} = 6 \times 24,9597 \text{ kmol} \times 36,465 \text{ kg/kmol} = 5460,9328 \text{ kg}$$

$$\text{CaCl}_2 = 3 \times 24,9597 \text{ kmol} \times 110,994 \text{ kg/kmol} = 8311,1308 \text{ kg}$$

$$\text{H}_3\text{PO}_4 = 2 \times 24,9597 \text{ kmol} \times 97,999 \text{ kg/kmol} = 4892,0513 \text{ kg}$$



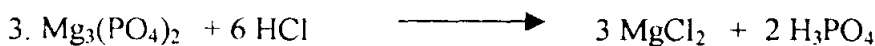
$$\text{CaCO}_3 = \frac{519,75}{100,91} = 5,1506 \text{ kmol}$$

$$\text{HCl} = 2 \times 5,1506 \text{ kmol} \times 36,465 \text{ kg/kmol} = 375,6333 \text{ kg}$$

$$\text{CaCl}_2 = 5,1506 \text{ kmol} \times 110,994 \text{ kg/kmol} = 571,6857 \text{ kg}$$

$$\text{H}_2\text{O} = 5,1506 \text{ kmol} \times 18,016 \text{ kg/kmol} = 92,7932 \text{ kg}$$

$$\text{CO}_2 = 5,1506 \text{ kmol} \times 44,011 \text{ kg/kmol} = 226,6831 \text{ kg}$$

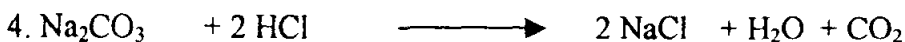


$$\text{Mg}_3(\text{PO}_4)_2 = \frac{276,75}{262,91} = 1,0526 \text{ kmol}$$

$$\text{HCl} = 6 \times 1,0526 \text{ kmol} \times 36,465 \text{ kg/kmol} = 230,2984 \text{ kg}$$

$$\text{MgCl}_2 = 3 \times 1,0526 \text{ kmol} \times 95,234 \text{ kg/kmol} = 300,7299 \text{ kg}$$

$$\text{H}_3\text{PO}_4 = 2 \times 1,0526 \text{ kmol} \times 97,999 \text{ kg/kmol} = 206,3075 \text{ kg}$$





Filtrat ( $\text{H}_3\text{PO}_4$ ,  $\text{NaCl}$ ,  $\text{CaCl}_2$

$\text{MgCl}_2$ ,  $\text{H}_2\text{O}$ )

Data - data : 1. Filtrat keluar dari Rotary Filter telah bebas impurities

2. Massa cake kering yang terkumpul dalam frame 60%-80% berat cake (Ulrich,1984)

3. Massa filtrat yang terikut 20% berat cake.

Massa cake kering = 4495,5000 kg

Massa filtrat yang terikut cake =  $\frac{20}{80} \times 4495,5 \text{ kg} = 1123,875 \text{ kg}$

Massa cake = 4495,5 kg + 1123,875 kg  
= 5619,375 kg

Massa filtrat = 136326,8652 kg – 1123,875 kg  
= 135202,9902 kg

Masuk :

Kolagen = 4495,5000 kg

$\text{H}_2\text{O}$  = 121531,2956 kg

$\text{CaCl}_2$  = 8882,8165 kg

$\text{MgCl}_2$  = 300,7299 kg

$\text{NaCl}$  = 513,6644 kg

$\text{H}_3\text{PO}_4$  = 5098,3588 kg

Keluar :

Cake ke tangki liming :

Kolagen = 4495,5000 kg

$\text{H}_2\text{O}$  = 1001,9346 kg

$\text{CaCl}_2$  = 73,1643 kg

$\text{MgCl}_2$  = 2,4725 kg

$\text{NaCl}$  = 4,2707 kg

$\text{H}_3\text{PO}_4$  = 42,0329 kg

Filtrat ke pembuangan :

$\text{H}_2\text{O}$  = 120529,3610 kg

$\text{CaCl}_2$  = 8809,6522 kg

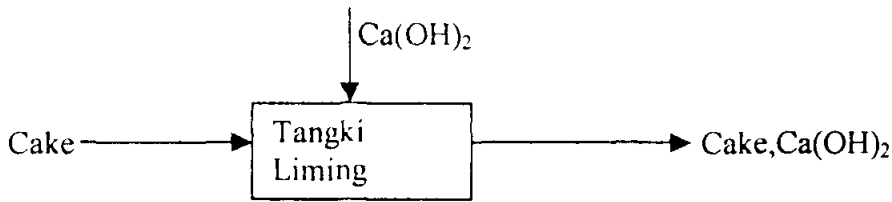
$\text{MgCl}_2$  = 298,2574 kg

$\text{NaCl}$  = 509,3937 kg

$\text{H}_3\text{PO}_4$  = 5056,3259 kg +

\_\_\_\_\_ +  
140822,3652 kg

140822,3652 kg

**Tangki Liming (F-150)**

Pemakaian air kapur : kolagen kering = 4 : 1 (Ward,1977)

$$\text{Berat air kapur} = \frac{4}{1} \times 4495,5 \text{ kg} = 17982 \text{ kg}$$

Masuk :

Cake :

Kolagen = 4495,5000 kg

H<sub>2</sub>O = 1001,9346 kg

CaCl<sub>2</sub> = 73,1643 kg

MgCl<sub>2</sub> = 2,4725 kg

NaCl = 4,2707 kg

H<sub>3</sub>PO<sub>4</sub> = 42,0329 kg

Ca(OH)<sub>2</sub> =  $\frac{17982,0000 \text{ kg}}{1} +$   
23601,3750 kg

Keluar :

Kolagen = 4495,5000 kg

H<sub>2</sub>O = 1001,9346 kg

CaCl<sub>2</sub> = 73,1643 kg

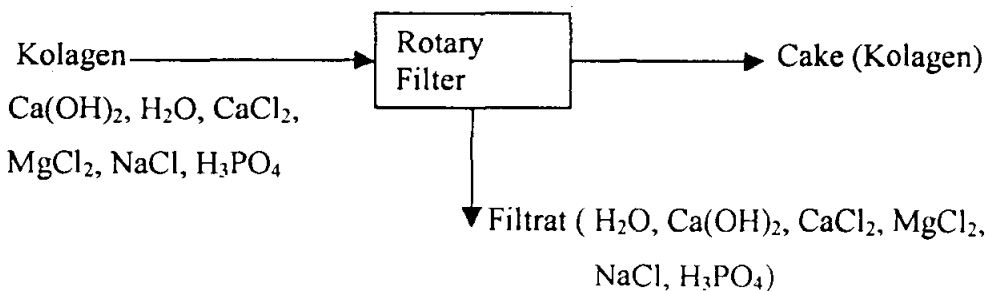
MgCl<sub>2</sub> = 2,4725 kg

NaCl = 4,2707 kg

H<sub>3</sub>PO<sub>4</sub> = 42,0329 kg

Ca(OH)<sub>2</sub> = 17982,0000 kg

$\frac{17982,0000 \text{ kg}}{1} +$   
23601,3750 kg

**Rotary Drum Filter (H-151)**

Data - data : 1. Filtrat keluar dari Rotary Filter telah bebas impurities

2. Massa cake kering yang terkumpul dalam frame 60%-80% berat cake (Ulrich,1984)

3. Massa filtrat yang terikut 20% berat cake.

Massa cake kering = 4495,5000 kg

$$\text{Massa filtrat terikut cake} = \frac{20}{80} \times 4495,5 \text{ kg} = 1123,875 \text{ kg}$$

$$\begin{aligned} \text{Massa cake} &= 4495,5 \text{ kg} + 1123,875 \text{ kg} \\ &= 5619,375 \text{ kg} \end{aligned}$$

$$\begin{aligned} \text{Massa filtrat} &= 1001,9346 + 73,1643 + 2,4725 + 4,2707 + 42,0329 + 17982 \\ &= 19105,875 \text{ kg} \end{aligned}$$

Masuk dari tangki liming :

$$\begin{aligned} \text{Kolagen} &= 4495,5000 \text{ kg} \\ \text{H}_2\text{O} &= 1001,9346 \text{ kg} \\ \text{CaCl}_2 &= 73,1643 \text{ kg} \\ \text{MgCl}_2 &= 2,4725 \text{ kg} \\ \text{NaCl} &= 4,2707 \text{ kg} \\ \text{H}_3\text{PO}_4 &= 42,0329 \text{ kg} \\ \text{Ca(OH)}_2 &= 17982,0000 \text{ kg} \end{aligned}$$

$$\begin{array}{r} \text{-----} + \\ 23601,3750 \text{ kg} \end{array}$$

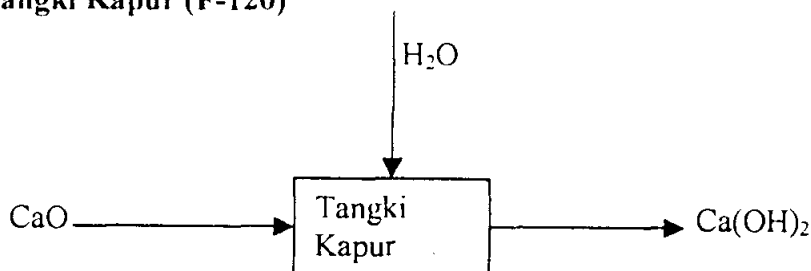
Keluar ke tangki netralisasi :

$$\begin{aligned} \text{Cake} &: \\ \text{Kolagen} &= 4495,5000 \text{ kg} \\ \text{H}_2\text{O} &= 59,0034 \text{ kg} \\ \text{CaCl}_2 &= 4,2707 \text{ kg} \\ \text{MgCl}_2 &= 0,1124 \text{ kg} \\ \text{NaCl} &= 0,2248 \text{ kg} \\ \text{H}_3\text{PO}_4 &= 2,4725 \text{ kg} \\ \text{Ca(OH)}_2 &= 1057,7912 \text{ kg} \end{aligned}$$

Filtrat ke pembuangan :

$$\begin{aligned} \text{H}_2\text{O} &= 942,9312 \text{ kg} \\ \text{CaCl}_2 &= 68,8936 \text{ kg} \\ \text{MgCl}_2 &= 2,3601 \text{ kg} \\ \text{NaCl} &= 4,0459 \text{ kg} \\ \text{H}_3\text{PO}_4 &= 39,5604 \text{ kg} \\ \text{Ca(OH)}_2 &= 16924,2088 \text{ kg} + \\ &23601,3750 \text{ kg} \end{aligned}$$

**Tangki Kapur (F-120)**



Kadar CaO = 10%

Air kapur 10% yang digunakan adalah = 17982 kg

s.g CaO = 3,32

(Perry ed.7, Tabel 2-1)

Berat CaO = 10% x 17982 kg = 1798,2 kg

Berat H<sub>2</sub>O = (17982-1798,2) kg = 16183,8 kg

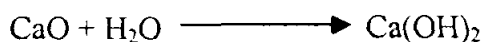
$\rho_{H_2O}$  pada T=30°C = 0,99568 kg/lit

(Geankoplis, 1983)

$\rho_{CaO} = s.g \text{ CaO} \times \rho_{H_2O} = 3,32 \times 0,99568 \text{ kg/lit} = 3,3057 \text{ kg/lit}$

$$\frac{1}{\rho_{campuran}} = \sum \frac{x_i}{\rho_i} = \frac{1798,2}{17982 \cdot 3,3057} + \frac{16183,8}{17982 \cdot 0,99568} = 0,9342$$

$\rho_{campuran} = 1,0705 \text{ kg/lit}$



CaO yang bereaksi =  $\frac{1798,2}{56,08} = 32,0649 \text{ kmol}$

H<sub>2</sub>O yang bereaksi = 32,0649 kmol x 18,016 kg/kmol  
= 577,6812 kg

Ca(OH)<sub>2</sub> yang terbentuk = 32,0649 kmol x 74,096 kg/kmol  
= 2375,8808 kg

Air yang tidak bereaksi = 16183,8 – 577,6812  
= 15606,1188 kg

Masuk :

CaO = 1798,2000 kg

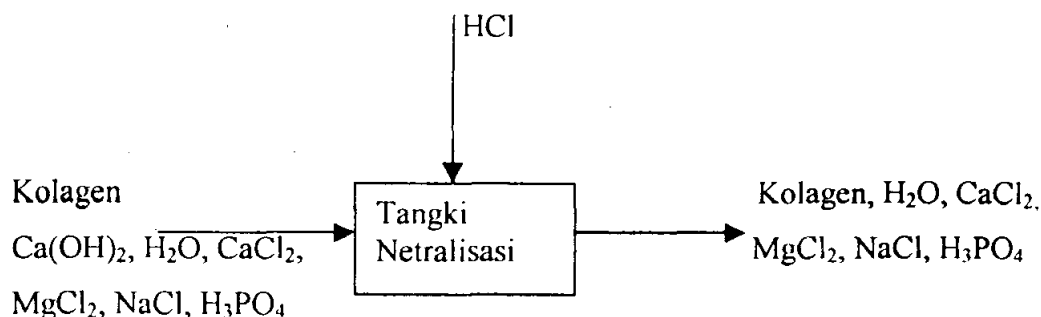
Air =  $\frac{16183,8000 \text{ kg} +}{17982,0000 \text{ kg}}$

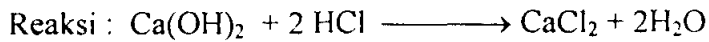
Keluar :

Ca(OH)<sub>2</sub> = 2375,8808 kg

Air =  $\frac{15606,1188 \text{ kg} +}{17982,0000 \text{ kg}}$

### Tangki Netralisasi (F-160)





$$\begin{aligned}\text{Ca(OH)}_2 \text{ yang terbawa masuk ke tangki netralisasi} &= \frac{1057,7912 \text{ kg}}{74,096 \text{ kg / kmol}} \\ &= 14,276 \text{ kmol}\end{aligned}$$

$$\begin{aligned}\text{HCl yang digunakan untuk netralisasi} &= 2 \times 14,276 \text{ kmol} \\ &= 28,552 \text{ kmol} \times 36,465 \text{ kg/kmol} \\ &= 1041,1487 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{H}_2\text{O} &= 2 \times 14,276 \text{ kmol} \\ &= 28,552 \text{ kmol} \times 18,016 \text{ kg/kmol} \\ &= 514,3928 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{CaCl}_2 &= 14,276 \text{ kmol} \times 110,994 \text{ kg/kmol} \\ &= 1584,5503 \text{ kg}\end{aligned}$$

$$\text{HCl 5\% yang digunakan} = \frac{1041,1487}{0,05} = 20822,974 \text{ kg}$$

$$\begin{aligned}\text{Air yang ditambahkan untuk membua HCl 5\%} &= 20822,974 - 1041,1487 \\ &= 19781,8253 \text{ kg}\end{aligned}$$

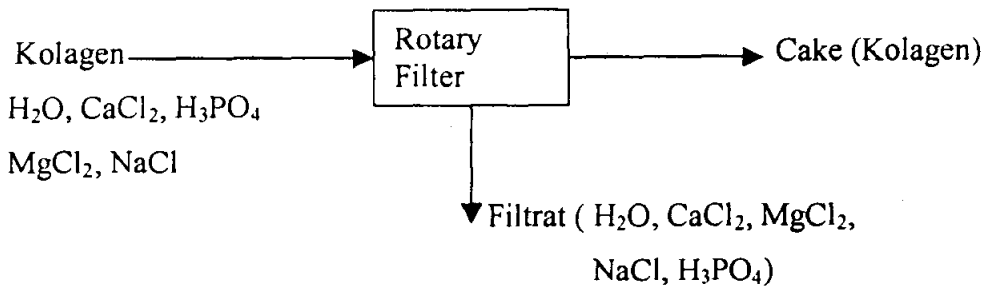
Masuk :

$$\begin{aligned}\text{Kolagen} &= 4495,5000 \text{ kg} \\ \text{H}_2\text{O} &= 19840,8287 \text{ kg} \\ \text{Ca(OH)}_2 &= 1057,7912 \text{ kg} \\ \text{HCl} &= 1041,1487 \text{ kg} \\ \text{CaCl}_2 &= 4,2707 \text{ kg} \\ \text{MgCl}_2 &= 0,1124 \text{ kg} \\ \text{NaCl} &= 0,2248 \text{ kg} \\ \text{H}_3\text{PO}_4 &= \underline{2,4725 \text{ kg} +} \\ &26442,3522 \text{ kg}\end{aligned}$$

Keluar :

$$\begin{aligned}\text{Kolagen} &= 4495,5000 \text{ kg} \\ \text{H}_2\text{O} &= 20355,2215 \text{ kg} \\ \text{CaCl}_2 &= 1588,8210 \text{ kg} \\ \text{MgCl}_2 &= 0,1124 \text{ kg} \\ \text{NaCl} &= 0,2248 \text{ kg} \\ \text{H}_3\text{PO}_4 &= 2,4725 \text{ kg} \\ &\underline{\hspace{1.5cm} +} \\ &26442,3522 \text{ kg}\end{aligned}$$



**Rotary Drum Filter (H-161)**

Data - data : 1. Filtrat keluar dari Rotary Filter telah bebas impurities

2. Massa cake kering yang terkumpul dalam frame 60%-80% berat cake (Ulrich,1984)

3. Massa filtrat yang terikut 20% berat cake.

Massa cake kering = 4495,5000 kg

Massa filtrat terikut cake =  $\frac{20}{80} \times 4495,5 \text{ kg} = 1123,875 \text{ kg}$

Massa cake = 4495,5 kg + 1123,875 kg  
= 5619,375 kg

Massa filtrat = 20355,2215 + 1588,821 + 0,1124 + 0,2248 + 2,4725  
= 21946,8522

Masuk :

Kolagen = 4495,5000 kg

H<sub>2</sub>O = 20355,2215 kg

CaCl<sub>2</sub> = 1588,8210 kg

MgCl<sub>2</sub> = 0,1124 kg

NaCl = 0,2248 kg

H<sub>3</sub>PO<sub>4</sub> = 2,4725 kg

Keluar :

Cake :

Kolagen = 4495,5000 kg

H<sub>2</sub>O = 1042,3694 kg

CaCl<sub>2</sub> = 81,3618 kg

MgCl<sub>2</sub> = 0,0056 kg

NaCl = 0,0112 kg

H<sub>3</sub>PO<sub>4</sub> = 0,1270 kg

Filtrat :

H<sub>2</sub>O = 19312,8521 kg

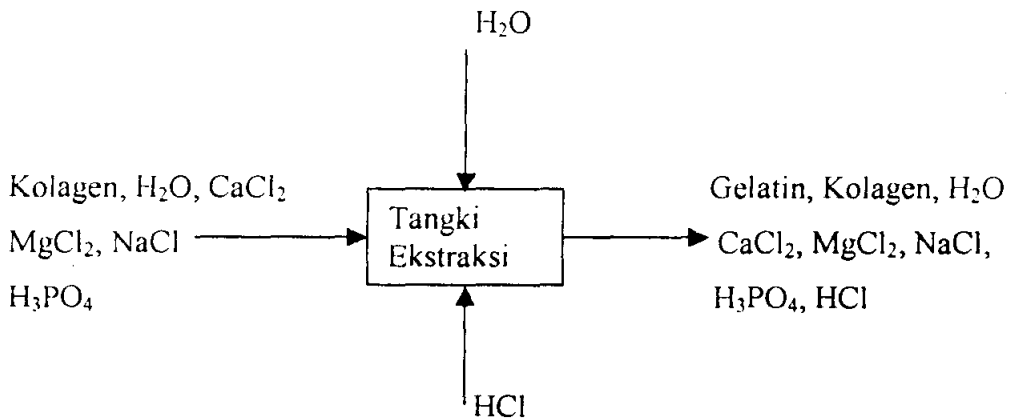
CaCl<sub>2</sub> = 1507,4592 kg

MgCl<sub>2</sub> = 0,1068 kg

NaCl = 0,2136 kg

$$\frac{\quad}{26442,3522 \text{ kg}} + \text{H}_3\text{PO}_4 = \frac{2,3455 \text{ kg}}{26442,3522 \text{ kg}}$$

## II. Unit Ekstraksi



Berat kolagen kering = 4495,5 kg.

Pada ekstraksi ini gelatin terekstrak dari kolagen dengan menggunakan pengeksrak air.

### **Tangki Ekstraksi I (F-210)**

Gelatin yang terekstrak 24% dari bahan kering (Ward, 1977)

$$= 24\% \times 4495,5 \text{ kg} = 1078,92 \text{ kg}$$

Kolagen yang belum terekstrak = (4495,5 – 1078,92) kg

$$= 3416,58 \text{ kg}$$

Air yang dibutuhkan untuk ekstraksi dibanding dengan kolagen kering = 2,3:1 (Ward, 1977)

$$= \frac{2,3}{1} \times 4495,5 \text{ kg} = 10339,65 \text{ kg}$$

Air yang ditambahkan = 10339,65 kg – 1042,3694 kg

$$= 9297,2806 \text{ kg}$$

pH ekstraksi = 2

$$\text{pH} = -\log [\text{H}^+]$$

$$C_{H^+} = C_{HCl} = 0,01 \text{ mol/l} \times 36,465 \text{ gr/mol} \\ = 0,36465 \text{ gr/l}$$

$$\text{Volume air yang dibutuhkan untuk ekstraksi} = \frac{10339,65 \text{ kg}}{1 \text{ kg/l}} = 10339,65 \text{ lt}$$

$$\text{Kebutuhan HCl} = 0,36465 \text{ gr/l} \times 10339,65 \text{ lt} \\ = 3770,3534 \text{ gr} = 3,7703534 \text{ kg}$$

$$\text{HCl 5\% yang digunakan} = \frac{3,7703534}{0,05} = 75,4071 \text{ kg}$$

Masuk :

$$\begin{array}{l} \text{Kolagen} = 4495,5000 \text{ kg} \\ \text{H}_2\text{O} = 10339,6500 \text{ kg} \\ \text{CaCl}_2 = 81,3618 \text{ kg} \\ \text{MgCl}_2 = 0,0056 \text{ kg} \\ \text{NaCl} = 0,0112 \text{ kg} \\ \text{H}_3\text{PO}_4 = 0,1270 \text{ kg} \\ \text{HCl 5\%} = 75,4071 \text{ kg} \\ \hline + \\ 14992,0627 \text{ kg} \end{array}$$

Keluar :

$$\begin{array}{l} \text{Gelatin} = 1078,9200 \text{ kg} \\ \text{Kolagen} = 3416,5800 \text{ kg} \\ \text{CaCl}_2 = 81,3618 \text{ kg} \\ \text{MgCl}_2 = 0,0056 \text{ kg} \\ \text{NaCl} = 0,0112 \text{ kg} \\ \text{H}_2\text{O} = 10339,6500 \text{ kg} \\ \text{H}_3\text{PO}_4 = 0,1270 \text{ kg} \\ \text{HCl 5\%} = \underline{75,4071 \text{ kg} +} \\ 14992,0627 \text{ kg} \end{array}$$

**Tangki Ekstraksi II (F-220)**

Gelatin yang terbentuk 19% dari bahan kering (Ward, 1977)

$$= 19\% \times 3416,5800 \text{ kg} = 649,1502 \text{ kg}$$

$$\text{Kolagen yang belum terekstrak} = (3416,58 - 649,1502) \text{ kg} \\ = 2767,4298 \text{ kg}$$

$$\text{Jumlah gelatin} = 1078,92 \text{ kg} + 649,1502 \text{ kg} \\ = 1728,0702 \text{ kg}$$

Masuk :

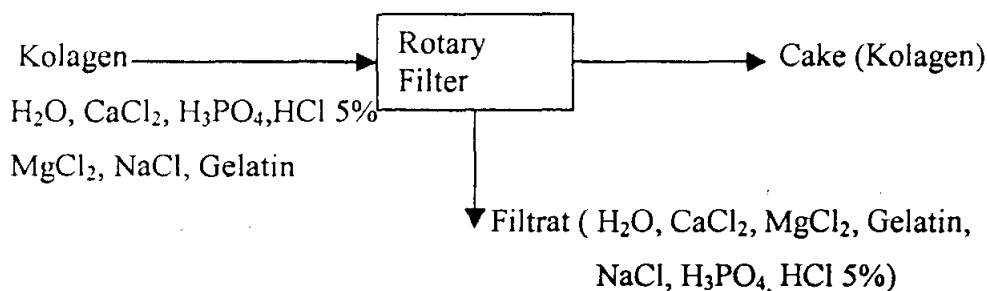
$$\text{Kolagen} = 3416,5800 \text{ kg}$$

Keluar :

$$\text{Gelatin} = 1728,0702 \text{ kg}$$

H <sub>2</sub> O	= 10339,6500 kg	Kolagen	= 2767,4298 kg
CaCl <sub>2</sub>	= 81,3618 kg	CaCl <sub>2</sub>	= 81,3618 kg
MgCl <sub>2</sub>	= 0,0056 kg	MgCl <sub>2</sub>	= 0,0056 kg
NaCl	= 0,0112 kg	NaCl	= 0,0112 kg
H <sub>3</sub> PO <sub>4</sub>	= 0,1270 kg	H <sub>2</sub> O	= 10339,6500 kg
HCl 5%	= 75,4071 kg	H <sub>3</sub> PO <sub>4</sub>	= 0,1270 kg
Gelatin	= <u>1078,9200 kg</u>	HCl 5%	= <u>75,4071 kg +</u>
	14992,0627 kg		14992,0627 kg

### Rotary Drum Filter (H-221)



- Data - data :
1. Filtrat keluar dari Rotary Filter telah bebas impurities
  2. Massa cake kering yang terkumpul dalam frame 60%-80% berat cake (Ulrich,1984)
  3. Massa filtrat yang terikut 20% berat cake.

Massa cake kering = 2767,4298 kg

Massa filtrat terikut cake =  $\frac{20}{80} \times 2767,4298 \text{ kg} = 691,8575 \text{ kg}$

Massa cake = 2767,4298 kg + 691,8575 kg  
= 3459,2873 kg

Masuk :

Kolagen	= 2767,4298 kg
H <sub>2</sub> O	= 10339,6500 kg
CaCl <sub>2</sub>	= 81,3618 kg
MgCl <sub>2</sub>	= 0,0056 kg
NaCl	= 0,0112 kg

Keluar :

Cake	:
Kolagen	= 2767,4298 kg
H <sub>2</sub> O	= 585,1762 kg
CaCl <sub>2</sub>	= 4,6047 kg
MgCl <sub>2</sub>	= 0,0003 kg



$$\text{Volume air yang dibutuhkan untuk ekstraksi} = \frac{6365,0885 \text{ kg}}{1 \text{ kg/l}} = 6365,0885 \text{ lt}$$

$$\begin{aligned} \text{Kebutuhan HCl} &= 0,1153 \text{ gr/lt} \times 6365,0885 \text{ lt} \\ &= 733,8947 \text{ gr} = 0,7338947 \text{ kg} \end{aligned}$$

$$\text{HCl 5\% yang digunakan} = \frac{0,7338947}{0,05} = 14,6779 \text{ kg}$$

$$\begin{aligned} \text{HCl 5\% yang ditambahkan} &= 14,6779 \text{ kg} - 4,2677 \text{ kg} \\ &= 10,4102 \text{ kg} \end{aligned}$$

Masuk :	Keluar :
Kolagen = 2767,4298 kg	Kolagen = 2435,3382 kg
Gelatin = 97,8008 kg	Gelatin = 429,8924 kg
H <sub>2</sub> O = 6365,0885 kg	H <sub>2</sub> O = 6365,0885 kg
CaCl <sub>2</sub> = 4,6047 kg	CaCl <sub>2</sub> = 4,6047 kg
MgCl <sub>2</sub> = 0,0003 kg	MgCl <sub>2</sub> = 0,0003 kg
NaCl = 0,0006 kg	NaCl = 0,0006 kg
H <sub>3</sub> PO <sub>4</sub> = 0,0072 kg	H <sub>3</sub> PO <sub>4</sub> = 0,0072 kg
HCl 5% = <u>14,6779 kg +</u> 9249,6098 kg	HCl 5% = <u>14,6779 kg +</u> 9249,6098 kg

#### Tangki Ekstraksi IV (F-240)

Gelatin yang terbentuk 8 % dari bahan kering (Ward, 1977)

$$= 8 \% \times 2435,3382 \text{ kg} = 194,8271 \text{ kg}$$

$$\text{Jumlah gelatin} = 429,8924 \text{ kg} + 194,8271 \text{ kg} = 624,7195 \text{ kg}$$

$$\begin{aligned} \text{Kolagen yang belum terekstraksi} &= (2435,3382 - 194,8271) \text{ kg} \\ &= 2240,5111 \text{ kg} \end{aligned}$$

Masuk :	Keluar :
Kolagen = 2435,3382 kg	Kolagen = 2240,5111 kg

Gelatin = 429,8924 kg

H<sub>2</sub>O = 6365,0885 kgCaCl<sub>2</sub> = 4,6047 kgMgCl<sub>2</sub> = 0,0003 kg

NaCl = 0,0006 kg

H<sub>3</sub>PO<sub>4</sub> = 0,0072 kgHCl 5% = 14,6779 kg +

9249,6098 kg

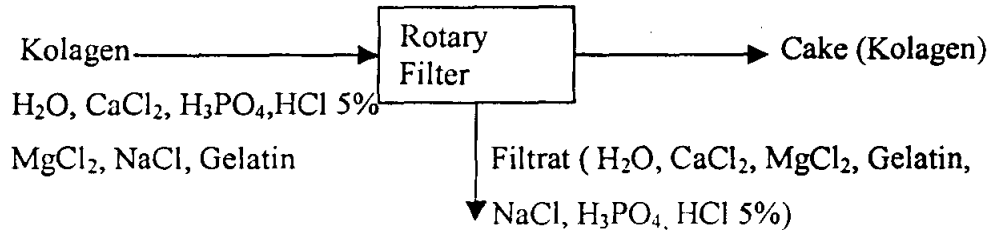
Gelatin = 624,7195 kg

H<sub>2</sub>O = 6365,0885 kgCaCl<sub>2</sub> = 4,6047 kgMgCl<sub>2</sub> = 0,0003 kg

NaCl = 0,0006 kg

H<sub>3</sub>PO<sub>4</sub> = 0,0072 kgHCl 5% = 14,6779 kg +

9249,6098 kg

**Rotary Drum Filter (H-241)**

Data - data : 1. Filtrat keluar dari Rotary Filter telah bebas impurities

2. Massa cake kering yang terkumpul dalam frame 60%-80% berat cake (Ulrich, 1984)

3. Massa filtrat yang terikut 20% berat cake.

Massa cake kering = 2240,5111 kg

$$\text{Massa filtrat terikut cake} = \frac{20}{80} \times 2240,5111 \text{ kg} = 560,1278 \text{ kg}$$

$$\begin{aligned} \text{Massa cake} &= 2240,5111 \text{ kg} + 560,1278 \text{ kg} \\ &= 2800,6389 \text{ kg} \end{aligned}$$

Masuk :

Kolagen = 2240,5111 kg

H<sub>2</sub>O = 6365,0885 kg

Keluar :

Cake :

Kolagen = 2240,51110 kg





$$C_{H^+} = C_{HCl} = 1,5849 \cdot 10^{-3} \text{ mol/l} \times 36,465 \text{ gr/mol} \\ = 0,0578 \text{ gr/l}$$

$$\text{Volume air yang dibutuhkan untuk hidrolisis} = \frac{5153,1755 \text{ kg}}{1 \text{ kg/l}} = 5153,1755 \text{ lt}$$

$$\text{Kebutuhan HCl} = 0,0578 \text{ gr/l} \times 5153,1755 \text{ lt} \\ = 297,8535 \text{ gr} = 0,29978535 \text{ kg}$$

$$\text{HCl 5\% yang digunakan} = \frac{0,29978535}{0,05} = 5,9957 \text{ kg}$$

$$\text{HCl 5\% yang ditambahkan} = 5,9957 \text{ kg} - 1,17298 \text{ kg} = 4,82272 \text{ kg}$$

Masuk :

Keluar :

$$\text{Kolagen} = 2240,51110 \text{ kg}$$

$$\text{Kolagen} = 1926,83950 \text{ kg}$$

$$\text{Gelatin} = 49,92407 \text{ kg}$$

$$\text{Gelatin} = 363,59567 \text{ kg}$$

$$\text{H}_2\text{O} = 5153,17550 \text{ kg}$$

$$\text{H}_2\text{O} = 5153,17550 \text{ kg}$$

$$\text{CaCl}_2 = 0,36798 \text{ kg}$$

$$\text{CaCl}_2 = 0,36798 \text{ kg}$$

$$\text{MgCl}_2 = 0,00002 \text{ kg}$$

$$\text{MgCl}_2 = 0,00002 \text{ kg}$$

$$\text{NaCl} = 0,00005 \text{ kg}$$

$$\text{NaCl} = 0,00005 \text{ kg}$$

$$\text{H}_3\text{PO}_4 = 0,00058 \text{ kg}$$

$$\text{H}_3\text{PO}_4 = 0,00058 \text{ kg}$$

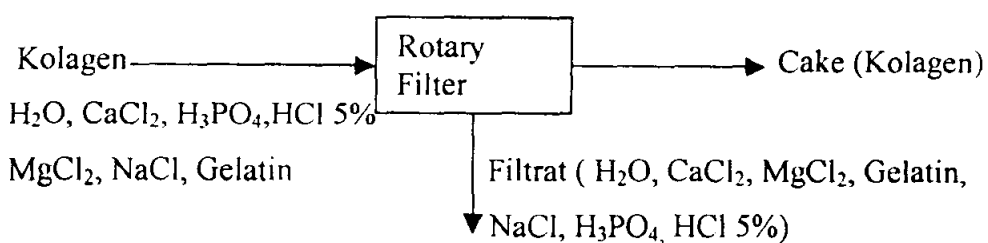
$$\text{HCl 5\%} = \underline{5,99570 \text{ kg} +}$$

$$\text{HCl 5\%} = \underline{5,99570 \text{ kg} +}$$

$$7449,97500 \text{ kg}$$

$$7449,97500 \text{ kg}$$

### Rotary Drum Filter (H-251)



Data - data : 1. Filtrat keluar dari Rotary Filter telah bebas impurities

2. Massa cake kering yang terkumpul dalam frame 60%-80% berat cake (Ulrich,1984)

3. Massa filtrat yang terikut 20% berat cake.

Massa cake kering = 1926,8395 kg

Massa filtrat terikut cake =  $\frac{20}{80} \times 1926,8395 \text{ kg} = 481,7099 \text{ kg}$

Massa cake = 1926,8395 kg + 481,7099 kg  
= 2408,5494 kg

Masuk :

Kolagen = 1926,83950 kg

H<sub>2</sub>O = 5153,17550 kg

CaCl<sub>2</sub> = 0,36798 kg

MgCl<sub>2</sub> = 0,00002 kg

NaCl = 0,00005 kg

H<sub>3</sub>PO<sub>4</sub> = 0,00058 kg

Gelatin = 363,59567 kg

HCl 5% = 5,99570 kg

+

7449,97500 kg

Keluar :

Cake :

Kolagen = 1926,839500 kg

H<sub>2</sub>O = 449,443193 kg

CaCl<sub>2</sub> = 0,032094 kg

MgCl<sub>2</sub> = 0,000002 kg

NaCl = 0,000004 kg

H<sub>3</sub>PO<sub>4</sub> = 0,000051 kg

HCl 5% = 0,522925 kg

Gelatin = 31,711631 kg

Filtrat :

Gelatin = 331,884039 kg

HCl 5% = 5,472775 kg

H<sub>2</sub>O = 4703,732307 kg

CaCl<sub>2</sub> = 0,335886 kg

MgCl<sub>2</sub> = 0,000018 kg

NaCl = 0,000046 kg

H<sub>3</sub>PO<sub>4</sub> = 0,000529 kg +

7449,975000 kg

**Tangki Penampung Hasil Ekstraksi (F-253)**

Berat total larutan gelatin = berat total gelatin + berat impurities + berat total air  
untuk ekstraksi + berat total HCl untuk ekstraksi

$$\begin{aligned}\text{Berat total gelatin} &= 1078,92 + 649,1502 + 332,0916 + 194,8271 + 313,6716 \\ &= 2568,6605 \text{ kg} - 31,711631 \text{ kg} \\ &= 2536,948869 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Berat impurities} &= (81,3618 + 0,0056 + 0,0112 + 0,127) - (0,032094 + 0,000002 \\ &\quad + 0,000004 + 0,000051) \\ &= 81,473449 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Berat total H}_2\text{O} &= 10339,65 + 5779,9123 + 4644,51338 - 449,443193 \\ &= 20314,63249 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Berat HCl 5\%} &= 75,4071 + 10,4102 + 4,82272 - 0,522925 \\ &= 90,117095 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Berat total larutan gelatin} &= 2536,948869 \text{ kg} + 81,473449 \text{ kg} + 20314,63249 \text{ kg} + \\ &\quad 90,117095 \text{ kg} \\ &= 23023,1719 \text{ kg}\end{aligned}$$

Masuk :

Keluar :

Dari rotary filter H-221 :

Larutan gelatin = 23023,1719 kg

Gelatin = 1630,2694 kg

HCl 5% = 71,1394 kg

H<sub>2</sub>O = 9754,4738 kg

CaCl<sub>2</sub> = 76,7571 kg

MgCl<sub>2</sub> = 0,0053 kg

NaCl = 0,0106 kg

H<sub>3</sub>PO<sub>4</sub> = 0,1198 kg

Dari rotary filter H-241 :

Gelatin = 574,79543 kg

HCl 5% = 13,50492 kg

H<sub>2</sub>O = 5856,42638 kg

$$\text{CaCl}_2 = 4,23672 \text{ kg}$$

$$\text{MgCl}_2 = 0,00028 \text{ kg}$$

$$\text{NaCl} = 0,00055 \text{ kg}$$

$$\text{H}_3\text{PO}_4 = 0,00662 \text{ kg}$$

Dari rotary filter H-251 :

$$\text{Gelatin} = 331,884039 \text{ kg}$$

$$\text{HCl 5\%} = 5,472775 \text{ kg}$$

$$\text{H}_2\text{O} = 4703,732307 \text{ kg}$$

$$\text{CaCl}_2 = 0,335886 \text{ kg}$$

$$\text{MgCl}_2 = 0,000018 \text{ kg}$$

$$\text{NaCl} = 0,000046 \text{ kg}$$

$$\begin{array}{rcl} \text{H}_3\text{PO}_4 & = & \underline{0,000529 \text{ kg} +} \\ & & 23023,1719 \text{ kg} \end{array} \qquad \qquad \qquad \underline{\qquad \qquad \qquad +} \qquad \qquad \qquad 23023,1719 \text{ kg}$$

### Tangki HCl 5% (F-110)

HCl yang dibutuhkan adalah 5%, sedangkan HCl yang tersedia adalah 36% jadi dibutuhkan tangki pengenceran HCl.

Data-data :

$$\rho_{\text{HCl 36\%}} \text{ pada } 20^\circ\text{C} = 1,1789 \text{ kg/lit}$$

$$\rho_{\text{HCl 5\%}} \text{ pada } 20^\circ\text{C} = 1,023 \text{ kg/lit}$$

HCl murni yang digunakan untuk proses :

$$\text{Demineralisasi} = 6387,3335 \text{ kg}$$

$$\text{Netralisasi} = 1041,1487 \text{ kg}$$

$$\begin{array}{rcl} \text{Ekstraksi} & = & \underline{4,5320 \text{ kg} +} \\ & & 7433,0142 \text{ kg} \end{array}$$

$$\text{HCl 36\% yang dibutuhkan} = \frac{7433,0142 \text{ kg}}{0,36}$$

$$= 20647,2617 \text{ kg}$$

$$\text{HCl 5\% yang dibutuhkan} = \frac{7433,0142 \text{ kg}}{0,05}$$



Dari persamaan (1) :

$$F = L + V$$

$$23023,1719 = 7249,0101 + V$$

$$V = 15774,1618 \text{ kg}$$

Masuk :

Feed larutan gelatin = 23023,1719 kg

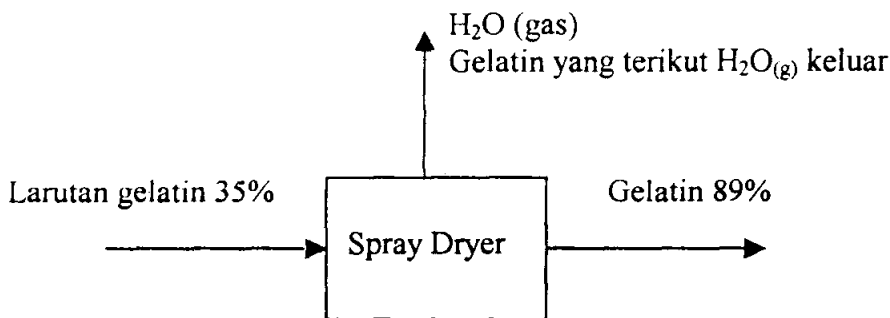
$$\begin{array}{r} \text{---} \\ + \\ \text{---} \\ 23023,1719 \text{ kg} \end{array}$$

Keluar :

Larutan gelatin = 7249,0101 kg

$$\begin{array}{r} \text{---} \\ + \text{ Uap} \\ \text{---} \\ = 15774,1618 \text{ kg} + \\ 23023,1719 \text{ kg} \end{array}$$

### Spray Dryer (B-330)



Effisiensi spray dryer = 99 %

(Perry 6<sup>th</sup>)

Larutan gelatin dengan kadar air 65% masuk spray dryer terdiri dari :

Gelatin = 2536,948869 kg

H<sub>2</sub>O = 4712,061231 kg

Gelatin keluar dari dryer dengan kadar air 11% :

Gelatin keluar = 99% . Gelatin masuk

$$= 99\% \cdot 2536,948869 \text{ kg}$$

$$= 2511,57938 \text{ kg}$$

H<sub>2</sub>O yang terkandung dalam gelatin :

$$= \frac{11\%}{(100 - 11)\%} \times 2511,57938 \text{ kg}$$

$$= 310,4199 \text{ kg}$$

Gelatin yang terikut H<sub>2</sub>O<sub>(gas)</sub> :

$$= (2536,948869 - 2511,57938) \text{ kg}$$

$$= 25,369489 \text{ kg}$$

H<sub>2</sub>O yang terkandung dalam gelatin :

$$= \frac{11\%}{(100 - 11)\%} \times 25,369489 \text{ kg}$$

$$= 3,1356 \text{ kg}$$

$$\text{H}_2\text{O yang menguap (H}_2\text{O gas)} = (4712,061231 - 310,4199 - 3,1356) \text{ kg}$$

$$= 4398,505731 \text{ kg}$$

Masuk :

$$\text{Gelatin} = 2536,948869 \text{ kg}$$

$$\text{H}_2\text{O} = 4712,061231 \text{ kg}$$

$$\underline{\hspace{2cm} + \hspace{2cm}}$$

$$7249,0101 \text{ kg}$$

Keluar :

$$\text{Gelatin} = 2511,579380 \text{ kg}$$

$$\text{H}_2\text{O} = 310,419900 \text{ kg}$$

$$\text{Gelatin yang terikut H}_2\text{O}_{(\text{gas})} :$$

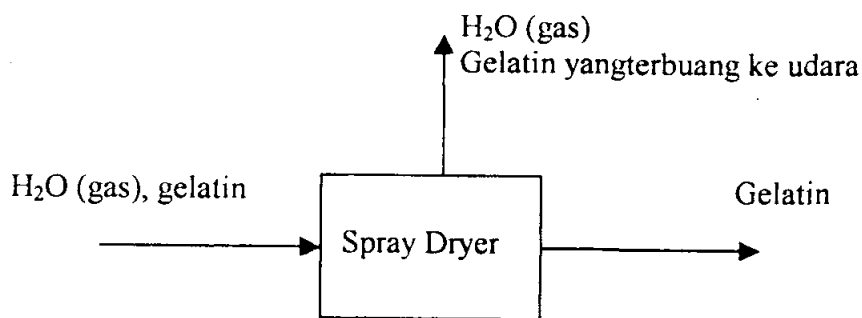
$$= 25,369489 \text{ kg}$$

$$\text{H}_2\text{O} = 3,135600 \text{ kg}$$

$$\text{H}_2\text{O}_{(\text{gas})} = \underline{4398,505731 \text{ kg} +}$$

$$7249,0101 \text{ kg}$$

### Cyclone (H-333)



$$\text{Effisiensi cyclone} = 95\%$$

(Van't land)

Bahan masuk :

$$\text{H}_2\text{O gas} = 4398,505731 \text{ kg}$$

$$\text{Gelatin} = 28,505089 \text{ kg}$$

$$\text{Gelatin keluar} = 95\% \cdot 28,505089 \text{ kg}$$

$$= 27,079835 \text{ kg}$$

$$\begin{aligned}\text{Gelatin yang terbang ke udara} &= (28,505089 - 27,079835) \text{ kg} \\ &= 1,425254 \text{ kg}\end{aligned}$$

$$\text{H}_2\text{O gas yang terbang ke udara} = 4398,505731 \text{ kg}$$

Masuk :

$$\text{Gelatin} = 28,505089 \text{ kg}$$

$$\text{H}_2\text{O gas} = 4398,505731 \text{ kg}$$

$$\begin{array}{r} \hline + \\ 4427,01082 \text{ kg} \end{array}$$

Keluar :

$$\text{Gelatin} = 27,079835 \text{ kg}$$

$$\begin{aligned}\text{Gelatin yang terbang ke udara :} \\ &= 1,425254 \text{ kg}\end{aligned}$$

$$\begin{array}{r} \text{H}_2\text{O}_{(\text{gas})} = \underline{4398,505731 \text{ kg} +} \\ 4427,01082 \text{ kg} \end{array}$$



## APPENDIKS B.

### PERHITUNGAN NERACA PANAS

## APPENDIX B

### PERHITUNGAN NERACA PANAS

Kapasitas produksi = 13500 kg/hari

Suhu referensi = 25 °C

Menentukan data-data  $\Delta H_f$  untuk setiap komponen yang didapat dari Himmelblau adalah :

$\Delta H_f \text{Ca}_3(\text{PO}_4)_2 = -988,4726 \text{ kkal/mol}$

$\Delta H_f \text{HCl} = -25,9113 \text{ kkal/mol}$

$\Delta H_f \text{CaCl}_2 = -189,9857 \text{ kkal/mol}$

$\Delta H_f \text{CaCO}_3 = -288,3284 \text{ kkal/mol}$

$\Delta H_f \text{H}_2\text{O} = -68,2782 \text{ kkal/mol}$

$\Delta H_f \text{CO}_2 = -94,0520 \text{ kkal/mol}$

$\Delta H_f \text{Mg}_3(\text{PO}_4)_2 = -961,5 \text{ kkal/mol}$

$\Delta H_f \text{MgCl}_2 = -153,333 \text{ kkal/mol}$

$\Delta H_f \text{H}_3\text{PO}_4 = -306,0309 \text{ kkal/mol}$

$\Delta H_f \text{Na}_2\text{CO}_3 = -269,9570 \text{ kkal/mol}$

$\Delta H_f \text{NaCl} = -98,1879 \text{ kkal/mol}$

$\Delta H_f \text{Ca}(\text{OH})_2 = -235,7950 \text{ kkal/mol}$

Menghitung  $C_p$  dari setiap komponen yang didapat dari App.E hal 676 dan hukum kopps hal 386 :

Komponen	Solid	Liquid
H	2,3	4,3
P	5,4	7,4
O	4,0	6,0
N	4,4	
F	5,0	7,0
C	1,8	2,8
Lainnya	6,2	8,0

Dimana  $C_p$  dari  $\text{Ca}_3(\text{PO}_4)_2$ ,  $\text{Mg}_3(\text{PO}_4)_2$ ,  $\text{Ca}(\text{OH})_2$ ,  $\text{HCl}$ , tulang ( $\text{C}_{102}\text{H}_{149}\text{N}_{31}\text{O}_{38}$ ), dan  $\text{H}_3\text{PO}_4$  menggunakan rumus hokum kopps:

$C_p \text{Ca}_3(\text{PO}_4)_2 : \text{Ca} = 6,2 \text{ kal/gmol}^\circ\text{C}$

$$P = 5,4 \text{ kal/gmo}$$

$$O = 4 \text{ kal/gmol}^{\circ}\text{C}$$

$$\begin{aligned}\text{Cp Ca}_3(\text{PO}_4)_2 &= (3 \times 6,2) + (2 \times 5,4) + (8 \times 4) \\ &= 61,4 \text{ kal/gmol }^{\circ}\text{C} \\ &= 0,1981 \text{ kal/gr }^{\circ}\text{C} \\ &= 0,1981 \text{ kkal/kg }^{\circ}\text{C}\end{aligned}$$

$$\text{Cp Mg}_3(\text{PO}_4)_2 : \text{Mg} = 6,2 \text{ kal/gmol }^{\circ}\text{C}$$

$$P = 5,4 \text{ kal/gmol }^{\circ}\text{C}$$

$$O = 4 \text{ kal/gmol }^{\circ}\text{C}$$

$$\begin{aligned}\text{Cp Mg}_3(\text{PO}_4)_2 &= (3 \times 6,2) + (2 \times 5,4) + (8 \times 4) \\ &= 61,4 \text{ kal/gmol }^{\circ}\text{C} \\ &= 0,2344 \text{ kal/gr }^{\circ}\text{C} \\ &= 0,2344 \text{ kkal/kg }^{\circ}\text{C}\end{aligned}$$

$$\text{Cp Ca(OH)}_2 : \text{Ca} = 8 \text{ kal/gmol }^{\circ}\text{C}$$

$$O = 6 \text{ kal/gmol }^{\circ}\text{C}$$

$$H = 4,3 \text{ kal/gmol }^{\circ}\text{C}$$

$$\begin{aligned}\text{Cp Ca(OH)}_2 &= (8 \times 1) + (6 \times 2) + (4,3 \times 2) \\ &= 28,6 \text{ kal/gmol }^{\circ}\text{C} \\ &= 0,2860 \text{ kkal/kg }^{\circ}\text{C}\end{aligned}$$

$$\text{Cp HCl 36\%} = 0,5818 \text{ kkal/kg }^{\circ}\text{C}$$

$$\text{Cp HCl 5\%} = 0,8377 \text{ kkal/kg }^{\circ}\text{C}$$

$$\text{Cp H}_3\text{PO}_4 : \text{H} = 4,3 \text{ kal/gmol }^{\circ}\text{C}$$

$$P = 7,4 \text{ kal/gmol }^{\circ}\text{C}$$

$$O = 6 \text{ kal/gmol }^{\circ}\text{C}$$

$$\begin{aligned}\text{Cp H}_3\text{PO}_4 &= (3 \times 4,3) + 7,4 + (4 \times 6) \\ &= 44,3 \text{ kal/gmol }^{\circ}\text{C} \\ &= 0,4520 \text{ kkal/kg }^{\circ}\text{C}\end{aligned}$$

$$\text{Cp tulang (C}_{102}\text{H}_{149}\text{N}_{31}\text{O}_{38}\text{)} : \quad \text{C} = 1,8 \text{ kal/gmol } ^\circ\text{C}$$

$$\text{H} = 2,3 \text{ kal/gmol } ^\circ\text{C}$$

$$\text{O} = 4 \text{ kal/gmol } ^\circ\text{C}$$

$$\text{N} = 4,4 \text{ kal/gmol } ^\circ\text{C}$$

$$\begin{aligned} \text{Cp tulang} &= (102 \times 1,8) + (149 \times 2,3) + (4,4 \times 31) + (38 \times 4) \\ &= 814,7 \text{ kal/gmol } ^\circ\text{C} \\ &= 0,3369 \text{ kkal/kg } ^\circ\text{C} \end{aligned}$$

$$\text{Cp MgCl}_2 : \text{Mg} = 8 \text{ kal/gmol } ^\circ\text{C}$$

$$\text{Cl} = 8 \text{ kal/gmol } ^\circ\text{C}$$

$$\begin{aligned} \text{Cp MgCl}_2 &= 8 + (8 \times 2) \\ &= 24 \text{ kal/gmol } ^\circ\text{C} \\ &= 0,2520 \text{ kkal/kg } ^\circ\text{C} \end{aligned}$$

$$\text{Cp CaCl}_2 : \quad \text{Ca} = 8 \text{ kal/gmol } ^\circ\text{C}$$

$$\text{Cl} = 8 \text{ kal/gmol } ^\circ\text{C}$$

$$\begin{aligned} \text{Cp CaCl}_2 &= 8 + (8 \times 2) \\ &= 24 \text{ kal/gmol } ^\circ\text{C} \\ &= 0,2162 \text{ kkal/kg } ^\circ\text{C} \end{aligned}$$

$$\text{Cp NaCl} : \quad \text{Na} = 8 \text{ kal/gmol } ^\circ\text{C}$$

$$\text{Cl} = 8 \text{ kal/gmol } ^\circ\text{C}$$

$$\begin{aligned} \text{Cp NaCl} &= 8 + 8 \\ &= 16 \text{ kal/gmol } ^\circ\text{C} \\ &= 0,2737 \text{ kkal/kg } ^\circ\text{C} \end{aligned}$$

Cp untuk  $\text{CO}_2$ ,  $\text{CaCO}_3$  dan  $\text{Na}_2\text{CO}_3$  dihitung dengan persamaan Cp :

$$\text{Untuk Cp CO}_2 \text{ dimana : } a = 36,11 \quad c = -2,887 \cdot 10^{-5}$$

$$b = 4,233 \cdot 10^{-2} \quad d = 7,464 \cdot 10^{-9}$$

$$\text{Cp CO}_2 = a + \frac{b}{2} \times (T_2 + T_1) + \frac{c}{3} \times [T_2^2 + (T_2 \cdot T_1) + T_1^2] + \frac{d}{4} \times (T_2^2 + T_1^2)(T_2 + T_1)$$

$$= 37,2524 \text{ J/gmol } ^\circ\text{C}$$

$$= 0,2023 \text{ kkal/kg } ^\circ\text{C}$$

Untuk  $C_p \text{ CaCO}_3$  dimana :  $a = 82,34$        $c = -12,87 \cdot 10^5$   
 $b = 4,975 \cdot 10^{-2}$

$$C_p \text{ CaCO}_3 = a + \frac{b}{2} (T_2 + T_1) + \frac{c}{T_2 \cdot T_1}$$

$$= 83,3037 \text{ J/gmol } ^\circ\text{C}$$

$$= 0,1989 \text{ kkal/kg } ^\circ\text{C}$$

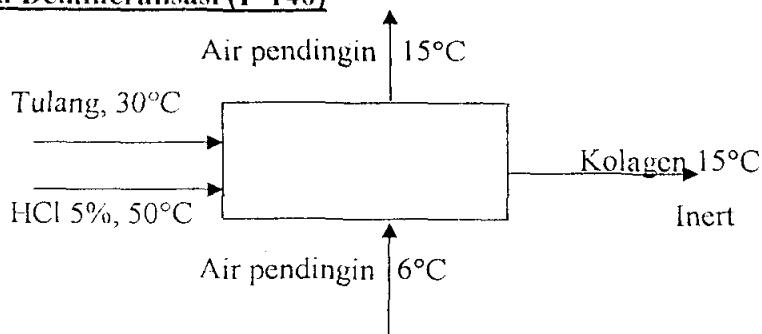
Untuk  $C_p \text{ Na}_2\text{CO}_3$  dimana :  $a = 121$

$$C_p \text{ Na}_2\text{CO}_3 = a + \frac{b}{2} (T_2 + T_1) + \frac{c}{3} [T_2^2 + (T_2 \cdot T_1) + T_1^2] + \frac{d}{4} (T_2^2 + T_1^2) (T_2 + T_1)$$

$$= 121 \text{ J/gmol } ^\circ\text{C}$$

$$= 0,2729 \text{ kkal/kg } ^\circ\text{C}$$

#### Tangki Demineralisasi (F-140)



$$\Delta H_{in} + \Delta H_r = \Delta H_{out} + \Delta H_{air \text{ pendingin}}$$

Dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

#### Menghitung panas dari bahan masuk ( $\Delta H_{in}$ )

Suhu bahan masuk =  $30^\circ\text{C}$

$\text{Ca}_3(\text{PO}_4)_2 = 7742,2500 \text{ kg/hari}$

$\text{Na}_2\text{CO}_3 = 465,7500 \text{ kg/hari}$

$\text{Mg}_3(\text{PO}_4)_2 = 276,7500 \text{ kg/hari}$

$\text{CaCO}_3 = 519,7500 \text{ kg/hari}$

$$\text{Kolagen} = 4495,5000 \text{ kg/hari}$$

$$\Delta H_{\text{in}} = \Delta H(\text{Ca}_3(\text{PO}_4)_2) + \Delta H(\text{Na}_2\text{CO}_3) + \Delta H(\text{Mg}_3(\text{PO}_4)_2) + \Delta H(\text{CaCO}_3) + \Delta H_{\text{kolagen}} + \Delta H_{\text{HCl}}$$

$$\begin{aligned}\Delta H(\text{Ca}_3(\text{PO}_4)_2) &= m \cdot C_p \cdot (T_1 - 25) \\ &= 7742,2500 \cdot 0,1981 \cdot (30 - 25) \\ &= 7668,698625 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{Na}_2\text{CO}_3) &= m \cdot C_p \cdot (T_1 - 25) \\ &= 465,7500 \cdot 0,2729 \cdot (30 - 25) \\ &= 635,515875 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{Mg}_3(\text{PO}_4)_2) &= m \cdot C_p \cdot (T_1 - 25) \\ &= 276,7500 \cdot 0,2344 \cdot (30 - 25) \\ &= 324,35100 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{CaCO}_3) &= m \cdot C_p \cdot (T_1 - 25) \\ &= 519,7500 \cdot 0,1989 \cdot (30 - 25) \\ &= 516,891375 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 4495,5000 \cdot 0,3369 \cdot (30 - 25)\end{aligned}$$

$$\Delta H_{\text{kolagen}} = 7572,66975 \text{ kkal}$$

Untuk larutan HCl 5% masuk pada suhu 50°C :

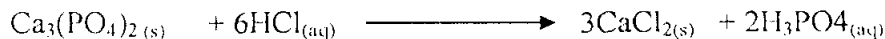
$$\begin{aligned}\Delta H_{\text{HCl}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= (5460,9328 + 375,6333 + 230,2984 + 320,4690) \cdot 0,8377 \cdot (50 - 25) \\ &= 133766,7318 \text{ kkal}\end{aligned}$$

$$\Delta H_{in} = 7668,698625 + 133766,7318 + 516,891375 + 324,351 + 635,515875 \\ + 7572,66975$$

$$\Delta H_{in} = 150484,8584 \text{ kkal/hari}$$

Menghitung panas reaksi dari setiap reaksi yang ada ( $\Delta H_r$ ) :

- untuk reaksi 1 :



$$\text{mol Ca}_3(\text{PO}_4)_2 = 24,9597 \text{ kmol/hari}$$

$$\text{mol HCl} = 149,7582 \text{ kmol/hari}$$

$$\text{mol CaCl}_2 = 74,8791 \text{ kmol/hari}$$

$$\text{mol H}_3\text{PO}_4 = 49,9194 \text{ kmol/hari}$$

$$\Delta H_{r1} = ((3 \times 74,8791 \times -189,9857) + (2 \times 49,9194 \times -306,0309)) \\ - ((24,9597 \times -988,4726) + (6 \times 149,7582 \times -25,9113)) \\ = -25277,0751 \text{ kkal/hari}$$

- untuk reaksi 2 :



$$\text{mol CaCO}_3 = 5,1506 \text{ kmol/hari}$$

$$\text{mol HCl} = 10,3012 \text{ kmol/hari}$$

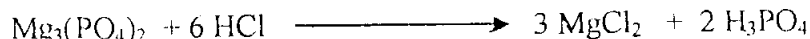
$$\text{mol CaCl}_2 = 5,1506 \text{ kmol/hari}$$

$$\text{mol H}_2\text{O} = 5,1506 \text{ kmol/hari}$$

$$\text{mol CO}_2 = 5,1506 \text{ kmol/hari}$$

$$\Delta H_{r2} = ((5,1506 \times -189,9857) + (5,1506 \times -68,2782) + (5,1506 \times -94,0520)) \\ - ((5,1506 \times -288,3284) + (6 \times 10,3012 \times -25,9113)) \\ = 204,2609 \text{ kkal/hari}$$

- untuk reaksi 3 :



$$\text{mol Mg}_3(\text{PO}_4)_2 = 1,0526 \text{ kmol/hari}$$

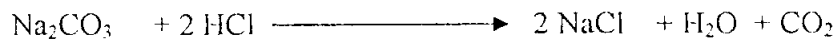
$$\text{mol HCl} = 6,3156 \text{ kmol/hari}$$

$$\text{mol MgCl}_2 = 3,1578 \text{ kmol/hari}$$

$$\text{mol H}_3\text{PO}_4 = 2,1052 \text{ kmol/hari}$$

$$\begin{aligned}\Delta H_{r3} &= ((3 \times 3,1578 \times -153,333) + (2 \times 2,1052 \times -306,0309)) \\ &\quad - ((6 \times 6,3156 \times -25,9113) + (1,0526 \times -961,5)) \\ &= -3324,175009 \text{ kkal/hari}\end{aligned}$$

- untuk reaksi 4 :



$$\text{mol Na}_2\text{CO}_3 = 4,3942 \text{ kmol/hari}$$

$$\text{mol HCl} = 8,7884 \text{ kmol/hari}$$

$$\text{mol NaCl} = 8,7884 \text{ kmol/hari}$$

$$\text{mol H}_2\text{O} = 4,3942 \text{ kmol/hari}$$

$$\text{mol CO}_2 = 4,3942 \text{ kmol/hari}$$

$$\begin{aligned}\Delta H_{r4} &= ((2 \times 8,7884 \times -98,1879) + (4,3942 \times -189,9857) + (4,3942 \times \\ &\quad -68,2782)) - ((4,3942 \times -269,9570) + (2 \times 8,7884 \times -25,9113)) \\ &= -797,4576583 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_r &= -25277,0751 + 204,2609 + -3324,175009 + -797,4576583 \\ &= -29194,4468 \text{ kkal/hari}\end{aligned}$$

Menghitung panas dari bahan keluar ( $\Delta H_{out}$ ) :

$$\text{Suhu bahan keluar} = 15^\circ\text{C}$$

$$\text{CaCl}_2 = 8882,8165 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 5098,3588 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 171,9591 \text{ kg/hari}$$

$$\text{CO}_2 = 420,0762 \text{ kg/hari}$$

$$\text{MgCl}_2 = 300,7299 \text{ kg/hari}$$

$$\text{NaCl} = 513,6644 \text{ kg/hari}$$

$$\text{Kolagen} = 4495,5000 \text{ kg/hari}$$

$$\begin{aligned}\Delta H_{out} &= \Delta H(\text{CaCl}_2) + \Delta H(\text{H}_3\text{PO}_4) + \Delta H(\text{H}_2\text{O}) + \Delta H(\text{CO}_2) + \Delta H(\text{MgCl}_2) \\ &\quad + \Delta H(\text{NaCl}) + \Delta H_{\text{kolagen}}\end{aligned}$$



$$\begin{aligned}\Delta H(\text{CaCl}_2) &= m \cdot C_p \cdot (T_2 - 25) \\ &= 8882,8165 \cdot 0,2162 \cdot (15 - 25) \\ &= -19204,6493 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{H}_3\text{PO}_4) &= m \cdot C_p \cdot (T_2 - 25) \\ &= 5098,3588 \cdot 0,452 \cdot (15 - 25) \\ &= -23044,5818 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{H}_2\text{O}) &= m \cdot C_p \cdot (T_2 - 25) \\ &= 171,9591 \cdot 1 \cdot (15 - 25) \\ &= -1719,591 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{CO}_2) &= m \cdot C_p \cdot (T_2 - 25) \\ &= 420,0762 \cdot 0,2033 \cdot (15 - 25) \\ &= -854,015 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{MgCl}_2) &= m \cdot C_p \cdot (T_2 - 25) \\ &= 300,7299 \cdot 0,2520 \cdot (15 - 25) \\ &= -757,8393 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H(\text{NaCl}) &= m \cdot C_p \cdot (T_2 - 25) \\ &= 513,6644 \cdot 0,2737 \cdot (15 - 25) \\ &= -1405,8995 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 4495,5000 \cdot 0,3369 \cdot (15 - 25) \\ &= -15145,3395 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{out}} &= -19204,6493 + -23044,5818 + -854,015 + -757,8393 + -1405,8995 \\ &\quad + -1719,591 + -15145,3395 \\ &= -62131,9154 \text{ kkal/hari}\end{aligned}$$

$$\Delta H_{\text{in}} + \Delta H_{\text{r}} = \Delta H_{\text{out}} + \Delta H_{\text{air pendingin}}$$

$$150484,8584 + 29194,4468 + 62131,9154 = \Delta H_{\text{air pendingin}}$$

$$\Delta H_{\text{air pendingin}} = 241811,2206 \text{ kkal/hari}$$

Suhu air masuk = 6 °C

Suhu air keluar = 15 °C

$$\begin{aligned} \text{Massa air pendingin yang dibutuhkan} &= \frac{241811,2206}{1 \times (15 - 6)} \\ &= 26867,9134 \text{ kg/hari} \end{aligned}$$

Masuk :

Panas masuk = 150484,8584 kkal/hari

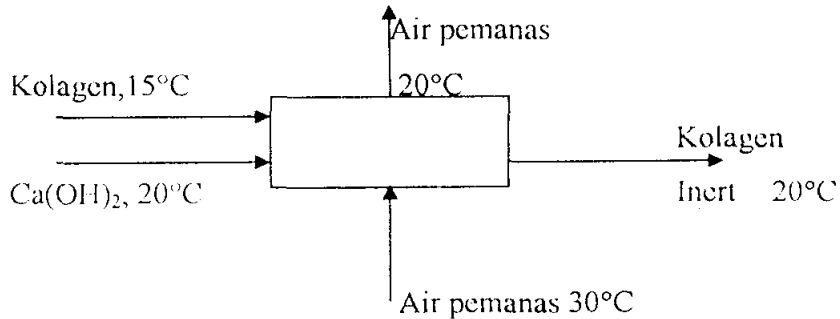
Panas reaksi = 29194,4468 kkal/hari

Panas keluar = 62131,9154 kkal/hari  
= 241811,2206 kkal/hari

Keluar :

Qpendingin = 241811,2206 kkal/hari  
= 241811,2206 kkal/hari

### Tangki Liming (F-150)



### Menghitung panas masuk ( $\Delta H_{\text{in}}$ ) :

Suhu bahan masuk = 15°C

Kolagen = 4495,5000 kg/hari

H<sub>2</sub>O = 1001,9346 kg/hari

CaCl<sub>2</sub> = 73,1643 kg/hari

MgCl<sub>2</sub> = 2,4725 kg/hari

NaCl = 4,2707 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 42,0329 kg/hari

$$\Delta H_{in} = \Delta H_{kolagen} + \Delta H_{H_2O} + \Delta H_{Ca(OH)_2} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4}$$

$$\begin{aligned}\Delta H_{kolagen} &= m.Cp.(T1-25) \\ &= 4495,5000 \cdot 0,3369 \cdot (15-25) \\ &= -15145,3395 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{H_2O} &= m.Cp.(T1-25) \\ &= 1001,9346 \cdot 1 \cdot (15-25) \\ &= -10019,346 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{HCl} &= 1041,1487 \cdot 0,8377 \cdot (15-25) \\ &= -8721,7027 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{CaCl_2} &= 4,2707 \cdot 0,2162 \cdot (15-25) \\ &= -9,2333 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{MgCl_2} &= 0,1124 \cdot 0,252 \cdot (15-25) \\ &= -0,2832 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{NaCl} &= 0,2248 \cdot 0,2737 \cdot (15-25) \\ &= -0,6153 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{H_3PO_4} &= 2,4725 \cdot 0,452 \cdot (15-25) \\ &= -11,1757 \text{ kkal/hari}\end{aligned}$$

Untuk  $Ca(OH)_2$  masuk pada suhu  $20^\circ C$  :

$$\begin{aligned}Ca(OH)_2 &= 17982 \text{ kg/hari} \\ \Delta H_{Ca(OH)_2} &= 1057,7912 \cdot 0,286 \cdot (20-25) \\ &= -1512,6414 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{in} &= -15145,3395 + -10019,346 + -8721,7027 + -1512,6414 + -9,2333 \\ &\quad + -0,2832 + -0,6153 + -11,1757 \\ &= -33907,6957 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 20°C

Kolagen = 4495,5000 kg/hari

H<sub>2</sub>O = 1001,9346 kg/hari

Ca(OH)<sub>2</sub> = 17982 kg/hari

CaCl<sub>2</sub> = 73,1643 kg/hari

MgCl<sub>2</sub> = 2,4725 kg/hari

NaCl = 4,2707 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 42,0329 kg/hari

$$\Delta H_{out} = \Delta H_{H_2O} + \Delta H_{kolagen} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4} + \Delta H_{Ca(OH)_2}$$

$$\begin{aligned}\Delta H_{kolagen} &= m.Cp.(T1-25) \\ &= 4495,5000 \cdot 0,3369 \cdot (20-25) \\ &= -7572,6698 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{H_2O} &= m.Cp.(T1-25) \\ &= 1001,9346 \cdot 1 \cdot (20-25) \\ &= -5009,673 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{HCl} &= 1041,1487 \cdot 0,8377 \cdot (20-25) \\ &= -4360,8513 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{Ca(OH)_2} &= 1057,7912 \cdot 0,286 \cdot (20-25) \\ &= -1512,6414 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{CaCl_2} &= 4,2707 \cdot 0,2162 \cdot (20-25) \\ &= -4,6166 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{MgCl}_2} &= 0,1124 \cdot 0,252 \cdot (20-25) \\ &= -0,1416 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{NaCl}} &= 0,2248 \cdot 0,2737 \cdot (20-25) \\ &= -0,3076 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_3\text{PO}_4} &= 2,4725 \cdot 0,452 \cdot (20-25) \\ &= -5,5879 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{out}} &= -7572,6698 + -5009,673 + -4360,8513 + -1512,6414 + -4,6166 + -0,1416 \\ &\quad + -0,3076 + -5,5879 \\ &= -18466,4892 \text{ kkal/hari}\end{aligned}$$

$$\Delta H_{\text{in}} + Q_{\text{pemanas}} = \Delta H_{\text{out}}$$

dimana :  $\Delta H_{\text{in}}$  = panas dari bahan masuk

$\Delta H_{\text{out}}$  = panas dari bahan keluar

$$\Delta H_{\text{in}} + Q_{\text{pemanas}} = \Delta H_{\text{out}}$$

$$18466,4892 + Q_{\text{pemanas}} = 33907,6957$$

$$Q_{\text{pemanas}} = 15441,2065 \text{ kkal/hari}$$

Suhu air masuk = 30°C

Suhu air keluar = 20°C

$$\begin{aligned}\text{Massa air pendingin yang dibutuhkan} &= \frac{15441,2065}{1 \cdot (30 - 20)} \\ &= 1544,1207 \text{ kg/hari}\end{aligned}$$

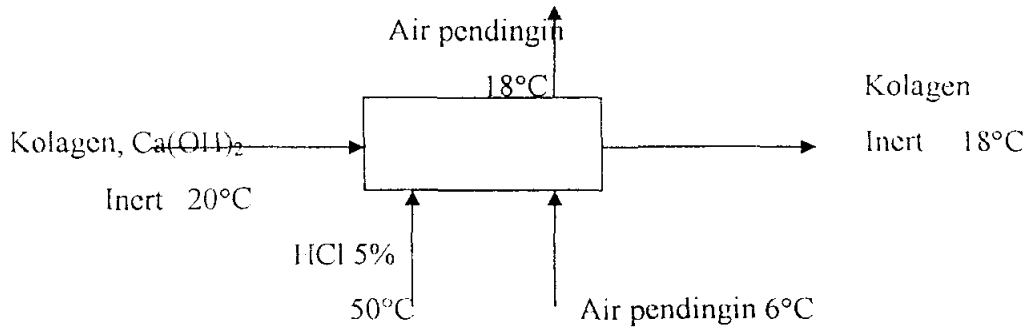
Masuk :

Keluar :

$$Q_{\text{pemanas}} = 15441,2065 \text{ kkal/hari}$$

$$\begin{aligned}\text{Panas keluar} &= 18466,4892 \text{ kkal/hari} \\ &= 33907,6957 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\text{Panas masuk} &= 33907,6957 \text{ kkal/hari} \\ &= 33907,6957 \text{ kkal/hari}\end{aligned}$$

Tangki Netralisasi (F-160)

$$\Delta H_{in} + \Delta H_r = \Delta H_{out} + \Delta H_{air \text{ pendingin}}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

Menghitung panas masuk ( $\Delta H_{in}$ ) :

Suhu bahan masuk = 20°C

Kolagen = 4495,5000 kg/hari

H<sub>2</sub>O = 19840,8287 kg/hari

Ca(OH)<sub>2</sub> = 1057,7912 kg/hari

CaCl<sub>2</sub> = 4,2707 kg/hari

MgCl<sub>2</sub> = 0,1124 kg/hari

NaCl = 0,2248 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 2,4725 kg/hari

$$\Delta H_{in} = \Delta H_{kolagen} + \Delta H_{H_2O} + \Delta H_{Ca(OH)_2} + \Delta H_{HCl} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4}$$

$$\Delta H_{kolagen} = m \cdot C_p \cdot (T_1 - 20)$$

$$= 4495,5000 \cdot 0,3369 \cdot (20 - 25)$$

$$= -7572,6698 \text{ kkal/hari}$$

$$\Delta H_{H_2O} = m \cdot C_p \cdot (25 - T_1)$$

$$= 19840,8287 \cdot 1 \cdot (20 - 25)$$

$$= -99204,1435 \text{ kkal/hari}$$

$$\begin{aligned}\Delta H_{\text{Ca(OH)}_2} &= 1057,7912 \cdot 0,286 \cdot (20-25) \\ &= -1512,6414 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{CaCl}_2} &= 4,2707 \cdot 0,2162 \cdot (20-25) \\ &= -4,6166 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{MgCl}_2} &= 0,1124 \cdot 0,252 \cdot (20-25) \\ &= -0,1416 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{NaCl}} &= 0,2248 \cdot 0,2737 \cdot (20-25) \\ &= -0,3076 \text{ kkal/hari}\end{aligned}$$

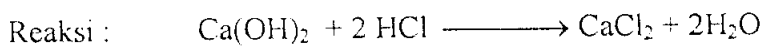
$$\begin{aligned}\Delta H_{\text{H}_3\text{PO}_4} &= 2,4725 \cdot 0,452 \cdot (20-25) \\ &= -5,5879 \text{ kkal/hari}\end{aligned}$$

Untuk larutan HCl 5% masuk pada suhu 50°C :

$$\begin{aligned}\text{HCl} &= 1041,1487 \text{ kg/hari} \\ \Delta H_{\text{HCl}} &= 1041,1487 \cdot 0,8377 \cdot (50-25) \\ &= 21804,2567 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{in}} &= -7572,6698 + -99204,1435 + 21804,2567 + -1512,6414 + -4,6166 \\ &\quad + -0,1416 + -0,3076 + -5,5879 \\ &= -86495,8517 \text{ kkal/hari}\end{aligned}$$

Menghitung panas reaksi ( $\Delta H_r$ ) :



mol  $\text{Ca(OH)}_2$  yang bereaksi = 14,276 kmol/hari

mol HCl yang bereaksi = 28,552 kmol/hari

mol  $\text{H}_2\text{O}$  yang bereaksi = 28,552 kmol/hari

mol  $\text{CaCl}_2$  yang bereaksi = 14,276 kmol/hari

$$\begin{aligned}\Delta H_r &= ((2 \times 28,552 \times -68,2782) + (14,276 \times -189,9857)) - ((14,276 \times -235,7950) \\ &\quad + (2 \times 28,552 \times -25,9113)) \\ &= -2496998,354 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

$$\text{Suhu bahan keluar} = 18^\circ\text{C}$$

$$\text{Kolagen} = 4495,5000 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 20355,2215 \text{ kg/hari}$$

$$\text{CaCl}_2 = 1588,8210 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,1124 \text{ kg/hari}$$

$$\text{NaCl} = 0,2248 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 2,4725 \text{ kg/hari}$$

$$\Delta H_{out} = \Delta H_{H_2O} + \Delta H_{kolagen} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4}$$

$$\begin{aligned}\Delta H_{kolagen} &= m.Cp.(T_2 - 25) \\ &= 4495,5000 \cdot 0,3369 \cdot (18 - 25) \\ &= -10601,7377 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{H_2O} &= m.Cp.(T_2 - 25) \\ &= 20355,2215 \cdot 1 \cdot (18 - 25) \\ &= -142486,5505 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{CaCl_2} &= m.Cp.(T_2 - 25) \\ &= 1588,8210 \cdot 0,2162 \cdot (18 - 25) \\ &= -2404,5217 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{MgCl_2} &= m.Cp.(T_2 - 25) \\ &= 0,1124 \cdot 0,252 \cdot (18 - 25) \\ &= -0,1983 \text{ kkal/hari}\end{aligned}$$



$$\begin{aligned}\Delta H_{\text{NaCl}} &= m \cdot \text{Cp} \cdot (\text{T2-25}) \\ &= 0,2248 \cdot 0,2737 \cdot (18-25) \\ &= -0,4307 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_3\text{PO}_4} &= \text{m.Cp. (T2-25)} \\ &= 2,4725 \cdot 0,452 \cdot (18-25) \\ &= -7,823 \text{ kkal/hari}\end{aligned}$$

$$\Delta H_{out} = -10601,7377 + -142486,5505 + -2404,5217 + -0,1983 + -0,4307 + -7,823$$

$$= -155501,2619 \text{ kkal/hari}$$

$$\begin{aligned}\Delta H_{in} + \Delta H_r &= \Delta H_{out} + \Delta H_{air \text{ pendingin}} \\ 155501,2619 + 2496998,354 &= 86495,8517 + \Delta H_{air \text{ pendingin}} \\ \Delta H_{air \text{ pendingin}} &= 2566003,764 \text{ kkal/hari}\end{aligned}$$

Suhu air masuk =  $6^{\circ}\text{C}$

Suhu air keluar =  $18^{\circ}\text{C}$

$$\begin{aligned}\text{Massa air pendingin yang dibutuhkan} &= \frac{2566003,764}{1 \times (18 - 6)} \\ &= 213833,647 \text{ kg/hari}\end{aligned}$$

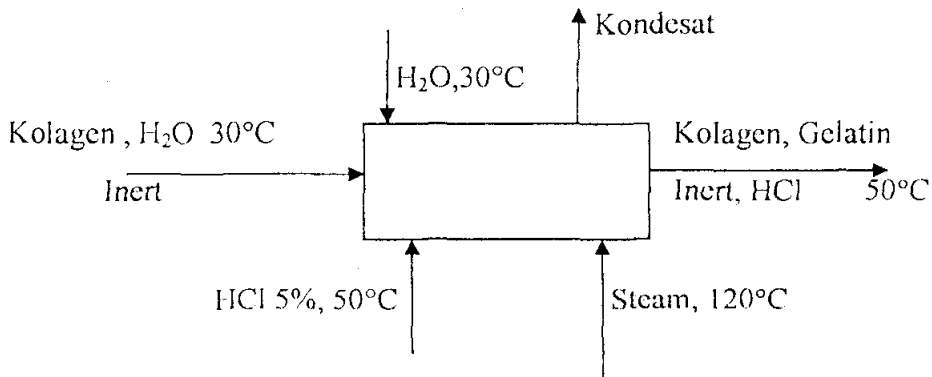
Masuk :

Keluar :

Panas reaksi = 2496998,354 kkal/hari      Qpendingin = 2566003,764 kkal/hari  
Panas keluar = 155501,2619 kkal/hari  


---

Panas masuk = 86495,8517 kkal/hari  
= 2652499,616 kkal/hari                      = 2652499,616 kkal/hari

Tangki Ekstraksi I (F-210)

Pada tangki ekstraksi ini digunakan steam suhu 120°C dan tekanan 198,53 kPa

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

$Q_{loss}$  = panas yang hilang = 0,05 dari suplai panas

Menghitung panas masuk ( $\Delta H_{in}$ ) :

Suhu bahan masuk = 30°C

Kolagen = 4495,5000 kg/hari

H<sub>2</sub>O = 10339,6500 kg/hari

CaCl<sub>2</sub> = 81,3618 kg/hari

MgCl<sub>2</sub> = 0,0056 kg/hari

NaCl = 0,0112 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,1270 kg/hari

$$\Delta H_{in} = \Delta H_{kolagen} + \Delta H_{H_2O} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4} + \Delta H_{HCl\ 5\%}$$

$$\Delta H_{kolagen} = m \cdot Cp \cdot (T_1 - 25)$$

$$= 4495,5000 \cdot 0,3369 \cdot (30 - 25)$$

$$= 7572,6698 \text{ kkal/hari}$$

$$\Delta H_{H_2O} = m \cdot Cp \cdot (T_1 - 25)$$

$$= 10339,6500 \cdot 1 \cdot (30 - 25)$$

$$= 51698,2500 \text{ kkal/hari}$$

$$\begin{aligned}
 \Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T1-25) \\
 &= 81,3618 \cdot 0,2162 \cdot (30-25) \\
 &= 87,9521 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T1-25) \\
 &= 0,0056 \cdot 0,252 \cdot (30-25) \\
 &= 0,0071 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T1-25) \\
 &= 0,0112 \cdot 0,2737 \cdot (30-25) \\
 &= 0,0153 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T1-25) \\
 &= 0,1270 \cdot 0,452 \cdot (30-25) \\
 &= 0,2870 \text{ kkal/hari}
 \end{aligned}$$

Untuk larutan HCl 5% masuk pada suhu 50°C :

$$\begin{aligned}
 \Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T1-25) \\
 &= 75,4071 \cdot 0,8377 \cdot (50-25) \\
 &= 1579,2132 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{in}} &= 7572,6698 + 51698,2500 + 87,9521 + 0,0071 + 0,0153 + 0,2870 \\
 &\quad + 1579,2132 \\
 &= 60938,3945 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas keluar ( $\Delta H_{\text{out}}$ ) :

Suhu bahan keluar = 50°C

Kolagen = 3416,5800 kg/hari

Gelatin = 1078,9200 kg/hari

$\text{CaCl}_2$  = 81,3618 kg/hari

$\text{MgCl}_2$  = 0,0056 kg/hari

$$\text{NaCl} = 0,0112 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 10339,6500 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,1270 \text{ kg/hari}$$

$$\text{HCl 5\%} = 75,4071 \text{ kg/hari}$$

$$\Delta H_{\text{out}} = \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_2\text{O}} + \Delta H_{\text{H}_3\text{PO}_4} \\ + \Delta H_{\text{HCl 5\%}}$$

$$\begin{aligned} \Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 3416,5800 \cdot 0,3369 \cdot (50 - 25) \\ &= 28776,1451 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 1078,9200 \cdot 0,3613 \cdot (50 - 25) \\ &= 9745,3449 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 81,3618 \cdot 0,2162 \cdot (50 - 25) \\ &= 439,7605 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 0,0056 \cdot 0,252 \cdot (50 - 25) \\ &= 0,0353 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 0,0112 \cdot 0,2737 \cdot (50 - 25) \\ &= 0,0766 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 0,1270 \cdot 0,452 \cdot (50 - 25) \end{aligned}$$

$$\Delta H_{\text{H}_3\text{PO}_4} = 1,4351 \text{ kkal/hari}$$

$$\begin{aligned}
 \Delta H_{H_2O} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 10339,6500 \cdot 1 \cdot (50 - 25) \\
 &= 258491,2500 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{HCl \text{ 5\%}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 75,4071 \cdot 0,8377 \cdot (50 - 25) \\
 &= 1822,7121 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{out} &= 28776,1451 + 9745,3449 + 439,7605 + 0,0353 + 0,0766 + 1,4351 \\
 &\quad + 258491,2500 + 1822,7121 \\
 &= 299276,7596 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{in} + Q_{steam} &= \Delta H_{out} + 0,05 \cdot Q_{steam} \\
 60938,3945 + Q_{steam} &= 299276,7596 + 0,05 \cdot Q_{steam} \\
 Q_{steam} &= 250882,4896 \text{ kkal/hari}
 \end{aligned}$$

Digunakan steam pada suhu 120°C dan tekanan 198,53 kPa didapat :

$$\begin{aligned}
 H_v &= 2706,3 \text{ kJ/kg} = 646,8057 \text{ kkal/kg} \\
 H_l &= 503,71 \text{ kJ/kg} = 120,3843 \text{ kkal/kg} \\
 \lambda &= H_v - H_l = (646,8057 - 120,3843) \text{ kkal/hari} = 526,4214 \text{ kkal/hari}
 \end{aligned}$$

$$\text{Jumlah steam yang digunakan} = \frac{250882,4896}{526,4214} = 476,5811 \text{ kg/hari}$$

$$\begin{aligned}
 \text{Panas yang hilang} &= 0,05 \times 250882,4896 \\
 &= 12544,1245 \text{ kkal/hari}
 \end{aligned}$$

Masuk :

Panas masuk = 60938,3945 kkal/hari

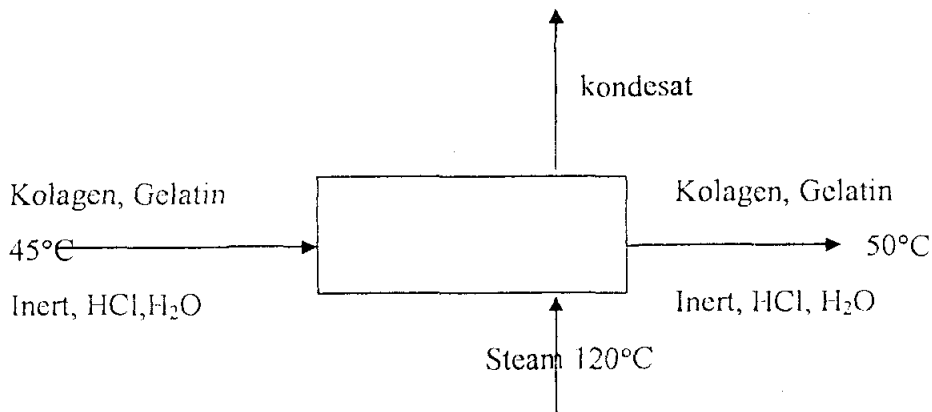
Panas Steam = 250882,4896 kkal/hari

$\frac{\text{Panas masuk} + \text{Panas Steam}}{\text{Panas hilang}} = \frac{60938,3945 + 250882,4896}{12544,1245}$   
 = 311820,8841 kkal/hari

Keluar :

Panas keluar = 299276,7596 kkal/hari

$\frac{\text{Panas keluar}}{\text{Panas hilang}} = \frac{299276,7596}{12544,1245}$   
 = 311820,8841 kkal/hari

**Tangki Ekstraksi II (F-220)**

Pada tangki ekstraksi ini digunakan steam suhu 120°C dan tekanan 198,53 kPa

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

$Q_{loss}$  = panas yang hilang = 0,05 dari suplai panas

Menghitung panas masuk ( $\Delta H_{in}$ ) :

Suhu bahan masuk = 45°C

Kolagen = 3416,5800 kg/hari

H<sub>2</sub>O = 10339,6500 kg/hari

CaCl<sub>2</sub> = 81,3618 kg/hari

MgCl<sub>2</sub> = 0,0056 kg/hari

NaCl = 0,0112 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,1270 kg/hari

HCl 5% = 75,4071 kg/hari

$$\text{Gelatin} = 1078,9200 \text{ kg/hari}$$

$$\Delta H_{\text{in}} = \Delta H_{\text{kolagen}} + \Delta H_{\text{H}_2\text{O}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_3\text{PO}_4} + \Delta H_{\text{HCl } 5\%} \\ + \Delta H_{\text{gelatin}}$$

$$\begin{aligned} \Delta H_{\text{kolagen}} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 3416,5800 \cdot 0,3369 \cdot (45 - 25) \\ &= 23020,9160 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{H}_2\text{O}} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 10339,6500 \cdot 1 \cdot (45 - 25) \\ &= 206793,0000 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{CaCl}_2} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 81,3618 \cdot 0,2162 \cdot (45 - 25) \\ &= 351,8084 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{MgCl}_2} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 0,0056 \cdot 0,252 \cdot (45 - 25) \\ &= 0,0282 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{NaCl}} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 0,0112 \cdot 0,2737 \cdot (45 - 25) \\ &= 0,0613 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 0,1270 \cdot 0,452 \cdot (45 - 25) \\ &= 1,1481 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{HCl } 5\%} &= m \cdot \text{Cp} \cdot (T_1 - 25) \\ &= 75,4071 \cdot 0,8377 \cdot (45 - 25) \end{aligned}$$

$$= 1458,1697 \text{ kkal/hari}$$

$$\begin{aligned}\Delta H_{\text{gelatin}} &= m.C_p.(T1-25) \\ &= 1078,9200 \cdot 0,3613 \cdot (45-25) \\ &= 7796,2759 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{in}} &= 23020,9160 + 206793,0000 + 1458,1697 + 351,8084 + 0,0282 + 0,0613 \\ &\quad + 1,1481 + 7796,2759 \\ &= 239421,4077 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{\text{out}}$ ) :

$$\text{Suhu bahan keluar} = 50^\circ\text{C}$$

$$\text{Kolagen} = 2767,4298 \text{ kg/hari}$$

$$\text{Gelatin} = 1728,0702 \text{ kg/hari}$$

$$\text{CaCl}_2 = 81,3618 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,0056 \text{ kg/hari}$$

$$\text{NaCl} = 0,00112 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 10339,6500 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,1270 \text{ kg/hari}$$

$$\text{HCl 5\%} = 75,4071 \text{ kg/hari}$$

$$\begin{aligned}\Delta H_{\text{out}} &= \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_2\text{O}} + \Delta H_{\text{H}_3\text{PO}_4} \\ &\quad + \Delta H_{\text{HCl 5\%}}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{kolagen}} &= m.C_p.(T2-25) \\ &= 2767,4298 \cdot 0,3369 \cdot (50-25) \\ &= 23308,6775 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{gelatin}} &= m.C_p.(T2-25) \\ &= 1728,0702 \cdot 0,3613 \cdot (50-25) \\ &= 15608,7941 \text{ kkal/hari}\end{aligned}$$



$$\begin{aligned}
 \Delta H_{CaCl_2} &= m.Cp.(T_2-25) \\
 &= 81,3618 \cdot 0,2162 \cdot (50-25) \\
 &= 439,7605 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{MgCl_2} &= m.Cp.(T_2-25) \\
 &= 0,0056 \cdot 0,252 \cdot (50-25) \\
 &= 0,0353 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{NaCl} &= m.Cp.(T_2-25) \\
 &= 0,00112 \cdot 0,2737 \cdot (50-25) \\
 &= 0,0077 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{H_3PO_4} &= m.Cp.(T_2-25) \\
 &= 0,1270 \cdot 0,452 \cdot (50-25) \\
 &= 1,4351 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{H_2O} &= m.Cp.(T_2-25) \\
 &= 10339,6500 \cdot 1 \cdot (50-25) \\
 &= 258491,2500 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{HCl \ 5\%} &= m.Cp.(T_2-25) \\
 &= 75,4071 \cdot 0,8377 \cdot (50-25) \\
 &= 1822,7121 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{out} &= 23308,6775 + 258491,2500 + 1822,7121 + 439,7605 + 0,0353 + 0,0077 \\
 &\quad + 1,4351 = 15608,7941 \\
 &= 299672,6723 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{in} + Q_{steam} &= \Delta H_{out} + \Delta H_r + 0,05 \cdot Q_{steam} \\
 239421,4077 + Q_{steam} &= 299672,6723 + 0,05 \cdot Q_{steam} \\
 Q_{steam} &= 63422,38377 \text{ kkal/hari}
 \end{aligned}$$

Digunakan steam pada suhu 120°C dan tekanan 198,53 kPa didapat :

$$H_v = 2706,3 \text{ kJ/kg} = 646,8057 \text{ kkal/kg}$$

$$H_l = 503,71 \text{ kJ/kg} = 120,3843 \text{ kkal/kg}$$

$$\lambda = H_v - H_l = (646,8057 - 120,3843) \text{ kkal/hari} = 526,4214 \text{ kkal/hari}$$

$$\text{Jumlah steam yang digunakan} = \frac{63422,3877}{526,4214} = 120,4948 \text{ kg/hari}$$

$$\begin{aligned} \text{Panas yang hilang} &= 0,05 \times 63422,3877 \\ &= 3171,1192 \text{ kkal/hari} \end{aligned}$$

Masuk :

$$\text{Panas masuk} = 239421,4077 \text{ kkal/hari}$$

$$\text{Panas Steam} = 63422,3877 \text{ kkal/hari}$$

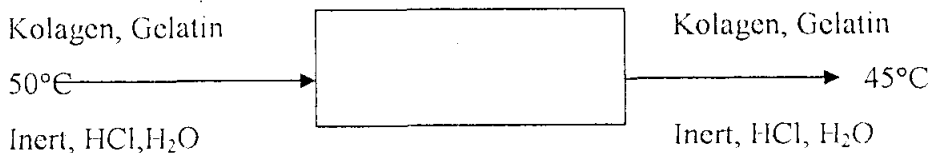
$$\begin{aligned} &= 302843,7954 \text{ kkal/hari} \end{aligned}$$

Keluar :

$$\text{Panas keluar} = 299672,6723 \text{ kkal/hari}$$

$$\begin{aligned} \text{Panas hilang} &= 3171,1192 \text{ kkal/hari} \\ &= 302843,7954 \text{ kkal/hari} \end{aligned}$$

### Rotary Drum Vacuum Filter II (H-221)



$$\Delta H_{in} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

Menghitung panas masuk ( $\Delta H_{in}$ ) :

$$\text{Suhu bahan masuk} = 50^\circ\text{C}$$

$$\text{Kolagen} = 2767,4298 \text{ kg/hari}$$

$$\text{Gelatin} = 649,1502 \text{ kg/hari}$$

$$\text{CaCl}_2 = 81,3618 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,0056 \text{ kg/hari}$$

$$\text{NaCl} = 0,0112 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,1270 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 10339,6500 \text{ kg/hari}$$

$$\text{HCl } 5\% = 75,4071 \text{ kg/hari}$$

$$\begin{aligned} \Delta H_{\text{lin}} &= \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_3\text{PO}_4} + \Delta H_{\text{H}_2\text{O}} \\ &+ \Delta H_{\text{HCl } 5\%} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T1-25) \\ &= 2767,4298 \cdot 0,3369 \cdot (50-25) \\ &= 23308,67749 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T1-25) \\ &= 649,1502 \cdot 0,3613 \cdot (50-25) \\ &= 5863,449182 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T1-25) \\ &= 81,3618 \cdot 0,2162 \cdot (50-25) \\ &= 439,760529 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T1-25) \\ &= 0,0056 \cdot 0,252 \cdot (50-25) \\ &= 0,03528 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T1-25) \\ &= 0,0112 \cdot 0,2737 \cdot (50-25) \\ &= 0,076636 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned}
 \Delta H_{H_3PO_4} &= m.Cp.(T1-25) \\
 &= 0,1270 \cdot 0,452 \cdot (50-25) \\
 &= 1,4351 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{H_2O} &= m.Cp.(T1-25) \\
 &= 10339,6500 \cdot 1 \cdot (50-25) \\
 &= 258491,25 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{HCl\ 5\%} &= m.Cp.(T1-25) \\
 &= 75,4071 \cdot 0,8377 \cdot (50-25) \\
 &= 1822,712144 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{in} &= 23308,67749 + 258491,25 + 1822,712144 + 439,760529 + 0,03528 \\
 &\quad + 0,076636 + 1,4351 + 5863,449182 \\
 &= 289927,3964 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 45°C

Cake keluar :

Kolagen = 2767,4298 kg/hari

H<sub>2</sub>O = 585,1762 kg/hari

CaCl<sub>2</sub> = 4,6047 kg/hari

MgCl<sub>2</sub> = 0,0003 kg/hari

NaCl = 0,0006 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,0072 kg/hari

HCl 5% = 4,2677 kg/hari

Gelatin = 97,8008 kg/hari

Filtrat keluar :

Gelatin = 1630,2694 kg/hari

H<sub>2</sub>O = 9754,4738 kg/hari

$$\text{HCl } 5\% = 71,1394 \text{ kg/hari}$$

$$\text{CaCl}_2 = 76,7571 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,0053 \text{ kg/hari}$$

$$\text{NaCl} = 0,0106 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,1198 \text{ kg/hari}$$

$$\Delta H_{\text{out}} = \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{H}_2\text{O}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{HCl } 5\%} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_3\text{PO}_4}$$

$$\begin{aligned}\Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 2767,4298 \cdot 0,3369 \cdot (45 - 25) \\ &= 18646,94199 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (97,8008 + 1630,2694) \cdot 0,3613 \cdot (45 - 25) \\ &= 12487,03527 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (585,1762 + 9754,4738) \cdot 1 \cdot (45 - 25) \\ &= 206793 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (4,6047 + 76,7571) \cdot 0,2162 \cdot (45 - 25) \\ &= 351,8084232 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (0,0003 + 0,0053) \cdot 0,252 \cdot (45 - 25) \\ &= 0,028224 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (0,0006 + 0,0106) \cdot 0,2737 \cdot (45 - 25) \\ &= 0,0613088 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{H_3PO_4} &= m.C_p.(T_2-25) \\ &= (0,0072 + 0,1198) . 0,452 . (45-25) \\ &= 1,14808 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{HCl\ 5\%} &= m.C_p.(T_2-25) \\ &= (4,2677 + 71,1394) . 0,8377 . (45-25) \\ &= 1458,169715 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{out} &= 18646,94199 + 206793 + 1458,169715 + 351,8084232 + 0,028224 \\ &\quad + 0,0613088 + 1,14808 + 12487,03527 \\ &= 239738,193 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{in} &= \Delta H_{out} + Q_{loss} \\ 289927,3964 &= 239738,193 + Q_{loss} \\ Q_{loss} &= 50189,20335 \text{ kkal/hari}\end{aligned}$$

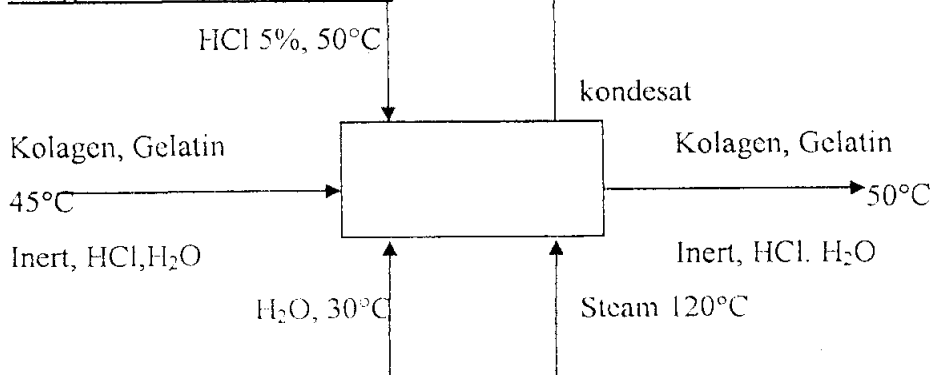
Masuk :

$$\begin{aligned}\text{Panas masuk} &= 289927,3964 \text{ kkal/hari} \\ &= 289927,3964 \text{ kkal/hari}\end{aligned}$$

Keluar :

$$\begin{aligned}\text{Panas keluar} &= 239738,193 \text{ kkal/hari} \\ \text{Panas hilang} &= 50189,20335 \text{ kkal/hari} \\ &= 289927,3964 \text{ kkal/hari}\end{aligned}$$

**Tangki Ekstraksi III (F-230)**



Pada tangki ekstraksi ini digunakan suhu steam 120°C dan tekanan 198,53 kPa

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

$Q_{loss}$  = panas yang hilang = 0,05 dari suplai panas

Menghitung panas masuk ( $\Delta H_{in}$ ) :

Suhu bahan masuk = 45°C

Kolagen = 2767,4298 kg/hari

H<sub>2</sub>O = 585,1762 kg/hari

CaCl<sub>2</sub> = 4,6047 kg/hari

MgCl<sub>2</sub> = 0,0003 kg/hari

NaCl = 0,0006 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,0072 kg/hari

Gelatin = 97,8008 kg/hari

HCl 5% = 4,2677 kg/hari

$$\Delta H_{in} = \Delta H_{kolagen} + \Delta H_{H_2O} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4} + \Delta H_{HCl\ 5\%} + \Delta H_{gelatin}$$

$$\begin{aligned}\Delta H_{kolagen} &= m.Cp.(T1-25) \\ &= 2767,4298 \cdot 0,3369 \cdot (45-25) \\ &= 18646,94199 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{H_2O} &= m.Cp.(T1-25) \\ &= 585,1762 \cdot 1 \cdot (45-25) \\ \Delta H_{H_2O} &= 11703,524 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{CaCl_2} &= m.Cp.(T1-25) \\ &= 4,6047 \cdot 0,2162 \cdot (45-25) \\ &= 19,91072 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{MgCl}_2} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 0,0003 \cdot 0,252 \cdot (45-25) \\
 &= 0,00151 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{NaCl}} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 0,0006 \cdot 0,2737 \cdot (45-25) \\
 &= 0,00328 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 0,0072 \cdot 0,452 \cdot (45-25) \\
 &= 0,06509 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{gelatin}} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 97,8008 \cdot 0,3613 \cdot (45-25) \\
 &= 706,70858 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{HCl } 5\%} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 4,2677 \cdot 0,8377 \cdot (45-25) \\
 &= 71,5010 \text{ kkal/hari}
 \end{aligned}$$

Untuk larutan HCl 5% masuk pada suhu 50°C:

$$\begin{aligned}
 \text{HCl } 5\% &= 10,4102 \text{ kg/hari} \\
 \Delta H_{\text{HCl } 5\%} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 10,4102 \cdot 0,8377 \cdot (50-25) \\
 &= 218,0156 \text{ kkal/hari}
 \end{aligned}$$

Untuk air yang ditambahkan pada tangki ini masuk pada suhu 30°C :

$$\begin{aligned}
 \text{H}_2\text{O} &= 5779,9123 \text{ kg/hari} \\
 \Delta H_{\text{H}_2\text{O}} &= m \cdot \text{Cp} \cdot (T1-25) \\
 &= 5779,9123 \cdot 1 \cdot (30-25) \\
 \Delta H_{\text{H}_2\text{O}} &= 28899,5615 \text{ kkal/hari}
 \end{aligned}$$



$$\begin{aligned}\Delta H_{in} &= 18646,94199 + 11703,524 + 218,0156 + 19,91072 + 0,00151 \\ &\quad + 0,00328 + 0,06509 + 706,70858 + 28899,5615 + 71,5010 \\ &= 60266,2333 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 50°C

Kolagen = 2435,3382 kg/hari

Gelatin = 429,8924 kg/hari

CaCl<sub>2</sub> = 4,60470 kg/hari

MgCl<sub>2</sub> = 0,0003 kg/hari

NaCl = 0,0006 kg/hari

H<sub>2</sub>O = 6365,0885 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,0072 kg/hari

HCl 5% = 14,67790 kg/hari

$$\begin{aligned}\Delta H_{out} &= \Delta H_{kolagen} + \Delta H_{gelatin} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_2O} + \Delta H_{H_3PO_4} \\ &\quad + \Delta H_{HCl \text{ 5\%}}\end{aligned}$$

$$\begin{aligned}\Delta H_{kolagen} &= m.Cp.(T_2 - T_1) \\ &= 2435,3382 \cdot 0,3369 \cdot (50 - 25) \\ &= 20511,6360 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{gelatin} &= m.Cp.(T_2 - T_1) \\ &= 429,8924 \cdot 0,3613 \cdot (50 - 25) \\ &= 3883,00310 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{CaCl_2} &= m.Cp.(T_2 - T_1) \\ &= 4,60470 \cdot 0,2162 \cdot (50 - 25) \\ &= 24,88840 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 0,0003 \cdot 0,252 \cdot (50 - 25) \\
 &= 0,00189 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 0,0006 \cdot 0,2737 \cdot (50 - 25) \\
 &= 0,00411 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 0,0072 \cdot 0,452 \cdot (50 - 25) \\
 &= 0,08136 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 6365,0885 \cdot 1 \cdot (50 - 25) \\
 &= 159127,2125 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 14,67790 \cdot 0,8377 \cdot (50 - 25) \\
 &= 354,7887 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{out}} &= 20511,6360 + 159127,2125 + 354,7887 + 24,88840 + 0,00189 + 0,00411 \\
 &\quad + 0,08136 + 3883,00310 \\
 &= 183876,6403 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{in}} + Q_{\text{steam}} &= \Delta H_{\text{out}} + \Delta H_r + 0,05 \cdot Q_{\text{steam}} \\
 60266,2333 + Q_{\text{steam}} &= 183876,6403 + 0,05 \cdot Q_{\text{steam}} \\
 Q_{\text{steam}} &= 130116,2179 \text{ kkal/hari}
 \end{aligned}$$

Digunakan steam pada suhu 120°C dan tekanan 198,53 kPa didapat :

$$\begin{aligned}
 H_v &= 2706,3 \text{ kJ/kg} = 646,8057 \text{ kkal/kg} \\
 H_l &= 503,71 \text{ kJ/kg} = 120,3843 \text{ kkal/kg}
 \end{aligned}$$

$$\lambda = H_v - H_L = (646,8057 - 120,3843) \text{ kkal/hari} = 526,4214 \text{ kkal/hari}$$

$$\text{Jumlah steam yang digunakan} = \frac{130116,2179}{526,4214} = 247,1712 \text{ kg/hari}$$

$$\begin{aligned} \text{Panas yang hilang} &= 0,05 \times 130116,2179 \\ &= 6505,8109 \text{ kkal/hari} \end{aligned}$$

Masuk :

$$\text{Panas masuk} = 60266,2333 \text{ kkal/hari}$$

$$\text{Panas Steam} = 130116,2179 \text{ kkal/hari}$$

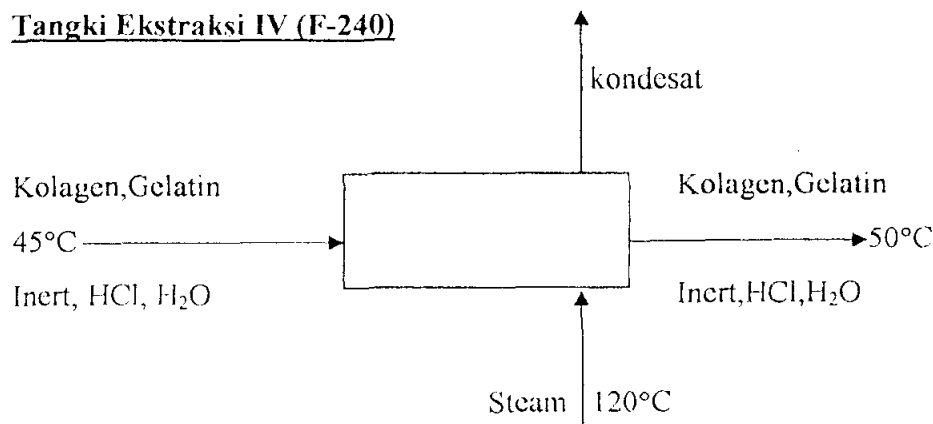
$$\begin{aligned} \text{Panas hilang} &= 6505,8109 \text{ kkal/hari} \\ &= 190382,4512 \text{ kkal/hari} \end{aligned}$$

Keluar :

$$\text{Panas keluar} = 183876,6403 \text{ kkal/hari}$$

$$\begin{aligned} \text{Panas hilang} &= 6505,8109 \text{ kkal/hari} \\ &= 190382,4512 \text{ kkal/hari} \end{aligned}$$

#### Tangki Ekstraksi IV (F-240)



Pada tangki ekstraksi ini digunakan steam pada suhu 120°C dan tekanan 198,53 kPa.

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + \Delta H_r + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

$Q_{loss}$  = panas yang hilang = 0,05 dari suplai panas

Menghitung panas masuk ( $\Delta H_{in}$ ) :

Suhu bahan masuk = 45°C

Kolagen = 2435,3382 kg/hari

$$H_2O = 6365,0885 \text{ kg/hari}$$

$$CaCl_2 = 4,60470 \text{ kg/hari}$$

$$MgCl_2 = 0,0003 \text{ kg/hari}$$

$$NaCl = 0,0006 \text{ kg/hari}$$

$$H_3PO_4 = 0,0072 \text{ kg/hari}$$

$$HCl \text{ 5\%} = 14,67790 \text{ kg/hari}$$

$$\text{Gelatin} = 429,8924 \text{ kg/hari}$$

$$\Delta H_{in} = \Delta H_{kolagen} + \Delta H_{H_2O} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_3PO_4} + \Delta H_{HCl \text{ 5\%}} + \Delta H_{gelatin}$$

$$\begin{aligned} \Delta H_{kolagen} &= m.Cp.(T1-25) \\ &= 2435,3382 \cdot 0,3369 \cdot (45-25) \\ &= 16409,3088 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{H_2O} &= m.Cp.(T1-25) \\ &= 6365,0885 \cdot 1 \cdot (45-25) \end{aligned}$$

$$\Delta H_{H_2O} = 127301,7700 \text{ kkal/hari}$$

$$\begin{aligned} \Delta H_{CaCl_2} &= m.Cp.(T1-25) \\ &= 4,60470 \cdot 0,2162 \cdot (45-25) \\ &= 19,9107 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{MgCl_2} &= m.Cp.(T1-25) \\ &= 0,0003 \cdot 0,252 \cdot (45-25) \\ &= 0,001512 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{NaCl} &= m.Cp.(T1-25) \\ &= 0,0006 \cdot 0,2737 \cdot (45-25) \\ &= 0,0032844 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned}
 \Delta H_{H_3PO_4} &= m.Cp.(T1-25) \\
 &= 0,0072 \cdot 0,452 \cdot (45-25) \\
 &= 0,0650880 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{HCl \ 5\%} &= m.Cp.(T1-25) \\
 &= 14,67790 \cdot 0,8377 \cdot (45-25) \\
 &= 283,8310 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{gelatin} &= m.Cp.(T1-25) \\
 &= 429,8924 \cdot 0,3613 \cdot (45-25) \\
 &= 3106,4025 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{in} &= 16409,3088 + 127301,7700 + 283,8310 + 19,9107 + 0,001512 \\
 &\quad + 0,0032844 + 0,0650880 + 3106,4025 \\
 &= 143994,9145 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

$$\begin{aligned}
 \text{Suhu bahan keluar} &= 50^\circ\text{C} \\
 \text{Kolagen} &= 2240,5111 \text{ kg/hari} \\
 \text{Gelatin} &= 624,7195 \text{ kg/hari} \\
 \text{CaCl}_2 &= 4,60470 \text{ kg/hari} \\
 \text{MgCl}_2 &= 0,0003 \text{ kg/hari} \\
 \text{NaCl} &= 0,0006 \text{ kg/hari} \\
 \text{H}_2\text{O} &= 6365,0885 \text{ kg/hari} \\
 \text{H}_3\text{PO}_4 &= 0,0072 \text{ kg/hari} \\
 \text{HCl 5\%} &= 14,6779 \text{ kg/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{out} &= \Delta H_{kolagen} + \Delta H_{gelatin} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_2O} + \Delta H_{H_3PO_4} \\
 &\quad + \Delta H_{HCl \ 5\%}
 \end{aligned}$$

$$\begin{aligned}\Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 2240,5111 \cdot 0,3369 \cdot (50 - 25) \\ &= 18870,70474 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 624,7195 \cdot 0,3613 \cdot (50 - 25) \\ &= 5642,7789 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 4,60470 \cdot 0,2162 \cdot (50 - 25) \\ &= 24,8884 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 0,0003 \cdot 0,252 \cdot (50 - 25) \\ &= 0,00189 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 0,0006 \cdot 0,2737 \cdot (50 - 25) \\ &= 0,004106 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 0,0072 \cdot 0,452 \cdot (50 - 25) \\ &= 0,081360 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 6365,0885 \cdot 1 \cdot (50 - 25) \\ &= 159127,2125 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T_2 - 25) \\ &= 14,6779 \cdot 0,8377 \cdot (50 - 25) \\ &= 354,7887 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{out} &= 18870,70474 + 159127,2125 + 354,7887 + 24,8884 + 0,00189 \\ &\quad + 0,004106 + 0,081360 + 5642,7789 \\ &= 183995,4848 \text{ kkal/hari}\end{aligned}$$

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + 0,05 \cdot Q_{steam}$$

$$\begin{aligned}143994,9145 + Q_{steam} &= 183995,4848 + 0,05 \cdot Q_{steam} \\ Q_{steam} &= 42105,86345 \text{ kkal/hari}\end{aligned}$$

Digunakan steam pada suhu 120°C dan tekanan 198,53 kPa didapat :

$$H_v = 2706,3 \text{ kJ/kg} = 646,8057 \text{ kkal/kg}$$

$$H_L = 503,71 \text{ kJ/kg} = 120,3843 \text{ kkal/kg}$$

$$\lambda = H_v - H_L = (646,8057 - 120,3843) \text{ kkal/hari} = 526,4214 \text{ kkal/hari}$$

$$\text{Jumlah steam yang digunakan} = \frac{42105,86345}{526,4214} = 79,9960 \text{ kg/hari}$$

$$\begin{aligned}\text{Panas yang hilang} &= 0,05 \times 42105,86345 \\ &= 2105,2932 \text{ kkal/hari}\end{aligned}$$

Masuk :

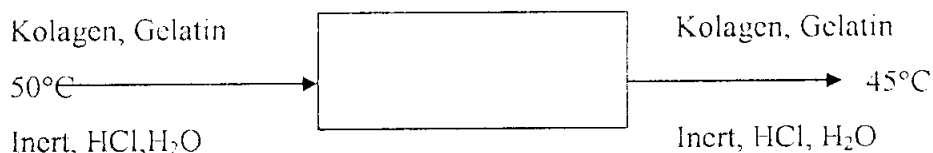
Keluar :

$$\text{Panas masuk} = 143994,9145 \text{ kkal/hari} \quad \text{Panas keluar} = 183995,4848 \text{ kkal/hari}$$

$$\text{Panas Steam} = 42105,86345 \text{ kkal/hari}$$

$$\begin{aligned}\text{Panas hilang} &= 2105,2932 \text{ kkal/hari} \\ &= 186100,778 \text{ kkal/hari}\end{aligned}$$

### Rotary Drum Vacuum Filter IV (H-241)



$$\Delta H_{in} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

Menghitung panas masuk ( $\Delta H_{in}$ ) :

$$\text{Suhu bahan masuk} = 50^{\circ}\text{C}$$

$$\text{Kolagen} = 2240,5111 \text{ kg/hari}$$

$$\text{Gelatin} = 624,7195 \text{ kg/hari}$$

$$\text{CaCl}_2 = 4,6047 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,0003 \text{ kg/hari}$$

$$\text{NaCl} = 0,0006 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,0072 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 6365,08850 \text{ kg/hari}$$

$$\text{HCl 5\%} = 14,6779 \text{ kg/hari}$$

$$\Delta H_{in} = \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_3\text{PO}_4} + \Delta H_{\text{H}_2\text{O}} \\ + \Delta H_{\text{HCl 5\%}}$$

$$\begin{aligned} \Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 2240,5111 \cdot 0,3369 \cdot (50 - 25) \\ &= 18870,70474 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 624,7195 \cdot 0,3613 \cdot (50 - 25) \\ &= 5642,778884 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 4,6047 \cdot 0,2162 \cdot (50 - 25) \\ &= 24,8884035 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 0,0003 \cdot 0,252 \cdot (50 - 25) \\ &= 0,00189 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 0,0006 \cdot 0,2737 \cdot (50 - 25) \\ &= 0,0041055 \text{ kkal/hari} \end{aligned}$$



$$\begin{aligned}\Delta H_{H_3PO_4} &= m.C_p.(T1-25) \\ &= 0,0072 \cdot 0,452 \cdot (50-25)\end{aligned}$$

$$\Delta H_{H_3PO_4} = 0,08136 \text{ kkal/hari}$$

$$\begin{aligned}\Delta H_{H_2O} &= m.C_p.(T1-25) \\ &= 6365,08850 \cdot 1 \cdot (50-25) \\ &= 159127,2125 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{HCl\ 5\%} &= m.C_p.(T1-25) \\ &= 14,6779 \cdot 0,8377 \cdot (50-25) \\ &= 354,7886946 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{in} &= 18870,70474 + 159127,2125 + 354,7886946 + 24,8884035 + 0,00189 \\ &\quad + 0,0041055 + 0,08136 + 5642,778884 \\ &= 184020,4606 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 45°C

Cake keluar :

Kolagen = 2240,5111 kg/hari

H<sub>2</sub>O = 508,66212 kg/hari

CaCl<sub>2</sub> = 0,36798 kg/hari

MgCl<sub>2</sub> = 0,00002 kg/hari

NaCl = 0,00005 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,00058 kg/hari

HCl 5% = 1,17298 kg/hari

Gelatin = 49,92407 kg/hari

Filtrat keluar :

Gelatin = 574,79543 kg/hari

H<sub>2</sub>O = 5856,42638 kg/hari

HCl 5% = 13,50492 kg/hari

$$\text{CaCl}_2 = 4,23672 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,00028 \text{ kg/hari}$$

$$\text{NaCl} = 0,00055 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,00662 \text{ kg/hari}$$

$$\Delta H_{\text{out}} = \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{H}_2\text{O}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{HCl 5\%}} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} \\ + \Delta H_{\text{H}_3\text{PO}_4}$$

$$\begin{aligned} \Delta H_{\text{kolagen}} &= m \cdot \text{Cp} \cdot (T_2 - 25) \\ &= 2240,5111 \cdot 0,3369 \cdot (45 - 25) \\ &= 15096,56379 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{gelatin}} &= m \cdot \text{Cp} \cdot (T_2 - 25) \\ &= (49,92407 + 574,79543) \cdot 0,3613 \cdot (45 - 25) \\ &= 4514,223107 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{H}_2\text{O}} &= m \cdot \text{Cp} \cdot (T_2 - 25) \\ &= (508,66212 + 5856,42638) \cdot 1 \cdot (45 - 25) \\ &= 127301,77 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{CaCl}_2} &= m \cdot \text{Cp} \cdot (T_2 - 25) \\ &= (4,23672 + 0,36798) \cdot 0,2162 \cdot (45 - 25) \\ &= 19,9107228 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{MgCl}_2} &= m \cdot \text{Cp} \cdot (T_2 - 25) \\ &= (0,0003 + 0,00002) \cdot 0,252 \cdot (45 - 25) \\ &= 0,001512 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{NaCl}} &= m \cdot \text{Cp} \cdot (T_2 - 25) \\ &= (0,0006 + 0,00005) \cdot 0,2737 \cdot (45 - 25) \\ &= 0,0032844 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned}\Delta H_{H_3PO_4} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (0,0072 + 0,00058) \cdot 0,452 \cdot (45 - 25) \\ &= 0,065088 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{HCl \text{ 5\%}} &= m \cdot C_p \cdot (T_2 - 25) \\ &= (1,17298 + 13,50492) \cdot 0,8377 \cdot (45 - 25) \\ &= 283,8309557 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{out} &= 15096,56379 + 127301,77 + 283,8309557 + 19,9107228 \\ &\quad + 0,001512 + 0,0032844 + 0,065088 + 4514,223107 \\ &= 147216,3685 \text{ kkal/hari}\end{aligned}$$

$$\Delta H_{in} = \Delta H_{out} + Q_{loss}$$

$$\begin{aligned}184020,4606 &= 147216,3685 + Q_{loss} \\ Q_{loss} &= 36804,09212 \text{ kkal/hari}\end{aligned}$$

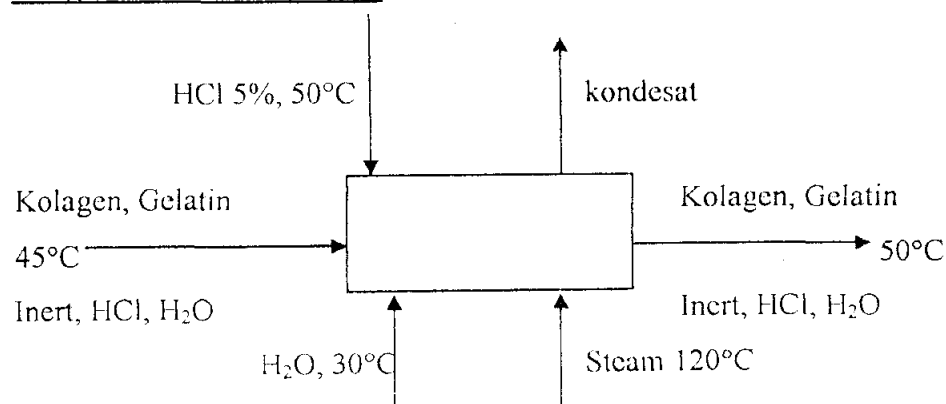
Masuk :

$$\begin{aligned}\text{Panas masuk} &= 184020,4606 \text{ kkal/hari} \\ &= 184020,4606 \text{ kkal/hari}\end{aligned}$$

Keluar :

$$\begin{aligned}\text{Panas keluar} &= 147216,3685 \text{ kkal/hari} \\ \text{Panas hilang} &= 36804,09212 \text{ kkal/hari} \\ &= 184020,4606 \text{ kkal/hari}\end{aligned}$$

### Tangki Ekstraksi V (F-250)



Pada tangki ekstraksi ini digunakan steam pada suhu  $120^{\circ}\text{C}$  dan tekanan 198,53 kPa.

$$\Delta H_{\text{in}} + Q_{\text{steam}} = \Delta H_{\text{out}} + \Delta H_{\text{r}} + Q_{\text{loss}}$$

dimana :  $\Delta H_{\text{in}}$  = panas dari bahan masuk

$\Delta H_{\text{out}}$  = panas dari bahan keluar

$Q_{\text{loss}}$  = panas yang hilang = 0,05 dari suplai panas

Menghitung panas masuk ( $\Delta H_{\text{in}}$ ) :

Suhu bahan masuk =  $45^{\circ}\text{C}$

Kolagen = 2240,5111 kg/hari

$\text{H}_2\text{O}$  = 508,66212 kg/hari

$\text{CaCl}_2$  = 0,36798 kg/hari

$\text{MgCl}_2$  = 0,00002 kg/hari

$\text{NaCl}$  = 0,00005 kg/hari

$\text{H}_3\text{PO}_4$  = 0,00058 kg/hari

$\text{HCl}$  5% = 1,17298 kg/hari

Gelatin = 22,1717140 kg/hari

$$\Delta H_{\text{in}} = \Delta H_{\text{kolagen}} + \Delta H_{\text{H}_2\text{O}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_3\text{PO}_4} + \Delta H_{\text{HCl 5\%}} + \Delta H_{\text{gelatin}}$$

$$\begin{aligned}\Delta H_{\text{kolagen}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 2240,5111 \cdot 0,3369 \cdot (45 - 25) \\ &= 15096,5638 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 508,66212 \cdot 1 \cdot (45 - 25)\end{aligned}$$

$$\Delta H_{\text{H}_2\text{O}} = 10173,2424 \text{ kkal/hari}$$

$$\begin{aligned}\Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T_1 - 25) \\ &= 0,36798 \cdot 0,2162 \cdot (45 - 25) \\ &= 1,5911 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T1-25) \\
 &= 0,00002 \cdot 0,252 \cdot (45-25) \\
 &= 0,0001008 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T1-25) \\
 &= 0,00005 \cdot 0,2737 \cdot (45-25) \\
 &= 0,0002737 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T1-25) \\
 &= 0,00058 \cdot 0,452 \cdot (45-25) \\
 &= 0,0052432 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T1-25) \\
 &= 1,17298 \cdot 0,8377 \cdot (45-25) \\
 &= 19,6521 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T1-25) \\
 &= 22,1717140 \cdot 0,3613 \cdot (45-25) \\
 &= 160,2128 \text{ kkal/hari}
 \end{aligned}$$

Untuk larutan HCl 5% yang ditambahkan masuk pada suhu 50°C :

$$\begin{aligned}
 \text{Massa HCl } 5\% &= 4,82272 \text{ kg/hari} \\
 \Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T1-25) \\
 &= 4,82272 \cdot 0,8377 \cdot (50-25) \\
 &= 100,9998 \text{ kkal/hari}
 \end{aligned}$$

Untuk air yang ditambahkan masuk pada suhu 30°C :

$$\begin{aligned}
 \text{Massa air} &= 4644,51338 \text{ kg/hari} \\
 \Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T1-25) \\
 &= 4644,51338 \cdot 1 \cdot (30-25) \\
 \Delta H_{\text{H}_2\text{O}} &= 23222,5669 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}\Delta H_{in} &= 15096,5638 + 10173,2424 + 19,6521 + 1,5911 + 0,0001008 \\ &\quad + 0,0002737 + 0,0052432 + 160,2128 + 100,9998 + 23222,5669 \\ &= 48774,8345 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 50°C

Kolagen = 1926,83950 kg/hari

Gelatin = 363,59567 kg/hari

CaCl<sub>2</sub> = 0,36798 kg/hari

MgCl<sub>2</sub> = 0,00002 kg/hari

NaCl = 0,00005 kg/hari

H<sub>2</sub>O = 5153,17550 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,00058 kg/hari

HCl 5% = 5,9957000 kg/hari

$$\begin{aligned}\Delta H_{out} &= \Delta H_{kolagen} + \Delta H_{gelatin} + \Delta H_{CaCl_2} + \Delta H_{MgCl_2} + \Delta H_{NaCl} + \Delta H_{H_2O} + \Delta H_{H_3PO_4} \\ &\quad + \Delta H_{HCl \text{ 5\%}}\end{aligned}$$

$$\begin{aligned}\Delta H_{kolagen} &= m.Cp.(T_2 - T_1) \\ &= 1926,83950 \cdot 0,3369 \cdot (50 - 25) \\ &= 16228,80569 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{gelatin} &= m.Cp.(T_2 - T_1) \\ &= 363,59567 \cdot 0,3613 \cdot (50 - 25) \\ &= 3284,1779 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{CaCl_2} &= m.Cp.(T_2 - T_1) \\ &= 0,36798 \cdot 0,2162 \cdot (50 - 25) \\ &= 1,9889 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{MgCl_2} &= m.Cp.(T_2 - T_1) \\ &= 0,00002 \cdot 0,252 \cdot (50 - 25) \\ &= 0,000126 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 0,00005 \cdot 0,2737 \cdot (50 - 25) \\
 &= 0,000342 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 0,00058 \cdot 0,452 \cdot (50 - 25) \\
 &= 0,006554 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 5153,17550 \cdot 1 \cdot (50 - 25) \\
 &= 128829,3875 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= 5,9957000 \cdot 0,8377 \cdot (50 - 25) \\
 &= 144,9258 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{out}} &= 16228,80569 + 128829,3875 + 144,9258 + 1,9889 + 0,000126 \\
 &\quad + 0,000342 + 0,006554 + 3284,1779 \\
 &= 148487,2969 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{in}} + Q_{\text{steam}} &= \Delta H_{\text{out}} + 0,05 \cdot Q_{\text{steam}} \\
 48774,8345 + Q_{\text{steam}} &= 148487,2969 + 0,05 \cdot Q_{\text{steam}} \\
 Q_{\text{steam}} &= 104960,4867 \text{ kkal/hari}
 \end{aligned}$$

Digunakan steam pada suhu 120°C dan tekanan 198,53 kPa didapat :

$$\begin{aligned}
 H_v &= 2706,3 \text{ kJ/kg} = 646,8057 \text{ kkal/kg} \\
 H_l &= 503,71 \text{ kJ/kg} = 120,3843 \text{ kkal/kg} \\
 \lambda &= H_v - H_l = (646,8057 - 120,3843) \text{ kkal/hari} = 526,4214 \text{ kkal/hari}
 \end{aligned}$$

$$\text{Jumlah steam yang digunakan} = \frac{104960,4867}{526,4214} = 199,3849 \text{ kg/hari}$$

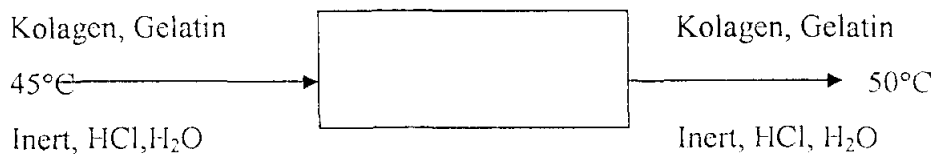
$$\begin{aligned}\text{Panas yang hilang} &= 0,05 \times 104960,4867 \\ &= 5248,0243 \text{ kkal/hari}\end{aligned}$$

Masuk :Keluar :

$$\text{Panas masuk} = 48774,8345 \text{ kkal/hari} \quad \text{Panas keluar} = 148487,2969 \text{ kkal/hari}$$

$$\text{Panas Steam} = 104960,4867 \text{ kkal/hari}$$

$$\begin{aligned}\text{Panas hilang} &= 5248,0243 \text{ kkal/hari} \\ &= 153735,3212 \text{ kkal/hari}\end{aligned}$$

**Rotary Drum Vaccum Filter V (H-251)**

$$\Delta H_{in} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas dari bahan masuk

$\Delta H_{out}$  = panas dari bahan keluar

Menghitung panas masuk ( $\Delta H_{in}$ ) :

$$\text{Suhu bahan masuk} = 50^\circ\text{C}$$

$$\text{Kolagen} = 1926,8395 \text{ kg/hari}$$

$$\text{Gelatin} = 363,59567 \text{ kg/hari}$$

$$\text{CaCl}_2 = 0,36798 \text{ kg/hari}$$

$$\text{MgCl}_2 = 0,00002 \text{ kg/hari}$$

$$\text{NaCl} = 0,00005 \text{ kg/hari}$$

$$\text{H}_3\text{PO}_4 = 0,00058 \text{ kg/hari}$$

$$\text{H}_2\text{O} = 5153,1755 \text{ kg/hari}$$

$$\text{HCl 5\%} = 5,9957 \text{ kg/hari}$$

$$\begin{aligned}\Delta H_{in} &= \Delta H_{\text{kolagen}} + \Delta H_{\text{gelatin}} + \Delta H_{\text{CaCl}_2} + \Delta H_{\text{MgCl}_2} + \Delta H_{\text{NaCl}} + \Delta H_{\text{H}_3\text{PO}_4} + \Delta H_{\text{H}_2\text{O}} \\ &\quad + \Delta H_{\text{HCl 5\%}}\end{aligned}$$



$$\begin{aligned}\Delta H_{\text{kolagen}} &= m.Cp.(T1-25) \\ &= 1926,8395 \cdot 0,3369 \cdot (50-25) \\ &= 16228,80569 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{gelatin}} &= m.Cp.(T1-25) \\ &= 363,59567 \cdot 0,3613 \cdot (50-25) \\ &= 3284,177889 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{CaCl}_2} &= m.Cp.(T1-25) \\ &= 0,36798 \cdot 0,2162 \cdot (50-25) \\ &= 1,9889319 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{MgCl}_2} &= m.Cp.(T1-25) \\ &= 0,00002 \cdot 0,252 \cdot (50-25) \\ &= 0,000126 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{NaCl}} &= m.Cp.(T1-25) \\ &= 0,00005 \cdot 0,2737 \cdot (50-25) \\ &= 0,000342125 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_3\text{PO}_4} &= m.Cp.(T1-25) \\ &= 0,00058 \cdot 0,452 \cdot (50-25) \\ &= 0,006554 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{H}_2\text{O}} &= m.Cp.(T1-25) \\ &= 5153,1755 \cdot 1 \cdot (50-25) \\ &= 128829,3875 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{\text{HCl } 5\%} &= m.Cp.(T1-25) \\ &= 5,9957 \cdot 0,8377 \cdot (50-25) \\ &= 144,925812 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}
 \Delta H_{in} &= 16228,80569 + 128829,3875 + 144,925812 + 1,9889319 \\
 &\quad + 0,000126 + 0,000342125 + 0,006554 + 3284,177889 \\
 &= 148489,2928 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 45°C

Cake keluar :

Kolagen = 1926,8395 kg/hari

H<sub>2</sub>O = 449,443193 kg/hari

CaCl<sub>2</sub> = 0,032094 kg/hari

MgCl<sub>2</sub> = 0,000002 kg/hari

NaCl = 0,000004 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,000051 kg/hari

HCl 5% = 0,522925 kg/hari

Gelatin = 31,711631 kg/hari

Filtrat keluar :

Gelatin = 331,884039 kg/hari

H<sub>2</sub>O = 4703,732307 kg/hari

HCl 5% = 5,472775 kg/hari

CaCl<sub>2</sub> = 0,335886 kg/hari

MgCl<sub>2</sub> = 0,000018 kg/hari

NaCl = 0,000046 kg/hari

H<sub>3</sub>PO<sub>4</sub> = 0,000529 kg/hari

$$\begin{aligned}
 \Delta H_{out} &= \Delta H_{kolagen} + \Delta H_{gelatin} + \Delta H_{H_2O} + \Delta H_{CaCl_2} + \Delta H_{HCl\ 5\%} + \Delta H_{MgCl_2} + \Delta H_{NaCl} \\
 &\quad + \Delta H_{H_3PO_4}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{kolagen} &= m \cdot Cp \cdot (T_2 - 25) \\
 &= 1926,8395 \cdot 0,3369 \cdot (45 - 25) \\
 &= 12983,04455 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{gelatin}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (31,711631 + 331,884039) \cdot 0,3613 \cdot (45 - 25) \\
 &= 2627,342311 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_2\text{O}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (449,443193 + 4703,732307) \cdot 1 \cdot (45 - 25) \\
 &= 103063,51 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{CaCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (0,032094 + 0,335886) \cdot 0,2162 \cdot (45 - 25) \\
 &= 1,59114552 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{MgCl}_2} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (0,000002 + 0,000018) \cdot 0,252 \cdot (45 - 25) \\
 &= 0,0001008 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{NaCl}} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (0,000004 + 0,000046) \cdot 0,2737 \cdot (45 - 25) \\
 &= 0,0002737 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{H}_3\text{PO}_4} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (0,000051 + 0,000529) \cdot 0,452 \cdot (45 - 25) \\
 &= 0,0052432 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{HCl } 5\%} &= m \cdot C_p \cdot (T_2 - 25) \\
 &= (0,522925 + 5,472775) \cdot 0,8377 \cdot (45 - 25) \\
 &= 115,9406496 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{out}} &= 12983,04455 + 103063,51 + 115,9406496 + 1,59114552 \\
 &\quad + 0,0001008 + 0,0002737 + 0,0052432 + 2627,342311 \\
 &= 118791,4343 \text{ kkal/hari}
 \end{aligned}$$

$$\Delta H_{in} = \Delta H_{out} + Q_{loss}$$

$$148489,2928 = 118791,4343 + Q_{loss}$$

$$Q_{loss} = 29697,85857 \text{ kkal/hari}$$

Masuk :

Keluar :

$$\text{Panas masuk} = 148489,2928 \text{ kkal/hari} \quad \text{Panas keluar} = 118791,4343 \text{ kkal/hari}$$

$$\begin{array}{rcl} & \underline{\hspace{1.5cm}} & \text{Panas hilang} = 29697,85857 \text{ kkal/hari} \\ = 148489,2928 \text{ kkal/hari} & & = 148489,2928 \text{ kkal/hari} \end{array}$$

### Tangki Penampung Ekstraksi (F-253)

$$\Delta H_{in} = \Delta H_{out} + Q_{loss}$$

dimana :  $\Delta H_{in}$  = Panas bahan masuk

$\Delta H_{out}$  = Panas bahan keluar

#### Menghitung panas masuk ( $\Delta H_{in}$ ) :

$$\text{Suhu bahan masuk} = 45^{\circ}\text{C}$$

$$\text{Massa total larutan gelatin} = 2536,948869 \text{ kg/hari}$$

$$\text{Massa total air} = 20314,63249 \text{ kg/hari}$$

$$\text{Massa total impurities} = 81,473449 \text{ kg/hari}$$

$$\text{Massa total HCl 5\%} = 90,117095 \text{ kg/hari}$$

$$\begin{aligned} \Delta H_{\text{larutan gelatin}} &= m_{\text{larutan gelatin}} \cdot C_{p_{\text{larutan gelatin}}} \cdot (T1-25) \\ &= 2536,948869 \cdot 0,92961526 \cdot (45-25) \\ &= 47167,72765 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned} \Delta H_{\text{air}} &= m_{\text{air}} \cdot C_{p_{\text{air}}} \cdot (T1-25) \\ &= 20314,63249 \cdot 1 \cdot (45-25) \\ &= 406292,6498 \text{ kkal/hari} \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{impurities}} &= m_{\text{impurities}} \cdot C_{p_{\text{impurities}}} \cdot (T_1 - 25) \\
 &= 81,473449 \cdot 1,1939 \cdot (45 - 25) \\
 &= 1945,423015 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{HCl 5\%}} &= m_{\text{HCl 5\%}} \cdot C_{p_{\text{HCl 5\%}}} \cdot (T_1 - 25) \\
 &= 90,117095 \cdot 0,8377 \cdot (45 - 25) \\
 &= 1509,8218 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{in}} &= 47167,72765 + 1945,423015 + 1509,8218 + 406292,6498 \\
 &= 457148,4218 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \text{Panas yang hilang} &= 0,05 \times 457148,4218 \\
 &= 22857,42109 \text{ kkal/hari}
 \end{aligned}$$

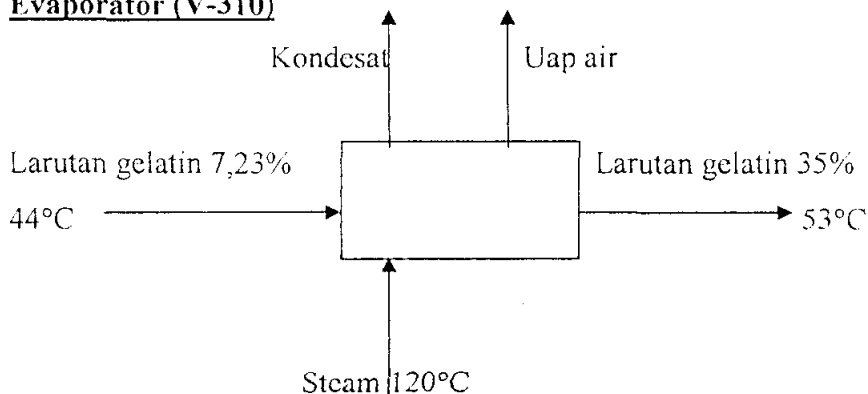
$$\begin{aligned}
 \Delta H_{\text{out}} &= \Delta H_{\text{in}} - \text{panas hilang} \\
 &= 457148,4218 - 22857,42109 \\
 &= 434291,0007 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \text{Suhu larutan gelatin keluar} &= \frac{434291,0007}{\left(\sum m \cdot C_p\right)} + 25 \\
 &= 44 \text{ } ^\circ\text{C}
 \end{aligned}$$

Masuk :

Keluar :

$$\begin{aligned}
 \text{Panas masuk} &= 457148,4218 \text{ kkal/hari} & \text{Panas keluar} &= 434291,0007 \text{ kkal/hari} \\
 & & \text{Panas hilang} &= \underline{22857,42109 \text{ kkal/hari}} \\
 &= 457148,4218 \text{ kkal/hari} & &= 457148,4218 \text{ kkal/hari}
 \end{aligned}$$

Evaporator (V-310)

Pada tangki evaporator ini digunakan steam pada suhu 120°C dan proses evaporasi beroperasi pada tekanan 4 inHg (= 13,5457 kPa = 0,13 atm).

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + Q_v + Q_{loss}$$

dimana :  $\Delta H_{in}$  = panas bahan masuk

$\Delta H_{out}$  = panas bahan keluar

$Q_v$  = panas penguapan

$Q_{loss} = 0,05 \cdot Q_s$

$$\begin{aligned} \text{Berat total gelatin} &= 1078,92 + 649,1502 + 332,0916 + 194,8271 + 313,6716 \\ &= 2568,6605 \text{ kg} - 31,711631 \text{ kg} \\ &= 2536,948869 \text{ kg} \end{aligned}$$

$$\begin{aligned} \text{Berat total H}_2\text{O} &= 10339,65 + 5779,9123 + 4644,51338 - 449,443193 \\ &= 20314,63249 \text{ kg} \end{aligned}$$

$$\Delta t_{b_{H_2O}} = K_b \cdot m_{H_2O}$$

dimana :  $\Delta t_b$  = kenaikan titik didih air

$K_b$  = tetapan kenaikan titik didih molal (untuk air,  $K_b = 0,52$ )

$m$  = molalitas air

$$\begin{aligned} \Delta t_{b_{H_2O}} &= 0,52 \times \left( \frac{2536,948869}{2435,578} \times \frac{1000}{20314,63249} \right) \\ &= 0,026662695 \end{aligned}$$

$$\begin{aligned}
 \text{Titik didih larutan} &= T_b \text{ air} + \Delta t_{\text{bair}} \\
 &= 53 + 0,026662695 \\
 &= 53,026662695 \text{ } ^\circ\text{C} \approx 53^\circ\text{C}
 \end{aligned}$$

$$\text{Titik didih larutan} = 53 \text{ } ^\circ\text{C}$$

$$\begin{aligned}
 \text{Cp gelatin (11,02\%)} &= (0,1102 \times 0,3613) + (0,8898 \times 1) \\
 &= 0,92961526 \text{ kkal/kg } ^\circ\text{C}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{in}} &= m_{\text{larutan gelatin}} \cdot \text{Cp} \cdot (T_1 - 25) \\
 &= 23023,1719 \cdot 0,92961526 \cdot (44 - 25) \\
 &= 406651,1467 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas penguapan (Qv) :

$$\text{Jumlah uap} = 15774,1618 \text{ kg/hari}$$

$$\Delta H_{\text{uap air jenuh pada } 53^\circ\text{C}} = 567,7605 \text{ kkal/kg}$$

$$\text{Cp}_{\text{H}_2\text{O}} \text{ pada } 53^\circ\text{C} =$$

$$a + \frac{b}{2} \cdot (T_2 + T_1) + \frac{c}{3} \cdot (T_2^2 + T_2 \cdot T_1 + T_1^2) + \frac{d}{4} \cdot (T_2 + T_1) \cdot (T_2^2 + T_1^2)$$

$$\text{dimana : } a = 33,46 \qquad c = 0,7604 \cdot 10^{-5}$$

$$b = 0,688 \cdot 10^{-2} \qquad d = -3,593 \cdot 10^{-9}$$

$$T_2 = 53^\circ\text{C} \qquad T_1 = 25^\circ\text{C}$$

$$\text{Cp}_{\text{H}_2\text{O}} \text{ pada } 53^\circ\text{C} = 0,4476 \text{ kkal/kg } ^\circ\text{C}$$

$$\begin{aligned}
 Q_{\text{VH}_2\text{O}} &= 15774,1618 \times (\Delta H_{\text{uap air jenuh}} + (\text{Cp}_{\text{H}_2\text{O}} \times \Delta t_{\text{bair}})) \\
 &= 15774,1618 \times (567,7605 + (0,4476 \times (53 - 25))) \\
 &= 9153640,406 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{out}} &= m_{\text{larutan gelatin keluar}} \cdot \text{Cp} \cdot (T_2 - 25) \\
 \text{Cp gelatin 35\%} &= (0,35 \times 0,3613) + (1 \times 0,65) \\
 &= 0,777 \text{ kkal/kg } ^\circ\text{C}
 \end{aligned}$$

$$\Delta H_{out} = 7249,0101 \cdot 0,777 \cdot (53-25)$$

$$= 157709,4637 \text{ kkal/hari}$$

$$\Delta H_{in} + Q_{steam} = \Delta H_{out} + Q_{v_{H_2O}} + Q_{loss}$$

$$406651,1467 + Q_{steam} = 9153640,406 + 157709,4637 + 0,05 \cdot Q_{steam}$$

$$Q_{steam} = 9373367,077 \text{ kkal/hari}$$

Digunakan steam pada suhu 120°C dan tekanan 198,53 kPa didapat :

$$H_v = 2706,3 \text{ kJ/kg} = 646,8057 \text{ kkal/kg}$$

$$H_L = 503,71 \text{ kJ/kg} = 120,3843 \text{ kkal/kg}$$

$$\lambda = H_v - H_L = (646,8057 - 120,3843) \text{ kkal/hari} = 526,4214 \text{ kkal/hari}$$

$$\text{Jumlah steam yang digunakan} = \frac{9373367,077}{526,4214} = 17805,8245 \text{ kg/hari}$$

$$\text{Panas hilang} = 0,05 \times 9373367,077$$

$$= 468668,3539 \text{ kkal/hari}$$

Masuk :

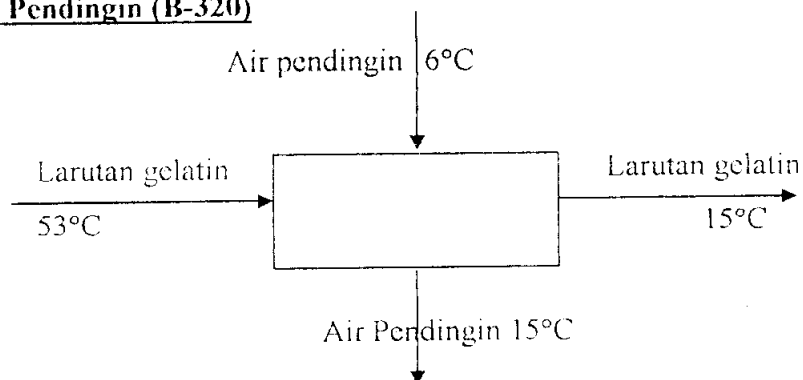
Keluar :

$$\text{Panas masuk} = 406651,1467 \text{ kkal/hari} \quad \text{Panas keluar} = 157709,4637 \text{ kkal/hari}$$

$$\text{Panas Steam} = 9373367,067 \text{ kkal/hari} \quad \text{Panas Penguapan} = 9153640,406 \text{ kkal/hari}$$

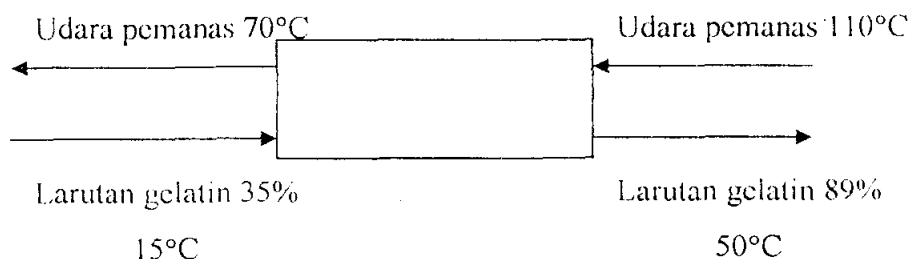
$$\begin{array}{rcl} \text{Panas hilang} & = & 468668,3539 \text{ kkal/hari} \\ \hline & = & 9780018,214 \text{ kkal/hari} \end{array}$$

### Alat Pendingin (B-320)







Spray Dryer (B-330)

$$\Delta H_{in} + Q_{pemanas} = \Delta H_{out} + \Delta H_{penguapan}$$

dimana :  $\Delta H_{in}$  = panas bahan masuk

$\Delta H_{out}$  = panas bahan keluar

$Q_{pemanas}$  = panas dari udara pemanas

$\Delta H_{penguapan}$  = panas penguapan air

Suhu larutan masuk = 15°C

Suhu larutan keluar = 50°C

Massa larutan masuk = 7249,0101 kg/hari

Massa larutan keluar = 2850,5043 kg/hari

Kadar air = 11%

Kadar gelatin = 89%

$C_{p_{gelatin\ 35\%}} = 0,7777 \text{ kkal/kg } ^\circ\text{C}$

$C_{p_{gelatin\ 89\%}} = (0,89 \times 0,3613) + (1 \times 0,11)$   
 $= 0,431557 \text{ kkal/kg } ^\circ\text{C}$

$$\begin{aligned}\Delta H_{in} &= m_{\text{larutan gelatin masuk}} \cdot C_p \cdot (T_1 - 25) \\ &= 7249,0101 \cdot 0,7777 \cdot (15 - 25) \\ &= -56375,55155 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{out} &= m_{\text{larutan gelatin keluar}} \cdot C_p \cdot (T_2 - 15) \\ &= 2850,50437 \cdot 0,431557 \cdot (50 - 15) \\ &= 43055,42866 \text{ kkal/hari}\end{aligned}$$

$$\text{mol air yang diuapkan} = \frac{4398,5058}{18,016} = 244,1444135 \text{ kmol/hari}$$

$$\begin{aligned}\Delta H_{\text{penguapan}} &= 244,1444135 \frac{\text{kmol}}{\text{hari}} \times 1000 \frac{\text{mol}}{\text{kmol}} \times 40,65 \frac{\text{kJoule}}{\text{mol}} \\ &= 9924470409 \text{ kJ/hari} = 9924470409 \frac{\text{kJ}}{\text{hari}} \times 0,239 \frac{\text{kcal}}{\text{kJ}} \\ &= 2371948,428 \text{ kkal/hari}\end{aligned}$$

$$\Delta H_{\text{in}} + Q_{\text{pemanas}} = \Delta H_{\text{out}} + \Delta H_{\text{penguapan}}$$

$$Q_{\text{pemanas}} = 43055,42866 + 2371948,428 + 56375,55155$$

$$Q_{\text{pemanas}} = 2471379,408 \text{ kkal/hari}$$

$$\text{Suhu udara masuk} = 110^{\circ}\text{C}$$

$$\text{Suhu udara keluar} = 70^{\circ}\text{C}$$

$$C_p (\text{N}_2 \text{ gas})$$

$$\begin{aligned}&= 29 + \left( \frac{2,199 \cdot 10^{-3}}{2} \times (110 + 70) \right) + \left( \frac{5,723 \cdot 10^{-6}}{3} \times (70^2 + (70 \cdot 110) + 110^2) \right) + \\ &\left( \frac{-2,871 \cdot 10^{-9}}{4} (70^2 + 110^2)(70 + 110) \right) \\ &= 29,2749 \text{ J/mol } ^{\circ}\text{C} = 0,2450 \text{ kkal/kg } ^{\circ}\text{C}\end{aligned}$$

$$C_p (\text{O}_2 \text{ gas})$$

$$\begin{aligned}&= 29,1 + \left( \frac{11,58 \cdot 10^{-3}}{2} \times (110 + 70) \right) + \left( \frac{-6,076 \cdot 10^{-6}}{3} \times (70^2 + (70 \cdot 110) + 110^2) \right) + \\ &\left( \frac{1,311 \cdot 10^{-9}}{4} (70^2 + 110^2) \times (70 + 110) \right) \\ &= 30,6198 \text{ J/mol } ^{\circ}\text{C} = 0,2287 \text{ kkal/kg } ^{\circ}\text{C}\end{aligned}$$

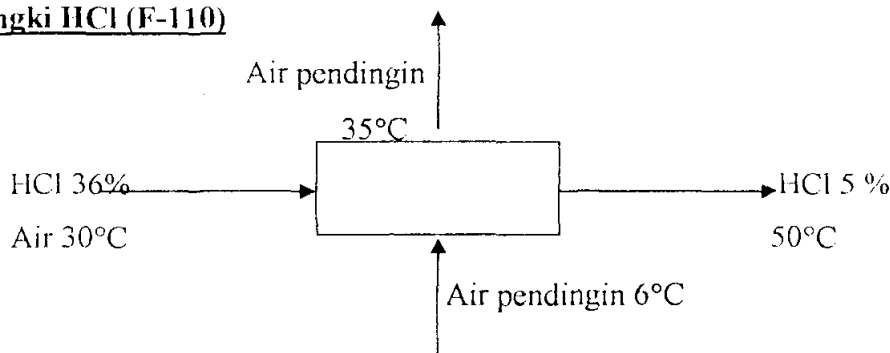
$$C_p \text{ udara} = (0,79 \times 0,2450) + (0,21 \times 0,2287)$$

$$= 0,2416 \text{ kkal/kg } ^{\circ}\text{C}$$

$$\begin{aligned}
 \text{Jumlah udara yang dibutuhkan} &= \frac{Q_{\text{pemanas}}}{C_{p_{\text{udara}}} \cdot (T_2 - T_1)} \\
 &= \frac{2471379,408}{0,2416 \cdot (110 - 70)} \\
 &= 91332,3161 \text{ kg/hari}
 \end{aligned}$$

Masuk :Keluar :

$$\begin{aligned}
 Q_{\text{pemanas}} &= 2471379,408 \text{ kkal/hari} & \text{Panas keluar} &= 18452,32657 \text{ kkal/hari} \\
 \text{Panas masuk} &= 56375,55155 \text{ kkal/hari} & \text{Panas penguapan} &= 2371948,428 \text{ kkal/hari} \\
 &= 2471379,408 \text{ kkal/hari} & &= 2471379,408 \text{ kkal/hari}
 \end{aligned}$$

Tangki HCl (F-110)

$$\Delta H_{\text{in}} = \Delta H_{\text{out}} + \text{Panas pengenceran} + Q_{\text{air pendingin}}$$

dimana :  $\Delta H_{\text{in}}$  = panas bahan masuk

$\Delta H_{\text{out}}$  = panas bahan keluar

Menghitung panas masuk ( $\Delta H_{\text{in}}$ ):

Suhu larutan masuk =  $30^\circ\text{C}$

Massa larutan HCl 36% = 20647,2617 kg/hari

Massa larutan air = 128013,0223 kg/hari

$C_{p_{\text{HCl 36\%}}} = 0,5818 \text{ kkal/kg}^\circ\text{C}$

$$\begin{aligned}
 \Delta H_{\text{HCl } 36\%} &= m \cdot C_p \cdot (T_1 - 25) \\
 &= 20647,2617 \cdot 0,5818 \cdot (30 - 25) \\
 &= 60062,8843 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{air}} &= m \cdot C_p \cdot (T_1 - 25) \\
 &= 128013,0223 \cdot 1 \cdot (30 - 25) \\
 &= 640337,0655 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{\text{in}} &= 60062,8843 + 640337,0655 \\
 &= 700399,9498 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas pengenceran ( $\Delta H_s$ ) :

$$\begin{aligned}
 \text{Pada HCl } 36\% \text{ didapat } \Delta H &= -48 \text{ kJ/mol} && (\text{Smith Van Ness}) \\
 &= -314,4 \text{ kkal/kg} \\
 \Delta H_1 &= -314,4 \times 20647,2617 \\
 &= -6491499,078 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \text{Pada HCl } 5\% \text{ didapat } \Delta H &= -70 \text{ kJ/mol} && (\text{Smith Van Ness}) \\
 &= -458,5 \text{ kkal/kg} \\
 \Delta H_2 &= -458,5 \times 148660,2840 \\
 &= -68160740,21 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_s &= -68160740,21 + 6491499,078 \\
 &= -61669241,13 \text{ kkal/hari}
 \end{aligned}$$

Menghitung panas keluar ( $\Delta H_{\text{out}}$ ) :

Suhu larutan keluar = 50°C

massa larutan HCl 5% = 148660,284 kg/hari

$C_{p\text{HCl } 5\%} = 0,8377 \text{ kkal/kg}^\circ\text{C}$

$$\begin{aligned}
 \Delta H_{out} &= \Delta H_{HCl\ 5\%} = m \cdot C_p \cdot (T_2 - 25) \\
 &= 148660,284 \cdot 0,8377 \cdot (50 - 25) \\
 &= 3113317,998 \text{ kkal/hari}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H_{in} &= \Delta H_{out} + \text{Panas pengenceran} + Q_{air\ pendingin} \\
 700399,9498 + 61669241,13 &= 3113317,998 + Q_{air\ pendingin} \\
 Q_{air\ pendingin} &= 59256323,08 \text{ kkal/hari}
 \end{aligned}$$

Suhu air masuk = 6°C

Suhu air keluar = 35°C

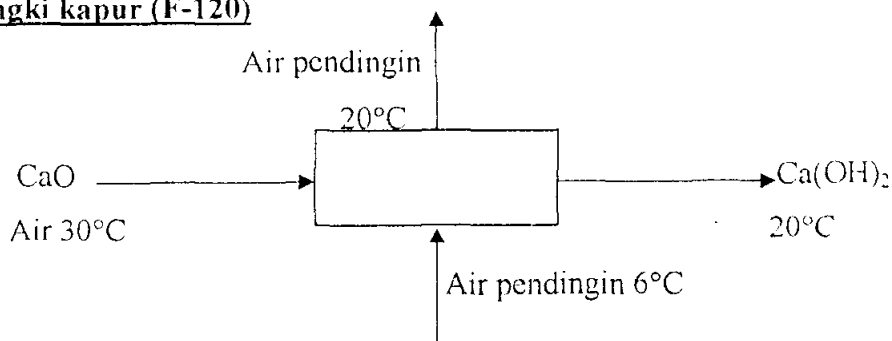
$$\begin{aligned}
 \text{Massa air pendingin yang dibutuhkan} &= \frac{59256323,08}{1 \cdot (35 - 6)} \\
 &= 2043321,486 \text{ kg/hari}
 \end{aligned}$$

Masuk :

$$\begin{aligned}
 \text{Panas masuk} &= 700399,9498 \text{ kkal/hari} & \text{Panas keluar} &= 3113317,998 \text{ kkal/hari} \\
 \text{Panas Pengencer} &= 61669241,13 \text{ kkal/hari} & Q_{air\ pendingin} &= 59256323,08 \text{ kkal/hari} \\
 &= 62369641,08 \text{ kkal/hari} & &= 62369641,08 \text{ kkal/hari}
 \end{aligned}$$

Keluar :

### Tangki kapur (F-120)



$$\begin{aligned}
 \Delta H_{in} + \Delta H_r &= \Delta H_{out} + Q_{air\ pendingin} \\
 \text{dimana : } \Delta H_r &= \text{panas reaksi} \\
 \Delta H_{in} &= \text{panas bahan masuk} \\
 \Delta H_{out} &= \text{panas bahan keluar}
 \end{aligned}$$

Menghitung panas masuk ( $\Delta H_{in}$ ) :

Suhu larutan masuk = 30°C

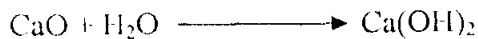
Massa CaO = 1798,2 kg/hari

Massa air = 16183,8000 kg/hari

$$\begin{aligned}\Delta H_{CaO} &= m_{CaO} \cdot C_{pCaO} \cdot (T1-25) \\ &= 1798,2 \cdot 0,2668 \cdot (30-25) \\ &= 2398,7988 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{air} &= m_{air} \cdot C_{pair} \cdot (T1-25) \\ &= 16183,8 \cdot 1 \cdot (30-25) \\ &= 80919 \text{ kkal/hari}\end{aligned}$$

$$\begin{aligned}\Delta H_{in} &= 80919 + 2398,7988 \\ &= 83818,28059 \text{ kkal/hari}\end{aligned}$$

Menghitung panas reaksi ( $\Delta H_r$ ) :

Mol CaO yang bereaksi = 32,0649 kmol/hari

Mol air yang bereaksi = 32,0649 kmol/hari

Mol Ca(OH)<sub>2</sub> = 32,0649 kmol/hari

$$\begin{aligned}\Delta H_r &= (32,0649 \times -235,7950) - ((32,0649 \times -68,2782) + (32,0649 \times -151,9084)) \\ &= -500,48179 \text{ kkal/hari}\end{aligned}$$

Menghitung panas keluar ( $\Delta H_{out}$ ) :

Suhu bahan keluar = 20°C

Massa Ca(OH)<sub>2</sub> = 2375,8808 kg/hari

Massa air = 15606,1188 kg/hari

$$\begin{aligned}\Delta H_{air} &= m_{air} \cdot C_{pair} \cdot (T2-25) \\ &= 15606,1188 \cdot 1 \cdot (20-25) \\ &= -78030,594 \text{ kkal/hari}\end{aligned}$$





## APPENDIKS C.

### PERHITUNGAN SPESIFIKASI PERALATAN

## APPENDIX C

### PERHITUNGAN SPESIFIKASI ALAT

#### 1. WAREHOUSE TULANG (F-130)

Fungsi : menyimpan bahan baku tulang

Tipe : gedung dengan bahan konstruksi concrete beton

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk padat

Kondisi operasi :  $T = 30^{\circ}\text{C}$

Jumlah : 1 buah

Perhitungan :

Rate tulang masuk = 13500 kg/hari

Persediaan tulang/bulan = 13500 kg/hari x 4 hari  
= 54000 kg = 119070 lbm

$\rho_{\text{tulang}} = 1,9 \text{ gr/cm}^3 \times 62,43$   
= 118,617 lbm/cuft

Volume tulang = 119070 lbm / (118,617 lbm/cuft)  
= 1003,819 cuft

Diambil volume ruang kosong = 15% x volume tulang  
= 0,15 x 1003,819 cuft  
= 150,5729 cuft

$\varepsilon = \text{void fraction} = \frac{\text{volume ruang kosong}}{\text{volume ruang kosong} + \text{volume tulang}}$   
=  $\frac{150,5729}{150,5729 + 1003,819} = 0,1304$

$\rho_{\text{tulang bulk}} = 118,617 \text{ lbm/cuft} \times (1 - 0,1304)$   
= 103,1493 lbm/cuft

Volume tulang = 119070 lbm / (103,1493 lbm/cuft)  
= 1154,3462 cuft

Diambil : volume warehouse = 2 x volume tulang  
= 2 x 1154,3462 cuft = 2308,6924 cuft

Warehouse dengan alas bujursangkar dengan ketentuan :

- Tinggi bangunan = 15 ft
- Panjang = lebar =  $\sqrt{\frac{2308,6924 \text{ cuft}}{15 \text{ ft}}} = 12,5 \text{ ft}$

Spesifikasi :

Kapasitas : 54000 kg = 119070 lbm

Waktu tinggal : 4 hari

Panjang : 12,5 ft

Lebar : 12,5 ft

Tinggi : 15 ft

Jumlah : 1 buah

## 2. TANGKI HCl (F-110)

Fungsi : menyiapkan bahan baku HCl 5 %

Tipe : silinder tegak dengan tutup atas dan tutup bawah berbentuk dished head dan dilengkapi dengan pengaduk dan coil pendingin.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk liquida

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Kebutuhan HCl 5 % = 148660,284 kg/hari x 2,2046  
= 327736,4621 lbm/hari

$\rho_{\text{HCl } 5\%} = 1,01185 \text{ gr/cm}^3 \times 62,43$   
= 63,1698 lbm/cuft

Rate volume =  $\frac{327736,4621 \text{ lbm} / \text{hari}}{63,1698 \text{ lbm} / \text{cuft}}$   
= 5188,1827 cuft/hari

Volume larutan HCl 5% per hari = 5188,1827 cuft

Diambil : tinggi shell ( $H_s$ ) = 1,5 . Diameter shell ( $D$ )

Volume shell =  $(\pi/4) \cdot D^2 \cdot H_s = (\pi/4) \cdot D^2 \cdot 1,5 \cdot D = 1,5 \cdot (\pi/4) \cdot D^3$

Volume torispherical dished head (cuft) = 0,000049 x  $D^3$  (inch)

(Brownell &amp; Young, pers.5.11,p.88)

Volume tangki penampung = vol.shell + (2.vol.dished head)

Diambil : volume tangki penampung = 1,2.volume larutan total

$$1,2 \cdot 5188,1827 \text{ cuft} = \frac{\pi}{4} \cdot 1,5 \cdot D^3 \cdot \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 + (2 \cdot 0,000049 \cdot D^3)$$

$$6225,8192 \text{ cuft} = 6,8177 \cdot 10^{-4} \cdot D^3 + 9,8 \cdot 10^{-5} \cdot D^3$$

$$6225,8192 \text{ cuft} = 7,7977 \cdot 10^{-4} \cdot D^3$$

$$D = 199,868 \text{ in} = 16,6557 \text{ ft} \approx 16,7 \text{ ft}$$

$$H_s = 1,5 \cdot D = 1,5 \cdot (16,7 \text{ ft}) = 25,05 \text{ ft}$$

$$\text{Volume larutan dalam dish} = 0,000049 \cdot D^3$$

$$= 0,000049 \cdot (199,868 \text{ in})^3$$

$$= 391,2244 \text{ cuft}$$

$$\text{Volume larutan dalam shell} = \text{vol. larutan total} - \text{vol. larutan dalam dish}$$

$$= (6225,8192 - 391,2244) \text{ cuft}$$

$$= 5834,5948 \text{ cuft}$$

$$\text{Tinggi larutan dalam shell (H)} = \frac{\text{Volume larutan dalam shell}}{\frac{\pi}{4} \cdot D^2}$$

$$= \frac{5834,5948}{\frac{\pi}{4} \cdot (16,7 \text{ ft})^2}$$

$$= 26,6372 \text{ ft}$$

$$P_{\text{operasi}} = P_{\text{hidrostatik}} = \left( \frac{\rho \cdot H}{144} \right) \text{ psi}$$

$$= \left( \frac{63,1698 \text{ lbm / cuft} \cdot 26,6372 \text{ ft}}{144} \right) \text{ psi}$$

$$= 11,6852 \text{ psi}$$

$$P_{\text{desain}} = 1,5 \cdot P_{\text{operasi}} = 1,5 \cdot (11,6852 \text{ psi}) = 17,5278 \text{ psi}$$

**Tebal shell**

$$t_s = \frac{P \cdot H D}{2(f \cdot E - 0,6 \cdot P)} + c$$

(Brownell &amp; Young, pers. 13.1)

dimana :

$$P = P_{\text{desain}} = 17,5278 \text{ psi}$$

$$ID = 16,7 \text{ ft} = 199,868 \text{ in}$$

Konstruksi :

Bahan konstruksi = High Alloy steel SA-240, grade M type 316

$$T = 50^{\circ}\text{C} (122^{\circ}\text{F})$$

$$f = \text{stress maksimum yang diijinkan} = 18750 \text{ psi}$$

(Brownell & Young, p 342-343)

tipe sambungan = *double-welded butt joint*, dengan

$$E = \text{welded-joint efficiency} = 0,8$$

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_s = \frac{(17,5278 \text{ psi})(199,868 \text{ in})}{2.(18750.0,8 - 0,6.17,5278) \text{ psi}} + \frac{1}{8} \text{ in}$$

$$t_s = 0,2419 \text{ in} \approx \frac{3}{8} \text{ in}$$

### Tebal dished head

$$t_s = \frac{3}{8} \text{ in}$$

$$OD = ID + 2.t_s$$

$$= 199,868 \text{ in} + (2. \frac{3}{8}) \text{ in}$$

$$= 200,618 \text{ in}$$

Dari table 5.7 Brownell & Young diperoleh :

$$OD \text{ standar} = 204 \text{ in dan } t_s = \frac{7}{8}$$

$$r \text{ (crown radius / radius of dish)} = 170 \text{ in}$$

$$icr \text{ (inside corner radius / knuckle radius)} = 12 \frac{1}{4} \text{ in}$$

$$W = \frac{1}{4} \left( 3 + \sqrt{\frac{r}{icr}} \right) \quad (\text{Brownell \& Young, pers. 7.76, p.138})$$

$$= \frac{1}{4} \left( 3 + \sqrt{\frac{170}{12,25}} \right)$$

$$= 1,6813$$

$$a = ID/2 = 199,868 \text{ in} / 2 = 99,934 \text{ in}$$

$$AB = ID/2 - icr = (99,934 - 12,25) \text{ in} = 87,684 \text{ in}$$

$$BC = r - icr = (170 - 12,25) \text{ in} = 157,75 \text{ in}$$

$$b = r - \sqrt{BC^2 - AB^2} = 170 - \sqrt{157,75^2 - 87,684^2} = 38,8643 \text{ in}$$

$$t_d = \frac{P \cdot r \cdot W}{2 \cdot f \cdot E - 0,2 \cdot P} + c \quad (\text{Brownell \& Young, pers. 7.77, p.138})$$

dimana :

$$P = P_{\text{desain}} = 14,4105 \text{ psi}$$

Konstruksi : bahan konstruksi = High Alloy steel SA-240, grade M type 316

$f$  = stress maksimum yang diijinkan = 18750 psi

(Brownell & Young, p.342-343)

tipe sambungan = *double-welded butt joint*, dengan

$E$  = *welded-joint efficiency* = 0,8

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_d = \frac{(14,4105 \text{ psi}) \cdot (170 \text{ in}) \cdot 1,6813}{2 \cdot (18750 \text{ psi}) \cdot (0,8) - 0,2 \cdot (14,4105 \text{ psi})} + \frac{1}{8} \text{ in}$$

$$= 0,1373 \text{ in} \approx \frac{3}{8} \text{ in}$$

Dari table 5.7 Brownell & Young diperoleh :

$$OD \text{ standar} = 204 \text{ in} \text{ dan } t_s = t_d = \frac{7}{8}$$

Dipilih panjang straight-flange ( $sf$ ) = 8 in

(Brownell & Young, table 5.8, p.93)

$$OA = t + b + sf$$

$$= (7/8 + 38,8643 + 8) \text{ in}$$

$$= 47,7393 \text{ in} = 3,9783 \text{ ft} \approx 4 \text{ ft}$$

Tinggi tangki keseluruhan = tinggi shell + (2 x tinggi dish)

$$= 25.05 \text{ ft} + (2 \times 4 \text{ ft})$$

$$= 33,05 \text{ ft}$$

### Perhitungan Pengaduk

Dipilih tipe pengaduk : flat six-blade turbine agitator with disk.

Kecepatan agitator antara 20 – 200 rpm (Geankoplis), diambil 200 rpm.

$$N = 50 \text{ rpm} = 0,8333 \text{ rps.}$$

$$\rho_{\text{larutan HCl 5\%}} = 1,01185 \text{ g/cm}^3 = 1011,85 \text{ kg/m}^3$$

$$\mu_{\text{larutan HCl 5\%}} = 0,6744 \text{ cp} = 0,6744 \cdot 10^{-3} \text{ kg/m.s} \quad (\text{Reid c., Robert})$$

Dari table 3.4-1, p.144 Geankoplis diperoleh :

$$Da/Dt = 0,3 \quad W/Da = 1/5 \quad L/Da = 1/4 \quad C/Dt = 1/3 \quad J/Dt = 1/12$$

dimana :

$$Dt = \text{diameter tangki} = 16,7 \text{ ft} / 3,2808 \text{ (m/ft)} = 5,0902 \text{ m}$$

$$Da = \text{diameter pengaduk} = 0,3 \times 16,7 \text{ ft} = 5,01 \text{ ft} = 1,5271 \text{ m}$$

$$W = \text{lebar blade} = (5,01 \text{ ft})/5 = 1,002 \text{ ft} = 0,3054 \text{ m}$$

$$L = \text{panjang blade} = (5,01 \text{ ft})/4 = 1,2525 \text{ ft} = 0,3818 \text{ m}$$

$$C = \text{jarak dari dasar tangki ke pusat pengaduk} = (16,7 \text{ ft})/3 = 5,5667 \text{ ft} \\ = 1,6968 \text{ m}$$

$$J = \text{lebar baffle} = \frac{1}{12} \times 5,0902 \text{ m} = 0,4242 \text{ m}$$

$$N_{Re} = \frac{Da^2 \cdot N \cdot \rho}{\mu} \\ = \frac{(1,5271)^2 \cdot 0,8333 \cdot 1011,85}{0,6744 \cdot 10^{-3}} \\ = 2915646,786 \text{ (turbulen)}$$

Dari fig. 3.4-4, p.145 Geankoplis, dengan memotong kurva 1 dengan  $N_{Re}$ , diperoleh  $N_p = 5$ .

$$N_p = \frac{P}{\rho \cdot N^3 \cdot D a^5} \quad (\text{Geankoplis, pers. 3.4-2, p.145})$$

Dimana :

$N_p$  = power number

$P$  = power pengaduk

$N$  = kecepatan pengaduk

$$P = N_p \cdot \rho \cdot N^3 \cdot D a^5$$

$$= 5 \cdot 1011,85 \text{ kg/m}^3 \cdot (0,8333 \text{ rps})^3 \cdot (1,5271 \text{ m})^5$$

$$= 24312,391 \text{ W} / 745,7 \text{ (hp/W)} = 32,6034 \text{ hp}$$

$$\text{Power input} = 110\% \times 32,6034 \text{ hp} = 35,8637 \text{ hp}$$

$$\text{Transmission system losses} = 20\% \text{ dari total hp}$$

$$35,8637 \text{ hp} + 0,2 \cdot \text{total hp} = \text{total hp}$$

$$\text{total hp} = 44,8296 \text{ hp}$$

$$\text{Efisiensi motor} = 88\% \quad (\text{Peter \& Timmerhaus, fig.14-38, p.521})$$

$$\text{Power yang dibutuhkan} = (44,8296 \text{ hp}) / 88\% = 50,9427 \text{ hp} \approx 51 \text{ hp}$$

### Perhitungan Coil Pendingin

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q = 59256323,08 \text{ kkal/hari} = 2869487,445 \text{ W} = 979113,333 \text{ btu/h}$
- ❖ Suhu bahan masuk tangki pada  $30^\circ\text{C}$  ( $86^\circ\text{F}$ ) yaitu HCl 36% dan air
- ❖ Suhu HCl 5% keluar  $50^\circ\text{C}$  ( $122^\circ\text{F}$ )
- ❖ Pendingin berupa air dengan suhu masuk  $6^\circ\text{C}$  ( $42,8^\circ\text{F}$ ) dan suhu keluar  $35^\circ\text{C}$  ( $95^\circ\text{F}$ )
- ❖ Massa air pendingin =  $2043321,486 \text{ kg/hari} = 187696,1062 \text{ lb/h}$

$$\rho_{\text{larutan HCl 5\%}} = 1,01185 \text{ g/cm}^3 = 1011,85 \text{ kg/m}^3 = 63,1698 \text{ lbm/cu ft}$$

$$\mu_{\text{larutan HCl 5\%}} = 0,6744 \text{ cp} = 0,6744 \cdot 10^{-3} \text{ kg/m.s}$$

$$C_{p\text{HCl 5\%}} = 0,966865 \text{ kkal/kg } ^\circ\text{C} \times 4,1868$$

$$= 4,0481 \text{ kJ/kg.K} = 4048,1 \text{ J/kg.K}$$

$$t_f = (T + t) / 2 = (86 + 42,8) / 2 = 64,4^\circ\text{F}$$

$$\Delta t = (T - t) = (86 - 42,8)^\circ\text{F} = 43,2^\circ\text{F}$$

Berdasarkan kern tabel 11, p.844, trial ukuran pipa coil = 10 in IPS, sch 40



$$D_o = 10,75 \text{ in}$$

$$D_i = 10,02 \text{ in}$$

$$a' = 78,8 \text{ in}^2$$

$$a'' = 2,814 \text{ ft}^2/\text{ft}$$

### Evaluasi Perpindahan Panas

Sisi bejana : fluida panas	Sisi pipa : air, fluida dingin
$\rho_{camp} = 63,1698 \text{ lb/ft}^3$ $\mu_{camp \text{ HCl } 5\%} = 1,6314 \text{ lb/ft.h}$ $D_a = 5,0101 \text{ ft}$ $N = 50 \text{ rpm} = 3000 \text{ rph}$ $N_{re} = \frac{\rho D_a^2 N}{\mu}$ $= \frac{63,1698 (5,0101)^2 \cdot 3000}{1,6314}$ $= 2915835,958$ $J_c = 2000 \quad (\text{kern fig 20-2, p-718})$ $h_o = J_c \cdot \frac{k}{D_{ivesse}} \left( \frac{C_p \mu}{k} \right)^{1/3} \left( \frac{\mu}{\mu_w} \right)^{0,14}$ Dimana : $D_i = 16,7 \text{ ft}$ $k = 0,45 \text{ W/m.K} = 0,26 \text{ Btu/hr.ft.}^\circ\text{F}$ $C_p \text{ rata-rata} = 4,0481 \text{ kJ/kg.K}$ $= 0,9669 \text{ Btu/lbm.}^\circ\text{F}$ $h_o = 2000 \cdot \frac{0,26}{16,7} \left( \frac{0,9669 \cdot 1,6314}{0,26} \right)^{1/3}$ $= 56,7906 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$	$A_p = 78,8 \text{ in}^2 = 0,5472 \text{ ft}^2$ $G_p = \frac{W}{a_p} = \frac{187696,1062}{0,5472} = 343017,2019 \text{ lb/h.ft}^2$ $N_{re} = \frac{D_p G_p}{\mu}$ Dimana : $D_p = 10,02 \text{ in} = 0,835 \text{ ft}$ $\mu = 3,5629 \text{ lb/ft.h}$ $N_{re} = \frac{343017,2019 \cdot 0,835}{3,5629}$ $= 80389,39167$ $J_H = 250 \quad (\text{Kern, fig.24, p.834})$ $h_i = J_H \cdot \frac{k}{D_{ivesse}} \left( \frac{C_p \mu}{k} \right)^{1/3} \left( \frac{\mu}{\mu_w} \right)^{0,14}$ dimana : $C_p = 4205 \text{ J/kg.K} = 1,0043 \text{ btu/lbm.}^\circ\text{F}$ $k = 0,5784 \text{ W/m.K} = 0,3342 \text{ btu/hr.ft.}^\circ\text{F}$ $(\text{Geankoplis, App.A.2-11, p.862})$ $h_i = 250 \cdot \frac{0,3342}{0,835} \left( \frac{1,0043 \cdot 1,6314}{0,3342} \right)^{1/3}$ $= 169,9805 \text{ btu/fr.ft}^2 \cdot ^\circ\text{F}$ $h_{io} = h_i \cdot \frac{d_i}{d_o}$ $= 169,9805 \cdot \frac{10,02}{10,75} = 158,4376 \text{ btu/fr.ft}^2 \cdot ^\circ\text{F}$

$$U_c = \frac{h_{io} h_o}{h_{io} + h_o} = \frac{158,4376 \cdot 56,7906}{158,4376 + 56,7906} = 41,8057 \text{ btu/hr.ft}^2 \cdot ^\circ\text{F}$$

$$R_d = \frac{U_c - U_d}{U_c \times U_d} \quad \text{Diambil } R_d = 0,004$$

$$0,004 = \frac{41,8057 - U_d}{41,8057 \times U_d} ; U_d = 35,8164 \text{ Btu/hr.ft}^2.\text{°F}$$

$$A_{\text{coil}} = \frac{Q}{U_d \Delta T_{\text{LMTD}}} = \frac{979113,333 \text{ btu/h}}{(35,8164 \text{ btu/hr.ft}^2.\text{°F}) 43,2 \text{ °F}} = 632,8012 \text{ ft}^2$$

$$L = \frac{A_{\text{coil}}}{a''} = \frac{632,8012 \text{ ft}^2}{2,814 \text{ ft}^2/\text{ft}} = 224,876 \text{ ft}$$

$$\begin{aligned} d_c &= \text{diameter coil} = 0,65 \text{ . di vessel} \\ &= 0,65 \text{ . (16,7 ft)} \\ &= 10,855 \text{ ft} \end{aligned}$$

$$n_c = \frac{L}{\pi \cdot d_c} = \frac{224,876 \text{ ft}}{\pi \cdot 10,855 \text{ ft}} = 6,5942 \text{ buah} \approx 7 \text{ buah}$$

sc = Spasi coil, diambil 2 in

$$\begin{aligned} h_c &= ((n_c - 1) \cdot (d_o + sc)) + d_o \\ &= ((7 - 1) \cdot (10,75 + 2)) + 10,75 \\ &= 87,25 \text{ in} = 7,2708 \text{ ft} \end{aligned}$$

Pengecekan :

Tinggi liquida dibagian silinder (Hs) = 26,6372 ft

$h_c < H_s$  (memenuhi)

### Spesifikasi :

*Tangki*

Kapasitas	: 6225,8192 cuft
Diameter	: 16,7 ft
Tinggi tutup atas (dished head)	: 4 ft
Tinggi tutup bawah (dished head)	: 4 ft
Tinggi shell	: 25,05 ft
Tinggi tangki total	: 33,05 ft
Tebal shell	: 7/8 in

Tebal tutup atas (dished head)	: 7/8 in
Tebal tutup bawah (dished head)	: 7/8 in
Bahan konstruksi	: High Alloy stell SA-240, grade M type 316
Jumlah tangki	: 1 buah

#### *Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 5,01 ft = 1,5271 m
Kecepatan pengaduk	: 50 rpm
Power motor	: 51 hp
Jumlah pengaduk	: 1 buah

#### *Coil pendingin*

Diameter	: 10,855 ft
Spasi coil	: 2 in
Tinggi coil	: 7,2708 ft

### 3. TANGKI KAPUR (F-120)

Fungsi	: menyiapkan bahan baku larutan kapur 10%
Tipe	: silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berebentuk konis yang dilengkapi dengan pengaduk coil pendingin.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk slurry

Kondisi operasi :  $T = 20^{\circ}\text{C}$

Kadar CaO = 10%

Air kapur 10% yang digunakan adalah = 17982 kg = 39643,1172lbm

s.g CaO = 3,32 (Perry ed.7, Tabel 2-1)

Berat CaO = 10% x 17982 kg = 1798,2 kg x 2,2046 = 3964,3117 lbm

Berat H<sub>2</sub>O = (17982-1798,2) kg = 16183,8 kg x 2,2046 = 35678,8055 lbm

$\rho_{\text{H}_2\text{O}}$  pada  $20^{\circ}\text{C}$  = 0,99823 g/cm<sup>3</sup> (= 62,3195 lbm/cuft) (Geankoplis, 1983)

$\rho_{\text{CaO}}$  = s.g CaO x  $\rho_{\text{H}_2\text{O}}$  = 3,32 x 62,3195 lbm/cuft

$$206,9007 \text{ lbm/cuft}$$

$$\frac{1}{\rho_{\text{campuran}}} = \sum \frac{x_i}{\rho_i} = \frac{1798,2}{17982.206,9007} + \frac{16183,8}{17982.62,3195} = 0,0149$$

$$\rho_{\text{campuran}} = 67,0015 \text{ lbm/cuft} = 1,0732 \text{ g/cm}^3 = 1073,2 \text{ kg/m}^3$$

$$\mu_{\text{campuran}} = \frac{\mu}{\phi_p} \quad (\text{Geankoplis, 1985})$$

$$\phi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{\frac{16183,8 \text{ kg}}{998,23 \text{ kg/m}^3}}{\frac{16183,8 \text{ kg}}{998,23 \text{ kg/m}^3} + \frac{1798,2 \text{ kg}}{1073,2 \text{ kg/m}^3}}$$

$$\varepsilon = 0,9063$$

$$\phi_p = \frac{1}{10^{1,82(1-0,9063)}}$$

$$\phi_p = 0,6753$$

$$\mu_{\text{air}} \text{ pada } 20^\circ\text{C} = 1,005 \cdot 10^{-3} \text{ kg/m.s} \quad (\text{Geankoplis, table A.2-4})$$

$$\mu_{\text{campuran}} = \frac{(1,005 \cdot 10^{-3}) \text{ kg/m.s}}{0,6753}$$

$$\mu_{\text{campuran}} = 1,4882 \cdot 10^{-3} \text{ kg/m.s}$$

$$\begin{aligned} \text{Volume air kapur 10\% per hari} &= \frac{39643,1172 \text{ lbm}}{67,0015 \text{ lbm/cuft}} \\ &= 591,6751 \text{ cuft} \end{aligned}$$

$$Q = \text{panas yang diserap} = 165749,8659 \text{ kkal/hari} = 80262,9198 \text{ W} = 273869,4503 \text{ btu/h}$$

Suhu bahan masuk tangki pada  $30^\circ\text{C}$  ( $=86^\circ\text{F}$ ) dan suhu air kapur keluar  $20^\circ\text{C}$

Pendingin berupa air dengan suhu masuk  $6^\circ\text{C}$  dan suhu keluar  $20^\circ\text{C}$  dengan

$$\text{massa air pendingin} = 11839,0619 \text{ kg/hari} = 1087,5333 \text{ lb/h}$$

$$C_{p\text{air kapur 10\%}} = 3,9186 \text{ kJ/kg.K}$$

Dengan cara perhitungan yang sama pada tangki HCl (F-110), maka didapat :

**Spesifikasi :**

*Tangki*

Kapasitas	: 710,0101 cuft
Diameter	: 8,1 ft
Tinggi tutup atas (dished head)	: 1,8 ft
Tinggi tutup bawah (konis)	: 6,6 ft
Tinggi shell	: 12,15 ft
Tinggi tangki total	: 20,6 ft
Tebal shell	: 5/16 in
Tebal tutup atas (dished head)	: 5/16 in
Tebal tutup bawah (konis)	: 5/16 in
Bahan konstruksi	: Carbon steel SA-283, grade D
Jumlah tangki	: 1 buah

*Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,7407 m
Kecepatan pengaduk	: 50 rpm
Power motor	: 1,75 hp
Jumlah pengaduk	: 1 buah

*Coil pendingin*

Diameter	: 5,265 ft
Spasi coil	: 2 in
Tinggi coil	: 8,4513 ft

### 3. BUCKET ELEVATOR (J-141)

Fungsi	: memindahkan tulang dari gudang bahan baku (F-130) ke crusher (C-142)
Tipe	: centrifugal discharge bucket.
Dasar pemilihan	: cocok untuk partikel kering
Kondisi operasi	: $T = 30^{\circ}\text{C}$

Jumlah : 1 buah

Perhitungan :

Rate tulang = 13500 kg/hari = 0,5625 ton/jam.

Jarak vertikal = 25 ft

Sudut elevasi =  $90^0$  (Perry 6<sup>th</sup> ed., table 7-9)

Dari Perry 6<sup>th</sup> ed. Table 7-9, diperoleh data-data untuk kapasitas 14 ton/jam sebagai berikut :

- Ukuran bucket :  $6 \times 4 \times 4 \frac{1}{4}$  (in)
- Spasi bucket : 12 in
- Kecepatan bucket : 225 ft/menit
- Putaran head shaft : 43 rpm
- Shaft diameter : *Head* =  $1 \frac{1}{8}$  in, *Tail* =  $1 \frac{1}{8}$  in
- Diameter of pulleys : *Head* = 20 in, *Tail* = 14 in
- Lebar belt : 7 in

Maka untuk kapasitas 0,5625 ton/jam diperoleh spesifikasi bucket sebagai berikut :

$$\text{Kecepatan bucket} = \frac{0,5625 \text{ ton/jam}}{14 \text{ ton/jam}} \times 225 \text{ ft/menit} = 9,04 \text{ ft/menit}$$

$$\text{Putaran head shaft} = \frac{0,5625 \text{ ton/jam}}{14 \text{ ton/jam}} \times 43 \text{ rpm} = 1,7277 \text{ rpm}$$

$$\text{Power bucket elevator (hp)} = \frac{TPH \times L}{500} \quad (\text{Perry 6}^{\text{th}} \text{ ed., table 7-13})$$

Dimana : TPH = kapasitas dalam ton/jam = 0,5625 ton/jam

L = tinggi elevasi bucket (diambil = 25 ft)

(Perry 6<sup>th</sup> ed., table 7-13)

$$\text{Power bucket elevator (hp)} = \frac{0,5625 \text{ ton/jam} \times 25 \text{ ft}}{500} = 0,0281 \text{ hp}$$

Efisiensi motor = 80% (Peters & Timmerhaus, 4<sup>th</sup> ed., fig. 14-38, p.521)

$$\text{Power motor yang dipakai} = \frac{0,0281 \text{ hp}}{0,8} = 0,0351 \text{ hp, dipilih 0,5 hp}$$

**Spesifikasi :**

Kapasitas	: 0,5625 ton/jam
Kecepatan bucket	: 9,04 ft/menit
Putaran head shaft	: 1,7277 rpm
Shaft diameter	: $Head = 1\frac{5}{16}$ in, $Tail = 1\frac{1}{16}$ in
Diameter of pulleys	: $Head = 20$ in, $Tail = 14$ in
Lebar belt	: 7 in
Tinggi elevator	: 50 ft
Ukuran bucket	: $6 \times 4 \times 4\frac{1}{4}$ (in)
Spasi bucket	: 12 in
Power motor	: 0,5 hp
Bahan konstruksi	:
	<ul style="list-style-type: none"> <li>• Driving head and boat : carbon steel</li> <li>• Roda : carbon steel</li> <li>• Bucket : cast iron</li> <li>• Belt : karet</li> </ul>
Jumlah	: 1 buah

**4. CRUSHER (C-142)**

Fungsi	: menghancurkan tulang dengan ukuran 1 – 8 cm
Tipe	: Jaw crusher
Kondisi operasi	: $T = 30^{\circ}\text{C}$
Tulang yang harus dihancurkan	$= 13500 \text{ kg/hari} = 0,5625 \text{ ton/jam} \approx 1 \text{ ton/jam}$
Dasar pemilihan	:

- Kapasitas 1 ton/jam dapat digunakan Dodge jaw crusher  
(Perry 6<sup>th</sup> ed., table 8-7)

**Spesifikasi :**

- Fungsi : menghancurkan tulang dengan ukuran 1-8 cm
- Tipe : Dodge jaw crusher
- Kecepatan putaran jaw : 275 rpm

- Power : 3 Hp
- Setting : 1 in
- Crusher size : 4 x 6 in
- Jaw motion :  $\frac{1}{2}$  in
- Jumlah : 1 buah

## 5. BUCKET ELEVATOR (J-143)

Fungsi : memindahkan tulang dari crusher (C-132) ke tangki demineralisasi (F-130)

Tipe : centrifugal discharge bucket.

Dasar pemilihan : cocok untuk partikel kering

Kondisi operasi :  $T = 30^{\circ}\text{C}$

Jumlah : 1 buah

Perhitungan :

Rate tulang = 13500 kg/hari = 0,5625 ton/jam.

Jarak vertikal = 25 ft

Sudut elevasi =  $90^{\circ}$  (Perry 6<sup>th</sup> ed., table 7-9)

Dari Perry 6<sup>th</sup> ed. Table 7-9, diperoleh data-data untuk kapasitas 14 ton/jam sebagai berikut :

- Ukuran bucket :  $6 \times 4 \times 4 \frac{1}{4}$  (in)
- Spasi bucket : 12 in
- Kecepatan bucket : 225 ft/menit
- Putaran head shaft : 43 rpm
- Shaft diameter : Head =  $1 \frac{1}{2}$  in, Tail =  $1 \frac{1}{2}$  in
- Diameter of pulleys : Head = 20 in, Tail = 14 in
- Lebar belt : 7 in

Maka untuk kapasitas 0,5625 ton/jam diperoleh spesifikasi bucket sebagai berikut :

$$\text{Kecepatan bucket} = \frac{0,5625 \text{ ton/jam}}{14 \text{ ton/jam}} \times 225 \text{ ft/menit} = 9,04 \text{ ft/menit}$$



$$\text{Putaran head shaft} = \frac{0,5625 \text{ ton / jam}}{14 \text{ ton / jam}} \times 43 \text{ rpm} = 1,7277 \text{ rpm}$$

$$\text{Power bucket elevator (hp)} = \frac{\text{TPH} \times L}{500} \quad (\text{Perry 6}^{\text{th}} \text{ ed., table 7-13})$$

Dimana : TPH = kapasitas dalam ton/jam = 0,5625 ton/jam

L = tinggi elevasi bucket (diambil = 25 ft)

(Perry 6<sup>th</sup> ed., table 7-13)

$$\text{Power bucket elevator (hp)} = \frac{0,5625 \text{ ton / jam} \times 25 \text{ ft}}{500} = 0,0281 \text{ hp}$$

Efisiensi motor = 80% (Peters & Timmerhaus, 4<sup>th</sup> ed., fig.14-38, p.521)

$$\text{Power motor yang dipakai} = \frac{0,0281 \text{ hp}}{0,8} = 0,0351 \text{ hp, dipilih 0,5 hp}$$

#### Spesifikasi :

Kapasitas : 0,5625 ton/jam

Kecepatan bucket : 9,04 ft/menit

Putaran head shaft : 1,7277 rpm

Shaft diameter : *Head* = 1  $\frac{1}{16}$  in, *Tail* = 1  $\frac{1}{16}$  in

Diameter of pulleys : *Head* = 20 in, *Tail* = 14 in

Lebar belt : 7 in

Tinggi elevator : 50 ft

Ukuran bucket : 6 x 4 x 4  $\frac{1}{4}$  (in)

Spasi bucket : 12 in

Power motor : 0,5 hp

Bahan konstruksi :

- Driving head and boat : carbon steel
- Roda : carbon steel
- Bucket : cast iron
- Belt : karet

Jumlah : 1 buah

## 6. TANGKI DEMINERALISASI (F-140)

Fungsi : Merendam tulang dengan HCl 5 % untuk mendapatkan kolagen

Tipe : silinder tegak dengan tutup atas dished head dan tutup bawah berbentuk konis yang dilengkapi dengan pengaduk dan coil pendingin.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk slurry

Kondisi operasi :  $T = 15^{\circ}\text{C}$

Berat hancuran tulang =  $13500 \text{ kg/hari} \times 2,2046 = 29762,1 \text{ lbm/hari}$

$\rho_{\text{tulang}} = 1,9 \text{ gr/cm}^3 \times 62,43 = 118,617 \text{ lbm/cuft}$

Rate volume tulang =  $\frac{29762,1 \text{ lbm / hari}}{118,617 \text{ lbm / cuft}} = 250,9092 \text{ cuft/hari}$

berat HCl 5% =  $127746,67 \text{ kg/hari} = 281630,3087 \text{ lbm/hari}$

$\rho_{\text{HCl 5\%}} = 1,023 \text{ gr/cm}^3 \times 62,43 = 63,8659 \text{ lbm/cuft} = 1023 \text{ kg/m}^3$

rate volume HCl 5% =  $\frac{281630,3087 \text{ lbm / hari}}{63,8659 \text{ lbm / cuft}} = 4409,7133 \text{ cuft/hari}$

Total rate volume =  $4409,7133 + 2780,0577 = 7189,771 \text{ cuft/hari}$

$\frac{1}{\rho_{\text{campuran}}} = \sum \frac{x_i}{\rho_i} = \frac{29762,1}{311392,4087 \cdot 118,617} + \frac{281630,3087}{311392,4087 \cdot 63,8659} = 0,015$

$\rho_{\text{campuran}} = 66,6667 \text{ lbm/cuft} = 1067,9005 \text{ kg/m}^3$

$\mu_{\text{larutan HCl 5\%}} = 1,035 \text{ cp} = 1,035 \cdot 10^{-3} \text{ kg/m.s}$

$C_p \text{ tulang} = 0,3368 \text{ kkal/kg.}^{\circ}\text{C}$

Jumlah impeller =  $\frac{sg.H}{Dt} = \frac{\left(\frac{66,6667}{62,43}\right) \cdot 27,9}{18,6}$

$= 1,6 \approx 2 \text{ impeller}$

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q = 241811,2206 \text{ kkal/hari} = 11709,7084 \text{ W} = 39955,3294 \text{ btu/h}$
- ❖ Suhu tulang masuk tangki pada  $30^{\circ}\text{C}$  ( $86^{\circ}\text{F}$ ) dan HCl 5% masuk pada suhu  $50^{\circ}\text{C}$  ( $122^{\circ}\text{F}$ )
- ❖ Suhu kolagen dan impurities keluar  $15^{\circ}\text{C}$  ( $59^{\circ}\text{F}$ )

- ❖ Pendingin berupa air dengan suhu masuk  $6^{\circ}\text{C}$  ( $42,8^{\circ}\text{F}$ ) dan suhu keluar  $15^{\circ}\text{C}$  ( $59^{\circ}\text{F}$ )
- ❖ Massa air pendingin =  $26867,9134 \text{ kg/hari} = 2468,08 \text{ lb/h}$

Dengan cara perhitungan yang sama pada tangki HCl ((F-110)), maka didapat :

**Spesifikasi :**

*Tangki*

Kapasitas	: 8627,7252 cuft
Diameter	: 18,6 ft
Tinggi tutup atas (dished head)	: 4,8 ft
Tinggi tutup bawah (konis)	: 15,7 ft
Tinggi shell	: 27,9 ft
Tinggi tangki total	: 48,4 ft
Tebal shell	: 1 in
Tebal tutup atas (dished head)	: 1 in
Tebal tutup bawah (konis)	: 1 in
Bahan konstruksi	: High Alloy stell SA-240, grade M type 316
Jumlah tangki	: 7 buah

*Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 1,7008 m = 5,58 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 89,0981 Hp
Jumlah pengaduk	: 2 buah

*Coil pendingin*

Diameter	: 12,09 ft
Spasi coil	: 2 in
Tinggi coil	: 0,635 ft

## 7. Rotary Vacuum Filter (H-144)

Fungsi	: Memisahkan kolagen dari impurities
Tipe	: Rotary vacuum drum filter

Dasar pemilihan : cocok digunakan untuk pemisahan dalam kondisi kontinyu

Kondisi operasi :  $T = 15^{\circ}\text{C}$

Rate Bahan masuk = 140822,3652 kg/hari

Fraksi masa bahan masuk rotary vacuum filter :

$$X_{\text{Kolagen}} = \frac{4495,5}{140822,3652} = 0,0319$$

$$X_{\text{MgCl}_2} = \frac{300,7299}{140822,3652} = 0,0021$$

$$X_{\text{Air}} = \frac{121531,2956}{140822,3652} = 0,8631$$

$$X_{\text{NaCl}} = \frac{513,6644}{140822,3652} = 0,0036$$

$$X_{\text{CaCl}_2} = \frac{8882,8165}{140822,3652} = 0,0631$$

$$X_{\text{H}_3\text{PO}_4} = \frac{5098,3588}{140822,3652} = 0,0362$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{Air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,0319}{945,1} + \frac{0,8631}{994,7} + \frac{0,0631}{1012} + \frac{0,0021}{1011,5} + \frac{0,0036}{1012,85} + \frac{0,0362}{1006,5}$$

$$\rho_{\text{campuran}} = 994,6287 \text{ kg/m}^3$$

$$\mu_{\text{campuran}} = \frac{\mu}{\phi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\phi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{\frac{121531,2956}{994,7}}{\frac{121531,2956}{994,7} + \frac{4495,5}{945,1} + \frac{8882,8165}{1012} + \frac{5098,3588}{1006,5} + \frac{300,7299}{1011,5} + \frac{513,6644}{1012,85}}$$

$$\varepsilon = 0,8629$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,8629)}}$$

$$= 0,563$$

$$\mu_{\text{air}} = 0,7679 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,7679 \cdot 10^{-3}}{0,563}$$

$$= 1,3639 \cdot 10^{-3} \text{ kg/m.s}$$

$$\text{padatan} = 4495,5 \text{ kg/hari}$$

$$\text{liquida} = 136326,8652 \text{ kg/hari}$$

$$\text{rate volume padatan} = \frac{4495,5 \text{ kg / hari}}{945,1 \text{ kg / m}^3} = 4,7566 \text{ m}^3 / \text{hari}$$

$$\begin{aligned} \text{Vol liquida} &= \frac{121531,2956}{994,7} + \frac{8882,8165}{1012} + \frac{5098,3588}{1006,5} + \frac{300,7299}{1011,5} + \frac{513,6644}{1012,85} \\ &= 136,8262 \text{ m}^3 / \text{hari} \end{aligned}$$

$$\text{rate volume slurry} = (136,8262 + 4,7566) \text{ m}^3 / \text{hari} = 141,5828 \text{ m}^3 / \text{hari}$$

$$C_x = \text{konsentrasi padatan} = \frac{4,7566}{141,5828} = 0,0323$$

$$\text{Massa slurry} = \frac{\text{wet cake}}{\text{dry cake}} = \frac{5619,375 \text{ kg / hari}}{4495,5 \text{ kg / hari}} = 1,25$$

$$C_s = \frac{\rho_{\text{air}} \cdot C_x}{1 - m \cdot C_x} = \frac{994,7 \cdot 0,0323}{1 - 1,25 \cdot 0,0323} = 33,4806 \frac{\text{kg solid}}{\text{m}^3 \text{ filtrat}}$$

$$\text{Filter cycle time (tc)} = 250 \text{ s}$$

$$\text{Filter rate} = 140822,3652 \text{ kg / hari} = 1,6299 \text{ kg/s}$$

$$\frac{V}{tc} = 1,6299 \cdot \frac{C_x}{C_s}$$

$$\frac{V}{tc} = 1,6299 \cdot \frac{0,0323}{33,4806} = 1,5724 \cdot 10^{-3} \text{ m}^3 \text{ filtrate/s}$$

fraksi drum yang tercelup dalam slurry : 33%

$$\Delta p \text{ dianggap} = 90 \text{ kPa} = 90 \cdot 10^3 \text{ Pa}$$

$$\alpha = (4,37 \cdot 10^9) \cdot (90 \cdot 10^3)^{0,3} = 1,3389 \cdot 10^{11} \text{ m/kg}$$

mensubstitusikan persamaan 14.2-24 (Geankoplis, 1993), dimana B=0 :

$$\frac{V}{A \cdot t_c} = \left( \frac{2 \cdot f \cdot (\Delta p)}{t_c \cdot \mu \cdot \alpha \cdot C \cdot s} \right)^{1/2}$$

$$\frac{1,5724 \cdot 10^{-3}}{A} = \left( \frac{2 \cdot 0,33 \cdot (90 \cdot 10^3)}{250 \cdot 1,3639 \cdot 10^{-3} \cdot 1,3389 \cdot 10^{11} \cdot 33,4806} \right)^{1/2}$$

$$A = 7,9763 \text{ m}^2 \approx 85,8541 \text{ ft}^2$$

$$L = 2 \cdot D$$

$$A = \pi \cdot D \cdot L$$

$$A = \pi \cdot 2 \cdot D^2$$

$$7,9763 = 2 \cdot \pi \cdot D^2$$

$$D = 1,1267 \text{ m}$$

$$\begin{aligned} \text{Power yang dibutuhkan} &= 0,005 \text{ hp /ft}^2 \text{ area filter} \\ &= 0,005 \cdot 85,8541 \\ &= 0,4293 \text{ hp} \approx 0,5 \text{ hp} \end{aligned}$$

#### Spesifikasi :

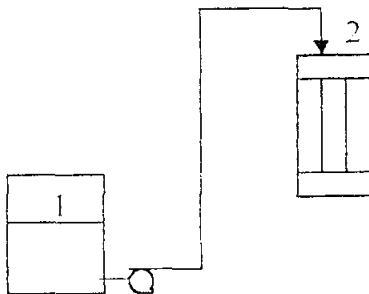
Tipe : Rotary vacuum drum filter

Filter rate : 1,6299 kg/s

Diameter : 1,1276 m = 3,6994 ft

Jumlah : 1 buah

#### 8. POMPA (L-145)



Fungsi : Memompa slurry ke tangki liming

Tipe : centrifugal pump

Dasar pemilihan : dapat menangani sampai tekanan 350 bar (Ulrich table 4-20, p.206)

Jumlah : 1 buah

Rate kolagen masuk = 5619,375 kg/hari

Fraaksi masa bahan masuk pompa :

$$X_{\text{Kolagen}} = \frac{4495,5}{5619,375} = 0,8$$

$$X_{\text{MgCl}_2} = \frac{2,4725}{5619,375} = 0,0004$$

$$X_{\text{Air}} = \frac{1001,9346}{5619,375} = 0,1783$$

$$X_{\text{NaCl}} = \frac{4,2707}{5619,375} = 0,0008$$

$$X_{\text{CaCl}_2} = \frac{73,1643}{5619,375} = 0,013$$

$$X_{\text{H}_3\text{PO}_4} = \frac{42,0329}{5619,375} = 0,0075$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{Air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,8}{945,1} + \frac{0,1783}{994,7} + \frac{0,013}{1012} + \frac{0,0004}{1011,5} + \frac{0,0008}{1012,85} + \frac{0,0075}{1006,5}$$

$$\rho_{\text{campuran}} = 994,6287 \text{ kg/m}^3 = 62,0947 \text{ lbm/ft}^3$$

$$\text{Rate volumetric kolagen} = \frac{5619,375 \text{ kg/hari}}{994,6287 \text{ kg/m}^3}$$

$$= 5,6497 \text{ m}^3/\text{hari} = 6,539 \cdot 10^{-5} \text{ m}^3/\text{s}$$

$$q_v = 2,3092 \cdot 10^{-3} \text{ cuft/s}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{1001,9346}{994,7} = \frac{1001,9346}{994,7 + \frac{4495,5}{945,1} + \frac{73,1643}{1012} + \frac{42,0329}{1006,5} + \frac{2,4725}{1011,5} + \frac{4,2707}{1012,85}}$$

$$\varepsilon = 0,1712$$

$$\phi_p = \frac{1}{10^{1,82(1-0,1712)}}$$

$$= 0,031$$

$$\mu_{\text{air}} = 0,7679 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,7679 \cdot 10^{-3}}{0,031}$$

$$= 0,0248 \text{ kg/m.s} = 0,0167 \text{ lbm/ft.s}$$

Diasumsikan aliran laminar, sehingga :

$$\begin{aligned} Di_{\text{opt}} &= 3,0 \cdot q_v^{0,36} \cdot \mu^{0,18} \quad (\text{Peters \& Timmerhaus 4}^{\text{ed}}, \text{ p.496}) \\ &= 3,0 \cdot (2,3092 \cdot 10^{-3} \text{ cuft/s})^{0,45} \cdot (0,0167 \text{ lbm/ft.s})^{0,13} \\ &= 0,1147 \text{ in} \end{aligned}$$

Dipilih steel-pipe (IPS) berukuran 1/8 in sch.80 :

$$ID = 0,215 \text{ in} = 0,0179 \text{ ft}$$

$$OD = 0,405 \text{ in} = 0,0338 \text{ ft}$$

$$A = 0,036 \text{ in}^2 = 0,25 \cdot 10^{-3} \text{ ft}^2 \quad (\text{Peters \& Timmerhaus 4}^{\text{ed}}, \text{ tabel 13, p.496})$$

$$\text{Kecepatan linear (v)} = (2,3092 \cdot 10^{-3} \text{ cuft/s}) / (0,25 \cdot 10^{-3} \text{ ft}^2) = 9,2368 \text{ ft/s}$$

$$N_{\text{Re}} = \frac{\rho \cdot v \cdot ID}{\mu} = \frac{(62,0947 \text{ lbm/ft}^3) \cdot (9,2368 \text{ ft/s}) \cdot (0,0179 \text{ ft})}{0,0167 \text{ lbm/ft.s}} = 614,77$$

(aliran laminar : asumsi benar)

Dengan menggunakan persamaan Bernoulli :

$$\frac{1}{2 \cdot \alpha \cdot g_c} \cdot (v_2^2 - v_1^2) + \frac{g}{g_c} (z_2 - z_1) + \frac{P_2 - P_1}{\rho} + \sum F + W_s = 0$$

(Geankoplis 3<sup>rd</sup> ed., pers. 2.7-28, p.64)

dimana  $\sum F$  merupakan total *friksional losses*, meliputi :

1. Losses karena kontraksi pada outlet rotary vacuum filter,  $h_c$ .
2. Losses karena friksi pada pipa lurus,  $F_t$ .



3. Losses karena friksi pada elbow dan valve,  $h_f$ .
4. Losses karena ekspansi pada inlet tangki.

Perhitungan  $\sum K$ :

1. Losses karena kontraksi pada outlet rotary vacuum filter,  $h_c$ .

$$K_c = 0,55 \cdot \left( 1 - \frac{A_2}{A_1} \right) \quad (\text{Geankoplis 3<sup>rd</sup> ed., pers.2.10-16, p.93})$$

Dimana :  $A_1$  = luas penampang tangki

$A_2$  = luas penampang pipa

Karena  $A_1 \gg A_2$ , maka  $(A_2/A_1)$  diabaikan.

$$K_c = 0,55 \cdot (1-0) = 0,55$$

Untuk aliran laminar,  $\alpha = 0,5$

$$h_c = K_c \cdot \left( \frac{v^2}{2 \cdot \alpha \cdot gc} \right) \quad (\text{Geankoplis 3<sup>rd</sup> rd., pers. 2.10-16, p.93})$$

$$= 0,55 \cdot \left( \frac{(9,2368 \text{ ft/s})^2}{2 \cdot 0,5 \cdot 32,174 \text{ lbm.ft/lbf.s}^2} \right)$$

$$= 1,4585 \text{ ft.lbf/lbm}$$

2. Losses karena friksi pada pipa lurus,  $F_t$ .

Digunakan pipa commercial steel,  $\epsilon = 0,00015 \text{ ft}$  ( Peters & Timmerhaus 4<sup>th</sup> ed., fig.14-1, p.482).

Dari Peters & Timmerhaus 4<sup>th</sup> ed., fig.14-1, p.482 diperoleh  $f = 0,027$

$$F_t = 4 \cdot f \cdot \frac{\Delta L}{D} \cdot \frac{v^2}{2 \cdot gc} \quad (\text{Geankoplis 3<sup>rd</sup> ed., pers.2.10-6, p.89})$$

$$= 4 \cdot 0,027 \cdot \frac{10 \text{ ft}}{0,0179 \text{ ft}} \cdot \frac{9,2368^2}{2 \cdot (32,174 \text{ lbm.ft/lbf.s}^2)}$$

$$= 79,9979 \text{ ft.lbf/lbm}$$

3. Losses karena friksi pada elbow dan valve,  $h_f$ .

Terdapat 3 elbow 90°, 1 gate valve, dan 1 globe valve.

Dari Geankoplis table 2.10-1, p.93 :

$$K_f = (3 \cdot 0,75) + 0,17 + 6 = 8,42$$

$$\begin{aligned}
 h_f &= K_f \frac{v^2}{2\alpha \cdot gc} && (\text{Geankoplis 3}^{\text{rd}} \text{ ed., per.2.10-17, p.94}) \\
 &= 8,42 \cdot \frac{(9,2368 \text{ ft/s})^2}{2,0,5 \cdot (32,174 \text{ lbm} \cdot \text{ft} / \text{lbf} \cdot \text{s}^2)} \\
 &= 22,328 \text{ ft} \cdot \text{lbf} / \text{lbm}
 \end{aligned}$$

4. Losses karena ekspansi pada inlet tangki

$$K_{ex} = \left(1 - \frac{A_1}{A_2}\right)^2 \quad (\text{Geankoplis 3}^{\text{rd}} \text{ ed., per.2.10-15, p.93})$$

Dimana :  $A_1$  = luas penampang pipa

$A_2$  = luas penampang tangki

Karena  $A_1 \ll A_2$ , maka  $(A_1/A_2)$  diabaikan.

$$K_{ex} = (1-0)^2 = 1$$

$$h_{ex} = K_{ex} \frac{v^2}{2\alpha \cdot gc} \quad (\text{Geankoplis 3}^{\text{rd}} \text{ ed., per.2.10-15, p.93})$$

$$\begin{aligned}
 &= 1 \cdot \frac{9,2368^2}{2,0,5 \cdot (32,174 \text{ lbm} \cdot \text{ft} / \text{lbf} \cdot \text{s}^2)} \\
 &= 2,6518 \text{ ft} \cdot \text{lbf} / \text{lbm}
 \end{aligned}$$

$$\begin{aligned}
 \sum F &= (2,6518 + 22,328 + 79,9979 + 1,4585) \text{ ft} \cdot \text{lbf} / \text{lbm} \\
 &= 106,4362 \text{ ft} \cdot \text{lbf} / \text{lbm}
 \end{aligned}$$

$$\begin{aligned}
 -Ws &= \frac{1}{2\alpha \cdot gc} \cdot (v_2^2 - v_1^2) + \frac{g}{gc} (z_2 - z_1) + \frac{P_2 - P_1}{\rho} + \sum F \\
 &= \frac{1}{2,0,5 \cdot (32,174 \text{ lbm} \cdot \text{ft} / \text{lbf} \cdot \text{s}^2)} \cdot (9,2368^2 - 0) + \\
 &\quad \frac{32,174 \text{ ft/s}^2}{32,174 \text{ ft} \cdot \text{lbm} / \text{lbf} \cdot \text{s}^2} (1) + 0 + 106,4362 \\
 &= 109,088 \text{ ft} \cdot \text{lbf} / \text{lbm}
 \end{aligned}$$

Efisiensi pompa ( $\eta$ ) = 5% (Peters & Timmerhaus 4<sup>th</sup> ed., fig 14.36, p.520)

$$\text{Brake hp} = \frac{-Ws \cdot m}{\eta \cdot 550} \quad (\text{Geankoplis 3}^{\text{rd}} \text{ ed., pers.3.3-2, p.134})$$

$$= \frac{(109,088 \text{ ft.lbf / lbm}).(0,1434 \text{ lbm / s})}{0,05.(550 \text{ ft.lbf / s})}$$

$$= 0,5688 \text{ hp}$$

Efisiensi motor = 80 % (Peters & Timmerhaus 4<sup>th</sup>ed., fig.14,38,p.521)

Power motor pompa yang dipakai = 0,5688 / 0,80

$$= 0,711 \text{ hp} \approx 0,75 \text{ hp}$$

### Spesifikasi :

Tipe : centrifugal pump.

Rate volumetric :  $2,3092.10^{-3} \text{ cuft/s}$

Dipilih steel-pipe (IPS) berukuran 1/8 in sch.80 :

ID = 0,215 in = 0,0179 ft

OD = 0,405 in = 0,0338 ft

A =  $0,036 \text{ in}^2 = 0,25.10^{-3} \text{ ft}^2$

Power motor : 0,75 hp

Jumlah : 7 buah

## 9. TANGKI LIMING (F-150)

Fungsi : Merendam kolagen dengan air kapur 10%

Tipe : silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berbentuk konis yang dilengkapi dengan pengaduk dan coil pemanas.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk slurry

Kondisi operasi :  $T = 20^{\circ}\text{C}$

Rate kolagen masuk = 5619,375 kg/hari

Rate  $\text{Ca}(\text{OH})_2$  masuk = 17982 kg/hari

Fraksi masa bahan masuk tangki liming :

$$X_{\text{Kolagen}} = \frac{4495,5}{23601,375} = 0,1905 \qquad X_{\text{MgCl}_2} = \frac{2,4725}{23601,375} = 0,0001$$

$$X_{\text{Air}} = \frac{1001,9346}{23601,375} = 0,0424 \qquad X_{\text{NaCl}} = \frac{4,2707}{23601,375} = 0,0002$$

$$X_{\text{CaCl}_2} = \frac{73,1643}{23601,375} = 0,0031$$

$$X_{\text{H}_3\text{PO}_4} = \frac{42,0329}{23601,375} = 0,0018$$

$$X_{\text{Ca(OH)}_2} = \frac{17982}{23601,375} = 0,7619$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\rho_{\text{Ca(OH)}_2} = 998,38 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}} + \frac{X_{\text{Ca(OH)}_2}}{\rho_{\text{Ca(OH)}_2}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,1905}{945,1} + \frac{0,0424}{994,7} + \frac{0,031}{1012} + \frac{0,0001}{1011,5} + \frac{0,0002}{1012,85} + \frac{0,0018}{1006,5} + \frac{0,7619}{998,38}$$

$$\rho_{\text{campuran}} = 961,4966 \text{ kg/m}^3 = 60,0262 \text{ lbm/ft}^3$$

$$\begin{aligned} \text{Rate volume kolagen dan Ca(OH)}_2 &= \frac{23601,375 \text{ kg/hari}}{961,4966 \text{ kg/m}^3} \\ &= 24,5464 \text{ m}^3/\text{hari} = 866,8432 \text{ cuft/hari} \end{aligned}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{\frac{1001,9346}{994,7}}{\frac{1001,9346}{994,7} + \frac{4495,5}{945,1} + \frac{73,1643}{1012} + \frac{42,0329}{1006,5} + \frac{2,4725}{1011,5} + \frac{4,2707}{1012,85} + \frac{17982}{998,38}}$$

$$\varepsilon = 0,0422$$

$$\begin{aligned} \varphi_p &= \frac{1}{10^{1,82(1-0,0422)}} \\ &= 0,0181 \end{aligned}$$

$$\mu_{\text{air}} = 0,7679 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,7679 \cdot 10^{-3}}{0,0181}$$

$$= 0,04243 \text{ kg/m.s} = 0,0285 \text{ lbm/ft.s}$$

Diambil : tinggi shell ( $H_s$ ) = 1,5 . Diameter shell ( $D$ )

$$\text{Volume shell} = (\pi/4) \cdot D^2 \cdot H_s = (\pi/4) \cdot D^2 \cdot 1,5 \cdot D = 1,5 \cdot (\pi/4) \cdot D^3$$

$$\text{Volume torispherical dished head (cuft)} = 0,000049 \times D^3 \text{ (inch)}$$

(Brownell & Young, pers.5.11,p.88)

Volume larutan dalam konis = volume konis

$$\begin{aligned} &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot H_t \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot Dn^2 \cdot Hn \right) \\ &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot \frac{D}{2 \cdot \text{tg} \alpha} \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot Dn^2 \cdot \frac{Dn}{2 \cdot \text{tg} \alpha} \right) \\ &= \left( \frac{\pi \cdot D^3}{24 \cdot \text{tg} \alpha} \right) - \left( \frac{\pi \cdot Dn^3}{24 \cdot \text{tg} \alpha} \right) \\ &= \frac{\pi}{24 \cdot \text{tg} \alpha} (D^3 - Dn^3) \end{aligned}$$

dimana :

$Dn$  = diameter lubang pengeluaran liquida = 0,5 ft (=6 in)

$D$  = diamete konis bagian atas = diameter shell

$H_t$  = tinggi konis

$H_n$  = tinggi konis terpancung

Volume tangki penampung = vol.shell + vol.dished head + vol.konis

Diambil : volume tangki penampung = 1,2.volume larutan total

$$1,2 \cdot 866,8432 \text{ cuft} = \frac{\pi}{4} \cdot 1,5 \cdot D^3 \cdot \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 + (0,000049 \cdot D^3)$$

$$1040,2118 \text{ cuft} = 6,8177 \cdot 10^{-4} \cdot D^3 + 4,9 \cdot 10^{-5} \cdot D^3 + \frac{\pi}{24 \cdot \text{tg} \alpha} (D^3 - Dn^3)$$

$$D = 110,0825 \text{ in} = 9,1735 \text{ ft} \approx 9,2 \text{ ft}$$

$$H_s = 1,5.D = 1,5.(9,2 \text{ ft}) = 13,8 \text{ ft}$$

$$\begin{aligned} \text{Volume larutan dalam konis} &= \frac{\pi}{24.Ig\alpha} (D^3 - (0,5 \text{ ft})^3) \\ &= \frac{\pi}{24.Ig30''} ((9,2 \text{ ft})^3 - (0,5 \text{ ft})^3) \\ &= 176,5196 \text{ cuft} \end{aligned}$$

$$\begin{aligned} \text{Volume larutan dalam shell} &= \text{vol. larutan total} - \text{vol. larutan dalam konis} \\ &= (866,8432 - 176,5196) \text{ cuft} \\ &= 690,3236 \text{ cuft} \end{aligned}$$

$$\begin{aligned} \text{Tinggi larutan dalam shell (H)} &= \frac{\text{volume larutan dalam shell}}{\frac{\pi}{4}.D^2} \\ &= \frac{690,3236 \text{ cuft}}{\frac{\pi}{4}.(9,2 \text{ ft})^2} \\ &= 10,3845 \text{ ft} \end{aligned}$$

$$\text{Tinggi larutan dalam konis} = \text{tinggi konis (Hc)}$$

$$\begin{aligned} H_c &= \frac{D}{2.Ig\alpha} - \frac{Dn}{2.Ig\alpha} \\ &= \frac{9,2 \text{ ft}}{2.Ig30''} - \frac{0,5 \text{ ft}}{2.Ig30''} \\ &= 7,5344 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{Tinggi larutan dalam tangki} &= H + H_c \\ &= (10,3845 + 7,5344) \text{ ft} \\ &= 17,9189 \text{ ft} \end{aligned}$$

$$\begin{aligned} P_{\text{operasi}} = P_{\text{hidrostatik}} &= \left( \frac{\rho.H}{144} \right) \text{ psi} \\ &= \left( \frac{60,0262 \text{ lbm/cuft} \cdot 17,9189 \text{ ft}}{144} \right) \text{ psi} \\ &= 7,4695 \text{ psi} \end{aligned}$$

$$P_{\text{desain}} = 1,5.P_{\text{operasi}} = 1,5.(7,4695 \text{ psi}) = 11,2043 \text{ psi}$$

**Tebal shell**

$$t_s = \frac{P \cdot ID}{2(f \cdot E - 0,6 \cdot P)} + c \quad (\text{Brownell \& Young, pers. 13.1})$$

dimana :

$$P = P_{\text{desain}} = 11,2043 \text{ psi}$$

$$ID = 9,2 \text{ ft} = 110,0825 \text{ in}$$

Konstruksi : bahan konstruksi = carbon stell SA-283, grade D, dengan

$f$  = stress maksimum yang diijinkan = 12650 psi

tipe sambungan = *double-welded butt joint*, dengan

$E$  = *welded-joint efficiency* = 0,8

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_s = \frac{(11,2043 \text{ psi}) \cdot (110,0825 \text{ in})}{2 \cdot (12650,8 - 0,6 \cdot 11,2043) \text{ psi}} + \frac{1}{8} \text{ in}$$

$$t_s = 0,186 \text{ in} \approx \frac{3}{8} \text{ in}$$

**Tebal dished head**

$$t_s = \frac{3}{8} \text{ in}$$

$$OD = ID + 2 \cdot t_s$$

$$= 110,0825 \text{ in} + (2 \cdot (3/8)) \text{ in}$$

$$= 110,8325 \text{ in}$$

Dari table 5.7 Brownell & Young diperoleh :

OD standar = 114 in dengan  $t_s = 3/8 \text{ in}$

$r$  (crown radius / radius of dish) = 108 in

$icr$  (inside corner radius / knuckle radius) =  $6 \frac{1}{8} \text{ in}$

$$W = \frac{1}{4} \left( 3 + \sqrt{\frac{r}{icr}} \right) \quad (\text{Brownell \& Young, pers. 7.76, p.138})$$

$$= \frac{1}{4} \left( 3 + \sqrt{\frac{108}{678}} \right)$$

$$= 1,7409$$

$$a = ID/2 = 110,0825 \text{ in} / 2 = 55,0413 \text{ in}$$

$$AB = ID/2 - icr = (55,0413 - 6,875) \text{ in} = 48,1663 \text{ in}$$

$$BC = r - icr = (108 - 6,875) \text{ in} = 101,125 \text{ in}$$

$$b = r - \sqrt{BC^2 - AB^2} = 108 - \sqrt{101,125^2 - 48,1663^2} = 19,0828 \text{ in}$$

$$t_d = \frac{P \cdot r \cdot W}{2 \cdot f \cdot E - 0,2 \cdot P} + c \quad (\text{Brownell \& Young, pers. 7.77, p.138})$$

dimana :

$$P = P_{\text{desain}} = 11,2043 \text{ psi}$$

Konstruksi : bahan konstruksi = carbon stell SA-283, grade D, dengan

$f$  = stress maksimum yang diijinkan = 12650 psi

tipe sambungan = *double-welded butt joint*, dengan

$E$  = *welded-joint efficiency* = 0,8

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_d = \frac{(11,2043 \text{ psi}) \cdot (108 \text{ in}) \cdot 1,7409}{2 \cdot (12650 \text{ psi}) \cdot (0,8) - 0,2 \cdot (11,2043 \text{ psi})} + \frac{1}{8} \text{ in}$$

$$= 0,2291 \text{ in} \approx \frac{3}{8} \text{ in}$$

Dipilih panjang straight-flange (sf) = 4,5 in

(Brownell & Young, table 5.8, p.93)

$$OA = t + b + sf$$

$$= (3/8 + 19,0828 + 4,5) \text{ in}$$

$$= 23,9578 \text{ in} = 1,9965 \text{ ft} \approx 2 \text{ ft}$$

Tinggi tangki keseluruhan = tinggi shell + tinggi dish + tinggi konis

$$= 13,8 \text{ ft} + 2 \text{ ft} + 7,6 \text{ ft}$$

$$= 23,4 \text{ ft}$$



**Perhitungan Pengaduk**

Dipilih tipe pengaduk : flat six-blade turbine agitator with disk.

Kecepatan agitator antara 20 – 200 rpm (Geankoplis), diambil 50 rpm.

$$N = 50 \text{ rpm} = 0,8333 \text{ rps.}$$

$$\mu_{\text{campuran}} = 0,04243 \text{ kg/m.s} = 0,0285 \text{ lbm/ft.s}$$

$$\rho_{\text{campuran}} = 961,4966 \text{ kg/m}^3 = 60,0262 \text{ lbm/ft}^3$$

Dari table 3.4, p.144 Geankoplis diperoleh :

$$Da/Dt = 0,3 \quad W/Da = 1/5 \quad L/Da = 1/4 \quad C/Dt = 1/3 \quad J/Dt = 1/12$$

dimana :

$$Dt = \text{diameter tangki} = 9,2 \text{ ft} = 2,8042 \text{ m}$$

$$Da = \text{diameter pengaduk} = 0,3 \times 2,8042 \text{ m} = 0,8413 \text{ m}$$

$$W = \text{lebar blade} = (0,8413 \text{ m})/5 = 0,1683 \text{ m}$$

$$L = \text{panjang blade} = (0,8413 \text{ m})/4 = 0,2103 \text{ m}$$

$$C = \text{jarak dari dasar tangki ke pusat pengaduk} = (2,8042 \text{ m})/3 = 0,9347 \text{ m}$$

$$J = \text{lebar bafflc} = \frac{1}{12} \times 2,8042 \text{ m} = 0,2337 \text{ m}$$

$$N_{Re} = \frac{Da^2 \cdot N \cdot \rho}{\mu}$$

$$= \frac{(0,8413)^2 \cdot 0,8333 \cdot 961,4966}{0,04243}$$

$$= 13365,2744 \text{ (turbulen)}$$

Dari fig. 3.4-4, p.155 Geankoplis 2<sup>nd</sup> ed., dengan memotong kurva 1 dengan

$$N_{Re}, \text{ diperoleh } Np = 5.$$

$$Np = \frac{P}{\rho \cdot N^3 \cdot Da^5} \quad (\text{Geankoplis, pers. 3.4-2, p.145})$$

Dimana :

$Np$  = power number

$P$  = power pengaduk

$N$  = kecepatan pengaduk

$$P = Np \cdot \rho \cdot N^3 \cdot Da^5$$

$$= 5 \cdot 961,4966 \text{ kg/m}^3 \cdot (0,8333 \text{ rps})^3 \cdot (0,8413 \text{ m})^5$$

$$= 1172,4015 \text{ W} = 1,1724015 \text{ kW} = 1,5722 \text{ Hp}$$

$$\text{Power input} = 110\% \times 1,5722 \text{ Hp} = 1,7294 \text{ Hp}$$

$$\text{Transmission system losses} = 20\% \text{ dari total Hp}$$

$$1,7294 \text{ Hp} + 0,2 \cdot \text{total Hp} = \text{total Hp}$$

$$\text{total Hp} = 2,1618 \text{ Hp}$$

$$\text{Efisiensi motor} = 82\% \quad (\text{Peter \& Timmerhaus, fig.14-38, p.521})$$

$$\text{Power yang dibutuhkan} = (2,1618 \text{ hp})/82\% = 2,6363 \text{ hp} \approx 3 \text{ hp}$$

### Perhitungan Coil Pemanas

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q = 15441,2065 \text{ kkal/hari} = 747,7404 \text{ W} = 2551,4055 \text{ btu/h}$
- ❖ Suhu bahan masuk tangki pada  $15^{\circ}\text{C}$  ( $59^{\circ}\text{F}$ ) yaitu kolagen, impurities dan  $\text{Ca(OH)}_2$  masuk pada suhu  $20^{\circ}\text{C}$  ( $68^{\circ}\text{F}$ )
- ❖ Suhu kolagen, impurities,  $\text{Ca(OH)}_2$  keluar  $20^{\circ}\text{C}$  ( $68^{\circ}\text{F}$ )
- ❖ Pemanas berupa air dengan suhu masuk  $30^{\circ}\text{C}$  ( $=86^{\circ}\text{F}$ ) dan suhu keluar  $20^{\circ}\text{C}$  ( $68^{\circ}\text{F}$ )
- ❖ Massa air =  $1544,1207 \text{ kg/hari} = 141,8426 \text{ lb/h}$

$$\mu_{\text{campuran}} = 0,04243 \text{ kg/m.s} = 0,0285 \text{ lbm/ft.s}$$

$$\rho_{\text{campuran}} = 961,4966 \text{ kg/m}^3 = 60,0262 \text{ lbm/ft}^3$$

$$C_{p_{\text{air kapur } 10\%}} = 0,2860 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p_{\text{CaCl}_2}} = 0,2162 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p_{\text{kolagen}}} = 0,3369 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p_{\text{MgCl}_2}} = 0,2520 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p_{\text{H}_3\text{PO}_4}} = 0,452 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p_{\text{NaCl}}} = 0,2737 \text{ kkal/kg}^{\circ}\text{C}$$

$$X_{\text{Kolagen}} = \frac{4495,5}{23601,375} = 0,1905$$

$$X_{\text{MgCl}_2} = \frac{2,4725}{23601,375} = 0,0001$$

$$X_{\text{Air}} = \frac{1001,9346}{23601,375} = 0,0424$$

$$X_{\text{NaCl}} = \frac{4,2707}{23601,375} = 0,0002$$

$$X_{\text{CaCl}_2} = \frac{73,1643}{23601,375} = 0,0031$$

$$X_{\text{H}_3\text{PO}_4} = \frac{42,0329}{23601,375} = 0,0018$$

$$X_{\text{air kapur } 10\%} = \frac{17982}{23601,375} = 0,7619$$

$$C_{p_{\text{campuran}}} = \sum x_i \cdot C_{p_i}$$

$$= (0,1905 \times 0,3369) + (0,0424 \times 1) + (0,0031 \times 0,2162) + (0,7619 \times 0,2860) + (0,0001 \times 0,252) + (0,0002 \times 0,2737) + (0,0018 \times 0,452)$$

$$C_{p\text{campuran}} = 0,326 \text{ kkal/kg} \cdot ^\circ\text{C} = 0,326 \text{ kal/g} \cdot ^\circ\text{C} \times 4,1868$$

$$= 1,3649 \text{ kJ/kg} \cdot \text{K} = 0,326 \text{ btu/lb}^\circ\text{F}$$

$$t_r = (T + t) / 2 = (86 + 63,5)^\circ\text{F} / 2 = 74,75^\circ\text{F}$$

$$\Delta t = (T - t) = (86 - 63,5)^\circ\text{F} = 22,5^\circ\text{F}$$

Berdasarkan kern tabel 11, p.844, trial ukuran pipa coil = 1/8 in IPS, sch 40

$$D_o = 0,405 \text{ in}$$

$$D_i = 0,269 \text{ in}$$

$$a' = 0,058 \text{ in}^2$$

$$a'' = 0,106 \text{ ft}^2/\text{ft}$$

## Evaluasi Perpindahan Panas

Sisi bejana : fluida panas	Sisi pipa : air, fluida dingin
$\rho_{camp} = 60,0262 \text{ lbm/cuft}$ $\mu_{campuran} = 102,6424 \text{ lb/ft.h}$ $Da = 2,7601 \text{ ft}$ $N = 50 \text{ rpm} = 3000 \text{ rph}$ $Nre = \frac{\rho Da^2 N}{\mu}$ $= \frac{60,0262 (2,7601)^2 \cdot 3000}{102,6424}$ $= 13365,4917$ $Jc = 340 \quad (\text{kern fig 20-2, p-718})$ $ho = Jc \cdot \frac{k}{D_{ivesseI}} \left( \frac{Cp \cdot \mu}{k} \right)^{1/3} \left( \frac{\mu}{\mu_w} \right)^{0,14}$ Dimana : $Di = 9,2 \text{ ft}$ $k = 0,45 \text{ W/m.K} = 0,26 \text{ Btu/hr.ft.}^\circ\text{F}$ $Cp \text{ rata-rata} = 0,326 \text{ Btu/lbm.}^\circ\text{F}$ $ho = 340 \cdot \frac{0,26}{9,2} \left( \frac{0,326 \cdot 102,6424}{0,26} \right)^{1/3}$ $= 48,5126 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$	$Ap = 0,058 \text{ in}^2 = 0,0004 \text{ ft}^2$ $Gp = \frac{W}{ap} = \frac{141.8426 \text{ lb/h}}{0,0004} = 354606,5 \text{ lb/h.ft}^2$ $Nre = \frac{Dp \cdot Gp}{\mu}$ Dimana : $Dp = 2,469 \text{ in} = 0,2058 \text{ ft}$ $\mu = 1,937 \text{ lb/ft.h}$ $Nre = \frac{354606,5 \cdot 0,2058}{1,937}$ $= 37675,7433$ $J_H = 115 \quad (\text{Kern, fig 24, p.834})$ $hi = J_H \cdot \frac{k}{Di} \left( \frac{Cp \cdot \mu}{k} \right)^{1/3} \left( \frac{\mu}{\mu_w} \right)^{0,14}$ dimana : $Cp = 0,9986 \text{ btu/lb.}^\circ\text{F}$ $k = 0,6158 \text{ W/m.K} = 0,3558 \text{ btu/h.ft.}^\circ\text{F}$ (Geankoplis, App.A.2-11, p.862) $hi = 115 \cdot \frac{0,6158}{0,2058} \left( \frac{0,9986 \cdot 1,937}{0,6158} \right)^{1/3}$ $= 503,9483 \text{ btu/fr.ft}^2 \cdot ^\circ\text{F}$ $hio = hi \cdot \frac{di}{do}$ $= 503,9483 \cdot \frac{0,269}{0,405} = 334,7212 \text{ btu/fr.ft}^2 \cdot ^\circ\text{F}$

$$Uc = \frac{hio x ho}{hio + ho} = \frac{334,7212 \times 48,5126}{334,7212 + 48,5126} = 42,3715 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$$

$$Rd = \frac{Uc - Ud}{Uc \times Ud} \quad \text{Diambil } Rd = 0,004$$

$$0,004 = \frac{42,3715 - U_d}{42,3715 \times U_d} ; U_d = 36,2309 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$$

$$A_{\text{coil}} = \frac{Q}{U_d \Delta T_{\text{LMTD}}} = \frac{2551,4055 \text{ btu/h}}{(36,2309 \text{ btu/hr.ft}^2 \cdot ^\circ\text{F}) \cdot 22,5^\circ\text{F}} = 3,1298 \text{ ft}^2$$

$$L = \frac{A_{\text{coil}}}{d''} = \frac{3,1298 \text{ ft}^2}{0,106 \text{ ft}^2/\text{ft}} = 29,5264 \text{ ft}$$

$$\begin{aligned} d_c &= \text{diameter coil} = 0,65 \cdot d_i \text{ vessel} \\ &= 0,65 \cdot (9,2 \text{ ft}) \\ &= 5,98 \text{ ft} \end{aligned}$$

$$n_c = \frac{L}{\pi \cdot d_c} = \frac{29,5264 \text{ ft}}{\pi \cdot 5,98 \text{ ft}} = 1,5717 \text{ buah} \approx 2 \text{ buah}$$

sc = Spasi coil, diambil 2 in

$$\begin{aligned} h_c &= ((n_c - 1) \cdot (d_o + sc)) + d_o \\ &= ((2 - 1) \cdot (0,405 + 2)) + 0,405 \\ &= 2,81 \text{ in} = 0,2342 \text{ ft} \end{aligned}$$

Pengecekan :

Tinggi liquida dibagian silinder ( $H_s$ ) = 10,3845 ft

$h_c < H_s$  (memenuhi)

### Spesifikasi :

*Tangki*

Kapasitas	: 1040,2118 cuft
Diameter	: 9,2 ft
Tinggi tutup atas (dished head)	: 2 ft
Tinggi tutup bawah (konis)	: 7,6 ft
Tinggi shell	: 13,8 ft
Tinggi tangki total	: 23,4 ft
Tebal shell	: 3/8 in
Tebal tutup atas (dished head)	: 3/8 in
Tebal tutup bawah (konis)	: 3/8 in

Bahan konstruksi	: Carbon steel SA-283, grade D
Jumlah tangki	: 7 buah
<i>Pengaduk</i>	
Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,8413 m = 2,7601 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 3 Hp
Jumlah pengaduk	: 1 buah
<i>Coil pemanas</i>	
Diameter	: 5,98 ft
Spasi coil	: 2 in
Tinggi coil	: 0,2342 ft

#### 10. Rotary Vacuum Filter (H-151)

Fungsi : Memisahkan kolagen dari impurities

Tipe : Rotary vacuum drum filter

Dasar pemilihan : cocok digunakan untuk pemisahan dalam kondisi kontinyu

Kondisi operasi : T = 20°C

Rate kolagen masuk = 23601,375 kg/hari

Fraksi masa bahan masuk tangki liming :

$$X_{\text{Kolagen}} = \frac{4495,5}{23601,375} = 0,1905 \quad X_{\text{MgCl}_2} = \frac{2,4725}{23601,375} = 0,0001$$

$$X_{\text{Air}} = \frac{1001,9346}{23601,375} = 0,0424 \quad X_{\text{NaCl}} = \frac{4,2707}{23601,375} = 0,0002$$

$$X_{\text{CaCl}_2} = \frac{73,1643}{23601,375} = 0,0031 \quad X_{\text{H}_3\text{PO}_4} = \frac{42,0329}{23601,375} = 0,0018$$

$$X_{\text{Ca(OH)}_2} = \frac{17982}{23601,375} = 0,7619$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\rho_{\text{Ca(OH)}_2} = 998,38 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}} + \frac{X_{\text{Ca(OH)}_2}}{\rho_{\text{Ca(OH)}_2}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,1905}{945,1} + \frac{0,0424}{994,7} + \frac{0,031}{1012} + \frac{0,0001}{1011,5} + \frac{0,0002}{1012,85} + \frac{0,0018}{1006,5} + \frac{0,7619}{998,38}$$

$$\rho_{\text{campuran}} = 961,4966 \text{ kg/m}^3 = 60,0262 \text{ lbm/ft}^3$$

$$\begin{aligned} \text{Rate volume kolagen} &= \frac{23601,375 \text{ kg/hari}}{961,4966 \text{ kg/m}^3} \\ &= 24,5464 \text{ m}^3/\text{hari} = 866,8432 \text{ cuft/hari} \end{aligned}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\phi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\phi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{1001,9346}{994,7 + \frac{4495,5}{945,1} + \frac{73,1643}{1012} + \frac{42,0329}{1006,5} + \frac{2,4725}{1011,5} + \frac{4,2707}{1012,85} + \frac{17982}{998,38}}$$

$$\varepsilon = 0,0422$$

$$\phi_p = \frac{1}{10^{1,82(1-0,0422)}}$$

$$= 0,0181$$

$$\mu_{\text{air}} = 0,7679 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,7679 \cdot 10^{-3}}{0,0181}$$

$$= 0,04243 \text{ kg/m.s} = 0,0285 \text{ lbm/ft.s}$$

$$\text{padatan} = 4495,5 \text{ kg/hari}$$

$$\text{liquida} = 19105,875 \text{ kg/hari}$$

$$\text{rate volume padatan} = \frac{4495,5 \text{ kg / hari}}{945,1 \text{ kg / m}^3} = 4,7566 \text{ m}^3 / \text{hari}$$

$$\text{Vol liquida} = \frac{1001,9346}{994,7} + \frac{73,1643}{1012} + \frac{42,0329}{1006,5} + \frac{2,4725}{1011,5} + \frac{4,2707}{1012,85} + \frac{17982}{998,38}$$

$$= 19,1392 \text{ m}^3 / \text{hari}$$

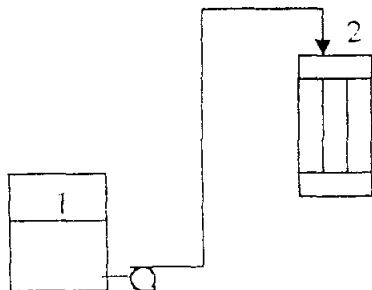
$$\text{rate volume slurry} = (19,1392 + 4,7566) \text{ m}^3 / \text{hari} = 23,8958 \text{ m}^3 / \text{hari}$$

Dengan cara perhitungan yang sama pada Rotary vacuum filter (11-144), maka didapat :

#### Spesifikasi :

Tipe	: Rotary vacuum drum filter
Filter rate	: 0,2732 kg/s
Diameter	: 1,604 m = 5,2624 ft
Jumlah	: 1 buah

#### 11. POMPA (L-152)



Fungsi : Memompa slurry ke tangki netralisasi

Tipe : centrifugal pump

Dasar pemilihan : dapat menangani sampai tekanan 350 bar (Ulrich table 4-20, p.206)

Jumlah : 1 buah

Rate kolagen masuk = 5619,375 kg/hari

Fraksi masa bahan masuk pompa :

$$X_{\text{Kolagen}} = \frac{4495,5}{5619,375} = 0,8$$

$$X_{\text{MgCl}_2} = \frac{0,1124}{5619,375} = 0,00002$$



$$X_{\text{Air}} = \frac{59,0034}{5619,375} = 0,0105$$

$$X_{\text{NaCl}} = \frac{0,2248}{5619,375} = 0,00004$$

$$X_{\text{CaCl}_2} = \frac{4,2707}{5619,375} = 0,00076$$

$$X_{\text{H}_3\text{PO}_4} = \frac{2,4725}{5619,375} = 0,00044$$

$$X_{\text{Ca(OH)}_2} = \frac{1057,7912}{5619,375} = 0,18824$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\rho_{\text{Ca(OH)}_2} = 998,38 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}} + \frac{X_{\text{Ca(OH)}_2}}{\rho_{\text{Ca(OH)}_2}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,8}{945,1} + \frac{0,0105}{994,7} + \frac{0,00076}{1012} + \frac{0,00002}{1011,5} + \frac{0,00004}{1012,85} + \frac{0,00044}{1006,5} + \frac{0,18824}{998,38}$$

$$\rho_{\text{campuran}} = 955,274 \text{ kg/m}^3 = 62,1328 \text{ lbm/ft}^3$$

$$\text{Rate volume kolagen} = \frac{5619,375 \text{ kg/hari}}{955,274 \text{ kg/m}^3}$$

$$= 5.8825 \text{ m}^3/\text{hari} = 207,7374 \text{ cuft/hari}$$

$$= 2,4044 \cdot 10^{-3} \text{ cuft/s}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\phi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\phi_p = \frac{1}{10^{(0,82(1-\epsilon))}}$$

$$\epsilon = \frac{\frac{59,0034}{994,7}}{\frac{59,0034}{994,7} + \frac{4495,5}{945,1} + \frac{4,2707}{1012} + \frac{2,4725}{1006,5} + \frac{0,1124}{1011,5} + \frac{0,2248}{1012,85} + \frac{1057,7912}{998,38}}$$

$$\varepsilon = 0,01$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,01)}}$$

$$= 0,0158$$

$$\mu_{\text{air}} = 0,7679 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,7679 \cdot 10^{-3}}{0,0158}$$

$$= 0,0486 \text{ kg/m.s} = 0,0327 \text{ lbm/ft.s}$$

Dengan cara yang sama pada perhitungan pompa (L-145), maka didapat :

**Spesifikasi :**

Tipe : centrifugal pump.

Rate volumetric :  $2,4044 \cdot 10^{-3} \text{ cuft/s}$

Dipilih steel-pipe (IPS) berukuran 1/8 in sch.80 :

$$ID = 0,215 \text{ in} = 0,0179 \text{ ft}$$

$$OD = 0,405 \text{ in} = 0,0338 \text{ ft}$$

$$A = 0,036 \text{ in}^2 = 0,25 \cdot 10^{-3} \text{ ft}^2$$

Power motor : 1,5 hp

Jumlah : 1 buah

## 12. TANGKI NETRALISASI (F-160)

Fungsi : Menetralkan kolagen setelah masuk dari tangki liming

Tipe : silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berbentuk konis yang dilengkapi dengan pengaduk dan coil pendingin.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk slurry.

Kondisi operasi :  $T = 18^{\circ}\text{C}$

Rate kolagen masuk = 5619,375 kg/hari

Rate HCl 5% masuk = 20822,974 kg/hari

Rate kolagen + HCl 5% masuk = 26442,349 kg/hari

Fraksi massa :

$$X_{\text{kolagen}} = \frac{4495,5}{5619,375} = 0,170011$$

$$X_{\text{MgCl}_2} = \frac{0,1124}{26442,349} = 0,000004$$

$$X_{\text{Air}} = \frac{59,0034}{5619,375} = 0,002231$$

$$X_{\text{NaCl}} = \frac{0,2248}{26442,349} = 0,000008$$

$$X_{\text{CaCl}_2} = \frac{4,2707}{5619,375} = 0,000162$$

$$X_{\text{H}_3\text{PO}_4} = \frac{2,4725}{26442,349} = 0,000094$$

$$X_{\text{Ca(OH)}_2} = \frac{1057,7912}{26442,349} = 0,040004$$

$$X_{\text{HCl}5\%} = \frac{20822,974}{26442,349} = 0,787486$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\rho_{\text{Ca(OH)}_2} = 998,38 \text{ kg/m}^3$$

$$\rho_{\text{HCl } 5\%} = 1023 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{Air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}} + \frac{X_{\text{Ca(OH)}_2}}{\rho_{\text{Ca(OH)}_2}} + \frac{X_{\text{HCl}5\%}}{\rho_{\text{HCl}5\%}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,170011}{945,1} + \frac{0,002231}{994,7} + \frac{0,000162}{1012} + \frac{0,000004}{1011,5} + \frac{0,000008}{1012,85} + \frac{0,000094}{1006,5} + \frac{0,040004}{998,38} + \frac{0,787486}{1023}$$

$$\rho_{\text{campuran}} = 1007,8157 \text{ kg/m}^3 = 62,9157 \text{ lbm/ft}^3$$

$$\text{Rate volume kolagen} = \frac{26442,349 \text{ kg/hari}}{1007,8157 \text{ kg/m}^3}$$

$$= 26,2373 \text{ m}^3/\text{hari} = 926,5565 \text{ cuft/hari}$$

$$= 0,0107 \text{ cuft/s}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{59,0034}{994,7} / \left( \frac{59,0034}{994,7} + \frac{4495,5}{945,1} + \frac{4,2707}{1012} + \frac{2,4725}{1006,5} + \frac{0,1124}{1011,5} + \frac{0,2248}{1012,85} \right. \\ \left. \frac{1057,7912}{998,38} + \frac{20822,974}{1023} \right)$$

$$\varepsilon = 0,0023$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,0023)}}$$

$$= 0,0153$$

$$\mu_{\text{air}} = 0,7679 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,7679 \cdot 10^{-3}}{0,0153}$$

$$= 0,0502 \text{ kg/m.s} = 0,0337 \text{ lbm/ft.s}$$

$$C_{p\text{HCl } 5\%} = 0,966865 \text{ kkal/kg}^\circ\text{C} \quad C_{p\text{CaCl}_2} = 0,2162 \text{ kkal/kg}^\circ\text{C}$$

$$C_{p\text{kolagen}} = 0,3369 \text{ kkal/kg}^\circ\text{C} \quad C_{p\text{MgCl}_2} = 0,2520 \text{ kkal/kg}^\circ\text{C}$$

$$C_{p\text{H}_3\text{PO}_4} = 0,452 \text{ kkal/kg}^\circ\text{C} \quad C_{p\text{NaCl}} = 0,2737 \text{ kkal/kg}^\circ\text{C}$$

$$C_{p\text{Ca(OH)}_2} = 0,286 \text{ kkal/kg}^\circ\text{C}$$

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q = 2566003,764 \text{ kkal/hari} = 124258,7323 \text{ W} = 423989,942 \text{ btu/h}$
- ❖ Suhu bahan masuk tangki pada  $20^\circ\text{C}$  ( $68^\circ\text{F}$ ) yaitu kolagen, impurities dan HCl 5% masuk pada suhu  $50^\circ\text{C}$  ( $122^\circ\text{F}$ )
- ❖ Suhu kolagen, impurities dan HCl keluar  $18^\circ\text{C}$  ( $64,4^\circ\text{F}$ )
- ❖ Pendingin berupa air dengan suhu masuk  $6^\circ\text{C}$  ( $42,8^\circ\text{F}$ ) dan suhu keluar  $18^\circ\text{C}$  ( $64,4^\circ\text{F}$ )
- ❖ Massa air pendingin =  $213833,647 \text{ kg/hari} = 19642,4024 \text{ lb/h}$

Dengan cara yang sama pada perhitungan tangki HCl (F-110), maka didapat :

**Spesifikasi :***Tangki*

Kapasitas	: 1111,8678 cuft
Diameter	: 9,4 ft
Tinggi tutup atas (dished head)	: 2,1 ft
Tinggi tutup bawah (konis)	: 7,8 ft
Tinggi shell	: 14,1 ft
Tinggi tangki total	: 24 ft
Tebal shell	: 3/8 in
Tebal tutup atas (dished head)	: 3/8 in
Tebal tutup bawah (konis)	: 3/8 in
Bahan konstruksi	: High Alloy stell SA-240, grade M type 316
Jumlah tangki	: 1 buah

*Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,8596 m = 2,8202 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 3,1 Hp
Jumlah pengaduk	: 1 buah

*Coil pendingin*

Diameter	: 6,11 ft
Spasi coil	: 2 in
Tinggi coil	: 10,0833 ft

**13. Rotary Vacuum Filter (H-161)**

Fungsi : Memisahkan kolagen dari impurities

Tipe : Rotary vacuum drum filter

Dasar pemilihan : cocok digunakan untuk pemisahan dalam kondisi kontinyu

Kondisi operasi :  $T = 18^{\circ}\text{C}$

Rate Bahan masuk = 26442,3522 kg/hari

Fraksi masa bahan masuk rotary vacuum filter :

$$X_{\text{Kolagen}} = \frac{4495,5}{26442,3522} = 0,170011$$

$$X_{\text{MgCl}_2} = \frac{0,1124}{26442,3522} = 0,000004$$

$$X_{\text{Air}} = \frac{20355,2215}{26442,3522} = 0,769796$$

$$X_{\text{NaCl}} = \frac{0,2248}{26442,3522} = 0,000009$$

$$X_{\text{CaCl}_2} = \frac{1588,821}{26442,3522} = 0,060086$$

$$X_{\text{H}_3\text{PO}_4} = \frac{2,4725}{26442,3522} = 0,000094$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{Air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,170011}{945,1} + \frac{0,769796}{994,7} + \frac{0,060086}{1012} + \frac{0,000004}{1011,5} + \frac{0,000009}{1012,85} + \frac{0,000094}{1006,5}$$

$$\rho_{\text{campuran}} = 986,9094 \text{ kg/m}^3 = 61,6128 \text{ lbm/cuft}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{\frac{20355,2215}{994,7}}{\frac{20355,2215}{994,7} + \frac{4495,5}{945,1} + \frac{1588,821}{1012} + \frac{2,4725}{1006,5} + \frac{0,1124}{1011,5} + \frac{0,2248}{1012,85}}$$

$$\varepsilon = 0,7638$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,7638)}}$$

$$= 0,3716$$

$$\mu_{\text{air}} \text{ pada } 30^\circ\text{C} = 0,8007 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{campuran} = \frac{0,8007 \cdot 10^{-3}}{0,3716}$$

$$= 2,1547 \cdot 10^{-3} \text{ kg/m.s} = 1,4479 \cdot 10^{-3} \text{ lbm/ft.s}$$

$$\text{padatan} = 4495,5 \text{ kg/hari}$$

$$\text{liquida} = 21946,8522 \text{ kg/hari}$$

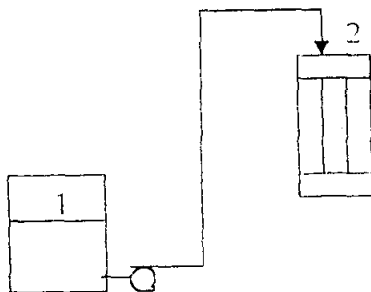
$$\text{rate volume slurry} = \frac{26442,3522 \text{ kg/hari}}{986,9094 \text{ kg/m}^3} = 26,7931 \text{ m}^3/\text{hari}$$

Dengan cara yang sama pada perhitungan Rotary vaccum filter (11-144), maka didapat :

**Spesifikasi :**

Tipe	: Rotary vacuum drum filter
Filter rate	: 0,306 kg/s
Diameter	: 0,7897 m = 2,5908 ft
Jumlah	: 1 buah

**14. POMPA (L-162)**



Fungsi : Memompa slurry dari tangki netralisasi

Tipe : centrifugal pump

Dasar pemilihan : dapat menangani sampai tekanan 350 bar (Ulrich table 4-20, p.206)

Jumlah : 1 buah

Rate kolagen masuk = 5619,375 kg/hari

Fraksi masa bahan masuk pompa :

$$X_{\text{Kolagen}} = \frac{4495,5}{5619,375} = 0,800000$$

$$X_{\text{MgCl}_2} = \frac{0,0056}{5619,375} = 0,000001$$

$$X_{\text{Air}} = \frac{1042,3694}{5619,375} = 0,185495$$

$$X_{\text{NaCl}} = \frac{0,0112}{5619,375} = 0,000002$$

$$X_{\text{CaCl}_2} = \frac{81,3618}{5619,375} = 0,014479$$

$$X_{\text{H}_3\text{PO}_4} = \frac{0,127}{5619,375} = 0,000023$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1012 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 1012,85 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,8}{945,1} + \frac{0,185495}{994,7} + \frac{0,014479}{1012} + \frac{0,000001}{1011,5} + \frac{0,000002}{1012,85} + \frac{0,000023}{1006,5}$$

$$\rho_{\text{campuran}} = 954,8474 \text{ kg/m}^3 = 59,6111 \text{ lbm/ft}^3$$

$$\text{Rate volumetric kolagen} = \frac{5619,375 \text{ kg/hari}}{954,8474 \text{ kg/m}^3}$$

$$= 5,8851 \text{ m}^3/\text{hari} = 6,811 \cdot 10^{-5} \text{ m}^3/\text{s}$$

$$q_f = 2,4052 \cdot 10^{-3} \text{ cuft/s}$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{\frac{1042,3694}{994,7}}{\frac{1042,3694}{994,7} + \frac{4495,5}{945,1} + \frac{81,3618}{1012} + \frac{0,127}{1006,5} + \frac{0,0056}{1011,5} + \frac{0,0112}{1012,85}}$$

$$\varepsilon = 0,1781$$



$$\varphi_p = \frac{1}{10^{1,82(1-0,1781)}}$$

$$= 0,0319$$

$$\mu_{\text{air}} = 0,8007 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,8007 \cdot 10^{-3}}{0,0319}$$

$$= 0,0251 \text{ kg/m.s} = 0,0169 \text{ lbm/ft.s}$$

Dengan cara yang sama pada perhitungan pompa (L-145), maka didapat :

**Spesifikasi :**

Tipe : centrifugal pump.

Rate volumetric :  $2,4052 \cdot 10^{-3} \text{ cuft/s}$

Dipilih steel-pipe (IPS) berukuran 1/8 in sch.80 :

$$\text{ID} = 0,215 \text{ in} = 0,0179 \text{ ft}$$

$$\text{OD} = 0,405 \text{ in} = 0,0338 \text{ ft}$$

$$A = 0,036 \text{ in}^2 = 0,25 \cdot 10^{-3} \text{ ft}^2$$

Power motor : 0,8 hp

Jumlah : 1 buah

## 15. STORAGE BIN (F-163)

Fungsi : tempat penyimpanan sementara kolagen yang akan di ekstraksi

Tipe : silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berbentuk konis

Dasar pemilihan : cocok digunakan untuk menampung slurry

Kondisi operasi :  $T = 30^{\circ}\text{C}$

Rate kolagen masuk = 5619,375 kg/hari

Fraksi masa bahan masuk :

$$X_{\text{Kolagen}} = \frac{4495,5}{5619,375} = 0,800000$$

$$X_{\text{MgCl}_2} = \frac{0,0056}{5619,375} = 0,000001$$

$$X_{\text{Air}} = \frac{1042,3694}{5619,375} = 0,185495$$

$$X_{\text{NaCl}} = \frac{0,0112}{5619,375} = 0,000002$$

$$X_{CaCl_2} = \frac{81,3618}{5619,375} = 0,014479$$

$$X_{H_3PO_4} = \frac{0,127}{5619,375} = 0,000023$$

$$\rho_{kolagen} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{air} = 994,7 \text{ kg/m}^3$$

$$\rho_{CaCl_2} = 1012 \text{ kg/m}^3$$

$$\rho_{MgCl_2} = 1011,5 \text{ kg/m}^3$$

$$\rho_{H_3PO_4} = 1006,5 \text{ kg/m}^3$$

$$\rho_{NaCl} = 1012,85 \text{ kg/m}^3$$

$$\frac{1}{\rho_{campuran}} = \frac{X_{kolagen}}{\rho_{kolagen}} + \frac{X_{Air}}{\rho_{air}} + \frac{X_{CaCl_2}}{\rho_{CaCl_2}} + \frac{X_{MgCl_2}}{\rho_{MgCl_2}} + \frac{X_{NaCl}}{\rho_{NaCl}} + \frac{X_{H_3PO_4}}{\rho_{H_3PO_4}}$$

$$\frac{1}{\rho_{campuran}} = \frac{0,8}{945,1} + \frac{0,185495}{994,7} + \frac{0,014479}{1012} + \frac{0,000001}{1011,5} + \frac{0,000002}{1012,85} + \frac{0,000023}{1006,5}$$

$$\rho_{campuran} = 954,8474 \text{ kg/m}^3 = 59,6111 \text{ lbm/ft}^3$$

$$\text{Rate volumetric kolagen} = \frac{5619,375 \text{ kg/hari}}{954,8474 \text{ kg/m}^3}$$

$$= 5,8851 \text{ m}^3/\text{hari} = 6,811 \cdot 10^{-3} \text{ m}^3/\text{s}$$

$$q_L = 2,4052 \cdot 10^{-3} \text{ cuft/s}$$

$$\mu_{campuran} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = \frac{\frac{1042,3694}{994,7}}{\frac{1042,3694}{994,7} + \frac{4495,5}{945,1} + \frac{81,3618}{1012} + \frac{0,127}{1006,5} + \frac{0,0056}{1011,5} + \frac{0,0112}{1012,85}}$$

$$\varepsilon = 0,1781$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,1781)}}$$

$$= 0,0319$$

$$\mu_{air} \text{ pada } 30^\circ\text{C} = 0,8007 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,8007 \cdot 10^{-3}}{0,0319}$$

$$= 0,0251 \text{ kg/m.s} = 0,0169 \text{ lbm/ft.s}$$

$$\text{Volume larutan kolagen} = \frac{5619,375 \text{ kg/hari}}{954,8474 \text{ kg/m}^3}$$

$$= 5,8851 \text{ m}^3/\text{hari} = 207,8205 \text{ cuft/hari}$$

Diambil : tinggi shell ( $H_s$ ) = 1,5 . Diameter shell ( $D$ )

$$\text{Volume shell} = (\pi/4) \cdot D^2 \cdot H_s = (\pi/4) \cdot D^2 \cdot 1,5 \cdot D = 1,5 \cdot (\pi/4) \cdot D^3$$

$$\text{Volume torispherical dished head (cuft)} = 0,000049 \times D^3 \text{ (inch)}$$

(Brownell & Young, pers.5.11,p.88)

Volume larutan dalam konis = volume konis

$$\begin{aligned} &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot H_t \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D_n^2 \cdot H_n \right) \\ &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot \frac{D}{2 \cdot \tan \alpha} \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D_n^2 \cdot \frac{D_n}{2 \cdot \tan \alpha} \right) \\ &= \left( \frac{\pi \cdot D^3}{24 \cdot \tan \alpha} \right) - \left( \frac{\pi \cdot D_n^3}{24 \cdot \tan \alpha} \right) \\ &= \frac{\pi}{24 \cdot \tan \alpha} (D^3 - D_n^3) \end{aligned}$$

dimana :

$D_n$  = diameter lubang pengeluaran liquida = 0,5 ft (=6 in)

$D$  = diameter konis bagian atas = diameter shell

$H_t$  = tinggi konis

$H_n$  = tinggi konis terpancung

Volume tangki penampung = vol.shell + vol.dished head + vol.konis

Diambil : volume tangki penampung = 1,2 . volume larutan total

$$1,2 \cdot 207,8205 \text{ cuft} = \frac{\pi}{4} \cdot 1,5 \cdot D^3 \cdot \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 + (0,000049 \cdot D^3)$$

$$+ \frac{\pi}{24.tg\alpha} \left( D^3 \left( \frac{1ft}{12in} \right)^3 - Dm^3 \right)$$

$$249,3846 \text{ cuft} = \frac{\pi}{4} . 1,5 . D^3 \cdot \left( \frac{1ft}{12in} \right)^3 + (0,000049 . D^3)$$

$$+ \frac{\pi}{24.tg\alpha} \left( D^3 \left( \frac{1ft}{12in} \right)^3 - (0,5ft)^3 \right)$$

$$249,3846 \text{ cuft} = 6,8177 . 10^{-4} . D^3 + 0,000049 . D^3 + 1,3121 . 10^{-4} . D^3 - 0,0283$$

$$249,3846 \text{ cuft} = 8,6198 . 10^{-4} . D^3$$

$$D = 66,139 \text{ in} = 5,5226 \text{ ft} \approx 5,6 \text{ ft}$$

$$Hs = 1,5 . D = 1,5 . (5,6 \text{ ft}) = 8,4 \text{ ft}$$

$$\text{Volume larutan dalam konis} = \frac{\pi}{24.tg\alpha} (D^3 - (0,5ft)^3)$$

$$= \frac{\pi}{24.tg30''} ((5,6ft)^3 - (0,5ft)^3)$$

$$= 39,7882 \text{ cuft}$$

$$\text{Volume larutan dalam shell} = \text{vol. larutan total} - \text{vol. larutan dalam konis}$$

$$= (249,3846 - 39,7882) \text{ cuft}$$

$$= 209,5964 \text{ cuft}$$

$$\text{Tinggi larutan dalam shell (H)} = \frac{\text{volume larutan dalam shell}}{\frac{\pi}{4} . D^2}$$

$$= \frac{209,5964}{\frac{\pi}{4} . (5,6ft)^2}$$

$$= 8,5098 \text{ ft} \approx 8,6 \text{ ft}$$

$$\text{Tinggi larutan dalam konis} = \text{tinggi konis (Hc)}$$

$$Hc = \frac{D}{2.tg\alpha} - \frac{Dn}{2.tg\alpha}$$

$$= \frac{5,6ft}{2.tg30''} - \frac{0,5ft}{2.tg30''}$$

$$\approx 4,4167 \text{ ft} \approx 4,5 \text{ ft}$$

Tinggi larutan dalam tangki =  $H + H_c$

$$\approx (8,6 + 4,5) \text{ ft}$$

$$\approx 13,1 \text{ ft}$$

$$P_{\text{operasi}} = P_{\text{hidrostatik}} = \left( \frac{\rho \cdot H}{144} \right) \text{psi}$$

$$\begin{aligned} P_{\text{operasi}} &= P_{\text{hidrostatik}} = \left( \frac{\rho \cdot H}{144} \right) \text{psi} \\ &= \left( \frac{59,6111 \text{ lbm / cuft} \cdot 13,1 \text{ ft}}{144} \right) \text{psi} \\ &\approx 5,423 \text{ psi} \end{aligned}$$

$$P_{\text{desain}} = 1,3 \cdot P_{\text{operasi}} = 1,3 \cdot (5,423 \text{ psi}) = 7,0499 \text{ psi}$$

### Tebal shell

$$t_s = \frac{P \cdot ID}{2(f \cdot E - 0,6 \cdot P)} + c \quad (\text{Brownell \& Young, pers. 13.1})$$

dimana :

$$P = P_{\text{desain}} = 7,0499 \text{ psi}$$

$$ID = 5,6 \text{ ft} = 66,139 \text{ in}$$

Konstruksi : bahan konstruksi = carbon stell SA-283, grade D, dengan

$$f = \text{stress maksimum yang diijinkan} = 12650 \text{ psi}$$

tipe sambungan = *double-welded butt joint*, dengan

$$E = \text{welded-joint efficiency} = 0,8$$

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_s = \frac{(7,0499 \text{ psi}) \cdot (66,139 \text{ in})}{2 \cdot (12650,8 - 0,6 \cdot 7,0499) \text{ psi}} + \frac{1}{8} \text{ in}$$

$$t_s = 0,148 \text{ in} \approx \frac{1}{4} \text{ in}$$

**Tebal dished head**

$$t_s = \frac{1}{4} \text{ in}$$

$$OD = ID + 2.t_s$$

$$= 66,139 \text{ in} + (2.(\frac{1}{4})) \text{ in}$$

$$= 66,639 \text{ in}$$

Dari table 5.7 Brownell & Young diperoleh :

$$OD \text{ standar} = 72 \text{ in dengan } t_s = \frac{1}{4} \text{ in}$$

$$r \text{ (crown radius / radius of dish)} = 72 \text{ in}$$

$$icr \text{ (inside corner radius / knuckle radius)} = 4 \frac{3}{8} \text{ in}$$

$$W = \frac{1}{4} \left( 3 + \sqrt{\frac{r}{icr}} \right) \quad (\text{Brownell \& Young, pers. 7.76, p.138})$$

$$= \frac{1}{4} \left( 3 + \sqrt{\frac{72}{4 \frac{3}{8}}} \right)$$

$$= 1,7642$$

$$a = ID/2 = 66,139 \text{ in} / 2 = 33,0695 \text{ in}$$

$$AB = ID/2 - icr = (33,0695 - 4,375) \text{ in} = 28,6945 \text{ in}$$

$$BC = r - icr = (72 - 4,375) \text{ in} = 67,625 \text{ in}$$

$$b = r - \sqrt{BC^2 - AB^2} = 72 - \sqrt{67,625^2 - 28,6945^2} = 10,7647 \text{ in}$$

$$t_d = \frac{P.rc.W}{2.f.E - 0,2.P} + c \quad (\text{Brownell \& Young, pers. 7.77, p.138})$$

dimana :

$$P = P_{\text{desain}} = 7,0499 \text{ psi}$$

Konstruksi : bahan konstruksi = carbon stell SA-283, grade D, dengan

$$f = \text{stress maksimum yang diijinkan} = 12650 \text{ psi}$$

$$\text{tipe sambungan} = \text{double-welded butt joint, dengan}$$

$$E = \text{welded-joint efficiency} = 0,8$$

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_d = \frac{(7,0499 \text{ psi}).(72 \text{ in}).1,7642}{2.(12650 \text{ psi}).(0,8) - 0,2.(7,0499 \text{ psi})} + \frac{1}{8} \text{ in}$$

$$= 0,1692 \text{ in} \approx \frac{1}{4} \text{ in}$$

Dipilih panjang straight-flange (sf) = 2 in

(Brownell & Young, table 5.8, p.93)

$$OA = t + b + sf$$

$$= (3/16 + 10,7647 + 2) \text{ in}$$

$$= 12,9522 \text{ in} = 1,0794 \text{ ft} \approx 1,1 \text{ ft}$$

Tinggi tangki keseluruhan = tinggi shell + tinggi dish + tinggi konis

$$= 8,4 \text{ ft} + 4,5 \text{ ft} + 1,1 \text{ ft}$$

$$= 14 \text{ ft}$$

### Tebal konis

$$t_c = \frac{P.HD}{2.\cos\alpha.(f.E - 0,6.P)} + c \quad (\text{Brownell \& Young, pers.6.154, p.118})$$

$$= \frac{(7,0499 \text{ psi}).(66,139 \text{ in})}{2.\cos 30^\circ.(12650.0,8 - 0,6.7,0499) \text{ psi}} + \frac{1}{8} \text{ in}$$

$$= 0,1516 \text{ in} \approx 3/16 \text{ in}$$

### Spesifikasi :

#### Tangki

Kapasitas	: 249,3846 cuft
Diameter	: 5,6 ft
Tinggi tutup atas (dished head)	: 1,1 ft
Tinggi tutup bawah (konikal)	: 1,1 ft
Tinggi shell	: 8,4 ft
Tinggi tangki total	: 14 ft
Tebal shell	: $\frac{1}{4}$ in
Tebal tutup atas (dished head)	: $\frac{1}{4}$ in
Tebal tutup bawah (konikal)	: $\frac{1}{4}$ in

Bahan konstruksi : Carbon steel SA-283, grade D  
 Jumlah tangki : 1 buah

## 16. TANGKI EKSTRAKSI I (F-210)

Fungsi : Mengekstrak gelatin dari larutan kolagen

Tipe : silinder tegak dengan tutup atas berbentuk dished head, tutup bawah berbentuk konis dan dilengkapi dengan pengaduk dan coil pemanas.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk slurry

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Perhitungan :

Rate kolagen masuk = 5619,375 kg/hari

Rate air masuk = 9297,2806 kg/hari

Rate HCl 5% masuk = 75,4071 kg/hari

Rate total bahan masuk tangki ekstraksi =  $(5619,375 + 9297,2806 + 75,4071)$   
 $= 15292,0627 \text{ kg/hari}$

Fraksi masa bahan masuk :

$$X_{\text{Kolagen}} = \frac{4495,5}{15292,0627} = 0,294$$

$$X_{\text{MgCl}_2} = \frac{0,0056}{15292,0627} = 3,6620 \cdot 10^{-7}$$

$$X_{\text{Air}} = \frac{10339,65}{15292,0627} = 0,6761$$

$$X_{\text{NaCl}} = \frac{0,0112}{15292,0627} = 7,3241 \cdot 10^{-7}$$

$$X_{\text{CaCl}_2} = \frac{81,3618}{15292,0627} = 5,3205 \cdot 10^{-3}$$

$$X_{\text{H}_3\text{PO}_4} = \frac{0,127}{15292,0627} = 8,305 \cdot 10^{-6}$$

$$X_{\text{HCl}5\%} = \frac{75,4071}{15292,0627} = 4,9311 \cdot 10^{-3}$$

$$\rho_{\text{kolagen}} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{gelatin}} = 988,4 \text{ kg/m}^3$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{HCl}5\%} = 1007,1 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1003,4 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1003,95 \text{ kg/m}^3$$



$$\rho_{H_3PO_4} = 999,3 \text{ kg/m}^3$$

$$\rho_{NaCl} = 994,94 \text{ kg/m}^3$$

$$\frac{1}{\rho_{campuran}} = \frac{X_{kolagen}}{\rho_{kolagen}} + \frac{X_{air}}{\rho_{air}} + \frac{X_{CaCl_2}}{\rho_{CaCl_2}} + \frac{X_{MgCl_2}}{\rho_{MgCl_2}} + \frac{X_{NaCl}}{\rho_{NaCl}} + \frac{X_{H_3PO_4}}{\rho_{H_3PO_4}}$$

$$\begin{aligned} \frac{1}{\rho_{campuran}} &= \frac{0,2948}{945,1} + \frac{0,6761}{988,07} + \frac{5,3205 \cdot 10^{-3}}{1003,4} + \frac{3,662 \cdot 10^{-7}}{1003,95} + \frac{7,3241 \cdot 10^{-7}}{994,94} \\ &\quad + \frac{8,305 \cdot 10^{-6}}{999,3} + \frac{4,9311 \cdot 10^{-3}}{1007,1} \end{aligned}$$

$$\rho_{campuran} = 993,6445 \text{ kg/m}^3 = 62,0311 \text{ lbm/ft}^3$$

$$\mu_{campuran} = \frac{\mu}{\phi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\phi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\begin{aligned} \varepsilon &= (10339,65/988,07) / ((10339,65/988,07) + (4495,5/945,1) + (81,3618/1003,4) \\ &\quad + (0,0056/1003,95) + (0,0112/994,94) + (0,127/999,3) + (75,4071/1007,1)) \end{aligned}$$

$$\varepsilon = 0,6805$$

$$\begin{aligned} \phi_p &= \frac{1}{10^{1,82(1-0,6805)}} \\ &= 0,2621 \end{aligned}$$

$$\mu_{air} \text{ pada } 50^\circ\text{C} = 0,5494 \cdot 10^{-3} \text{ kg/m.s}$$

$$\begin{aligned} \mu_{campuran} &= \frac{0,5494 \cdot 10^{-3}}{0,2621} \\ &= 2,09615 \cdot 10^{-3} \text{ kg/m.s} = 1,4054 \cdot 10^{-3} \text{ lbm/ft.s} \end{aligned}$$

$$\begin{aligned} \text{Volume larutan kolagen} &= \frac{15292,0627 \text{ kg/hari}}{993,6445 \text{ kg/m}^3} \\ &= 15,3899 \text{ m}^3/\text{hari} = 543,4635 \text{ cuft/hari} \end{aligned}$$

waktu tinggal = 1 jam maka :

$$\begin{aligned} \text{Volume larutan kolagen} &= \frac{543,4635 \text{ cuft / hari}}{24 \text{ jam / hari}} \times 1 \text{ jam} \\ &= 22,6443 \text{ cuft} \end{aligned}$$

Diambil : tinggi shell ( $H_s$ ) = 1,5 . Diameter shell ( $D$ )

$$\text{Volume shell} = (\pi/4).D^2.H_s = (\pi/4).D^2.1,5.D = 1,5.(\pi/4).D^3$$

$$\text{Volume torispherical dished head (cuft)} = 0,000049 \times D^3 \text{ (inch)}$$

(Brownell & Young, pers.5.11,p.88)

Volume larutan dalam konis = volume konis

$$\begin{aligned} &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot H_t \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot Dn^2 \cdot H_n \right) \\ &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot \frac{D}{2 \cdot \text{tg} \alpha} \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot Dn^2 \cdot \frac{Dn}{2 \cdot \text{tg} \alpha} \right) \\ &= \left( \frac{\pi \cdot D^3}{24 \cdot \text{tg} \alpha} \right) - \left( \frac{\pi \cdot Dn^3}{24 \cdot \text{tg} \alpha} \right) \\ &= \frac{\pi}{24 \cdot \text{tg} \alpha} (D^3 - Dn^3) \end{aligned}$$

dimana :

$Dn$  = diameter lubang pengeluaran liquida = 0,5 ft (=6 in)

$D$  = diamete konis bagian atas = diameter shell

$H_t$  = tinggi konis

$H_n$  = tinggi konis terpancung

Volume tangki penampung = vol.shell + vol.dished head + vol.konis

Diambil : volume tangki penampung = 1,2.volume larutan total

$$1,2. 22,6443 \text{ cuft} = \frac{\pi}{4} \cdot 1,5 \cdot D^3 \cdot \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 + (0,000049 \cdot D^3) + \frac{\pi}{24 \cdot \text{tg} \alpha} (D^3 - Dn^3)$$

$$27,1732 \text{ cuft} = 6,8177 \cdot 10^{-4} \cdot D^3 + 4,9 \cdot 10^{-5} \cdot D^3 + \frac{\pi}{24 \cdot \text{tg} \alpha} (D^3 - Dn^3)$$

$$D = 32,6632 \text{ in} = 2,7219 \text{ ft} \approx 2,8 \text{ ft}$$

$$H_s = 1,5 \cdot D = 1,5 \cdot (2,8 \text{ ft}) = 4,2 \text{ ft}$$

$$\text{Volume larutan dalam konis} = \frac{\pi}{24 \cdot \text{tg} \alpha} (D^3 - (0,5 \text{ ft})^3)$$

$$= \frac{\pi}{24 \cdot \text{tg} 30^\circ} ((2,8 \text{ ft})^3 - (0,5 \text{ ft})^3)$$

$$= 4,9487 \text{ cuft}$$

$$\begin{aligned}
 \text{Volume larutan dalam shell} &= \text{vol. larutan total} - \text{vol. larutan dalam konis} \\
 &= (27,1732 - 4,9487) \text{ cuft} \\
 &= 22,2245 \text{ cuft}
 \end{aligned}$$

$$\begin{aligned}
 \text{Tinggi larutan dalam shell (H)} &= \frac{\text{volume larutan dalam shell}}{\frac{\pi}{4} D^2} \\
 &= \frac{22,2245}{\frac{\pi}{4} (2,8 \text{ ft})^2} \\
 &= 3,6093 \text{ ft}
 \end{aligned}$$

$$\text{Tinggi larutan dalam konis} = \text{tinggi konis (Hc)}$$

$$\begin{aligned}
 Hc &= \frac{D}{2 \tan \alpha} - \frac{Dn}{2 \tan \alpha} \\
 &= \frac{2,8 \text{ ft}}{2 \tan 30^\circ} - \frac{0,5 \text{ ft}}{2 \tan 30^\circ} \\
 &= 1,9919 \text{ ft}
 \end{aligned}$$

$$\begin{aligned}
 \text{Tinggi larutan dalam tangki} &= H + Hc \\
 &= (3,6093 + 1,9919) \text{ ft} \\
 &= 5,6012 \text{ ft}
 \end{aligned}$$

$$\begin{aligned}
 P_{\text{operasi}} = P_{\text{hidrostatik}} &= \left( \frac{\rho H}{144} \right) \text{ psi} \\
 &= \left( \frac{62,0311 \text{ lbm/cuft} \cdot 5,6012 \text{ ft}}{144} \right) \text{ psi} \\
 &= 2,4128 \text{ psi}
 \end{aligned}$$

$$P_{\text{desain}} = 1,5 \cdot P_{\text{operasi}} = 1,5 \cdot (2,4128 \text{ psi}) = 3,6193 \text{ psi}$$

### Tebal shell

$$t_s = \frac{P \cdot D}{2(f \cdot E - 0,6 \cdot P)} + c \quad (\text{Brownell \& Young, pers. 13.1})$$

dimana :

$$P = P_{\text{desain}} = 3,6193 \text{ psi}$$

$$ID = 32,6632 \text{ in} = 2,7219 \text{ ft} \approx 2,8 \text{ ft}$$

Konstruksi : bahan konstruksi = High Alloy steel SA-240, grade M type 316

$f$  = stress maksimum yang diijinkan = 18750 psi

tipe sambungan = *double-welded butt joint*, dengan

$E$  = *welded-joint efficiency* = 0,8

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_s = \frac{(3,6193 \text{ psi}) \cdot (32,6632 \text{ in})}{2 \cdot (18750,8 - 0,6 \cdot 3,6193) \text{ psi}} + \frac{1}{8} \text{ in}$$

$$t_s = 0,1289 \approx 3/16 \text{ in}$$

#### Tebal dished head

$$t_s = 3/16 \text{ in}$$

$$\begin{aligned} OD &= ID + 2 \cdot t_s \\ &= 32,6632 \text{ in} + (2 \cdot (3/16)) \text{ in} \\ &= 33,0382 \text{ in} \end{aligned}$$

Dari table 5.7 Brownell & Young diperoleh :

OD standar = 34 in dengan  $t_s = 3/16 \text{ in}$

$r$  (crown radius / radius of dish) = 34 in

$icr$  (inside corner radius / knuckle radius) =  $2 \frac{1}{8} \text{ in}$

$$W = \frac{1}{4} \left( 3 + \sqrt{\frac{r}{icr}} \right) \quad (\text{Brownell \& Young, pers. 7.76, p.138})$$

$$= \frac{1}{4} \left( 3 + \sqrt{\frac{34}{2 \frac{1}{8}}} \right) = 1,75$$

$$a = ID/2 = 32,6632 \text{ in} / 2 = 16,3316 \text{ in}$$

$$AB = ID/2 - icr = (16,3316 - 2,125) \text{ in} = 14,2066 \text{ in}$$

$$BC = r - icr = (34 - 2,125) \text{ in} = 31,875 \text{ in}$$

$$b = r - \sqrt{BC^2 - AB^2} = 34 - \sqrt{31,875^2 - 14,2066^2} = 5,466 \text{ in}$$

$$t_d = \frac{P \cdot r \cdot W}{2 \cdot f \cdot E - 0,2 \cdot P} + c \quad (\text{Brownell \& Young, pers. 7.77, p.138})$$

dimana :

$$P = P_{\text{desain}} = 3,6193 \text{ psi}$$

Konstruksi : bahan konstruksi = High Alloy steel SA-240, grade M type 316

$f$  = stress maksimum yang diijinkan = 18750 psi

tipe sambungan = *double-welded butt joint*, dengan

$E$  = *welded-joint efficiency* = 0,8

(Brownell & Young, table 13.2)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_d = \frac{(3,6193 \text{ psi}) \cdot (34 \text{ in}) \cdot 1,75}{2 \cdot (18750 \text{ psi}) \cdot (0,8) - 0,2 \cdot (3,6193 \text{ psi})} + \frac{1}{8} \text{ in}$$

$$= 0,1322 \text{ in} \approx 3/16 \text{ in}$$

Dipilih panjang straight-flange (sf) = 2 in

(Brownell & Young, table 5.8, p.93)

$$OA = t + b + sf$$

$$= (3/16 + 5,466 + 2) \text{ in}$$

$$= 7,6535 \text{ in} = 0,6378 \text{ ft} \approx 0,7 \text{ ft}$$

Tinggi tangki keseluruhan = tinggi shell + tinggi dish + tinggi konis

$$= 4,2 \text{ ft} + 0,7 \text{ ft} + 2 \text{ ft}$$

$$= 6,9 \text{ ft}$$

### Perhitungan Pengaduk

Dipilih tipe pengaduk : flat six-blade turbine agitator with disk.

Kecepatan agitator antara 20 – 200 rpm (Geankoplis), diambil 50 rpm.

$$N = 50 \text{ rpm} = 0,8333 \text{ rps.}$$

$$\rho_{\text{campuran}} = 993,6445 \text{ kg/m}^3 = 62,0311 \text{ lbm/ft}^3$$

$$\mu_{\text{campuran}} = 2,09615 \cdot 10^{-3} \text{ kg/m.s} = 1,4054 \cdot 10^{-3} \text{ lbm/ft.s}$$

Dari table 3.4, p.144 Geankoplis diperoleh :

$$Da/Dt = 0,3 \quad W/Da = 1/5 \quad L/Da = 1/4 \quad C/Dt = 1/3 \quad J/Dt = 1/12$$

dimana :

$$Dt = \text{diameter tangki} = 2,8 \text{ ft} = 0,8535 \text{ m}$$

$$Da = \text{diameter pengaduk} = 0,3 \times 0,8535 \text{ m} = 0,2561 \text{ m}$$

$$W = \text{lebar blade} = (0,2561 \text{ m})/5 = 0,0512 \text{ m}$$

$$L = \text{panjang blade} = (0,2561 \text{ m})/4 = 0,064 \text{ m}$$

$$C = \text{jarak dari dasar tangki ke pusat pengaduk} = (0,8535 \text{ m})/3 = 0,2845 \text{ m}$$

$$J = \text{lebar baffle} = \frac{1}{12} \times 0,8535 \text{ m} = 0,0711 \text{ m}$$

$$N_{Re} = \frac{Da^2 \cdot N \cdot \rho}{\mu}$$

$$= \frac{(0,2561)^2 \cdot 0,8333 \cdot 993,6445}{2,09615 \cdot 10^{-3}}$$

$$= 25907,7212 \text{ (turbulen)}$$

Dari fig. 3.4-4, p.145 Geankoplis, dengan memotong kurva 1 dengan  $N_{Re}$ , diperoleh  $N_p = 5$ .

$$N_p = \frac{P}{\rho \cdot N^3 \cdot Da^5} \quad (\text{Geankoplis, pers. 3.4-2, p.145})$$

Dimana :

$N_p$  = power number

$P$  = power pengaduk

$N$  = kecepatan pengaduk

$$P = N_p \cdot \rho \cdot N^3 \cdot Da^5$$

$$= 5 \cdot 993,6445 \text{ kg/m}^3 \cdot (0,8333 \text{ rps})^3 \cdot (0,2561 \text{ m})^5$$

$$= 3,167 \text{ W} = 0,0042 \text{ hp}$$

$$\text{Power input} = 110\% \times 0,0042 \text{ hp} = 0,0047 \text{ hp}$$

$$\text{Transmission system losses} = 20\% \text{ dari total hp}$$

$$0,0047 \text{ Hp} + 0,2 \cdot \text{total hp} = \text{total hp}$$

$$\text{total hp} = 0,0058 \text{ hp}$$

$$\text{Efisiensi motor} = 80\% \quad (\text{Peter \& Timmerhaus, fig.14-38, p.521})$$

$$\text{Power yang dibutuhkan} = (0,0058 \text{ hp})/80\% = 0,0073 \text{ hp} \approx 0,25 \text{ hp}$$

### Perhitungan Coil Pemanas

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q = 250882,4896 \text{ kkal/hari} = 12148,9846 \text{ W} = 41454,2075 \text{ btu/h}$
- ❖ Suhu bahan masuk tangki pada  $30^{\circ}\text{C}$  ( $=86^{\circ}\text{F}$ ) yaitu kolagen, impurities dan HCl 5% masuk pada suhu  $50^{\circ}\text{C}$  ( $=122^{\circ}\text{F}$ )
- ❖ Suhu kolagen, impurities, gelatin dan HCl keluar  $50^{\circ}\text{C}$  ( $=122^{\circ}\text{F}$ )
- ❖ Pemanas berupa steam dengan suhu masuk  $120^{\circ}\text{C}$  ( $=248^{\circ}\text{F}$ ) dan suhu keluar  $120^{\circ}\text{C}$
- ❖ Massa steam  $= 476,5811 \text{ kg/hari} = 5,516 \cdot 10^{-3} \text{ kg/s}$

$$\rho_{\text{campuran}} = 993,6445 \text{ kg/m}^3 = 62,0311 \text{ lbm/ft}^3$$

$$\mu_{\text{campuran}} = 2,09615 \cdot 10^{-3} \text{ kg/m.s} = 5,0708 \text{ lbm/ft.h}$$

$$C_{p\text{HCl } 5\%} = 0,966865 \text{ kkal/kg}^{\circ}\text{C} \quad C_{p\text{CaCl}_2} = 0,2162 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p\text{kolagen}} = 0,3369 \text{ kkal/kg}^{\circ}\text{C} \quad C_{p\text{MgCl}_2} = 0,2520 \text{ kkal/kg}^{\circ}\text{C}$$

$$C_{p\text{H}_3\text{PO}_4} = 0,452 \text{ kkal/kg}^{\circ}\text{C} \quad C_{p\text{NaCl}} = 0,2737 \text{ kkal/kg}^{\circ}\text{C}$$

$$X_{\text{Kolagen}} = 0,294 \quad X_{\text{MgCl}_2} = 3,6620 \cdot 10^{-7}$$

$$X_{\text{Air}} = 0,6761 \quad X_{\text{NaCl}} = 7,3241 \cdot 10^{-7}$$

$$X_{\text{CaCl}_2} = 5,3205 \cdot 10^{-3} \quad X_{\text{H}_3\text{PO}_4} = 8,305 \cdot 10^{-6}$$

$$X_{\text{HCl } 5\%} = 4,9311 \cdot 10^{-3}$$

$$C_{p\text{campuran}} = \sum x_i \cdot C_{p_i}$$

$$= (0,294 \times 0,3369) + (0,6761 \times 1) + (5,3205 \cdot 10^{-3} \times 0,2162) + \\ (4,9311 \cdot 10^{-3} \times 0,966865) + (3,6620 \cdot 10^{-7} \times 0,252) + (7,3241 \cdot 10^{-7} \times \\ 0,2737) + (8,305 \cdot 10^{-6} \times 0,452)$$

$$C_{p\text{campuran}} = 0,7811 \text{ kkal/kg}^{\circ}\text{C} = 0,7811 \text{ kal/g}^{\circ}\text{C} \times 4,1868$$

$$= 3,2703 \text{ kJ/kg.K}$$

$$\Delta T_{\text{LMTD}} = \frac{(T_1 - t_2) - (T_2 - t_1)}{\ln \frac{(T_1 - t_2)}{(T_2 - t_1)}} = \frac{(248 - 122) - (248 - 104)}{\ln \frac{(248 - 122)}{(248 - 104)}} = 134,7998^{\circ}\text{F}$$

$$T_c = \frac{1}{2} \cdot (T_1 + T_2) = \frac{1}{2} \cdot (248 + 248) = 248^{\circ}\text{F}$$

$t_c = \frac{1}{2} \cdot (t_1 + t_2) = \frac{1}{2} \cdot (122 + 104) = 113^{\circ}\text{F}$

Berdasarkan kern tabel 11, p.844, trial ukuran pipa coil = 1/8 in IPS, sch 40

$D_o = 0,405 \text{ in}$

$D_i = 0,269 \text{ in}$

$a' = 0,058 \text{ in}$

$a'' = 0,106 \text{ ft}^2/\text{ft}$

Evaluasi Perpindahan Panas

Sisi bejana : fluida dingin	Sisi pipa : Steam, fluida panas
$\rho_{\text{campuran}} = 62,0311 \text{ lbm/ft}^3$ $D_a = 0,8402 \text{ ft}$ $N = 50 \text{ rpm} = 3000 \text{ rph}$ $\mu_{\text{camp}} = 5,0708 \text{ lbm/ft.h}$ $N_{\text{re}} = \frac{\rho \cdot D_a^2 \cdot N}{\mu}$ $= \frac{62,0311 \cdot (0,8402)^2 \cdot 3000}{5,0708 \text{ lbm/ft.h}}$ $= 25907,1482$ $J_c = 520 \quad (\text{kern fig 20-2, p-718})$ $h_o = J_c \cdot \frac{k}{D_{\text{ivesse}} l} \left( \frac{C_p \cdot \mu}{k} \right)^{\frac{1}{3}} \left( \frac{\mu}{\mu_w} \right)^{0,14}$ Dimana : $D_i = 2,8 \text{ ft}$ $k = 0,45 \text{ W/m.K} = 0,26 \text{ Btu/hr.ft.}^{\circ}\text{F}$ $C_p \text{ rata-rata} = 3,2703 \text{ kJ/kg.K}$ $= 0,7811 \text{ Btu/lbm.}^{\circ}\text{F}$ $h_o = 520 \cdot \frac{0,26}{2,8} \left( \frac{0,7811 \cdot 5,0708}{0,26} \right)^{\frac{1}{3}}$ $= 119,6985 \text{ Btu/hr.ft}^2.^{\circ}\text{F}$	$h_{io} = 1500 \text{ Btu/hr.ft.}^{\circ}\text{F} \text{ (untuk steam)}$

$$U_c = \frac{h_{io} \times h_o}{h_{io} + h_o} = \frac{1500 \times 119,6985}{1500 + 119,6985} = 110,8526 \text{ Btu/hr ft}^2.^{\circ}\text{F}$$

$$R_d = \frac{U_c - U_d}{U_c \times U_d} \qquad \text{Diambil } R_d = 0,004$$



$$0,004 = \frac{110,8526 - U_d}{110,8526 \times U_d} ; U_d = 76,7991 \text{ Btu/hr.ft}^2.\text{°F}$$

$$A_{\text{coil}} = \frac{Q}{U_d \Delta T_{\text{LMTD}}} = \frac{41454,2075 \text{ btu/h}}{76,7991 \cdot 134,7998} = 4,0043 \text{ ft}^2$$

$$L = \frac{A_{\text{coil}}}{a''} = \frac{4,0043 \text{ ft}^2}{0,106 \text{ ft}^2/\text{ft}} = 37,7764 \text{ ft}$$

$$\begin{aligned} D_c &= \text{diameter coil} = 0,65 \cdot d_i \text{ vessel} \\ &= 0,65 \cdot (2,8) \\ &= 1,82 \text{ ft} \end{aligned}$$

$$n_c = \frac{L}{\pi \cdot d_c} = \frac{37,7764}{\pi \cdot 1,82} = 6,6069 \text{ buah} \approx 7 \text{ buah}$$

sc = Spasi coil, diambil 2 in

$$\begin{aligned} h_c &= ((n_c - 1) \cdot (d_o + sc)) + d_o \\ &= ((7 - 1) \cdot (0,405 + 2)) + 0,405 \\ &= 14,835 \text{ in} = 1,2363 \text{ ft} \end{aligned}$$

Pengecekan :

Tinggi liquida dibagian silinder (Hs) = 3,6093 ft

$h_c < H_s$  (memenuhi)

### Spesifikasi :

*Tangki*

Kapasitas	: 27,1732 cuft
Diameter	: 2,8 ft
Tinggi tutup atas (dished head)	: 0,7 ft
Tinggi tutup bawah (konikal)	: 2 ft
Tinggi shell	: 4,2 ft
Tinggi tangki total	: 6,9 ft
Tebal shell	: 3/16 in
Tebal tutup atas (dished head)	: 3/16 in
Tebal tutup bawah (konikal)	: 3/16 in

Bahan konstruksi	: High Alloy stell SA-240, grade M type 316
Jumlah tangki	: 1 buah
<i>Pengaduk</i>	
Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,8402 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 0,25 Hp
Jumlah pengaduk	: 1 buah
<i>Coil pemanas</i>	
Diameter	: 1,82 ft
Spasi coil	: 2 in
Tinggi coil	: 1,2363 ft

## 17. TANGKI EKSTRAKSI II (F-220)

Fungsi : Mengekstrak gelatin dari larutan kolagen

Tipe : silinder tegak dengan tutup atas berbentuk dished head, tutup bawah berbentuk konis dan dilengkapi dengan pengaduk dan jaket pemanas.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk liquida

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Perhitungan :

Rate total bahan masuk tangki ekstraksi = 15292,0627 kg/hari

Fraaksi masa bahan masuk :

$$X_{\text{Kolagen}} = \frac{3416,58}{15292,0627} = 0,2234$$

$$X_{\text{MgCl}_2} = \frac{0,0056}{15292,0627} = 3,6620 \cdot 10^{-7}$$

$$X_{\text{Air}} = \frac{10339,65}{15292,0627} = 0,6761$$

$$X_{\text{NaCl}} = \frac{0,0112}{15292,0627} = 7,3241 \cdot 10^{-7}$$

$$X_{\text{CaCl}_2} = \frac{81,3618}{15292,0627} = 5,3205 \cdot 10^{-3}$$

$$X_{\text{H}_3\text{PO}_4} = \frac{0,127}{15292,0627} = 8,305 \cdot 10^{-6}$$

$$X_{\text{HCl}5\%} = \frac{75,4071}{15292,0627} = 4,9311 \cdot 10^{-3} \quad X_{\text{gelatin}} = \frac{1078,92}{15292,0627} = 0,0706$$

$$\rho_{\text{kolagen}} = 945,1 \text{ kg/m}^3 \quad (\text{Ward}, 1977)$$

$$\rho_{\text{gelatin}} = 988,4 \text{ kg/m}^3$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{HCl}5\%} = 1007,1 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1003,4 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1003,95 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 999,3 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 994,94 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}}$$

$$\begin{aligned} \frac{1}{\rho_{\text{campuran}}} &= \frac{0,2234}{945,1} + \frac{0,6761}{988,07} + \frac{5,3205 \cdot 10^{-3}}{1003,4} + \frac{3,662 \cdot 10^{-7}}{1003,95} + \frac{7,3241 \cdot 10^{-7}}{994,94} \\ &\quad + \frac{8,305 \cdot 10^{-6}}{999,3} + \frac{4,9311 \cdot 10^{-3}}{1007,1} + \frac{0,0706}{988,4} \end{aligned}$$

$$\rho_{\text{campuran}} = 997,728 \text{ kg/m}^3 = 62,286 \text{ lbm/ft}^3$$

$$\mu_{\text{campuran}} = \frac{\mu}{\phi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\phi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\begin{aligned} \varepsilon &= (10339,65/988,07) / ((10339,65/988,07) + (3416,58/945,1) + \\ &\quad (81,3618/1003,4) + (0,0056/1003,95) + (0,0112/994,94) + (0,127/999,3) \\ &\quad + (75,4071/1007,1) + (1078,92/988,4)) \end{aligned}$$

$$\varepsilon = 0,6827$$

$$\phi_p = \frac{1}{10^{1,82(1-0,6827)}}$$

$$= 0,2646$$

$$\mu_{\text{air}} \text{ pada } 50^\circ\text{C} = 0,5494 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,5494 \cdot 10^{-3}}{0,2646}$$

$$= 2,0763 \cdot 10^{-3} \text{ kg/m.s} = 1,3952 \cdot 10^{-3} \text{ lbm/ft.s}$$

$$\text{Volume larutan kolagen} = \frac{14992,0627 \text{ kg/hari}}{997,728 \text{ kg/m}^3}$$

$$= 15,3269 \text{ m}^3/\text{hari} = 541,2614 \text{ cuft/hari}$$

waktu tinggal = 1 jam maka :

$$\text{Volume larutan kolagen} = \frac{541,2614 \text{ cuft / hari}}{24 \text{ jam / hari}} \times 1 \text{ jam}$$

$$= 22,5526 \text{ cuft}$$

$$C_{p\text{HCl } 5\%} = 0,966865 \text{ kkal/kg}^\circ\text{C} \quad C_{p\text{CaCl}_2} = 0,2162 \text{ kkal/kg}^\circ\text{C}$$

$$C_{p\text{kolagen}} = 0,3369 \text{ kkal/kg}^\circ\text{C} \quad C_{p\text{MgCl}_2} = 0,2520 \text{ kkal/kg}^\circ\text{C}$$

$$C_{p\text{H}_3\text{PO}_4} = 0,452 \text{ kkal/kg}^\circ\text{C} \quad C_{p\text{NaCl}} = 0,2737 \text{ kkal/kg}^\circ\text{C}$$

$$C_{p\text{gelatin}} = 0,3613 \text{ kkal/kg}^\circ\text{C}$$

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q = 63422,3877 \text{ kkal/hari} = 44043,3248 \text{ W} = 150282,611 \text{ btu/h}$
- ❖ Suhu bahan masuk tangki pada  $45^\circ\text{C}$  ( $113^\circ\text{F}$ ) yaitu kolagen, gelatin, impurities dan HCl 5%
- ❖ Suhu kolagen, impurities, gelatin dan HCl keluar  $50^\circ\text{C}$  ( $122^\circ\text{F}$ )
- ❖ Pemanas berupa steam dengan suhu masuk  $120^\circ\text{C}$  ( $248^\circ\text{F}$ ) dan suhu keluar  $120^\circ\text{C}$
- ❖ Massa steam =  $120,4948 \text{ kg/hari} = 11,0685 \text{ btu/h}$

Dengan cara yang sama pada perhitungan tangki ekstraksi I (F-210), maka didapat:

#### Spesifikasi :

##### *Tangki*

Kapasitas : 27,0631 cuft

Diameter : 2,8 ft

Tinggi tutup atas (dished head) : 0,7 ft

Tinggi tutup bawah (konikal) : 2 ft

Tinggi shell : 4,2 ft

Tinggi tangki total : 6,9 ft

Tebal shell	: 3/16 in
Tebal tutup atas (dished head)	: 3/16 in
Tebal tutup bawah (konikal)	: 3/16 in
Bahan konstruksi	: High Alloy stell SA-240, grade M type 316
Jumlah tangki	: 1 buah

#### *Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,8402 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 0,25 Hp
Jumlah pengaduk	: 1 buah

#### *Coil pemanas*

Diameter	: 1,82 ft
Spasi coil	: 2 in
Tinggi coil	: 2,375 ft

### 18. Rotary Vacuum Filter (H-221)

Fungsi : Memisahkan kolagen dari larutan gelatin

Tipe : Rotary vacuum drum filter

Dasar pemilihan : cocok digunakan untuk pemisahan dalam kondisi kontinyu

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Rate total bahan masuk rotary filter = 15292,0627 kg/hari

Fraksi masa bahan masuk :

$$\begin{aligned}
 X_{\text{Kolagen}} &= \frac{3416,58}{15292,0627} = 0,2234 & X_{\text{MgCl}_2} &= \frac{0,0056}{15292,0627} = 3,6620 \cdot 10^{-7} \\
 X_{\text{Air}} &= \frac{10339,65}{15292,0627} = 0,6761 & X_{\text{NaCl}} &= \frac{0,0112}{15292,0627} = 7,3241 \cdot 10^{-7} \\
 X_{\text{CaCl}_2} &= \frac{81,3618}{15292,0627} = 5,3205 \cdot 10^{-3} & X_{\text{H}_3\text{PO}_4} &= \frac{0,127}{15292,0627} = 8,305 \cdot 10^{-6} \\
 X_{\text{HCl}5\%} &= \frac{75,4071}{15292,0627} = 4,9311 \cdot 10^{-3} & X_{\text{gelatin}} &= \frac{1078,92}{15292,0627} = 0,0706
 \end{aligned}$$

$$\rho_{\text{kolagen}} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{gelatin}} = 988,4 \text{ kg/m}^3$$

$$\rho_{\text{air}} = 994,7 \text{ kg/m}^3$$

$$\rho_{\text{HCl 15\%}} = 1007,1 \text{ kg/m}^3$$

$$\rho_{\text{CaCl}_2} = 1003,4 \text{ kg/m}^3$$

$$\rho_{\text{MgCl}_2} = 1003,95 \text{ kg/m}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 999,3 \text{ kg/m}^3$$

$$\rho_{\text{NaCl}} = 994,94 \text{ kg/m}^3$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{X_{\text{kolagen}}}{\rho_{\text{kolagen}}} + \frac{X_{\text{Air}}}{\rho_{\text{air}}} + \frac{X_{\text{CaCl}_2}}{\rho_{\text{CaCl}_2}} + \frac{X_{\text{MgCl}_2}}{\rho_{\text{MgCl}_2}} + \frac{X_{\text{NaCl}}}{\rho_{\text{NaCl}}} + \frac{X_{\text{H}_3\text{PO}_4}}{\rho_{\text{H}_3\text{PO}_4}}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,2234}{945,1} + \frac{0,6761}{988,07} + \frac{5,3205 \cdot 10^{-3}}{1003,4} + \frac{3,662 \cdot 10^{-7}}{1003,95} + \frac{7,3241 \cdot 10^{-7}}{994,94}$$

$$+ \frac{8,305 \cdot 10^{-6}}{999,3} + \frac{4,9311 \cdot 10^{-3}}{1007,1} + \frac{0,0706}{988,4}$$

$$\rho_{\text{campuran}} = 997,728 \text{ kg/m}^3 = 62,286 \text{ lbm/ft}^3$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p} \quad (\text{Geankoplis 2 ed., 1985})$$

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

$$\varepsilon = (10339,65/988,07) / ((10339,65/988,07) + (3416,58/945,1) + (81,3618/1003,4) + (0,0056/1003,95) + (0,0112/994,94) + (0,127/999,3) + (75,4071/1007,1) + (1078,92/988,4))$$

$$\varepsilon = 0,6827$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,6827)}}$$

$$= 0,2646$$

$$\mu_{\text{air}} \text{ pada } 50^\circ\text{C} = 0,5494 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,5494 \cdot 10^{-3}}{0,2646}$$

$$= 2,0763 \cdot 10^{-3} \text{ kg/m.s} = 1,3952 \cdot 10^{-3} \text{ lbm/ft.s}$$

$$\begin{aligned}\text{Volume larutan slurry} &= \frac{15292,0627 \text{ kg/hari}}{997,728 \text{ kg/m}^3} \\ &= 15,3269 \text{ m}^3/\text{hari} = 541,2614 \text{ cuft/hari}\end{aligned}$$

Dengan cara yang sama pada perhitungan Rotary vacuum filter (H-144), maka didapat :

**Spesifikasi :**

Type	: Rotary vacuum drum filter
Filter rate	: 0,177 kg/s
Diameter	: 0,5852 m = 1,9199 ft
Jumlah	: 1 buah

### 19. TANGKI EKSTRAKSI III (F-230)

Fungsi : Mengekstrak gelatin dari larutan kolagen.

Type : silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berbentuk konis serta dilengkapi dengan pengaduk dan coil pemanas.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk liquid

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Kapasitas =  $9249,6098 \text{ kg/hari} \times 2,2046 = 20391,68977 \text{ lbm/hari}$

Perhitungan :

Waktu tinggal = 120 menit = 2 jam

Rate larutan masuk = 849,6436 lbm/jam

Rate larutan masuk =  $849,6436 \text{ lbm/jam} \times 2 \text{ jam}$

= 1699,2872 lbm

$\rho \text{ HCl } 5\% = 1,0071 \text{ gr/cm}^3$

$\rho \text{ CaCl}_2 = 1,0034 \text{ gr/cm}^3$

$\rho \text{ H}_2\text{O} = 0,98807 \text{ gr/cm}^3$

$\rho \text{ MgCl}_2 = 1,00395 \text{ gr/cm}^3$

$\rho \text{ H}_3\text{PO}_4 = 0,9993 \text{ gr/cm}^3$

$\rho \text{ NaCl} = 0,99494 \text{ gr/cm}^3$

$\rho \text{ gelatin} = 0,9884 \text{ gr/cm}^3$

(Ward, A.G)

$X_{\text{kolagen}} = 0,2992$

$X_{\text{HCl } 5\%} = 0,00159$

$X_{\text{H}_2\text{O}} = 0,6881$

$X_{\text{CaCl}_2} = 4,978 \cdot 10^{-4}$

$$X_{H_3PO_4} = 7,784 \cdot 10^{-7}$$

$$X_{MgCl_2} = 3,243 \cdot 10^{-8}$$

$$X_{NaCl} = 6,487 \cdot 10^{-8}$$

$$X_{gelatin} = 0,0106$$

$$\frac{1}{\rho_{campuran}} = \sum \frac{x_i}{\rho_i}$$

$$\frac{1}{\rho_{campuran}} =$$

$$\frac{0,00159}{1,0071} + \frac{0,6881}{0,98807} + \frac{0,0106}{0,9884} + \frac{4,978 \cdot 10^{-4}}{1,0034} + \frac{3,243 \cdot 10^{-8}}{1,00395} + \frac{6,487 \cdot 10^{-8}}{0,99494} \\ + \frac{7,784 \cdot 10^{-7}}{0,9993} + \frac{0,2992}{0,9451}$$

$$\rho_{campuran} = 0,9749 \text{ gr/cm}^3 = 60,863 \text{ lbm/ft}^3$$

$$\text{Volume larutan total per hari} = \frac{1699,2872}{60,863} = 27,9199 \text{ ft}^3$$

$$C_{pHCl 5\%} = 0,966865 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{pCaCl_2} = 0,2162 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{pKolagen} = 0,3369 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{pMgCl_2} = 0,2520 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{pgelatin} = 0,3613 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{pNaCl} = 0,2737 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{pH_3PO_4} = 0,4520 \text{ kkal/kg } ^\circ\text{C}$$

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖ Q = panas yang disuplai = 130116,2179 kkal/hari = 6301,1496 W = 21500,4934 Btu/hr
- ❖ Suhu bahan masuk tangki pada 47,5<sup>0</sup>C yaitu larutan kolagen, gelatin, HCl 5% yang ditambahkan dan H<sub>2</sub>O yang ditambahkan.
- ❖ Suhu kolagen dan impurities keluar 50<sup>0</sup>C
- ❖ Pemanas berupa steam dengan suhu masuk 120<sup>0</sup>C dan suhu keluar 120<sup>0</sup>C
- ❖ Massa steam = 247,1712 kg/hari = 2,86.10<sup>-3</sup> kg/s = 22,7047 lbm/jam

Dengan cara yang sama ~~pada~~ perhitungan tangki ekstraksi I (F-210), maka didapat:



**Spesifikasi :***Tangki :*

Kapasitas	: 27,9199 cuft
Diameter	: 2,9 ft
Tinggi tutup atas (dished head)	: 0,7 ft
Tinggi tutup bawah (konis)	: 2,1 ft
Tinggi shell	: 4,35 ft
Tinggi tangki total	: 7,15 ft
Tebal shell	: 3/16 in
Tebal tutup atas (dished head)	: 3/16 in
Tebal tutup bawah (dished head)	: 3/16 in
Bahan konstruksi	: High alloy stell SA-240 grade M, type316
Jumlah tangki	: 2 buah

*Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,264 m = 0,866 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 0,5 Hp
Jumlah pengaduk	: 1 buah

*Coil pemanas*

Diameter dalam pipa	: 0,269 in
Spasi coil	: 2 in
Tinggi coil	: 0,635 ft
Diameter luar pipa	: 0,405 in

**20. TANGKI EKSTRAKSI IV (F-240)**

Fungsi : Mengekstrak gelatin dari larutan kolagen.

Tipe : silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berbentuk konis serta dilengkapi dengan pengaduk dan coil pemanas.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk  
liquida

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Kapasitas =  $9249,6098 \text{ kg/hari} \times 2,2046 = 20391,68977 \text{ lbm/hari}$

Perhitungan :

Waktu tinggal = 120 menit = 2 jam

Rate larutan masuk =  $849,6436 \text{ lbm/jam} \times 2 \text{ jam}$

= 1699,2872 lbm

$\rho_{\text{HCl } 5\%} = 1,0071 \text{ gr/cm}^3$

$\rho_{\text{CaCl}_2} = 1,0034 \text{ gr/cm}^3$

$\rho_{\text{H}_2\text{O}} = 0,98807 \text{ gr/cm}^3$

$\rho_{\text{MgCl}_2} = 1,00395 \text{ gr/cm}^3$

$\rho_{\text{H}_3\text{PO}_4} = 0,9993 \text{ gr/cm}^3$

$\rho_{\text{NaCl}} = 0,99494 \text{ gr/cm}^3$

$\rho_{\text{gelatin}} = 0,9884 \text{ gr/cm}^3$

(Ward, A.G)

$X_{\text{kolagen}} = 0,2633$

$X_{\text{HCl } 5\%} = 0,00159$

$X_{\text{H}_2\text{O}} = 0,6881$

$X_{\text{CaCl}_2} = 4,978 \cdot 10^{-4}$

$X_{\text{H}_3\text{PO}_4} = 7,784 \cdot 10^{-7}$

$X_{\text{MgCl}_2} = 3,243 \cdot 10^{-8}$

$X_{\text{NaCl}} = 6,487 \cdot 10^{-8}$

$X_{\text{gelatin}} = 0,0465$

$$\frac{1}{\rho_{\text{campuran}}} = \sum \frac{x_i}{\rho_i}$$

$$\frac{1}{\rho_{\text{campuran}}}$$

$$\frac{0,00159}{1,0071} + \frac{0,6881}{0,98807} + \frac{0,0465}{0,9884} + \frac{4,978 \cdot 10^{-4}}{1,0034} + \frac{3,243 \cdot 10^{-8}}{1,00395} + \frac{6,487 \cdot 10^{-8}}{0,99494} + \frac{7,784 \cdot 10^{-7}}{0,9993} + \frac{0,2633}{0,9451}$$

$$\rho_{\text{campuran}} = 0,9764 \text{ gr/cm}^3 = 60,96 \text{ lbm/ft}^3$$

$$\text{Volume larutan total per hari} = \frac{1699,2872}{60,96} = 27,8754 \text{ ft}^3$$

$\text{Cp}_{\text{HCl } 5\%} = 0,966865 \text{ kkal/kg } ^{\circ}\text{C}$

$\text{Cp}_{\text{CaCl}_2} = 0,2162 \text{ kkal/kg } ^{\circ}\text{C}$

$\text{Cp}_{\text{Kolagen}} = 0,3369 \text{ kkal/kg } ^{\circ}\text{C}$

$\text{Cp}_{\text{MgCl}_2} = 0,2520 \text{ kkal/kg } ^{\circ}\text{C}$

$\text{Cp}_{\text{gelatin}} = 0,3613 \text{ kkal/kg } ^{\circ}\text{C}$

$\text{Cp}_{\text{NaCl}} = 0,2737 \text{ kkal/kg } ^{\circ}\text{C}$

$\text{Cp}_{\text{H}_3\text{PO}_4} = 0,4520 \text{ kkal/kg } ^{\circ}\text{C}$

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖  $Q$  = panas yang disuplai = 42105,86345 kkal/hari = 2039,064 W = 6957,6 Btu/hr
- ❖ Suhu bahan masuk tangki pada 45<sup>0</sup>C yaitu larutan kolagen, impuritis dan gelatin
- ❖ Suhu kolagen, gelatin dan impurities keluar 50<sup>0</sup>C
- ❖ Pemanas berupa steam dengan suhu masuk 120<sup>0</sup>C dan suhu keluar 120<sup>0</sup>C
- ❖ Massa steam = 79,9960 kg/hari =  $9,3 \cdot 10^{-4}$  kg/s = 7,35 lbm/jam

Dengan cara yang sama pada perhitungan tangki ekstraksi I (F-210), maka didapat:

**Spesifikasi :**

*Tangki :*

Kapasitas	: 27,8754 cuft
Diameter	: 2,9 ft
Tinggi tutup atas (dished head)	: 0,7 ft
Tinggi tutup bawah (konis)	: 2,1 ft
Tinggi shell	: 4,35 ft
Tinggi tangki total	: 7,15 ft
Tebal shell	: 3/16 in
Tebal tutup atas (dished head)	: 3/16 in
Tebal tutup bawah (dished head)	: 3/16 in
Bahan konstruksi	: High alloy stell SA-240 grade M type 316
Jumlah tangki	: 2 buah

*Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,264 m = 0,866 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 0,5 Hp
Jumlah pengaduk	: 1 buah

*Coil pemanas*

Diameter dalam pipa	: 0,269 in
Diameter luar pipa	: 0,405 in
Spasi coil	: 2 in
Tinggi coil	: 0,2342 ft

## 21. Tangki Ekstraksi V (F-250)

Fungsi : Mengekstrak gelatin dari larutan kolagen.

Tipe : silinder tegak dengan tutup atas berbentuk dished head dan tutup bawah berbentuk konis serta dilengkapi dengan pengaduk dan coil pemanas.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk liquida

Kondisi operasi :  $T = 50^{\circ}\text{C}$

Kapasitas =  $7449,975 \text{ kg/hari} \times 2,2046 = 16424,215 \text{ lbm/hari}$

Perhitungan :

Waktu tinggal = 240 menit = 4 jam

Rate larutan masuk =  $684,3423 \text{ lbm/jam} \times 4 \text{ jam}$   
 $= 2737,3692 \text{ lbm}$

$\rho \text{ HCl } 5\% = 1,0071 \text{ gr/cm}^3$

$\rho \text{ CaCl}_2 = 1,0034 \text{ gr/cm}^3$

$\rho \text{ H}_2\text{O} = 0,98807 \text{ gr/cm}^3$

$\rho \text{ MgCl}_2 = 1,00395 \text{ gr/cm}^3$

$\rho \text{ H}_3\text{PO}_4 = 0,9993 \text{ gr/cm}^3$

$\rho \text{ NaCl} = 0,99494 \text{ gr/cm}^3$

$\rho \text{ gelatin} = 0,9884 \text{ gr/cm}^3$

(Ward, A.G)

$X_{\text{kolagen}} = 0,30$

$X_{\text{HCl } 5\%} = 8,05 \cdot 10^{-4}$

$X_{\text{H}_2\text{O}} = 0,6917$

$X_{\text{CaCl}_2} = 4,94 \cdot 10^{-5}$

$X_{\text{H}_3\text{PO}_4} = 7,79 \cdot 10^{-8}$

$X_{\text{MgCl}_2} = 2,68 \cdot 10^{-9}$

$X_{\text{NaCl}} = 6,711 \cdot 10^{-9}$

$X_{\text{gelatin}} = 6,7 \cdot 10^{-3}$

$$\frac{1}{\rho_{\text{campuran}}} = \sum \frac{x_i}{\rho_i}$$

$$\frac{1}{\rho_{\text{campuran}}} = \frac{8,05 \cdot 10^{-4}}{1,0071} + \frac{0,6917}{0,98807} + \frac{6,7 \cdot 10^{-3}}{0,9884} + \frac{4,94 \cdot 10^{-3}}{1,0034} + \frac{2,68 \cdot 10^{-9}}{1,00395} + \frac{6,711 \cdot 10^{-9}}{0,99494} + \frac{7,79 \cdot 10^{-8}}{0,9993} + \frac{0,3}{0,9451}$$

$$\rho_{\text{campuran}} = 0,9755 \text{ gr/cm}^3 = 60,90 \text{ lbm/ft}^3$$

$$\text{Volume larutan total} = \frac{2737,3692}{60,90} = 44,95 \text{ ft}^3$$

$$C_{p\text{HCl } 5\%} = 0,966865 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{p\text{CaCl}_2} = 0,2162 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{p\text{Kolagen}} = 0,3369 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{p\text{MgCl}_2} = 0,2520 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{p\text{gelatin}} = 0,3613 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{p\text{NaCl}} = 0,2737 \text{ kkal/kg } ^\circ\text{C}$$

$$C_{p\text{H}_3\text{PO}_4} = 0,4520 \text{ kkal/kg } ^\circ\text{C}$$

Dari perhitungan neraca panas didapatkan data sebagai berikut :

- ❖ Q = panas yang disuplai = 104960,4867 kkal/hari = 5082,931 W = 17343,744 Btu/jam
- ❖ Suhu bahan masuk tangki pada 47,5<sup>0</sup>C yaitu larutan kolagen, impuritis, gelatin, HCl 5% yang ditambahkan dan H<sub>2</sub>O yang ditambahkan.
- ❖ Suhu kolagen, gelatin dan impurities keluar 50<sup>0</sup>C
- ❖ Pemanas berupa steam dengan suhu masuk 120<sup>0</sup>C dan suhu keluar 120<sup>0</sup>C
- ❖ Massa steam = 199,3849 kg/hari = 2,31.10<sup>-3</sup> kg/s = 18,3152 lbm/jam

Dengan cara yang sama pada perhitungan tangki ekstraksi I (F-210), maka didapat:

#### Spesifikasi :

##### Tangki :

Kapasitas	: 44,95 cuft
Diameter	: 3,4 ft
Tinggi tutup atas (dished head)	: 0,8 ft
Tinggi tutup bawah (konis)	: 2,6 ft
Tinggi shell	: 5,1 ft

Tinggi tangki total	: 8,5 ft
Tebal shell	: 3/16 in
Tebal tutup atas (dished head)	: 3/16 in
Tebal tutup bawah (dished head)	: 3/16 in
Bahan konstruksi	: High alloy stell SA-240 grade M, type 316
Jumlah tangki	: 4 buah

#### *Pengaduk*

Tipe	: flat six-blade turbine agitator with disk
Diameter	: 0,311 m = 1,02 ft
Kecepatan pengaduk	: 50 rpm
Power motor	: 0,5 Hp
Jumlah pengaduk	: 1 buah

#### *Coil Pemanas*

Diameter dalam pipa	: 0,269 in
Spasi coil	: 2 in
Diameter luar pipa	: 0,405 in
Tinggi coil	: 0,4346 ft

## 22. Rotary Vacuum Filter V (II-241)

Fungsi : Memisahkan kolagen dari larutan gelatin hasil ekstraksi pada tangki IV

Tipe : Rotary vacuum drum filter

Dasar pemilihan : cocok digunakan untuk pemisahan dalam kondisi kontinyu

Kondisi operasi :  $T = 45^{\circ}\text{C}$

Rate Bahan masuk = 9249,6098 kg/hari

Fraksi masa bahan masuk rotary vacuum filter :

$$X_{\text{Kolagen}} = \frac{2240,5111}{9249,6098} = 0,2422$$

$$X_{\text{MgCl}_2} = \frac{0,0003}{9249,6098} = 3,24 \cdot 10^{-8}$$

$$X_{\text{Air}} = \frac{6365,0885}{9249,6098} = 0,6881$$

$$X_{\text{NaCl}} = \frac{0,0006}{9249,6098} = 6,5 \cdot 10^{-8}$$

$$X_{\text{CaCl}_2} = \frac{4,0647}{9249,6098} = 4,39 \cdot 10^{-4}$$

$$X_{\text{H}_3\text{PO}_4} = \frac{0,0072}{9249,6098} = 7,78 \cdot 10^{-7}$$

$$X_{\text{gelatin}} = \frac{624,7195}{9249,6098} = 0,0675$$

$$X_{\text{HCl 5\%}} = \frac{14,6779}{9249,6098} = 1,59 \cdot 10^{-3}$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3$$

(Ward, 1977)

$$\rho_{\text{HCl 5\%}} = 1,0071 \text{ gr/cm}^3$$

$$\rho_{\text{CaCl}_2} = 1,0034 \text{ gr/cm}^3$$

$$\rho_{\text{H}_2\text{O}} = 0,98807 \text{ gr/cm}^3$$

$$\rho_{\text{MgCl}_2} = 1,00395 \text{ gr/cm}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 0,9993 \text{ gr/cm}^3$$

$$\rho_{\text{NaCl}} = 0,99494 \text{ gr/cm}^3$$

$$\rho_{\text{gelatin}} = 0,9884 \text{ gr/cm}^3$$

(Ward, A.G)

$$\frac{1}{\rho_{\text{campuran}}} = \sum \frac{x_i}{\rho_i}$$

$$\begin{aligned} \frac{1}{\rho_{\text{campuran}}} &= \frac{0,00159}{1,0071} + \frac{0,6881}{0,98807} + \frac{0,0675}{0,9884} + \frac{4,39 \cdot 10^{-4}}{1,0034} + \frac{3,24 \cdot 10^{-8}}{1,00395} + \frac{6,5 \cdot 10^{-8}}{0,99494} \\ &+ \frac{7,78 \cdot 10^{-7}}{0,9993} + \frac{0,2422}{0,9451} \end{aligned}$$

$$\rho_{\text{campuran}} = 0,9881 \text{ gr/cm}^3 = 61,69 \text{ lbm/ft}^3$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p}$$

(Geankoplis 2 ed., 1985)

$$\varphi_p = \frac{1}{10^{1,82(1-\varepsilon)}}$$

c—

$$\begin{aligned} &\frac{6365,0885}{988,07} + \frac{2240,5111}{945,1} + \frac{624,7195}{988,4} + \frac{4,067}{1003,4} + \frac{0,0003}{1003,95} + \frac{0,0006}{994,94} + \frac{0,0072}{999,3} + \frac{14,6779}{1007,1} \\ c &= 0,804 \end{aligned}$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,804)}}$$

$$= 0,4398$$

$$\mu_{\text{air}} \text{ pada } 50^\circ\text{C} = 0,5494 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,5494 \cdot 10^{-3}}{0,4398}$$

$$= 1,25 \cdot 10^{-3} \text{ kg/m.s}$$

$$\text{padatan} = 2240,5111 \text{ kg/hari}$$

$$\text{liquida} = 7009,0987 \text{ kg/hari}$$

$$\text{rate volume padatan} = \frac{2240,5111 \text{ kg / hari}}{945,1 \text{ kg / m}^3} = 2,3707 \text{ m}^3 / \text{hari}$$

$$\begin{aligned} \text{Vol liquida} &= \frac{6365,0885}{998,07} + \frac{624,7195}{988,4} + \frac{4,067}{1003,4} + \frac{0,0003}{1003,95} + \frac{0,0006}{994,94} + \frac{0,0072}{999,3} \\ &\quad + \frac{14,7669}{1007,1} \\ &= 7,0282 \text{ m}^3 / \text{hari} \end{aligned}$$

$$\text{rate volume slurry} = (7,0282 + 2,3707) \text{ m}^3 / \text{hari} = 9,3989 \text{ m}^3 / \text{hari}$$

Dengan cara yang sama pada perhitungan Rotary vaccum filter (H-144), maka didapat :

**Spesifikasi :**

Tipe	: Rotary vacuum drum filter
Filter rate	: 0,1071 kg/s
Diameter	: 0,1882 m
Jumlah	: 1 buah

### 23. Rotary Vacuum Filter VI (H-251)

Fungsi : Memisahkan kolagen dari larutan gelatin hasil ekstraksi pada tangki V.

Tipe : Rotary vacuum drum filter

Dasar pemilihan : cocok digunakan untuk pemisahan dalam kondisi kontinyu

Kondisi operasi :  $T = 45^{\circ}\text{C}$

Rate Bahan masuk = 7449,97500 kg/hari

Fraksi masa bahan masuk rotary vacuum filter :



$$X_{\text{Kolagen}} = \frac{1926,8395}{7449,975} = 0,2586$$

$$X_{\text{MgCl}_2} = \frac{0,00002}{7449,975} = 2,68 \cdot 10^{-9}$$

$$X_{\text{Air}} = \frac{5153,1755}{7449,975} = 0,6917$$

$$X_{\text{NaCl}} = \frac{0,00005}{7449,975} = 6,71 \cdot 10^{-9}$$

$$X_{\text{CaCl}_2} = \frac{0,36798}{7449,975} = 4,94 \cdot 10^{-5}$$

$$X_{\text{H}_3\text{PO}_4} = \frac{0,00058}{7449,975} = 7,78 \cdot 10^{-8}$$

$$X_{\text{gelatin}} = \frac{363,59567}{7449,975} = 0,0488$$

$$X_{\text{HCl 5\%}} = \frac{5,99570}{7449,975} = 8,048 \cdot 10^{-4}$$

$$\rho_{\text{kolagen}} = 59 \text{ lb/cuft} = 945,1 \text{ kg/m}^3 \quad (\text{Ward, 1977})$$

$$\rho_{\text{HCl 5\%}} = 1,0071 \text{ gr/cm}^3$$

$$\rho_{\text{CaCl}_2} = 1,0034 \text{ gr/cm}^3$$

$$\rho_{\text{H}_2\text{O}} = 0,98807 \text{ gr/cm}^3$$

$$\rho_{\text{MgCl}_2} = 1,00395 \text{ gr/cm}^3$$

$$\rho_{\text{H}_3\text{PO}_4} = 0,9993 \text{ gr/cm}^3$$

$$\rho_{\text{NaCl}} = 0,99494 \text{ gr/cm}^3$$

$$\rho_{\text{gelatin}} = 0,9884 \text{ gr/cm}^3$$

(Ward, A.G)

$$\frac{1}{\rho_{\text{campuran}}} = \sum \frac{x_i}{\rho_i}$$

$$\frac{1}{\rho_{\text{campuran}}} =$$

$$\frac{8,048 \cdot 10^{-4}}{1,0071} + \frac{0,6917}{0,98807} + \frac{0,0488}{0,9884} + \frac{4,94 \cdot 10^{-5}}{1,0034} + \frac{2,68 \cdot 10^{-9}}{1,00395} + \frac{6,71 \cdot 10^{-9}}{0,99494} \\ + \frac{7,78 \cdot 10^{-8}}{0,9993} + \frac{0,2586}{0,9451}$$

$$\rho_{\text{campuran}} = 0,9901 \text{ gr/cm}^3 = 61,81 \text{ lbm/ft}^3$$

$$\mu_{\text{campuran}} = \frac{\mu}{\varphi_p}$$

(Geankoplis 2 ed., 1985)

$$\varphi_p = \frac{1}{10^{1,82(1-c)}}$$

c=

$$c = \frac{\frac{5153,1755}{988,07} + \frac{1926,8395}{945,1} + \frac{363,59567}{988,4} + \frac{0,36798}{1003,4} + \frac{0,00002}{1003,95} + \frac{0,00005}{994,94} + \frac{0,00058}{999,3} + \frac{5,9957}{1007,1}}{0,756}$$

$$\varphi_p = \frac{1}{10^{1,82(1-0,756)}}$$

$$= 0,4441$$

$$\mu_{\text{air}} \text{ pada } 50^\circ\text{C} = 0,5494 \cdot 10^{-3} \text{ kg/m.s}$$

$$\mu_{\text{campuran}} = \frac{0,5494 \cdot 10^{-3}}{0,4441}$$

$$= 1,24 \cdot 10^{-3} \text{ kg/m.s}$$

$$\text{padatan} = 1926,8395 \text{ kg/hari}$$

$$\text{liquida} = 5523,1355 \text{ kg/hari}$$

$$\text{rate volume padatan} = \frac{1926,8395 \text{ kg / hari}}{945,1 \text{ kg / m}^3} = 2,039 \text{ m}^3 / \text{hari}$$

$$\text{Vol} \quad \text{liquida} \quad =$$

$$\frac{5153,1755}{998,07} + \frac{363,59567}{988,4} + \frac{0,36798}{1003,4} + \frac{0,00002}{1003,95} + \frac{0,00005}{994,94} + \frac{0,00058}{999,3}$$

$$+ \frac{5,9957}{1007,1}$$

$$= 6,8564 \text{ m}^3/\text{hari}$$

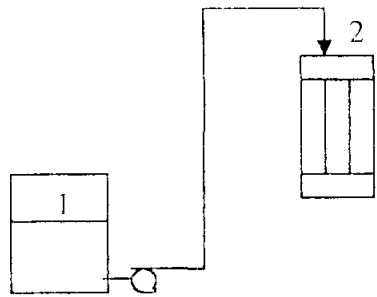
$$\text{rate volume slurry} = (6,8564 + 2,039) \text{ m}^3/\text{hari} = 8,8954 \text{ m}^3/\text{hari}$$

Dengan cara yang sama pada perhitungan Rotary vaccum filter (H-144), maka didapat :

#### Spesifikasi :

Tipe	: Rotary vacuum drum filter
Filter rate	: 0,086 kg/s
Diameter	: 0,3828 m
Jumlah	: 1 buah

24. POMPA (L-254)



Fungsi : untuk mengalirkan larutan gelatin 11,02% menuju evaporator  
Kebutuhan total larutan gelatin = 23023,1719 kg/ hari = 2114,903 lb/jam  
Pada P = 1 atm dan pada suhu 44°C :

$\rho \text{ HCl } 5\% = 1,0071 \text{ gr/cm}^3$   
 $\rho \text{ H}_2\text{O} = 0,98807 \text{ gr/cm}^3$   
 $\rho \text{ H}_3\text{PO}_4 = 0,9993 \text{ gr/cm}^3$   
 $\rho \text{ gelatin} = 0,9884 \text{ gr/cm}^3$

$\rho \text{ CaCl}_2 = 1,0034 \text{ gr/cm}^3$   
 $\rho \text{ MgCl}_2 = 1,00395 \text{ gr/cm}^3$   
 $\rho \text{ NaCl} = 0,99494 \text{ gr/cm}^3$   
(Ward, A.G)

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,0039}{1,0071} + \frac{0,8824}{0,98807} + \frac{0,1102}{0,9884} + \frac{0,0035}{1,0034} + \frac{0,0056}{1,00395} + \frac{4,86 \cdot 10^{-7}}{0,99494}$$
$$+ \frac{5,51 \cdot 10^{-6}}{0,9993}$$
$$= 1,0176$$
$$\rho \text{ campuran} = 0,9827 \text{ gr/cm}^3 = 61,35 \text{ lb/ft}^3$$
$$\text{volume} = \frac{2114,903}{61,35} = 34,4727 \text{ ft}^3/\text{jam} = 9,576 \cdot 10^{-3} \text{ ft}^3/\text{s} = 4,3 \text{ gpm}$$
$$\mu \text{ gelatin pada } 44^\circ\text{C} = 0,1 \text{ poises} = 6,7197 \cdot 10^{-3} \text{ lb/ft} \cdot \text{s} \quad (\text{Ward,A.G})$$

Dengan cara yang sama pada perhitungan pompa (L-145), maka didapat :

Spesifikasi :

Tipe	: centrifugal pump
Rate aliran pompa	: $9,576 \cdot 10^{-3} \text{ ft}^3/\text{s}$
Power motor	: 0,1 Hp
Bahan konstruksi	: stainless steel
Jumlah	: 1 buah

## 25. TANKI PENAMPUNG HASIL EKSTRAKSI (F-253)

Fungsi : untuk menyimpan larutan gelatin hasil ekstraksi

Bentuk : tanki silinder vertikal dengan tutup atas dished head dan tutup bawah berbentuk konis.

Dasar pemilihan : cocok digunakan untuk menampung bahan yang berbentuk liquida

Kondisi operasi :  $T = 50^{\circ} \text{C}$

$$P = 14,7 \text{ psia} = 1 \text{ atm} \quad (\text{Perry } 7^{\text{ed}}, \text{ hal 2-88})$$

Kebutuhan larutan gelatin = 23023,1719 kg/ hari = 2114,903 lb/jam

Pada  $P = 1 \text{ atm}$  :

$$\rho \text{ HCl } 5\% = 1,0071 \text{ gr/cm}^3$$

$$\rho \text{ CaCl}_2 = 1,0034 \text{ gr/cm}^3$$

$$\rho \text{ H}_2\text{O} = 0,98807 \text{ gr/cm}^3$$

$$\rho \text{ MgCl}_2 = 1,00395 \text{ gr/cm}^3$$

$$\rho \text{ H}_3\text{PO}_4 = 0,9993 \text{ gr/cm}^3$$

$$\rho \text{ NaCl} = 0,99494 \text{ gr/cm}^3$$

$$\rho \text{ gelatin} = 0,9884 \text{ gr/cm}^3$$

(Ward, A.G)

$$\begin{aligned} \frac{1}{\rho_{\text{campuran}}} &= \frac{0,0039}{1,0071} + \frac{0,8824}{0,98807} + \frac{0,1102}{0,9884} + \frac{0,0035}{1,0034} + \frac{0,0056}{1,00395} + \frac{4,86 \cdot 10^{-7}}{0,99494} \\ &\quad + \frac{5,51 \cdot 10^{-6}}{0,9993} \\ &= 1,0176 \end{aligned}$$

$$\rho_{\text{campuran}} = 0,9827 \text{ gr/cm}^3 = 61,35 \text{ lb/ft}^3$$

$$\text{Rate volumetrik} = \frac{2114,903}{61,35} = 34,4727 \text{ ft}^3/\text{jam}$$

Dirancang waktu tinggal = 6 jam

$$\text{Volume larutan gelatin} = 34,4727 \text{ cuft/jam} \times 6 \text{ jam} = 206,8362 \text{ cuft}$$

Diambil : tinggi shell ( $H_s$ ) = 1,5 . Diameter shell ( $D$ )

$$\text{Volume shell} = (\pi/4) \cdot D^2 \cdot H_s = (\pi/4) \cdot D^2 \cdot 1,5 \cdot D = 1,5 \cdot (\pi/4) \cdot D^3$$

$$\text{Volume torispherical dished head (cuft)} = 0,000049 \times D^3 \text{ (inch)}$$

(Brownell & Young, pers.5.11,p.88)

$$\text{Volume larutan dalam konis} = \text{volume konis}$$

$$\begin{aligned}
 &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot H_t \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D_n^2 \cdot H_n \right) \\
 &= \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D^2 \cdot \frac{D}{2 \cdot \tan \alpha} \right) - \left( \frac{1}{3} \cdot \frac{\pi}{4} \cdot D_n^2 \cdot \frac{D_n}{2 \cdot \tan \alpha} \right) \\
 &= \left( \frac{\pi \cdot D^3}{24 \cdot \tan \alpha} \right) - \left( \frac{\pi \cdot D_n^3}{24 \cdot \tan \alpha} \right) \\
 &= \frac{\pi}{24 \cdot \tan \alpha} (D^3 - D_n^3)
 \end{aligned}$$

dimana :

$D_n$  = diameter lubang pengeluaran liquida = 0,5 ft (=6 in)

$D$  = diameter konis bagian atas = diameter shell

$H_t$  = tinggi konis

$H_n$  = tinggi konis terpancung

Volume tangki penampung = vol.shell + vol.dished head + vol.konis

Diambil : volume tangki penampung = 1,2.volume larutan total

$$1,2 \cdot 206,8362 \text{ cuft} = \frac{\pi}{4} \cdot 1,5 \cdot D^3 \cdot \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 + (0,000049 \cdot D^3)$$

$$+ \frac{\pi}{24 \cdot \tan \alpha} \left( D^3 \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 - D_n^3 \right)$$

$$248,2034 \text{ cuft} = \frac{\pi}{4} \cdot 1,5 \cdot D^3 \cdot \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 + (0,000049 \cdot D^3)$$

$$+ \frac{\pi}{24 \cdot \tan \alpha} \left( D^3 \left( \frac{1 \text{ ft}}{12 \text{ in}} \right)^3 - (0,5 \text{ ft})^3 \right)$$

$$248,2034 \text{ cuft} = 6,8177 \cdot 10^{-4} \cdot D^3 + 0,000049 \cdot D^3 + 1,3121 \cdot 10^{-4} \cdot D^3 - 0,0283$$

$$248,2034 \text{ cuft} - 8,6198 \cdot 10^{-4} \cdot D^3$$

$$D = 66,03 \text{ in} = 5,5 \text{ ft}$$

$$H_s = 1,5 \cdot D = 1,5 \cdot (5,5 \text{ ft}) = 8,25 \text{ ft}$$

$$\text{Volume larutan dalam konis} = \frac{\pi}{24 \cdot \tan \alpha} (D^3 - (0,5 \text{ ft})^3)$$

$$= \frac{\pi}{24.1g30''} ((5,5\text{ ft})^3 - (0,5\text{ ft})^3)$$

$$= 37,69 \text{ cuft}$$

Volume larutan dalam shell = vol. larutan total – vol. larutan dalam konis

$$= (206,8362 - 37,69) \text{ cuft}$$

$$= 169,1462 \text{ cuft}$$

Tinggi larutan dalam shell (H) =  $\frac{\text{volume larutan dalam shell}}{\frac{\pi}{4} \cdot D^2}$

$$= \frac{169,1462}{\frac{\pi}{4} \cdot (5,5\text{ ft})^2}$$

$$= 7,12 \text{ ft} \approx 7,2 \text{ ft}$$

Tinggi larutan dalam konis = tinggi konis (Hc)

$$Hc = \frac{D}{2.1g\alpha} - \frac{Dn}{2.1g\alpha}$$

$$= \frac{5,5\text{ ft}}{2.1g30''} - \frac{0,5\text{ ft}}{2.1g30''}$$

$$= 4,33 \text{ ft} \approx 4,34 \text{ ft}$$

Tinggi larutan dalam tangki = H + Hc

$$= (7,2 + 4,34) \text{ ft}$$

$$= 11,54 \text{ ft}$$

$$P_{\text{operasi}} = P_{\text{hidrostatik}} = \left( \frac{\rho \cdot H}{144} \right) \text{ psi}$$

$$= \left( \frac{61,35 \text{ lbm/cuft} \cdot 11,54 \text{ ft}}{144} \right) \text{ psi}$$

$$= 4,92 \text{ psi}$$

$$P_{\text{desain}} = 1,5 \cdot P_{\text{operasi}} = 1,5 \cdot (4,92 \text{ psi}) = 7,38 \text{ psi}$$

Tebal shell

$$t_s = \frac{P.ID}{2(f.E - 0,6.P)} + c \quad (\text{Brownell \& Young, pers. 13.1})$$

dimana :

$$P = P_{\text{desain}} = 7,38 \text{ psi}$$

$$ID = 5,5 \text{ ft} = 66 \text{ in}$$

Konstruksi : bahan konstruksi High Alloy stell SA-240 grade M, type 316

$$f = \text{stress maksimum yang diijinkan} = 18750 \text{ psi}$$

type sambungan = *double-welded butt joint*, dengan

$$E = \text{welded-joint efficiency} = 0,8$$

(Brownell & Young, hal 343)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_s = \frac{(7,38 \text{ psi}).(66 \text{ in})}{2.(18750.0,8 - 0,6.7,38) \text{ psi}} + \frac{1}{8} \text{ in}$$

$$t_s = 0,1412 \text{ in} \approx \frac{1}{4} \text{ in}$$

Tebal dished head

$$t_s = \frac{1}{4} \text{ in}$$

$$OD = ID + 2.t_s$$

$$= 66 \text{ in} + (2. \frac{1}{4}) \text{ in}$$

$$= 66,5 \text{ in}$$

Dari table 5.7 Brownell & Young diperoleh :

$$OD \text{ standar} = 72 \text{ in}$$

$$r \text{ (crown radius / radius of dish)} = 72 \text{ in}$$

$$icr \text{ (inside corner radius / knuckle radius)} = 4 \frac{3}{8} \text{ in}$$

$$W = \frac{1}{4} \left( 3 + \sqrt{\frac{r}{icr}} \right) \quad (\text{Brownell \& Young, pers. 7.76, p.138})$$

$$= \frac{1}{4} \left( 3 + \sqrt{\frac{72}{4 \frac{3}{8}}} \right)$$

$$= 1,76$$

$$a = ID/2 = 66 \text{ in} / 2 = 33 \text{ in}$$

$$AB = ID/2 - icr = (33 - 4 \frac{3}{8}) \text{ in} = 28,625 \text{ in}$$

$$BC = r - icr = (72 - 4 \frac{3}{8}) \text{ in} = 67,625 \text{ in}$$

$$b = r - \sqrt{BC^2 - AB^2} = 72 - \sqrt{67,625^2 - 28,625^2} = 10,73 \text{ in}$$

$$t_d = \frac{P.rc.W}{2.f.E - 0,2.P} + c \quad (\text{Brownell \& Young, pers. 7.77, p.138})$$

dimana :

$$P = P_{\text{desain}} = 7,38 \text{ psi}$$

Konstruksi : bahan konstruksi High alloy stell SA-240 grade M, type 316

$$f = \text{stress maksimum yang diijinkan} = 18750 \text{ psi}$$

$$\text{tipe sambungan} = \text{double-welded butt joint, dengan}$$

$$E = \text{welded-joint efficiency} = 0,8$$

(Brownell & Young, hal 343)

$$c = \text{corrosion allowance} = \frac{1}{8} \text{ in}$$

$$t_d = \frac{(7,38 \text{ psi}).(72 \text{ in}).1,76}{2.(12650 \text{ psi}).(0,8) - 0,2.(7,38 \text{ psi})} + \frac{1}{8} \text{ in}$$

$$= 0,171 \text{ in} \approx \frac{1}{4} \text{ in}$$

$$\text{Dipilih panjang straight-flange (sf)} = 2 \text{ in}$$

(Brownell & Young, table 5.8, p.93)

$$OA = t + b + sf$$

$$= (1/4 + 10,73 + 2) \text{ in}$$

$$= 12,98 \text{ in} = 1,08 \text{ ft} \approx 1,1 \text{ ft}$$

$$\text{Tinggi tangki keseluruhan} = \text{tinggi shell} + \text{tinggi dish} + \text{tinggi konis}$$

$$= 7,2 \text{ ft} + 1,1 \text{ ft} + 4,34 \text{ ft}$$

$$= 12,64 \text{ ft}$$

Tebal konis

$$t_c = \frac{P.ID}{2.\cos\alpha.(f.E - 0,6.P)} + c \quad (\text{Brownell \& Young, pers.6.154, p.118})$$



$$= \frac{(7,38 \text{ psi}) \cdot (66 \text{ in})}{2 \cdot \cos 30^\circ \cdot (18750,0,8 - 0,6 \cdot 7,38) \text{ psi}} + \frac{1}{8} \text{ in}$$

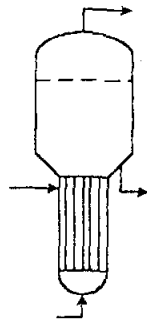
$$= 0,1438 \text{ in} \approx \frac{1}{4} \text{ in}$$

*Spesifikasi :*

Tangki :

Kapasitas	: 206,8362 cuft
Diameter	: 6,6 ft
Tinggi tutup atas (dished head)	: 1,1 ft
Tinggi tutup bawah (konis)	: 4,34 ft
Tinggi shell	: 7,2 ft
Tinggi tangki total	: 12,64 ft
Tebal shell	: $\frac{1}{4}$ in
Tebal tutup atas (dished head)	: $\frac{1}{4}$ in
Tebal tutup bawah (dished head)	: $\frac{1}{4}$ in
Bahan konstruksi	: High alloy steel SA-240 grade M, type 316
Jumlah tangki	: 1 buah

## 26. Evaporator (V-310)



Fungsi : untuk memekatkan konsentrasi larutan gelatin dari 11,02% hingga 35%.

Tipe : Single Effect Vertical Forced Circulation Evaporator.

Dasar pemilihan : - beroperasi secara kontinyu  
 - forced circulation supaya koefisien perpindahan panas besar

- single effect karena kapasitas kecil

Kondisi operasi :

➤ T larutan (fluida dingin) :  $t_1 = 44^\circ\text{C} = 111,2^\circ\text{F}$ ;  $t_2 = 53^\circ\text{C} = 127,4^\circ\text{F}$

➤ T steam (fluida panas) :  $T = 120^\circ\text{C} = 248^\circ\text{F}$

➤  $T_{\text{operasi}} = 53^\circ\text{C} = 127,4^\circ\text{F}$  dan  $P_{\text{operasi}} = 4 \text{ inHg} = 13,5457 \text{ kPa} = 0,13 \text{ atm}$

➤ Dari appendix B didapat : kebutuhan panas =  $9373367,067 \text{ kkal/hari}$   
 $= 1548845,817 \text{ Btu/jam}$

laju steam =  $17085,8245 \text{ kg/hari}$

$= 1569,4754 \text{ lb/jam}$

massa larutan =  $7249,0101 \text{ kg/hari}$

$= 665,882 \text{ lb/jam}$

Perhitungan

$$\Delta T = 248 - 127,4 = 120,6^\circ\text{F}$$

$$T_d \text{ larutan} = 53^\circ\text{C} = 127,4^\circ\text{F}$$

Dari Badger fig.5-26 hal 209, diperoleh :

$$U_{11} = 510 \text{ Btu/ft}^2 \cdot \text{jam} \cdot ^\circ\text{F}$$

$$Q = U_{11} \cdot A \cdot \Delta T \rightarrow A = 1548845,817 / (510 \cdot 120,6) = 25,182 \text{ ft}^2$$

Untuk evaporator jenis Vertical Forced Circulation, tube ditentukan :

$$\text{OD} < 2 \text{ in (Kern, hal 406)}$$

$$L = 48 \text{ in} = 4 \text{ ft (Badger, hal 208)}$$

Ditentukan ukuran tube pipa diameter nominal (Kern, tabel 11) :

$$1\frac{1}{2} \text{ in sch.40}$$

$$\text{OD} = 1,9 \text{ in}$$

$$\text{ID} = 1,610 \text{ in}$$

$$a_t = 0,498 \text{ ft}^2/\text{ft}$$

$$A = N_t \cdot a_t \cdot L \rightarrow N_t = 25,182 / (0,498 \cdot 4) = 12,64 \approx 13$$

$$\rho_l = 982,7 \text{ kg/m}^3$$

$$\rho_v = 0,6381 \text{ kg/m}^3$$

Dari steam table untuk superheated vapor pada  $T = 127,4^{\circ}\text{F}$  dan  $P = 13,5457 \text{ kPa}$ ,

diperoleh : spesifik volum =  $v = 281,9542 \text{ ft}^3/\text{lb}$

Laju volumetrik uap =  $1569,4754 \text{ lb/jam} \times 281,9542 \text{ ft}^3/\text{lb}$

$$= 450939,3614 \text{ ft}^3/\text{jam} = 3,5471 \text{ m}^3/\text{s}$$

Kecepatan uap maksimum =  $V_v = 0,035 \cdot (\rho_l/\rho_v)^{0,5} = 0,035 \cdot (982,7/0,6381)^{0,5}$

$$= 1,3735 \text{ m/s}$$

$$Q = V_v \cdot A_s \rightarrow A_s = 3,5471/1,3735 = 2,5825 \text{ m}^2 = \pi/4 \cdot D_s^2 \rightarrow D_s = 1,813 \text{ m} = 5,9481 \text{ ft}$$

#### Menentukan tinggi evaporator

Tinggi badan silinder = 1,5 – 2 dari panjang tube (Hugot, hal 500)

Ditentukan  $H_s = 2 \cdot L = 2 \cdot 4 = 8 \text{ ft}$

#### *Menghitung tebal shell*

Asumsi :  $t_s = 1/4 \text{ in}$

$$ID = 5,9481 \text{ ft} = 71,3772 \text{ in}$$

$$OD = ID + 2 \cdot t_s = 71,3772 + 2 \cdot 1/4 = 71,8772 \text{ in}$$

$$H_s/OD = (8 \times 12)/71,8772 = 1,3356$$

$$OD/t_s = 71,8772/(1/4) = 287,5088$$

Dari fig. 8.8 dan fig 8.7 brownell & young pada suhu  $127,4^{\circ}\text{F}$ , diperoleh :

$$A = 0,00022$$

$$B = 3900 \text{ psi}$$

$$P \text{ allowable} = \frac{4 \cdot B}{3 \left( \frac{OD}{t_s} \right)} = \frac{4 \cdot 3900}{3 \cdot 287,5088} = 18,0864 \text{ psi} > 14,7 \text{ psi (memenuhi)}$$

$$t_s' = t_s + c = 1/4 + 1/8 = 0,375 \text{ in}$$

Digunakan tebal plate =  $3/8 \text{ in}$

#### *Spesifikasi :*

Fungsi : untuk memekatkan konsentrasi larutan gelatin hingga 35%

Tipe : Vertical Forced Circulation Evaporator

Kondisi operasi : -  $T_{\text{larutan}}$  (fluida dingin)

$$t_1 = 44^{\circ}\text{C} = 111,2^{\circ}\text{F}$$

$$t_2 = 53^{\circ}\text{C} = 127,4^{\circ}\text{F}$$

-  $T_{\text{steam}}$  (fluida panas)

$$T = 120^{\circ}\text{C} = 248^{\circ}\text{F}$$

Luas perpindahan panas =  $24,6235 \text{ ft}^2$

Jumlah = 1 buah

## 27. Barometric Kondenser (E-311)

Fungsi : mengembunkan uap air dari evaporator

Tipe : Counter Current Condenser

Dasar Pemilihan : operasinya mudah

Perhitungan

Laju massa uap masuk =  $15774,1618 \text{ kg/hari} = 1449,0107 \text{ lb/jam}$

Suhu uap masuk =  $53^{\circ}\text{C} = 127,4^{\circ}\text{F}$

Suhu air pendingin =  $30^{\circ}\text{C} = 86^{\circ}\text{F}$

Catatan : (Ludwig, hal 211)

Non condensable gas tidak melebihi 1% dari total uap air yang akan dikondensasi.

Jadi laju uap yang akan dikondensasi =  $0,99 \times 1449,0107 = 1434,5206 \text{ lb/jam}$

Non condensable gas =  $1449,0107 - 1434,5206 = 14,4901 \text{ lb/jam}$

Temperatur uap jenuh pada  $13,5457 \text{ kPa}$  (dari neraca panas) =  $53^{\circ}\text{C} = 127,4^{\circ}\text{F}$

Terminal difference =  $5^{\circ}\text{F}$

Temperatur air keluar barometrik kondenser =  $127,4 - 5 = 122,4^{\circ}\text{F}$

Temperatur air masuk =  $86^{\circ}\text{F}$

Kenaikan suhu air =  $(122,4 - 86) = 36,4^{\circ}\text{F}$

Temperatur udara (non condensable gas) meninggalkan barometrik kondenser  
=  $86 + 5 = 91^{\circ}\text{F}$

Dari Ludwig hal 211, pers. 6.8 :

Gpm air pendingin yang dibutuhkan :

$$G_{pm} = \frac{W_{s.L}}{T_{w.500}}$$

di mana :  $W_s$  = jumlah uap yang dikondensasi (lb)

$L$  = panas laten penguapan pada  $T_{sat} = 127,4^\circ F = 1021,3$  Btu/lb

$$Gpm = \frac{1434,5206 \cdot 1021,3}{36,4 \cdot 500} = 80,4987 \text{ gpm}$$

Tinggi barometrik kondenser

$P_a$  = tekanan pada permukaan liquid dalam kaki barometer

$P_b$  = tekanan pada permukaan liquid

$P_h$  = tekanan hidrostatik

$$P_A = P_B$$

$$P_h + P_a = P_b$$

$$\rho_{air} \cdot h_{air} \cdot g + 0 = \rho_{liq} \cdot h_{liq} \cdot g$$

$$h_{air} = (13,6 \cdot 76) / 1 = 1033,6 \text{ cm} = 10,34 \text{ m}$$

Pada keadaan ideal :  $P_a = 0$

*Spesifikasi :*

Fungsi : mengembunkan uap air dari evaporator

Tipe : Counter Current Condenser

Kapasitas : 1449,0107 lb/jam

Laju air pendingin: 80,4987 gpm

Tinggi barometrik : 10,34 m

## 28. Hot Well (F-310)

Fungsi : untuk menampung kondensasi dari barometrik kondensor.

Perhitungan

$$\text{Laju massa} = 1434,5206 \text{ lb/jam} = 650,6842 \text{ kg/jam}$$

$$\text{Waktu tinggal} = 10 \text{ menit}$$

$$\text{Kapasitas} = 650,6842 \cdot 10 / 60 = 108,4474 \text{ kg}$$

$$\rho_{air} = 1000 \text{ kg/m}^3$$

$$\text{Volume air} = 108,4474 / 1000 = 0,1084 \text{ m}^3$$

$$\text{Volume air} = 80\% \text{ dari volume hot well}$$

$$\text{Volume hot well} = 0,1084 / 0,8 = 0,1355 \text{ m}^3$$

$$\text{Bentuk} = \text{persegi}$$

Ukuran = p : 0,52 m; l : 0,52 m; t : 0,5 m

*Spesifikasi:*

Fungsi : untuk menampung kondensasi dari barometrik kondensor.

Volume : 0.1355 m<sup>3</sup>

Bentuk : persegi

Ukuran : p = 0,52 m; l = 0,52 m; t = 0,5 m

## 29. Ejektor (G-312)

Fungsi : untuk memvakumkan evaporator

Tipe : Single stage steam ejector

Dasar Pemilihan : kondisi vakum cukup besar

Perhitungan

Tekanan masuk ejektor = P operasi = 4 inHg abs

Uap yang masuk = non condensable gas = 14,4901 lb/jam

Dari Ludwig vol I, hal 230, didapat :

$W_s = W_{s90} \cdot f$  di mana :  $W_s$  = total steam (lb/jam)

$W_{s90}$  = total steam untuk steam 90 psig

$f$  = steam pressure factor

Dari Ludwig vol I, hal 227, didapat :

Pada P suction 3 – 5 inHg, diambil udara leakage = 22,5 lb/jam

$W'm$  = total campuran = uap + udara leakage

= 26,13 + 22,5 = 48,63 lb/jam

Dari Ludwig fig. 6-26A, hal 230, didapat :

P suction = 4 inHg abs

$W'm$  = 48,63 lb/jam

Ukuran ejektor = 2 in L ejektor

$W_{s90}$  = 360 lb/jam

Dari Ludwig fig. 6-26B, hal 230, didapat :

Pada tekanan steam = 125 psig (Ludwig, hal 230)

$F = 0,88$   $W_s = 360 \cdot 0,88 = 316,8$  lb/jam

Evacuation : sistem volume = 300 ft<sup>3</sup> (Ludwig, hal 230)

$$E = 1,3 \quad V = 300 \quad W'm = 48,63 \text{ lb/jam}$$

Dari Ludwig pers. 6-7, hal 230, didapat :

$$W'm = E.V/t$$

$$t = (1,3.300)/48,63 = 8,02 \text{ menit}$$

$$\text{Konsumsi air} = 0,06.W_s = 0,06 \times 316,8 = 19,008 \text{ gpm} \quad (\text{Ludwig, hal 232})$$

*Spesifikasi:*

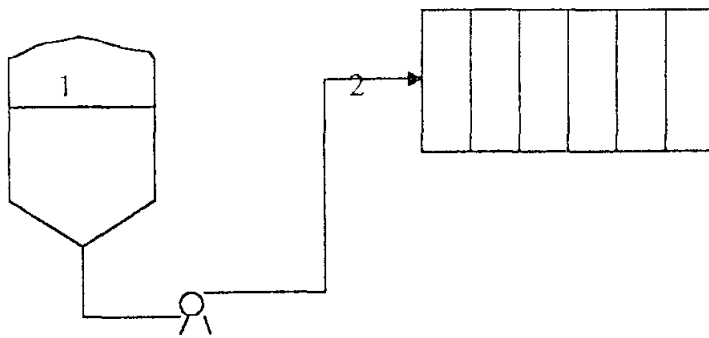
Fungsi : untuk memvakumkan evaporator

Tipe : Single stage steam ejector

Laju steam: 316,8 lb/jam

Air pendingin : 19,008 gpm

### 30. POMPA (L-314)



Fungsi : untuk mengalirkan larutan gelatin 35% menuju cooler

Kebutuhan total larutan gelatin = 7249,0101 kg/ hari = 665,882 lb/jam

Pada  $P = 1,96 \text{ psia}$  ( $0,13 \text{ atm}$ ) dan suhu operasi  $53^\circ\text{C}$  ( $127,4^\circ\text{F}$ ) :

$$\rho \text{ CaCl}_2 = 1,0034 \text{ gr/cm}^3$$

$$\rho \text{ H}_2\text{O} = 0,98565 \text{ gr/cm}^3$$

$$\rho \text{ H}_3\text{PO}_4 = 0,9993 \text{ gr/cm}^3$$

$$\rho \text{ gelatin} = 0,9866 \text{ gr/cm}^3$$

$$\rho \text{ MgCl}_2 = 1,00395 \text{ gr/cm}^3$$

$$\rho \text{ NaCl} = 0,99494 \text{ gr/cm}^3$$

(Ward, A.G)

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,6409}{0,98565} + \frac{0,35}{0,9866} + \frac{0,0035}{1,0034} + \frac{0,0056}{1,00395} + \frac{4,86 \cdot 10^{-7}}{0,99494} + \frac{5,51 \cdot 10^{-6}}{0,9993}$$

$$= 1,0141$$

$$\rho \text{ campuran} = 0,9861 \text{ gr/cm}^3 = 61,55 \text{ lb/ft}^3$$

$$\text{volume} = \frac{665,882}{61,55} = 10,8186 \text{ ft}^3/\text{jam} = 3 \cdot 10^{-3} \text{ ft}^3/\text{s} = 1,35 \text{ gpm}$$

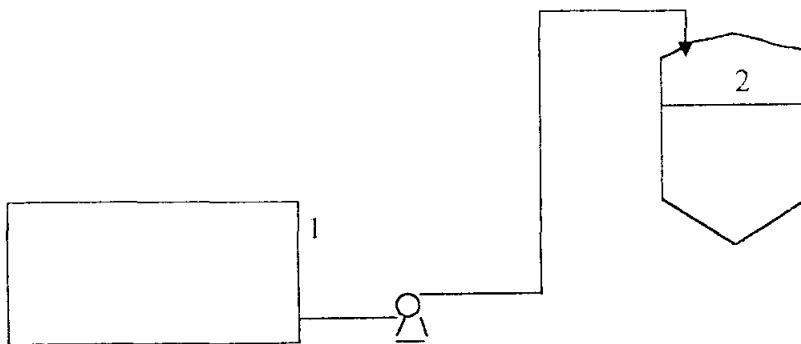
$$\mu \text{ gelatin pada } 53^\circ\text{C} = 0,9 \text{ poises} = 0,0605 \text{ lb/ft} \cdot \text{s} \quad (\text{Ward, A.G})$$

Dengan cara yang sama pada perhitungan pompa (L-145), maka didapat :

**Spesifikasi :**

Tipe	: centrifugal pump
Rate aliran pompa	: $3 \cdot 10^{-3} \text{ ft}^3/\text{s}$
Power motor	: 0,1 Hp
Bahan konstruksi	: stainless steel
Jumlah	: 1 buah

### 31. POMPA (L-321)



Fungsi : untuk mengalirkan larutan gelatin 35% menuju spray dryer

Kebutuhan total larutan gelatin = 7249,0101 kg/ hari = 665,882 lb/jam

Suhu operasi  $15^\circ\text{C}$  ( $59^\circ\text{F}$ ) dan  $P = 1 \text{ atm}$  :

$$\rho \text{ CaCl}_2 = 1,0041 \text{ gr/cm}^3$$

$$\rho \text{ H}_2\text{O} = 0,99930 \text{ gr/cm}^3$$

$$\rho \text{ H}_3\text{PO}_4 = 0,9996 \text{ gr/cm}^3$$

$$\rho \text{ gelatin} = 0,9868 \text{ gr/cm}^3$$

$$\rho \text{ MgCl}_2 = 1,0035 \text{ gr/cm}^3$$

$$\rho \text{ NaCl} = 0,9944 \text{ gr/cm}^3$$

(Ward, A.G)

$$\frac{1}{\rho_{\text{campuran}}} = \frac{0,6409}{0,99930} + \frac{0,35}{0,9868} + \frac{0,0035}{1,0041} + \frac{0,0056}{1,0035} + \frac{4,86 \cdot 10^{-7}}{0,9944}$$



$$+ \frac{5,51 \cdot 10^{-6}}{0,9996}$$

$$= 1,013$$

$$\rho \text{ campuran} = 0,9872 \text{ gr/cm}^3 = 61,63 \text{ lb/ft}^3$$

$$\text{volume} = \frac{665,882}{61,63} = 10,8045 \text{ ft}^3/\text{jam} = 3 \cdot 10^{-3} \text{ ft}^3/\text{s} = 1,35 \text{ gpm}$$

$$\mu \text{ gelatin pada } 15^\circ\text{C} = 0,5 \text{ poises} = 0,034 \text{ lb/ft} \cdot \text{s (Ward, A.G)}$$

Dengan cara yang sama pada perhitungan pompa (L-145), maka didapat :

**Spesifikasi :**

Tipe	: Displacement pump
Rate aliran pompa	: $3 \cdot 10^{-3} \text{ ft}^3/\text{s}$
Power motor	: 0,1 Hp
Bahan konstruksi	: stainless steel
Jumlah	: 1 buah

### 32. Alat Pendingin (B-320)

Fungsi : untuk mendinginkan larutan gelatin

Tipe : Double pipe Heat Exchanger

Perhitungan :

$$R_d = 0,003$$

Dipilih ukuran pipa  $2 \frac{1}{2} \times 1 \frac{1}{4}$  in IPS dengan 20 ft-hairpains

$$\text{Suhu air masuk} = 4^\circ\text{C} = 39,2^\circ\text{F}$$

$$\text{Suhu larutan masuk} = 53^\circ\text{C} = 127,4^\circ\text{F}$$

$$\text{Suhu air keluar} = 15^\circ\text{C} = 59^\circ\text{F}$$

$$\text{Suhu larutan keluar} = 15^\circ\text{C} = 59^\circ\text{F}$$

Neraca panas :

$$\text{Massa air} = 9212,173836 \text{ kg/hari} = 846,215 \text{ lb/jam}$$

$$\text{Massa larutan gelatin} = 7249,0101 \text{ kg/hari} = 665,882 \text{ lb/jam}$$

$$Q = 16744,3145 \text{ Btu/jam}$$

$$\Delta T_{LMTD} = \frac{(127,4 - 59) - (59 - 39,2)}{\ln \frac{(127,4 - 59)}{(59 - 39,2)}} = 39,19^{\circ}\text{F}$$

$$\text{Dalam } ^{\circ}\text{F} : T_c = \frac{127,4 + 59}{2} = 93,2^{\circ}\text{F} ; \quad t_c = \frac{59 + 39,2}{2} = 49,1^{\circ}\text{F}$$

$$\text{Dalam } ^{\circ}\text{C} : T_c = \frac{53 + 15}{2} = 34^{\circ}\text{C} ; \quad t_c = \frac{15 + 4}{2} = 9,5^{\circ}\text{C}$$

Hot fluid : annulus, larutan gelatin 35%	Cold fluid : inner pipe, air
- Dari table 11 :	- $D = 1,38/12 = 0,115 \text{ ft}$
$D_2 = \frac{2,469}{12} = 0,20575 \text{ ft}$	$a_p = \frac{\pi D^2}{4} = \frac{\pi (0,115)^2}{4} = 0,0104 \text{ ft}^2$
$D_1 = \frac{1,66}{12} = 0,138 \text{ ft}$	- $Gp = \frac{W}{a_p} = \frac{846,215}{0,0104} = 81366,83$
$a_a = \frac{\pi (D_2^2 - D_1^2)}{4}$	- pada $9,5^{\circ}\text{C}$ , $\mu = 3,2108 \text{ lbm/ft.hr}$
$a_a = \frac{\pi (0,20575^2 - 0,138^2)}{4}$	$Rep = \frac{D \cdot Gp}{\mu} = \frac{81366,83 \cdot 0,115}{3,2108}$
$= 0,0183 \text{ ft}^2$	$Rep = 2914,285$
$De = \frac{(D_2^2 - D_1^2)}{D_1}$	- $J_{11} = 15$
$= \frac{(0,20575^2 - 0,138^2)}{0,138}$	- pada $49,1^{\circ}\text{F}$ , $k = 0,3378^{\circ}\text{F}$
$De = 0,1688 \text{ ft}$	$\left( \frac{Cp \cdot \mu}{k} \right)^{1/3} = \left( \frac{1,3,2108}{0,3378} \right)^{1/3} = 2,118$
- $Ga = \frac{Wa}{a_a} = \frac{665,882}{0,0183} = 36386,995$	- $hi = J_{11} \cdot \frac{k}{D} \cdot \left( \frac{Cp \cdot \mu}{k} \right)^{1/3} \cdot \left( \frac{\mu}{\mu_w} \right)^{0,14}$
- pada $34^{\circ}\text{C}$ , $\mu = 0,4015 \text{ lbm/ft.hr}$	$= \frac{15 \times 0,3378 \times 2,118}{0,115}$
$Rea = \frac{De \cdot G_a}{\mu} = \frac{36386,995 \cdot 0,1688}{0,4015}$	$= 93,3209$
$= 15297,9446$	- $hio = hi \times \frac{ID}{OD}$
- $J_{11} = 65$ didapat dari fig. 34 (Kern)	

- Cp gelatin 35% = 0,777 kkal/kg.°C = 1,39768 Btu/lbm.°F	= 93,3209 x $\frac{1,38}{1,66}$
- k = 0,359 pada 93,2°F	= 77,58
$\left(\frac{Cp \cdot \mu}{k}\right)^{1/3} = \left(\frac{1,39768 \cdot 0,4015}{0,359}\right)^{1/3}$ = 1,1606	
- ho = $J_H \cdot \frac{k}{De} \left(\frac{Cp \cdot \mu}{k}\right)^{1/3} \left(\frac{\mu}{\mu_w}\right)^{0,14}$ = $\frac{65 \times 0,359 \times 1,1606}{0,1688}$ = 160,4419	

$$- U_c = \frac{h_o \cdot h_o}{h_o + h_o} = \frac{160,4419 \cdot 77,58}{160,4419 + 77,58} = 52,2939$$

- Design Overall coefficient

$$\frac{1}{U_D} = \frac{1}{U_c} + R_D$$

$$\frac{1}{U_D} = \frac{1}{52,2939} + 0,003$$

$$U_D = 45,2025$$

- Luas perpindahan panas

$$Q = U_D \cdot A \cdot \Delta T$$

$$A = \frac{Q}{U_D \cdot \Delta T} = \frac{16744,3145}{45,2025 \cdot 39,19} = 9,45 \text{ ft}^2$$

Dari table 11 kern untuk  $1 \frac{1}{4}$  in IPS didapat :

$$a'' = 0,435 \text{ ft}^2/\text{ft}$$

$$\text{panjang} = \frac{9,45}{0,435} = 21,72 \text{ ft}$$

$$20 \text{ ft-hairpain} : \frac{21,72}{20} = 1,086 \text{ pipa} \approx 2 \text{ pipa}$$

- Luas perpindahan actual = 40 x 0,435

$$= 17,4 \text{ ft}^2$$

$$U_D = \frac{16744,3145}{17,4 \cdot 39,19} = 24,555$$

$$Rd = \frac{U_c - U_D}{U_c U_D} = \frac{52,2939 - 24,555}{52,2939 \times 24,555} = 0,007$$

$\Delta P$

Hot fluid : annulus, larutan gelatin 35%	Cold fluid : inner pipe, air
$- Dc' = (D_2 - D_1)$ $= (0,20575 - 0,138)$ $= 0,06775 \text{ ft}$ $Rea' = \frac{De' \cdot Ga}{\mu} = \frac{0,06775 \cdot 36386,995}{0,4015}$ $= 6140,02$ $f = 0,0035 + \frac{0,264}{(6140,02)^{0,42}} = 0,0103$ $- \text{pada } 34^\circ\text{C}, \rho_{\text{gelatin}} = 62,1326 \text{ lb/ft}^3$ $\Delta Fa = \frac{4 \cdot f \cdot Ga^2 \cdot L}{2 \cdot g \cdot \rho^2 \cdot De'}$ $--$ $\frac{4 \times 0,0103 \times (36386,995)^2 \cdot 40}{2 \times (62,1326)^2 \times 4,18 \cdot 10^8 \times 0,06775}$ $\Delta Fa = 9,979 \cdot 10^{-3}$ $- v = \frac{G}{3600 \cdot \rho} = \frac{36386,995}{3600 \cdot 62,1326} = 0,1627$ $F_l = 3 \times \left( \frac{v^2}{2 \cdot g'} \right) = 3 \times \left( \frac{0,1627^2}{2 \times 32,2} \right)$ $F_l = 1,23 \cdot 10^{-3}$ $\Delta Pa = \frac{(9,979 \cdot 10^{-3} + 1,23 \cdot 10^{-3}) \times 61,1326}{144}$	$- Rep' = 2914,285$ $f = 0,0035 + \frac{0,264}{(2914,285)^{0,42}}$ $= 0,0128$ $- \text{pada } tc = 49,1^\circ\text{F}$ $sg = 1$ $\text{pair} = 62,5 \times 1 = 62,5$ $\Delta Fp = \frac{4 \cdot f \cdot Gp^2 \cdot L}{2 \cdot g \cdot \rho^2 \cdot D}$ $= \frac{4 \cdot 0,0128 \cdot 81366,83^2 \cdot 40}{2 \cdot 4,18 \cdot 10^8 \cdot 62,5^2 \cdot 0,115}$ $\Delta Fp = 0,1033 \text{ ft}$ $\Delta Pp = \frac{0,1033 \times 62,5}{144} = 0,045$

$\approx 4,76 \cdot 10^{-3}$	
------------------------------	--

**33. Blower I (G-332)**

Fungsi : Menghembuskan udara masuk ke pemanas udara (HE)

Data : Kapasitas udara = 91332,3161 kg/hari = 1,0571 kg/s

$$T_{\text{udara}} = 30^{\circ}\text{C} = 546^{\circ}\text{R}$$

$$P_{\text{masuk}} = 1 \text{ atm}$$

$$P_{\text{keluar}} = 2 \text{ atm}$$

$$-W_s = \frac{\gamma}{\gamma - 1} \times \frac{R \cdot T_1}{M} \times \left[ \left( \frac{P_2}{P_1} \right)^{\frac{(\gamma-1)}{\gamma}} - 1 \right] \quad \text{Geankoplis 3<sup>rd</sup>.ed, p.139, eq(3.3-14)}$$

di mana :  $R = 8314,3 \text{ j/kmol}$

$\gamma = 1,4$  (untuk udara, geankoplis, 3<sup>rd</sup>.ed, p.139)

$$T_1 = 30^{\circ}\text{C}$$

$$M = 28,84 \text{ kg udara/mol}$$

$$P_2 = 2 \text{ atm}$$

$$P_1 = 1 \text{ atm}$$

$$\begin{aligned} -W_s &= \frac{1,4}{1,4 - 1} \times \frac{8314,3 \cdot 30}{28,84} \times \left[ \left( \frac{2}{1} \right)^{\frac{(1,4-1)}{1,4}} - 1 \right] \\ &= 6629,6549 \text{ j/kg} \end{aligned}$$

$$\text{brake kw} = \frac{-W_s \cdot m}{\eta \cdot 1000};$$

di mana :  $m = \text{kapasitas} = 1,0571 \text{ kg/s}$

$$\eta = 0,8$$

$$\text{brake kw} = \frac{6629,6549 \times 1,0571}{0,8 \times 1000} = 8,76 \text{ kw}$$

$$= 8,76 \text{ kw} \times \frac{1 \text{ Hp}}{0,7457 \text{ kw}} = 11,75 \text{ Hp}$$

bahan konstruksi = wrought iron

jumlah = 1 buah

**Spesifikasi alat:**

Fungsi: untuk menghembuskan udara ke dalam alat pemanas udara

Type: Centrifugal blower

Power : 11,75 Hp

Kapasitas udara: 3805,5131 kg/jam

Kondisi operasi: T udara masuk = 30 °C

P masuk = 1 atm

P keluar = 2 atm

Bahan konstruksi = wrought iron

Jumlah : 1 buah

**34. Heat Exchanger (E-331)**

Fungsi : Untuk memanaskan udara yang dipakai pada spray dryer

Type : Shell and Tube

Kondisi operasi : T steam masuk,  $T_1 = 120^\circ\text{C}$

t udara masuk,  $t_1 = 30^\circ\text{C}$

t udara keluar,  $t_2 = 110^\circ\text{C}$

1. Neraca panas.

Dari geankoplis App. A-5, untuk steam pada suhu  $120^\circ\text{C}$ , didapatkan harga

$$\lambda = 526,4214 \text{ kkal/kg} = 946,9511 \text{ btu/lbm}$$

$$Q = m \cdot \lambda$$

$$= 431,2454 \cdot 946,9511 = 408368,2662 \text{ btu/jam}$$

$$2. \Delta T_{\text{LMTD}} = \frac{(120 - 30) - (120 - 110)}{\ln \frac{(120 - 30)}{(120 - 110)}} = 36,41^\circ\text{C}$$

$$3. T_c = \frac{T_2 + T_1}{2} = \frac{120 + 120}{2} = 120^\circ\text{C}$$

$$t_c = \frac{t_2 + t_1}{2} = \frac{110 + 30}{2} = 70^\circ\text{C}$$

Trial Ud

Dari table 8 hal 840 kern, untuk proses heater, jika digunakan liquida panas berupa steam dan pada liquida dingin berupa gas (udara), maka Ud overall berkisar antara 5-50 btu/jam.ft<sup>2</sup>.°F.

Dicoba trial Ud = 20 Btu/jam.ft<sup>2</sup>.°F , maka :

$$A = \frac{Q}{Ud.\Delta T'_{LMTD}} = \frac{408368,2662}{20.36,41} = 560,79 \text{ ft}^2$$

Asumsi dipakai pipa ¾ in OD, 16 BWG, panjang 16 ft

Dari Kern, p843 , table 10 diperoleh data :

$a' = 0,302 \text{ in}^2$

$a'' = 0,19633 \text{ ft}^2/\text{ft}$

$$\text{Jumlah tube} = \frac{A}{a''.L} = \frac{560,79}{0,19633.16} = 178,55$$

Standarisasi, Nt = 184 (Kern, p.842, table 9, 1 in OD on 1¼ square pitch , 6 passes)

Bagian Shell

ID shell = 23,25 in

Ns = 1

Bagian Tube

OD = 1 in, 16 BWG

ID = 0,870 in = 0,072 ft

L = 16 ft, pt = 1,25 in

Nt = 184

N = 6 passes

Evaluasi Perpindahan Panas (Rd)

Shell, udara	Tube, Steam
4. ID shell = 23,25 in ; B = 12 in Pt = 1,25 in C' = 1,25 – 1 = 0,25 in $As = \frac{ID.C'.B}{144.Pt} = \frac{23,25.0,25.12}{144.1,25}$ $= 0,3875 \text{ ft}^2$	$4'. \text{ at} = \frac{Nts \tan dar.a't}{144.n}$ $= \frac{184x0,594}{144x6} = 0,1265 \text{ ft}^2$
5. $Gs = \frac{M}{as} = \frac{91332,3161x2,2046/24}{0,3875}$	$5'. Gt = \frac{m}{at} = \frac{431,2454}{0,1265}$ $= 3409,055 \text{ lb/ft}^2.\text{jam}$
	6'. ID = 0,870 in = 0,072 ft

$= 21650,67 \text{ lb/jam.ft}^2$ <p>6. pada <math>t_c = 70^\circ\text{C}</math> (Kern, fig 15) didapatkan :</p> $\mu = 0,02 \times 2,42 = 0,0484 \text{ lb/j.ft}$ <p>dari fig 28 didapat : <math>De = 0,95 \text{ in}</math>  <math>= 0,079 \text{ ft}</math></p> $\text{Res} = \frac{De.Gs}{\mu} = \frac{0,079.21650,67}{0,0484}$ $= 35338,90$ <p>7. <math>jH = 130</math></p> <p>8. <math>\left(\frac{Cp.\mu}{k}\right)^{1/3} = \left(\frac{0,25.0,0484}{0,01548}\right)^{1/3}</math>  <math>= 0,9212</math></p> <p>9. <math>ho = jH \cdot \frac{k}{De} \cdot \left(\frac{Cp.\mu}{k}\right)^{1/3}</math>  <math>= 130 \cdot \frac{0,01548}{0,079} \cdot 0,9212</math>  <math>= 23,47 \text{ btu/jam.ft}^2.\text{°F}</math></p>	$\text{Ret} = \frac{ID.Gt}{\mu} = \frac{0,072.3409,055}{0,013.2,42}$ $= 7792,1257$ <p>7'. <math>hio = 1500 \text{ btu/jam.ft}^2.\text{°F}</math></p>
--	--

$$10. U_c = \frac{hio.ho}{hio + ho} = \frac{1500 \times 23,47}{1500 + 23,47} = 23,11 \text{ Btu/j.ft}^2.\text{°F}$$

$$A \text{ terkoreksi} = Nt.L.a'' = 184.16.0,2618 \text{ ft}^2/\text{ft} = 770,7392 \text{ ft}^2$$

$$U_D \text{ terkoreksi} = \frac{Q}{A \text{ terkoreksi} \cdot \Delta T_{LMTD}} = 14,55 \text{ Btu/j.ft}^2.\text{°F}$$

$$14. R_d - \frac{1}{U_d} - \frac{1}{U_c} = \frac{1}{14,55} - \frac{1}{23,11} = 0,025 \text{ (memenuhi)}$$

EVALUASI  $\Delta P$ 

Shell, udara	Tube
1. $\text{Res} = 35338,90$ $f_s = 0,0016$ (Kern, p.839, fig 29)	1'. $\text{Ret} = 7792,1257$ $f_t = 0,00029$



$N+1 = \frac{12.L}{B} \text{ (eq 7.43, Kern)}$ $= \frac{12.16}{12} = 16$	$2'. \Delta Pt = \frac{f.Gt^2.Dt.L.n}{5,22.10^{10}.Dt.\theta_i}$ $= 6,198.10^{-6}$
$2. \Delta Ps = \frac{f.Gs^2.Ds.(N+1)}{5,22.10^{10}.Des.\theta_s}$ $= 5,761 \text{ psia} < 10 \text{ psi (memenuhi)}$	$3'. Ret = 7792,1257$ $V^2/2g' = 0,0008 \text{ (Kem,p.837,fig 27)}$ $\Delta Pr = \frac{4.n}{s} \cdot \frac{V^2}{2.g'} = 0,4973$ $\Delta P_{total} = \Delta Pt + \Delta Pr = 0,4973$
$\Delta P = 5,761 + 6,198.10^{-6} + 0,4973 = 6,2583 \text{ psi (memenuhi)}$	

### 35. Spray Dryer (B-330)

Fungsi : sebagai alat pengering slurry gelatin sehingga diperoleh powder gelatin dalam bentuk granular

Tipe : Bejana silinder dengan bagian bawah berbentuk konis dan tutup atas berbentuk dished head.

Kapasitas : 7249,0101 kg/hari = 302,0421 kg/jam

Kondisi : Suhu udara masuk = 110°C

Suhu feed masuk = 15°C

Tekanan operasi = 1 atm

Suhu operasi = 50°C

Perhitungan :

Menghitung ukuran dryer :

Feed masuk terdiri dari :

Padatan : 2537,1535 kg/hari = 105,7147 kg/jam

Air : 4711,8566 kg/hari = 196,3274 kg/jam

Kadar air mula-mula = 65%

Dari perhitungan Appendix A :

H<sub>2</sub>O yang menguap = 4398,5058 kg/hari = 183,2711 kg/jam

Gelatin keluar sebagai produk = 2850,5043 kg/hari = 118,771 kg/hari

Kadar air dalam produk = 11 %

Laju pengeringan = air yang menguap = 183,2711 kg/jam

Suhu udara masuk = 110 °C

Suhu udara keluar = 70 °C

Dari Perry ed 5.p. 20-63 :

Volume chamber = 6000 ft<sup>3</sup>

Diameter = 22 ft = 6,705 m

Lubang pengeluaran = 12 in (Hesse p.85)

Tinggi shell = 0,4 D (fig 20.72 Perry ed 5) = 8,8 ft = 2,68 m

Volume silinder =  $\pi/4 \cdot D^2 \cdot H_s = \pi/4 \cdot 22^2 \cdot 8,8 = 3345,167 \text{ ft}^3$

Volume konis = Volume chamber – Volume silinder

$$= 6000 - 3345,167$$

$$= 2465,833 \text{ ft}^3$$

$$\text{Tinggi konis} = \frac{0,5 \cdot D}{\lg \alpha} = \frac{0,5 \cdot 22}{\lg 60} = 6,3502 \text{ ft} = 1,9356 \text{ m}$$

Tinggi total = Hs + hc = 4,6156 m

Waktu pengeringan total : T = 17.(L/S) (Foust,p.548)

L/S = Perbandingan massa dari air terhadap solid dalam feed

$$T = 17 \times (196,3274/105,7147) = 31,5714 \text{ detik}$$

Volume yang dibutuhkan untuk pengeringan tersebut :

Volume = waktu tinggal x total rate gas

$\rho$  udara pada suhu 110°C = 0,901 kg/m<sup>3</sup> (geankoplis app. A3-3)

$$\begin{aligned} \text{Rate volume udara} &= 91332,3161 \text{ kg/hari}/0,901 = 101367,7204 \text{ m}^3/\text{hari} \\ &= 4223,6550 \text{ m}^3/\text{jam} \end{aligned}$$

Air yang diuapkan = 183,2711 kg/jam

$\rho$  uap air pada suhu 110°C = 0,5507 kg/m<sup>3</sup> (Geankoplis A.2-12)

$$\text{Rate volumetric air yang menguap} = 183,2711/0,5507 = 332,7966 \text{ m}^3/\text{jam}$$

Total aliran gas meninggalkan dryer = rate udara + rate uap air

$$= 4223,6550 + 332,7966$$

$$= 4556,4516 \text{ m}^3/\text{jam}$$

$$= 0,053 \text{ m}^3/\text{s}$$

$$\text{Volume} = 31,5714 \times 0,053 \text{ m}^3/\text{s} = 1,673 \text{ m}^3$$

Pengecekan terhadap volume chamber:

Volume chamber =  $6000 \text{ ft}^3 = 169,9 \text{ m}^3 > 1,673 \text{ m}^3$  kadar air yang diinginkan dapat tercapai

**.Spesifikasi dryer :**

Tinggi silinder	: 2,68 m
Tinggi konis	: 1,9356 m
Diameter	: 6,705 m

Menghitung tebal shell

Untuk shell tutup bagian bawah dan tutup bagian atas dipilih bahan konstruksi carbon steel SA-283 grade C (Brownell, App D item 4)

$$f = 12650 \text{ psia}$$

$$E = 0,85 \text{ (single welded butt join)}$$

$$C = 1/8 \text{ in} = 0,125 \text{ in}$$

$$R_i = 11 \text{ ft} = 132 \text{ in}$$

$$P_{\text{desain}} = 1,1 \cdot P_{\text{operasi}}$$

$$= 1,1 \cdot 14,7$$

$$= 16,17 \text{ psi}$$

Tebal shell

$$\begin{aligned} t_s &= \frac{P \cdot r_i}{f \cdot E - 0,6 \cdot P} + C \\ &= \frac{16,17 \cdot 132}{(12650 \cdot 0,85) - (0,6 \cdot 16,17)} + \frac{1}{8} \text{ in} \\ &= 0,3237 \text{ in} \approx \frac{7}{16} \text{ in} \end{aligned}$$

Menghitung tebal konis

$$\text{Tebal konis} = \frac{P \cdot L \cdot W}{2 \cdot f \cdot E - (0,2 \cdot P)} + C \quad (\text{B \& Y pers 7.77})$$

$$L = d_i / 2 \cos \alpha \quad (\text{B \& Y pers 13.17})$$

$$d_i = D - 2 \cdot i \cdot \cos \alpha \quad (\text{B \& Y p. 260})$$

$$D = 22 \text{ ft} = 132 \text{ in}$$

Dari table 5.6 B&Y untuk  $t = 7/16 \text{ in}$  didapat  $icr = 8$

$$di = 132 - 2.8.(1 - \cos 30) = 129,856 \text{ in}$$

$$L = \frac{129,856}{2 \cdot \cos 30} = 74,9724 \text{ in}$$

$$W = \frac{1}{4} \cdot \left\{ 3 + \sqrt{L/icr} \right\} = \frac{1}{4} \cdot \left\{ 3 + \sqrt{\frac{74,9724}{8}} \right\} = 1,52 \text{ in}$$

$$t = \frac{16,17 \cdot 74,9724 \cdot 1,52}{(2 \cdot 12650 \cdot 0,85) - (0,2 \cdot 16,17)} + \frac{1}{8} \text{ in}$$

$$= 0,2107 \text{ in} \approx \frac{3}{8} \text{ in}$$

#### Menghitung tutup atas (dished head)

$$\text{Tebal dished} = \frac{0,885 \cdot P \cdot rc}{f \cdot E - 0,1 \cdot P} + C \quad (\text{Brownell, pers 13-12})$$

Dimana :  $Rc = ID = 22 \text{ ft} = 132 \text{ in}$

$$t = \frac{0,885 \cdot 16,17 \cdot 132}{(12650 \cdot 0,85) - (0,1 \cdot 16,17)} + \frac{1}{8} \text{ in}$$

$$= 0,3 \text{ in} \approx \frac{3}{8} \text{ in}$$

Dari table 5.6 Brownell didapat, untuk tebal dished =  $\frac{3}{8} \text{ in}$ ,  $sf = 1 \frac{1}{2} - 3 \text{ in}$ ,

$icr = 1 \frac{1}{8} \text{ in}$ , diambil  $sf = 2 \text{ in}$

$$BC = r - icr \text{ (B\&Y, p.87)}$$

$$= 132 - 1,125 \text{ in} = 130,875 \text{ in} = 10,90625 \text{ ft} = 3,32 \text{ m}$$

$$a = ID/2 = 132/2 = 66 \text{ in} = 5,5 \text{ ft} = 1,7 \text{ m}$$

$$AB = a - icr = 1,7 - 1,125 = 0,575 \text{ m}$$

$$b = r - \sqrt{BC^2 - AC^2} = 132 - \sqrt{130,875^2 - 22,6375^2} = 3,1 \text{ in} = 0,079 \text{ m}$$

$$OA = t + b + sf = \frac{3}{8} + 3,1 + 2 = 5,475 \text{ in} = 0,13291 \text{ m}$$

$$OD = ID + 2 \cdot t = 132 + 2 \cdot \frac{3}{8} = 132,75 \text{ in} = 3,4 \text{ m}$$

#### Atomizer

Menentukan kecepatan atomizer

$$D_{50} = \frac{k.M^a}{N^b.dh^c.(n.h)^d} \cdot 10^{-4} \mu m \quad (\text{Van't land, pers. 8-1})$$

$D_{50}$  = median diameter,  $\mu m$  (=152  $\mu m$ )

K = konstanta

M = atomizer wheel feed, kg/j = 7249,0101 kg/jam = 266,3528 lb/menit

N = kecepatan putaran, rpm

di = diameter roda atomizer, m (=0,25 m = 0,8202 ft (Van't land hal 163))

n = banyaknya vane dalam roda (=45)

h = tinggi vane roda atomizer (= 30 mm = 0,03 m)

Vane liquid loading = r = feed masuk / Lw

Dimana Lw = wetted dish peripheral

$$= \pi . Di = \pi . 0,25 = 0,7854 \text{ m}$$

$$r = 7249,0101 \text{ kg/jam} / 0,7854 \text{ m} = 9229,705 \text{ kg/jam.m}$$

Dari table 8.1 Van't Land didapat :

$$a = 0,12$$

$$b = 0,8$$

$$c = 0,6$$

$$d = 0,12$$

$$k = 1,2$$

$$152 = \frac{1,2.(7249,0101)^{0,12}}{N^{0,8}.(0,25)^{0,6}.(45.0,03)^{0,12}} \cdot 10^{-4} \mu m$$

$$N = 2456,0526 \text{ rpm} \approx 2500 \text{ rpm}$$

Menghitung power yang dibutuhkan :

$$P = 1,04.10^{-8} . (r.N)^2 . W$$

Dimana : P = Hp netto, Hp

r = jari-jari roda atomizer, ft

N = putara disk, rpm

W = rate feed, lbm/menit

$$P = 1,04.10^{-8} . (0,4101.2500)^2 . 266,3528 \text{ lbm/menit}$$

$$P = 2,9 \text{ Hp} \approx 3 \text{ Hp}$$

### 36. Cyclone Spray Dryer (H-333)

Fungsi : Memisahkan debu yang terbawa oleh gas yang keluar dari Spray Dryer

Perhitungan :

Kapasitas : 4427,01082 kg/hari = 184,4588 kg/jam

Nozzle gas masuk ukuran 15" (fig 14-2 Peter and Timmerhaus)

Standar pipe ukuran nozzle : 15,25 in (Tabel 13,p.888 Peter and Timmerhaus)

Luas penampang nozzle :  $\frac{1}{4} \pi D_i^2 = \frac{1}{4} \pi 15,25^2 = 182,65 \text{ in}^2 = A_c$

Dari Perry edisi 6 hal 20-82, didapatkan penampang gas masuk :

$$A_c = B_c \times H_c$$

$$H_c = 2 \times B_c$$

$$A_c = 2 \times B_c^2$$

$$B_c = \sqrt{\frac{A_c}{2}} = \sqrt{\frac{182,654}{2}} = 9,55 \text{ in}$$

$$D_c = B_c \times 4 = 9,55 \text{ in} \times 4 = 38,226 \text{ in}$$

$$D_e = D_c/2 = 38,226 \text{ in} / 2 = 19,113 \text{ in}$$

$$H_c = D_c/2 = 38,226 \text{ in} / 2 = 19,113 \text{ in}$$

$$L_c = 2 \times D_c = 2 \times 38,226 \text{ in} = 76,45 \text{ in}$$

$$S_c = D_c/8 = 38,226 \text{ in} / 8 = 4,778 \text{ in}$$

$$Z_c = 2 \times D_c = 2 \times 38,226 \text{ in} = 76,452 \text{ in}$$

$$J_c = D_c/4 = 38,226 \text{ in} / 4 = 9,5565 \text{ in}$$

Dimana :

$A_c$  = luas penampang gas masuk,  $\text{in}^2$

$D_c$  = diameter cyclone, in

$D_e$  = diameter lubang pengeluaran gas

$H_c$  = diameter lubang masuk, in

$L_c$  = tinggi cyclone bagian silinder, in

$Z_c$  = tinggi cyclone bagian kerucut, in

$J_c$  = diameter lubang pengeluaran partikel, in

#### Spesifikasi :

1. Type : Effluent Dust Cyclone
2. Kapasitas : 1074,177 lb/menit

3. Ukuran :  $A_c = 182,65 \text{ in}^2$

$B_c = 9,55 \text{ in}$

$D_c = 38,226 \text{ in}$

$D_e = 19,113 \text{ in}$

$H_c = 19,113 \text{ in}$

$L_c = 76,45 \text{ in}$

$S_c = 4,778 \text{ in}$

$Z_c = 76,45 \text{ in}$

$J_c = 5,9965 \text{ in}$

### 37. CRUSHER (C-340)

Fungsi : menghancurkan gelatin dengan ukuran 6 mm

Tipe : Jaw crusher

Kondisi operasi :  $T = 30^0\text{C}$

Tulang yang harus dihancurkan =  $7249,0101 \text{ kg/hari} = 0,302 \text{ ton/jam} \approx 1 \text{ ton/jam}$ .

Dasar pemilihan :

- Kapasitas 1 ton/jam dapat digunakan Dodge jaw crusher

(Perry 6<sup>th</sup> ed., table 8-7)

Spesifikasi :

- Fungsi : menghancurkan gelatin dengan ukuran 6 mm
- Tipe : Dodge jaw crusher
- Kecepatan putaran jaw : 275 rpm
- Power : 3 Hp
- Setting : 1 in
- Crusher size : 4 x 6 in
- Jaw motion :  $\frac{1}{2} \text{ in}$
- Jumlah : 1 buah

## APPENDIKS D.

### PERHITUNGAN ANALISA EKONOMI



## APPENDIX D

### PERHITUNGAN ANALISA EKONOMI

Harga alat akan berubah setiap saat tergantung pada kondisi ekonomi dan politik, untuk itu dibutuhkan suatu metode yang dapat dipakai untuk mengkonversi harga alat pada beberapa tahun yang lalu agar dapat diperoleh harga alat yang ekuivalen untuk waktu sekarang.

Ekivalensi itu dapat dihitung dengan menggunakan persamaan :

$$\text{Harga alat saat ini} = \frac{\text{Indeks harga saat ini}}{\text{Indeks harga tahun X}} \times \text{Harga alat tahun X}$$

Harga alat yang digunakan dalam pra rencana ini didasarkan pada harga alat yang terdapat pada pustaka:

- a. Garret, D.E., 1989, "Chemical Engineering Economics", Van Nostrand Reinhold, New York.
- b. Peters, M.S., Timmerhouse, K.D., 1991, "Plant Design and Economics for Chemical Engineers", 4<sup>th</sup> ed., McGraw Hill Co, Singapore
- c. Ulrich, G.D., 1984, "A Guide To Chemical Engineering Process Design and Economic" John Wiley and Sons, Singapore

Dalam perhitungan ini digunakan indeks harga sebagai berikut:

➤ Marshall & Swift equipment cost index

Tahun X ( Januari 1990) = 904 (Peters & Timmerhaus, tabel 3, p.163)

Saat ini (Juni 2002) = 1073,5 (Chemical Engineering, Januari 2002)

➤ Chemical engineering plant cost index

Tahun X (awal 1987) = 320 (Garret, p. 255)

Saat ini (awal 2002) = 394,3 (Chemical Engineering, Januari 2002)

#### A. Perhitungan Harga Peralatan

Contoh perhitungan :

Nama alat : Bucket elevator (J-141)

Fungsi : memindahkan tulang dari gudang bahan baku (F-130) ke crusher (C-142)

Tipe : centrifugal discharge bucket.

Jumlah : 1 buah

Harga tahun 1990 : \$6500 (Peters & Timmerhaus, fig. 14-90, p.569)

Harga alat saat ini :  $\frac{1073,5}{904} \times \$ 6500 = \$ 7718,75$

Dengan cara yang sama didapatkan harga alat yang lain seperti pada tabel D.1.

Tabel D.1. Harga Peralatan Proses

Nama alat	Kode	Jumlah	Harga tahun 2002, @ \$	Total \$
Tangki HCl	F-110	1	123437,5	123437,5
Tangki Kapur	F-120	1	42323,121	42323,121
Tangki Demineralisasi	F-140	7	132661,7056	928631,9393
Bucket elevator	J-141	1	7718,75	7718,75
Crusher	C-142	1	14786,25	14786,25
Bucket elevator	J-143	1	7718,75	7718,75
Rotary vacuum filter	H-144	1	59375	59375
Pompa	L-145	7	111,11	777,77
Tangki Liming	F-150	7	71250	498750
Rotary vacuum filter	H-151	1	77187,5	77187,5
Pompa	L-152	1	200	200
Tangki Netralisasi	F-160	1	77187,5	77187,5
Rotary vacuum filter	H-161	1	45125	45125
Pompa	L-162	1	111,11	111,11
Storage bin	F-163	1	1500	1500
Tangki ekstraksi I	F-210	1	10093,75	10093,75
Tangki ekstraksi II	F-220	1	10006,25	10006,25
Rotary vacuum filter	H-221	1	34437,5	34437,5
Tangki ekstraksi III	F-230	1	11281,25	11281,25
Tangki ekstraksi IV	F-240	1	11340,625	11340,625
Rotary vacuum filter	H-241	1	27312,5	27312,5
Tangki ekstraksi V	F-250	1	14250	14250

Rotary vacuum filter	H-251	1	24937,5	24937,5
Tangki penampung hasil ekstraksi	F-252	1	1971,5	1971,5
Pompa	L-253	1	111,11	111,11
Evaporator	V-310	1	77187,5	77187,5
Barometric condensor	E-311	1	1306,25	1306,25
Steam ejector	G-312	1	593,75	593,75
Hot well	F-313	1	415,625	415,625
Pompa	L-314	1	111,11	111,11
Coller	B-320	1	21375	21375
Pompa	L-321	1	111,11	111,11
Spray dryer	B-330	1	110896,875	110896,875
Shell & tube	E-331	1	17218,75	17218,75
Blower	G-332	1	25875,9375	25875,9375
Cyclone	H-333	1	2800	2800
Crusher	C-340	1	14786,25	14786,25
Total				2.303.250,333

Tabel D.2. Harga Peralatan Utilitas

No	Nama alat	Kode	Jml	Total US \$
1	Filter	H-413	1	1710
2	Tangki demineralisasi	H-418	1	52563,12
3	Tangki pengisi boiler	F-422	1	3176
4	Pompa air sumur	L-410	1	111,11
5	Pompa air ke filter	L-415	1	111,11
6	Pompa air sanitasi	L-412	1	111,11
7	Pompa tangki demineralisasi	L-417	1	111,11
8	Pompa air pendingin	L-420	1	111,11
9	Pompa air boiler	L-426	1	111,11
10	Pompa air refrigerant	L-425	1	111,11
11	Pompa air proses	L-424	1	111,11
11	Boiler	-	1	12842
12	Generator	-	1	21620
13	Tangki penampung IDO	-	1	14750
14	Refrigeration	-	1	27623,9733
Total				135173,9733

Tabel D.3 Bak Utilitas

No	Nama Alat	Kode	Jml	Luas/m <sup>2</sup>
1	Bak penampung air sumur	F-411	1	68,04
2	Bak penampung air bersih	F-414	1	68,2889
3	Bak penampung air demineralisasi	F-419	1	109,44
4	Bak air pendingin	F-423	1	27,3597
5	Bak air sanitasi	F-416	1	18,4157
6	Bak penampung air proses	F-421	1	46,0649
	Total			337,6092

Harga bak = Rp. 275.000,- /m<sup>2</sup>

Total harga bak utilitas = 337,6092m<sup>2</sup>.( Rp. 275.000,- /m<sup>2</sup>)

= Rp. 92842530,-

Total harga peralatan = (( \$2303250,333 + \$135173.9733 ) x Rp.9000,-/\$)

+Rp.92842530,-

= Rp. 22.038.661.290,-

#### B. Perhitungan Harga Tanah dan Bangunan

Luas tanah = 10.000 m<sup>2</sup>

Luas pabrik dan gudang = 1.875 m<sup>2</sup>

Luas bangunan kantor = 400 m<sup>2</sup>

Luas bangunan lain-lain = 2.091 m<sup>2</sup>

Harga tanah = Rp. 150.000,- /m<sup>2</sup>

Harga bangunan pabrik dan gudang = Rp. 1.000.000,-/m<sup>2</sup>

Harga bangunan kantor = Rp. 1.250.000,-/m<sup>2</sup>

Harga bangunan lain-lain = Rp. 750.000,-/m<sup>2</sup>

Jadi :

Harga tanah = 10.000 m<sup>2</sup> x Rp.150.000,-/m<sup>2</sup> = Rp.1.500.000.000,-

Harga bangunan pabrik dan gudang = 1.875 m<sup>2</sup> x Rp. 1.000.000,-/m<sup>2</sup>  
= Rp.1.875.000.000,-/m<sup>2</sup>

Harga bangunan kantor = 400 m<sup>2</sup> x Rp. 1.250.000,-/m<sup>2</sup>  
= Rp.500.000.000,-

Harga bangunan lain-lain = 2.091 m<sup>2</sup> x Rp. 750.000,-/m<sup>2</sup>

$$= \text{Rp.} 1.568.250.000,-$$

$$\begin{aligned} \text{Total harga tanah dan bangunan} &= \text{Rp.} 1.568.250.000,- + \text{Rp.} 1.875.000.000,- + \\ &\quad \text{Rp.} 500.000.000,- + \text{Rp.} 1.500.000.000,- \\ &= \text{Rp.} 5.443.250.000,- \end{aligned}$$

### C. Perhitungan Harga Bahan Baku dan Harga Jual Produk

#### ➤ Harga Bahan Baku

##### 1. Tulang

$$\begin{aligned} \text{Harga} &: \text{Rp.} 1000,-/\text{kg} \\ \text{Kebutuhan} &: 13500 \text{ kg/hari} \\ \text{Total} &: \text{Rp.} 4.455.000.000,- / \text{tahun} \end{aligned}$$

##### 2. HCl

$$\begin{aligned} \text{Harga} &: \text{Rp.} 2500,-/\text{liter} \\ \text{Kebutuhan} &: 20647,2617 \text{ kg/hari} \\ \text{Total} &: \text{Rp.} 14.449.054.900,- / \text{tahun} \end{aligned}$$

##### 3. CaO

$$\begin{aligned} \text{Harga} &: \text{Rp.} 1000,-/\text{kg} \\ \text{Kebutuhan} &: 1798,2 \text{ kg/hari} \\ \text{Total} &: \text{Rp.} 593.406.000,- / \text{tahun} \end{aligned}$$

Total harga bahan baku per tahun :

$$\begin{aligned} &= \text{Rp.} 4.455.000.000,- + \text{Rp.} 14.449.054.900,- + \text{Rp.} 593.406.000,- \\ &= \text{Rp.} 19.497.460.900,- \end{aligned}$$

#### ➤ Harga Jual Produk

$$\begin{aligned} \text{Harga} &: \text{Rp.} 70000,- / \text{kg} \\ \text{Produksi} &: 2849,079115 \text{ kg/hari} \\ \text{Total} &: \text{Rp.} 65.813.727.500,- \end{aligned}$$

### D. Perhitungan Gaji Karyawan

Perincian gaji karyawan tiap bulan dapat dilihat pada tabel D.4 berikut ini.

Tabel D.4 Gaji Karyawan

No	Jabatan	Jumlah	Gaji/bulan (Rp)	Total (Rp)
1	Direktur Utama	1	7.000.000,00	7.000.000,00
2	Direktur Teknik & Produksi	1	5.000.000,00	5.000.000,00
3	Direktur Administrasi & Keuangan	1	5.000.000,00	5.000.000,00
4	Sekretaris	3	1.000.000,00	3.000.000,00
5	Kabag Produksi	1	2.500.000,00	2.500.000,00
6	Kabag Teknik	1	2.500.000,00	2.500.000,00
7	Kabag Keuangan	1	2.500.000,00	2.500.000,00
8	Kabag Pemasaran	1	2.500.000,00	2.500.000,00
9	Kabag Personalia & Umum	1	2.500.000,00	2.500.000,00
10	Kasie Proses	1	1.500.000,00	1.500.000,00
11	Kasie Penelitian & Pengembangan	1	1.500.000,00	1.500.000,00
12	Kasie Utilitas	1	1.500.000,00	1.500.000,00
13	Kasie Pemeliharaan & Perbaikan	1	1.500.000,00	1.500.000,00
14	Kasie Laboratorium & QC	1	1.500.000,00	1.500.000,00
15	Kasie Promosi & Penjualan	1	1.500.000,00	1.500.000,00
16	Kasie Keuangan	1	1.500.000,00	1.500.000,00
17	Kasie Pembelian	1	1.500.000,00	1.500.000,00
18	Kasie Gudang	1	1.500.000,00	1.500.000,00
19	Kasie Personalia	1	1.500.000,00	1.500.000,00
20	Kasie Keamanan	1	1.500.000,00	1.500.000,00
21	Seksi Proses	20	1.000.000,00	20.000.000,00
22	Seksi Penelitian & Pengembangan	2	1.000.000,00	2.000.000,00
23	Seksi Utilitas	8	1.000.000,00	8.000.000,00
24	Seksi Pemeliharaan & Perbaikan	6	1.000.000,00	6.000.000,00
25	Seksi Laboratorium & QC	4	1.000.000,00	4.000.000,00
26	Seksi Promosi & Penjualan	4	1.000.000,00	4.000.000,00
27	Seksi Keuangan	2	1.000.000,00	2.000.000,00
28	Seksi Pembelian	2	1.000.000,00	2.000.000,00
29	Seksi Personalia	2	1.000.000,00	2.000.000,00
30	Seksi Gudang	6	800.000,00	4.800.000,00
31	Seksi Keamanan	10	800.000,00	8.000.000,00
32	Sopir & Pesuruh	8	600.000,00	4.800.000,00
TOTAL		96		116.600.000,00

Total gaji karyawan per bulan = Rp. 116.600.000,00

Ditetapkan 1 tahun produksi adalah 12 bulan + 1 bulan tunjangan, jadi gaji

karyawan per tahun :

$$= \text{Rp. } 116.600.000,00 \times 13$$

$$= \text{Rp. } 1.515.800.000,00$$

### E. Pehitungan Biaya Utilitas

#### 1. Kebutuhan Air

Air yang harus disuplai per hari =  $1028,1907 \text{ m}^3/\text{hari}$

Harga pengolahan air sumur =  $\text{Rp.}1000,- / \text{m}^3$

Biaya air =  $1028,1907 \text{ m}^3/\text{hari} \times \text{Rp.}1000,- / \text{m}^3 \times 330 \text{ hari/tahun}$   
 $= \text{Rp.} 339.302.931,- / \text{tahun}$

#### 2. Kebutuhan Listrik

Kebutuhan listrik untuk penerangan =  $29,690 \text{ kVA}$

Kebutuhan listrik untuk motor penggerak peralatan & utilitas, perkantoran  
 $= 258,4686 \text{ kVA}$

Total kebutuhan listrik =  $288,1586 \text{ kVA}$

Beban listrik terpasang =  $1,25 \times 288,1586 = 360,2 \text{ kVA}$

Biaya beban per bulan =  $\text{Rp.}25.000,-$

Biaya beban per tahun =  $12 \text{ bulan} \times \text{Rp.}25.000,-/\text{bulan} = \text{Rp.}300.000,-$

Biaya listrik : WBP =  $\text{Rp.} 455,-/\text{kVA} (\text{pk. } 18.00 - 22.00)$

LWBP =  $\text{Rp.} 350,-/\text{kVA} (\text{pk. } 22.00 - 18.00)$

Biaya listrik terpakai per tahun :

$= [((4 \text{ jam} \times 288,1586 \text{ kVA} \times \text{Rp.} 455,-/\text{kVA}) + (20 \text{ jam} \times 288,1586 \text{ kVA} \times \text{Rp.} 350,-/\text{kVA})) \times 330 \text{ hari/tahun}] + [30 \text{ hari} \times 29,690 \text{ kVA} \times (4 \text{ jam} \times \text{Rp.} 455,-/\text{kVA} + 20 \text{ jam} \times \text{Rp.} 350,-/\text{kVA})]$   
 $= \text{Rp.}846.570.395,-$

Biaya listrik total per tahun = biaya beban + biaya listrik terpakai

$= \text{Rp.}300.000,- + \text{Rp.}846.570.395,-$

$= \text{Rp.}846.870.395,-$

#### 3. Kebutuhan bahan bakar

Kebutuhan bahan bakar per bulan =  $63996 \text{ lt}$

Harga bahan bakar per liter =  $\text{Rp.}1440,-$

Biaya bahan bakar per tahun =  $63996 \text{ lt/bln} \times 12 \text{ bln/th} \times \text{Rp.}1440,-$   
 $= \text{Rp.}1.105.850.880,-$

Total biaya utilitas per tahun =  $\text{Rp.}1.105.850.880,- + \text{Rp.}846.870.395,- +$   
 $\text{Rp.}339.302.931,-$

= Rp. 2.292.024.206,-

PERPUSTAKAAN  
Universitas Katolik Widya Mandala  
SURABAYA